MICROSTRUCTURAL CHARACTERIZATION OF SUPERALLOY 718 WITH BORON AND PHOSPHORUS ADDITIONS

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ABSTRACT

Boron and phosphorus additions are known to improve the stress rupture properties of IN-718. One possible mechanism to explain this property improvement relies on the boron and phosphorus additions slowing down the growth of γ'' and γ' precipitates during high temperature service or aging. However, atom probe analysis found no segregation of boron and phosphorus to $\gamma-\gamma''$ or to $\gamma-\gamma'$ interfaces in the alloys with the high boron and high phosphorous levels. No difference in growth rates were found by transmission electron microscopy in the sizes of the γ'' or γ' in alloys with high phosphorus and high boron as compared to commercial alloys and to alloys with even lower levels of phosphorus and boron. Atom probe analysis further found that much of the phosphorus, boron, and carbon segregated to grain boundaries. Creep curves comparing the alloys with high levels of phosphorus and boron and alloys with low levels of phosphorus and boron show a large difference in strain rate in the first hours of the test. These results suggest that the boron and phosphorus may have a direct effect on dislocation mobility by some pinning mechanism.

INTRODUCTION

Optimization of the phosphorus, boron and carbon levels in Inconel alloy 718 has been found by Cao and Kennedy to significantly improve the stress-rupture properties [1-2]. For example, a commercial grade 718 with 0.03 wt.% C, 0.008% P, and 0.004% B has a stress rupture life of 200 h at 690 MPa at 650°C, while if the phosphorus and boron are increased to 0.022% P and 0.011% B with a carbon level of 0.005%, then the stress rupture life increases to 750 h. This study not only found an individual improvement from higher levels of phosphorus or boron, but found an even greater effect when both phosphorus and boron levels were increased.

One possible mechanism to explain this property improvement relies on the boron and phosphorus additions slowing down the growth of γ'' and γ' precipitates during high temperature service or aging. Growth of γ'' and γ' precipitates during service has been thoroughly documented by Wlodek and Field [3] for times up to 50,000 h and temperatures up to 704°C. In this study, the growth rates of the γ'' and γ' precipitates with low levels, commercial levels, and high levels of phosphorus and boron were compared by transmission electron microscopy (TEM) on crept and statically annealed specimens. Creep curves were generated and the segregation behavior of boron, phosphorus, and carbon was investigated by atom probe field ion microscopy (APFIM).

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Table I Alloy Compositions in wt.%

	IN-718	High Purity	Commercial	Microalloyed
	Specification	G988	G457	G727
Ni	50-55	balance	balance	balance
Cr	17-21	17.9	17.8	17.9 (19.9 at.%)
Fe	balance	19.4	19.4	19.3 (20.0 at.%)
Nb+Ta	4.75-5.5	5.22	5.06	5.05 (3.15 at.%)
Mo	2.8-3.3	2.95	2.87	2.86 (1.73 at.%)
Ti	0.65-1.15	0.92	0.92	0.92 (1.11 at.%)
Al	0.2-0.8	0.47	0.63	0.58 (1.25 at.%)
В	0.006 max.	< 0.001	0.004	0.011 (0.059 at.%
P	0.015 max.	< 0.001	0.006	0.022 (0.041 at.%
C	0.08 max.	0.008	0.031	0.009 (0.043 at.%
Si	0.35 max.	0.01	0.1	0.01 (0.02 at.%)
N	-	0.015	0.0037	0.002 (0.08 at.%
S	0.015 max.	0.0008	0.0004	0.0004 (0.001 at.9

EXPERIMENTAL PROCEDURE

Table I lists the compositions of the three alloys studied (Teledyne Allvac alloy numbers G988, G457, and G727) as measured by Teledyne. Wrought bars were obtained from Teledyne with a solution heat treatment at 954°C for 1 h followed by an air cool and aging at 718°C for 8 h, furnace cooling at 55°C/h to 621°C, followed by an air cool. Static anneals were subsequently carried out at ORNL on slices encapsulated in evacuated quartz capsules at a temperature of 704°C for periods up to 6 months. Creep tests were conducted in air on 3 mm dia. specimens with an 18 mm gage length at temperatures of 649 or 704°C and various loads, as listed in Table II. TEM was performed on both crept and statically annealed specimens at 300 kV. Dark field images along 100 poles around the inner ring of superlattice reflections with indices of 010, 0 1 1/2, 011, 0 1/2 1, 001, etc. were made using the 10 channel dark field capability. This allowed for all three variants of the γ'' from one area to be imaged. Exposure times were typically 45 s to 1 min. with 10 min. in the D19 developer. Measurements of the precipitate size and area fraction were made with Buehler's Omnimet image analysis system. Atom probe analysis was performed with the ORNL energy-compensated atom probe at a specimen temperature of 50 K and a pulse fraction of 20%. Field ion micrographs were made at a specimen temperature of 50-60 K with neon as the imaging gas. Further details on the atom probe analyses were previously published [4].

Table II Creep Conditions

Alloy	Creep Temp, °C	Load, MPa (ksi)	Rupture Time, hr	Elongation at Failure	Min. Creep Rate, %/s
G457	649	620 (90)	108	26.3%	1.9 × 10 ⁻⁶
G727	649	690 (100)	1436	7.4%	2.4×10^{-7}
G727	704	379 (55)	>>32	stopped at 0.6%	1.9×10^{-6}
G988	704	379 (55)	37	8.85%	1.9×10^{-5}

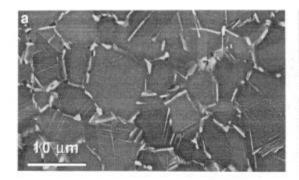




Fig. 1. Backscattered electron images of as-polished (no etching) of (a) G727 and (b) G988.

RESULTS AND DISCUSSION

No obvious differences were found by optical microscopy between the high purity and microalloyed alloys in the phase distribution of the larger precipitates present in alloy 718 both before and after heat treatments or creep experiments [1-2]. Analyses of these phases have been extensively described [3,5] and so was not investigated here. The emphasis in the current study concerned the stability of the much finer distribution of γ'' and γ' precipitates and the segregation pattern of the phosphorus and the boron. Figure 1 shows scanning electron microscope images using backscattered electrons of as-polished, i.e. with no chemical etching, of specimens showing the starting microstructures. Again no differences in the precipitation was apparent. Grain size was ASTM 11.

Transmission Electron Microscopy

Mean sizes of γ'' precipitates and typical diameters of the smaller γ' are given in Table III. Figures 2a, 2b, 3a, and 3b show the γ'' precipitates on edge. γ'' precipitates listed in Table III as variant 1 were in the plan view orientation, i.e. the disks were flat. Note in Fig 2c and 3c that these disks are not all that round, especially in Fig. 3c where after the longest anneal, the γ'' precipitates are rather rectangular shaped. γ' precipitates are visible in (a) and (c) as the smaller circular precipitates. Areas were chosen for analysis that were away from the denuded zones near some grain boundaries. The same precipitate measuring technique was applied to all the specimens. Although hundreds of precipitates were measured, there are a

Table III Sizes of γ' and γ"

	Y'	γ" v	ariant 1	γ" ν	ariant 2	γ''ν	ariant 3
Alloy	typical dia,	area fraction,	mean breadth × length, nm	area fraction,	mean breadth × length, nm	area fraction, %	mean breadth × length, nm
			Crept	at 649°C			
G457	obscured	9	9 × 21	11	10 × 26	9	7×24
G727		17	10 × 28	6	8 × 30	8	8×28
			Crept	at 704°C			
G727	10-20	14	10 × 36	4	7 × 34		
G988	10-20	19	12 × 49	2	8 × 32		
			Static Anneal,	406 hrs at 7	94°C		
G727	25-35	6	50 × 90	5	9 × 54		
G988	25-35	6	50×100	6	12 × 72		
		5	satic Anneal, 6	months at	704°C		
G457	40-60	6	100 × 300	1	20 × 160	1	30×240
G727	40-60	14	150×320	4	17× 130	2	14× 130
G988	40-60	7	85 × 260	3	22 × 140	2	20 × 150

number of experimental variables which adversely affect the accuracy of the results. These include inhomogeneities in the specimens, slight differences in diffracting conditions, and slight differences in photographic exposure conditions. The final conclusion reached here was that, in comparing the alloys with different phosphorus and boron levels, there was no significant difference in precipitate sizes in either γ'' or γ' precipitates after creep or after the static anneals used here.

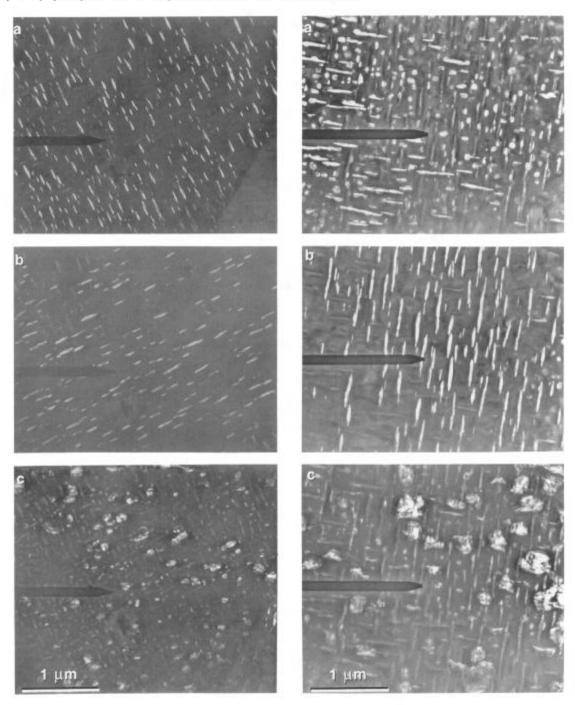


Fig. 2 Alloy G727 after 406 hrs at 704°C showing the three variants of γ'' . γ' precipitates are more visible in (a) and (c) as the smaller circular precipitates. Both Fig. 2 and 3 were made at a [100] zone axis using (a)-001 (b)-01/21 (c)-011 reflections.

Fig. 3 Alloy G727 after 6 mon. at 704°C. γ" are disk-shaped precipitates viewed on edge in (a) and (b) and viewed flat in (c). The exact same area is shown in a, b, and c in both Figs. 2 and 3.

Table IV Compositions as Measured by Atom Probe

	G988 High Purity Alloy (at.%)			G727 Microalloyed (at.%)		
	γ	γ΄	γ″	γ	γ΄	γ″
Cr	21.7	0.3	1.1	21.4	0.7	1.3
Fe	23.7	2.1	1.6	24.0	1.9	1.2
Nb	0.9	8.9	20.9	1.1	6.4	19.8
Mo	2.3	0.2	1.5	1.8	0.08	0.7
Ti	0.3	10.0	4.7	0.4	9.6	5.2
A1	0.8	9.0	0.15	0.4	8.8	0.4
В	-	-	-	0.02	-	-
P	-	-	-	-	-	-
N/Si	0.07	-	-	0.05	-	-

Atom Probe-Field Ion Microscopy

No solute enrichments, including boron and phosphorus, were observed by field ion microscopy at either the γ'' or γ' precipitate/matrix interfaces in either the microalloyed or the high purity alloy. No evidence of solute enrichments were obtained from atom probe composition profiles across the γ'' or γ' precipitate/matrix interfaces. The compositions of the γ matrix and the γ'' and γ' precipitates are given in Table IV. Further details can be found in ref. [4]. In the microalloyed specimen, note the depletion of phosphorus to below the detectability limits in all three phases and the depletion of boron to 1/3 the nominal value in the matrix and to less than the detection limits in the γ'' or γ' phases. The amount of phosphorus added to the alloy was 0.04 at.%. The critical amount needed for a beneficial effect is 0.02 at.%. The detection limit for the atom probe measurements in the matrix was 0.01 at.%. The amount measured in the matrix was below this level. Based on these compositions, a summary of the partitioning behavior is presented in Table V.

Table V Partitioning Behavior Based on Atom Probe Compositions

Alloying Element	Had highest concentration in	Had the 2nd highest level in	Had the 3rd highest level in
Nb	Phosphide	γ"	γ′
Mo	Phosphide	γ	GB
Ti	γ′	γ″	Phosphide
Al	γ΄		
В	GB	Phosphide	
P	GB	Phosphide	
Ni	γ″	γ'	
Cr	γ		
Fe	γ		

Table VI Solute Grain Boundary Coverage in Alloy G727

Solute	Monolayer Coverage
Boron	61%
Phosphorus	37%
Carbon	5.7%
Nitrogen/Silicon	1.9%

In the microalloyed specimens, a significant segregation of boron and phosphorus to grain boundaries was found. In the atom probe, a cylindrical core of material is sampled along a grain boundary. In effect, as the evaporation proceeds, the grain boundary traverses the probe hole. A statistical technique was used to determine the level at the boundary (assuming an effective grain boundary thickness of 0.2 nm width) even though the probe hole is slightly wider. The monolayer calculation then assumes a normal metallic atomic spacing, so that a coverage of 61% would mean that there was 61% as many boron atoms as metal atoms in a volume 0.2 nm thick, see Table VI. No indication could be made on whether the boron is present interstitially or not. A typical field ion micrograph of alloy G727 is shown in Fig. 4. This particular image shows brightly imaging spots on a grain boundary and also includes a phosphide particle near the grain boundary.

The observation of boron and carbon segregation to grain boundaries is consistent with several previous atom probe investigations in nickel-based superalloys. Buchon *et al.* [6] reported high boron levels and occasionally carbon enrichment at decorated (with brightly imaging spots) γ – γ boundaries in an NA4 nickel-based superalloy containing 0.11 at.% B and 0.14% C. Boron and carbon segregation was found by Stiller [7] in Alloy 600 with 0.005 at.% B and 0.18 at.% C and in Alloy 690 with 0.005 at.% B and 0.09 at.% C. Letellier *et al.* [8] found a 2 nm thick boron and molybdenum enriched zone along γ/γ , γ'/γ' interfaces in an Astroloy containing 0.11 at.% B.

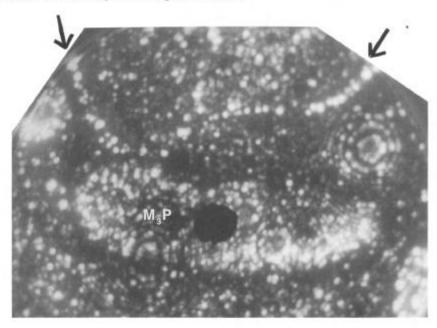


Fig. 4 Field ion micrograph of a grain boundary in alloy G727 showing bright spot decoration along a grain boundary (arrowed) and a neighboring oval-shaped phosphide precipitate.

Creep Tests

Typical creep curves for the alloys used in this study are shown in Fig. 5. Analyses of the fracture modes indicated no differences in the fracture morphology between the three alloys; rupture occurred in a ductile manner. The microalloyed G727 alloy clearly shows superior creep strength as compared to the other two alloys. Comparison of the curves for the three alloys suggests that whatever strengthening mechanism is operative in the G727 alloy is present at the beginning of the creep test and is not the result of any extensive aging during the test. For example, after 10 h of creep testing at 704°C, a period of time far too low for any aging or growth of phases to have occurred, the high-purity G988 alloy has already entered tertiary creep and is almost to failure, while the G727 alloy has only just reached the minimum creep rate regime. TEM examinations showed no significant differences in precipitate sizes after creep tests, Table III.

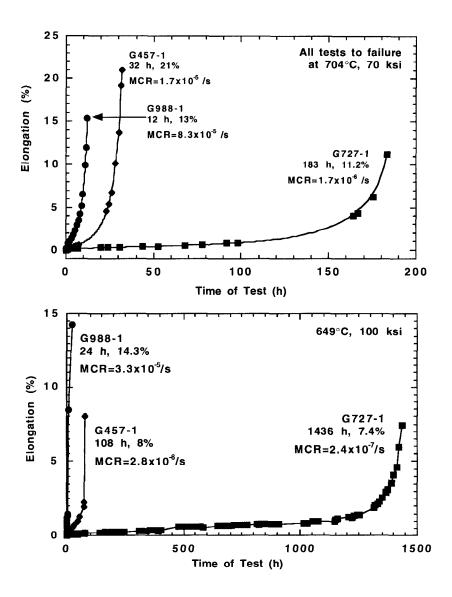


Fig. 5. Creep curves for the commercial alloy G457, the high purity (low levels of P & B) G988 alloy, and microalloyed (high levels of P & B) G727. Tests were performed in air.

SUMMARY AND CONCLUSIONS

Phosphorus and boron additions improve the stress rupture life and creep rates of Inconel 718. Since the principal strengthening mechanism in Alloy 718 is the presence of γ'' and γ' precipitates, it was first suspected that the improvement in creep properties in alloy G727 microalloyed with phosphorus and boron was a result of boron and phosphorus pinning the precipitates and slowing their growth. However, no segregation of phosphorus and boron was found by atom probe at γ'' and γ' interfaces and no difference in γ'' or γ' precipitate stability was observed by TEM after creep tests or after long term static anneals at 704°C. The phosphorus and boron were found strongly segregated to γ - γ grain boundaries. The creep curves show that differences in the creep strength of the alloys are apparent at the beginning of the tests and do not require long aging times. Aside from grain boundary effects, this suggests as another possibility, that the beneficial effects of phosphorus and boron result from a dislocation pinning mechanism.

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