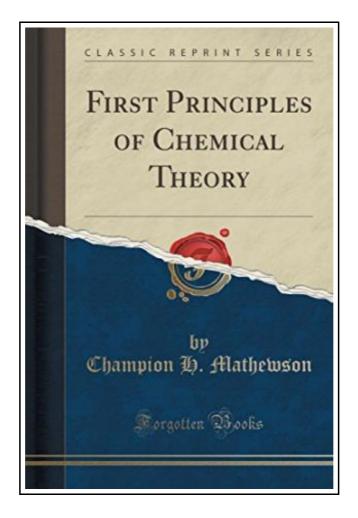
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Reviews

The publication is great and fantastic. It really is simplistic but surprises within the 50 % from the publication. Your daily life span will be change when you comprehensive reading this article book. (Althea Aufderhar)

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Forgotten Books, United States, 2015. Paperback. Book Condition: New. 229 x 152 mm. Language: English . Brand New Book ***** Print on Demand *****. Excerpt from First Principles of Chemical Theory This small volume has been prepared for the use of first year students at the Sheffield Scientific School, as reference text in connection with a short course of lectures on Chemical Theory. A period of six weeks immediately following the first four months instruction in General Chemistry is devoted to work of this nature. General principles and theoretical topics arc discussed with the utmost simplicity, and in particular view of their continued application. During this time recitations and laboratory exercises arc adjusted to the particular task of explaining and emphasizing the lecture subjects. It is not the intention to segregate and summarily dispose of much important material by untimely or unduly restricted discussion under the above heading. The purpose is rather to offer very early presentation of leading principles which arc of material assistance in teaching the beginner to properly explain and correlate his experimental results. Such preliminary preparation permits very open class room discussion of the specific chemical phenomena, which are gradually developed in the laboratory. Every opportunity for illustrating and applying these principles is improved as the actual chemical experience of the student increases. The advisability of using the Electrolytic Dissociation Theory and the Mass Action Law in first year work is no longer questioned by most teachers. It is rather a question of when and how these subjects should be introduced. As soon as the student has acquired practical familiarity with the molecular and atomic theory and is able to fully comprehend a few of the more general types of chemical change, no particular difficulty will be met in studying the characteristic behavior of electrolytes in aqueous...



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