

# Arrhenius-Kooij Fits to Experimentally Generated Thermal Rate Coefficients for C + H<sub>3</sub><sup>+</sup> forming CH<sup>+</sup> and CH<sub>2</sub><sup>+</sup>

WILLIAM OSTLING <sup>1</sup>, PIERRE-MICHEL HILLENBRAND <sup>2</sup>, DMITRY IVANOV <sup>1</sup>, AODH O’CONNOR <sup>1</sup>,  
SHREYAS VISSAPRAGADA <sup>1</sup>, XAVIER URBAIN <sup>3</sup> AND DANIEL W. SAVIN <sup>1</sup>

<sup>1</sup>*Columbia Astrophysics Laboratory, Columbia University, New York, NY 10027, USA*

<sup>2</sup>*GSI Helmholtzzentrum für Schwerionenforschung, 64291 Darmstadt, Germany*

<sup>3</sup>*Institute of Condensed Matter and Nanosciences, Université Catholique de Louvain, Louvain-la-Neuve B-1348, Belgium*

## ABSTRACT

The reaction of C + H<sub>3</sub><sup>+</sup> forming CH<sup>+</sup> and CH<sub>2</sub><sup>+</sup> is an important early link in the chain of astrochemical reactions leading to the formation of organic molecules in molecular clouds. In O’Connor et al. (2015), using a merged fast-beams apparatus, we measured the cross section for each reaction channel and generated thermal rate coefficients with an estimated accuracy of 15%. Here, we present fits to those thermal rate coefficients using the Arrhenius-Kooij equation, which can be readily incorporated into astrochemical models. The fits reproduce the thermal rate coefficient to better than 0.6% for temperatures from 10 to 10,000 K.

## 1. INTRODUCTION

One of the first steps in the astrochemical formation of organic molecules began when neutral atomic C was fixed into hydrocarbons in interstellar clouds (van Dishoeck et al. 1999; Herbst & van Dishoeck 2009). The initial chemical network involved is rather simple, primarily consisting of C reacting with H<sub>3</sub><sup>+</sup>. The cosmic pathway from there to more complex hydrocarbons and other organic molecules passes through the molecular cations CH<sub>n</sub><sup>+</sup> ( $n = 1 - 3$ ). In our studies, we have investigated the reactions



The resulting CH<sup>+</sup> and CH<sub>2</sub><sup>+</sup> will then rapidly undergo sequential hydrogen abstraction with the abundant H<sub>2</sub> to form CH<sub>3</sub><sup>+</sup>. In O’Connor et al. (2015), using a merged fast-beams apparatus, we measured the integral cross sections for reactions (1a) and (1b). Using these data, we generated thermal rate coefficients for each reaction channel for the temperature range of 10 – 10,000 K, with an estimated accuracy of  $\sim 15\%$  at an estimated  $1\sigma$  confidence level. In Vissapragada et al. (2016), we fit the thermal rate coefficients of O’Connor et al. (2015) using the equation proposed by Novotný et al. (2013). One advantage of that fitting formula is that it is continuously differential and can accurately fit for complicated temperature dependencies. However, this equation is not currently supported by astrochemical databases such as KIDA (Wakelam et al. 2012) and UMIST (Millar et al. 2024), which use the mathematically simpler Arrhenius-Kooij (AK) formula. The disadvantage of the AK formula is that reactions with complicated temperature dependencies must be handled by dividing the temperature range into smaller intervals and fitting each interval separately. Here, we present fits to the thermal rate coefficient data of O’Connor et al. (2015) using the AK formula. In addition, since that work, we have developed a more reliable quantitative determination of the internal excitation of the reactant H<sub>3</sub><sup>+</sup> ions (Hillenbrand et al. 2019, Ivanov et al, in preparations). In Sec. 2, we show that this internal excitation has an insignificant effect on our measurements. Section 3 then presents our AK fits.

## 2. EXPERIMENTAL CONSIDERATIONS

The measurements of O’Connor et al. (2015) were performed with statistically populated C(<sup>2</sup>P<sub>J</sub>) in the ground term and internally excited H<sub>3</sub><sup>+</sup>. Thermal rate coefficients were generated using theoretical considerations to account for the temperature dependence of the fine-structure  $J$  levels. As for the internal excitation of the H<sub>3</sub><sup>+</sup>, very good agreement

was found between the O'Connor et al. (2015) results and the mass-scaled results of Savic et al. (2005) for statistically populated C reacting with  $D_3^+$  at an internal temperature of 77 K. Based on this agreement, the internal excitation of the  $H_3^+$  was inferred to be unimportant, to within the level of the mutual experimental uncertainties. Here, we further explore any potential effects due to the internal excitation of the  $H_3^+$ .

The  $H_3^+$  beam of O'Connor et al. (2015) was generated using a duoplasmatron ion source adjusted to maximize the  $H_3^+$  current. There, we estimated the internal excitation of the  $H_3^+$  from the measured relative collision energy at which the first competing endoergic reaction channel opened up, namely



This occurred at  $\langle E_{\text{int}} \rangle \approx 0.61$  eV below the theoretical threshold energy of  $E_{\text{th}} = 1.98$  eV. This was attributed to the  $H_3^+$  internal excitation. Here, we have refit the merged-beams rate coefficient results of O'Connor et al. (2015) to more accurately determine  $\langle E_{\text{int}} \rangle$ . We find that  $\langle E_{\text{int}} \rangle = 0.61_{-0.28}^{+0.21}$  eV, where the uncertainty is quoted at a  $2\sigma$  confidence level. Using the partition function of Kylänpää & Rantala (2011), this corresponds to an internal temperature of  $T_{\text{int}} = 2415_{-765}^{+500}$  K.

Subsequent work by Hillenbrand et al. (2019) and Ivanov et al. (in preparation), on the same apparatus as that of O'Connor et al. (2015), has more accurately determined the internal excitation of the  $H_3^+$  using the endoergic reaction



This reaction has a barrier of  $E_b \approx 68$  meV. Any signal measured for collision energies below  $E_b$  is due solely to internal excitation of the  $H_3^+$ , enabling a more reliable determination of  $T_{\text{int}}$ . For ion source conditions adjusted to optimize the  $H_3^+$  current, Those recent works find that  $T_{\text{int}} < 1600$  K, which corresponds to ( $\langle E_{\text{int}} \rangle < 0.28$  eV).

Based on our more reliable determination of  $T_{\text{int}}$ , we can re-investigate if the internal excitation of the  $H_3^+$  has any effect on the experimentally generated thermal rate coefficient of O'Connor et al. (2015). The fraction of  $H_3^+$  with sufficient internal energy to access competing endoergic reactions such as reaction (2) and those at higher energies is given by Hillenbrand et al. (2022) as

$$f_{E_{\text{int}} > E_{\text{th}}} = \exp\left(-\frac{E_{\text{th}}}{\langle E_{\text{int}} \rangle}\right). \quad (4)$$

Using this equation, we find that only 0.085% of the  $H_3^+$  ions have an internal energy greater than  $E_{\text{th}}$ . As a worst-case scenario, we can assume that all reactions of  $H_3^+$  ions with  $E_{\text{int}} > E_{\text{th}}$  do not contribute to reactions (1a) or (1b). This would mean that the cross sections and rate coefficients reported by O'Connor et al. (2015) need to be scaled up by a factor of 1.085. This scaling lies well within the total experimental uncertainty limits. Moreover, it affects both channels in the same direction and therefore does not affect the branching ratios.

### 3. RECOMMENDED ARRHENIUS-KOOIJ FIT PARAMETERS

We used the AK formula to model the rate coefficients of both reaction channels. However, with one such function, we cannot model all the experimentally observed features in the rate coefficient correctly. Following the procedure described in Paul et al. (2022), we provide a representation of the thermal rate coefficient  $\alpha$  using  $m$  piecewise-joined AK equations ( $T \in [T_i, T_{i+1}]$ ,  $i \in 0, \dots, m$ ) with

$$\alpha(T)[\text{cm}^3 \text{ s}^{-1}] = A_i \left(\frac{T[K]}{300}\right)^{\beta_i} \exp\left(-\frac{\gamma_i}{T[K]}\right), \quad (5)$$

where  $T$  is the gas temperature and  $A_i$ ,  $\beta_i$ , and  $\gamma_i$  are fitting parameters. Following KIDA conventions, the uncertainty is described by a log-normal factor  $F = \exp(\Delta\alpha/\alpha)$  for each interval with

$$F(T) = F_i \exp\left(g_i \left(\frac{1}{T[K]} - \frac{1}{300}\right)\right), \quad (6)$$

where  $F_i$  and  $g_i$  are the fit parameters. We construct the data for  $F(T)$  as average of  $F_{\text{high}}(T) = \exp(\Delta_{\text{high}}\alpha/\alpha)$  and  $F_{\text{low}}(T) = \exp(\Delta_{\text{low}}\alpha/\alpha)$ , where  $\Delta_{\text{high}}\alpha$  and  $\Delta_{\text{low}}\alpha$  correspond to the upper and lower one-sigma errors, calculated from the quadrature sum of the systematic uncertainty and the original thermal rate fitting accuracy. For astrochemical

**Table 1.** Fit Parameters for the Thermal Rate Coefficients for reaction (1a) and Equation (1b) And Its Relative Uncertainty, Using Equation (7) and (8).

Parameter	C + H <sub>3</sub> <sup>+</sup> → CH <sup>+</sup> + H <sub>2</sub>			
	10 - 17.4 K	17.4 - 169 K	169 - 4440 K	4440 - 10000 K
<i>i</i>	1	2	3	4
<i>A<sub>i</sub></i>	$5.54 \times 10^{-10}$	$6.77 \times 10^{-10}$	$6.44 \times 10^{-10}$	$1.28 \times 10^{-9}$
<i>β<sub>i</sub></i>	$-9.90 \times 10^{-2}$	$9.73 \times 10^{-3}$	$1.30 \times 10^{-1}$	$-5.52 \times 10^{-2}$
<i>γ<sub>i</sub></i>	$-7.85 \times 10^{-2}$	$-1.97 \times 10^0$	$-2.23 \times 10^1$	$8.01 \times 10^2$
<i>F<sub>i</sub></i>	$1.13 \times 10^0$	$1.13 \times 10^0$	$1.13 \times 10^0$	$1.15 \times 10^0$
<i>g<sub>i</sub></i>	$-1.09 \times 10^{-4}$	$-7.93 \times 10^{-4}$	$2.90 \times 10^{-2}$	$4.77 \times 10^0$
Parameter	C + H <sub>3</sub> <sup>+</sup> → CH <sub>2</sub> <sup>+</sup> + H			
	10 - 283 K	283 - 1330 K	1330 - 3760 K	3760 - 10000 K
<i>i</i>	1	2	3	4
<i>A<sub>i</sub></i>	$3.28 \times 10^{-10}$	$3.36 \times 10^{-10}$	$4.71 \times 10^{-10}$	$1.56 \times 10^{-9}$
<i>β<sub>i</sub></i>	$-1.66 \times 10^{-1}$	$-1.94 \times 10^{-1}$	$-3.30 \times 10^{-1}$	$-6.70 \times 10^{-1}$
<i>γ<sub>i</sub></i>	$1.58 \times 10^{-3}$	$7.39 \times 10^0$	$1.89 \times 10^2$	$1.47 \times 10^3$
<i>F<sub>i</sub></i>	$1.18 \times 10^0$	$1.18 \times 10^0$	$1.18 \times 10^0$	$1.17 \times 10^0$
<i>g<sub>i</sub></i>	$-7.90 \times 10^{-4}$	$-2.40 \times 10^{-1}$	$-1.39 \times 10^0$	$-3.75 \times 10^0$

calculations, a continuous behavior of the functions at the interval borders is typically desired, so we introduce a dependency of the *A<sub>i</sub>* and *F<sub>i</sub>* parameters in accordance with Paul et al. (2022) to obtain the continuous fit functions

$$\alpha(T)[\text{cm}^3 \text{ s}^{-1}] = A_0 \left( \frac{T[K]}{300} \right)^{\beta_i} \exp \left( -\frac{\gamma_i}{T[K]} \right) \times \prod_{j=1}^i \left( \frac{T_j}{300} \right)^{\beta_{j-1} - \beta_j} \exp \left( \frac{\gamma_j - \gamma_{j-1}}{T_j} \right) \quad (7)$$

and

$$F(T) = F_0 \exp \left( \frac{g_i - g_0}{300} \right) \exp \left( g_i \left( \frac{1}{T[K]} - \frac{1}{300} \right) \right) \times \prod_{j=1}^i \exp \left( \frac{g_{j-1} - g_j}{T_j} \right), \quad (8)$$

for the rate coefficient and uncertainty fits, respectively. We note that for the first temperature interval (*i* = 0) the product terms evaluate to unity and that the functions show the same dependence on *T<sub>k</sub>* as given by Equations (5) and (6), respectively. For the fit with Equation (7), we let the interval borders *T<sub>i</sub>* vary as free parameters and subsequently fixed them for the fit with Equation (8). The Levenberg-Marquardt algorithm was implemented for a more robust and accurate fit as per Moré (1978).

The resulting parameters for Equations 5, using *m* = 4 temperature intervals, are listed in Table 1 for (1a) and (1b) For each reaction, the relative deviations of the fit functions to the rate coefficients are below 0.6% at all temperatures.

This research was supported, in part, by the NSF Division of Astronomical Sciences Astronomy and Astrophysics Grants program under AST-2002461.

## REFERENCES

- Herbst, E., & van Dishoeck, E. F. 2009, ARA&A, 47, 427, doi: [10.1146/annurev-astro-082708-101654](https://doi.org/10.1146/annurev-astro-082708-101654)
- Hillenbrand, P. M., Bowen, K. P., Liévin, J., Urbain, X., & Savin, D. W. 2019, ApJ, 877, 38, doi: [10.3847/1538-4357/ab16dc](https://doi.org/10.3847/1538-4357/ab16dc)
- Hillenbrand, P.-M., de Ruette, N., Urbain, X., & Savin, D. W. 2022, ApJ, 927, 47, doi: [10.3847/1538-4357/ac41ce](https://doi.org/10.3847/1538-4357/ac41ce)
- Kylänpää, I., & Rantala, T. T. 2011, JChPh, 135, 104310, doi: [10.1063/1.3633516](https://doi.org/10.1063/1.3633516)
- Millar, T. J., Walsh, C., Van de Sande, M., & Markwick, A. J. 2024, A&A, 682, A109, doi: [10.1051/0004-6361/202346908](https://doi.org/10.1051/0004-6361/202346908)
- Moré, J. J. 1978, Numerical Analysis, 105, doi: <https://doi.org/10.1007/bfb0067700>

- 107 Novotný, O., Becker, A., Buhr, H., et al. 2013, ApJ, 777,  
108 54, doi: [10.1088/0004-637X/777/1/54](https://doi.org/10.1088/0004-637X/777/1/54)
- 109 O'Connor, A. P., Urbain, X., Stützel, J., et al. 2015, ApJS,  
110 219, 6, doi: [10.1088/0067-0049/219/1/6](https://doi.org/10.1088/0067-0049/219/1/6)
- 111 Paul, D., Grieser, M., Grussie, F., et al. 2022, ApJ, 939,  
112 122, doi: [10.3847/1538-4357/ac8e02](https://doi.org/10.3847/1538-4357/ac8e02)
- 113 Savic, I., Cermák, I., & Gerlich, D. 2005, International  
114 Journal of Mass Spectrometry, 240, 139,  
115 doi: [10.1016/j.ijms.2004.10.015](https://doi.org/10.1016/j.ijms.2004.10.015)
- 116 van Dishoeck, E. F., Wright, C. M., Thi, W. F., &  
117 Boonman, A. M. S. 1999, in Astronomical Society of the  
118 Pacific Conference Series, Vol. 177, Astrophysics with  
119 Infrared Surveys: A Prelude to SIRTf, ed. M. D. Bica,  
120 R. M. Cutri, & B. F. Madore, 288
- 121 Vissapragada, S., Buzard, C. F., Miller, K. A., et al. 2016,  
122 ApJ, 832, 31, doi: [10.3847/0004-637X/832/1/31](https://doi.org/10.3847/0004-637X/832/1/31)
- 123 Wakelam, V., Herbst, E., Loison, J. C., et al. 2012, ApJS,  
124 199, 21, doi: [10.1088/0067-0049/199/1/21](https://doi.org/10.1088/0067-0049/199/1/21)