# 練習実験報告

肖宇笑 2024年6月11日

# Galvano Sepctrum

#### REMPI scan

Selected peaks

Peak assignments

Speed correction

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REMPI scan

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Peak assignments

Speed correction

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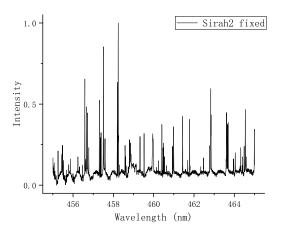


Fig. 1: Wavelen. correction

# Galvano Sepctrum

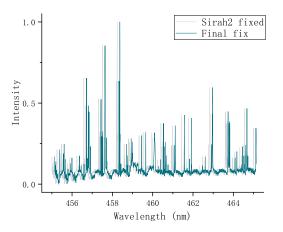


Fig. 1: Wavelen. correction

# Galvano Sepctrum

#### Calibration

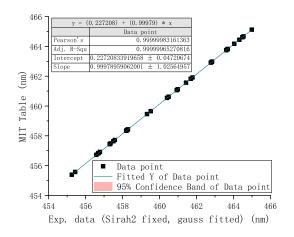


Fig. 2: Correction function

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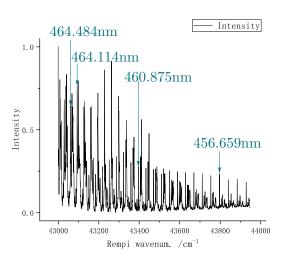
#### REMPI scan

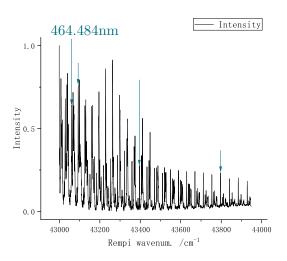
Selected peaks

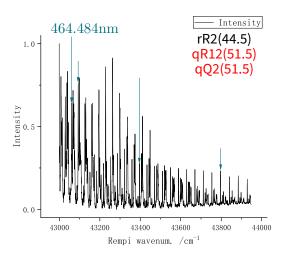
Peak assignments

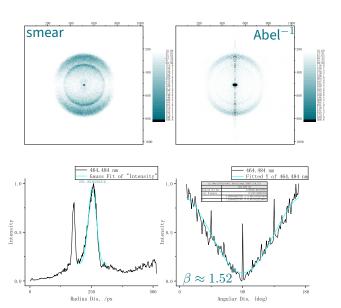
Speed correction

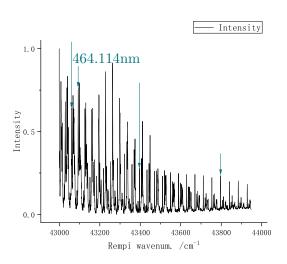
# Selected peaks

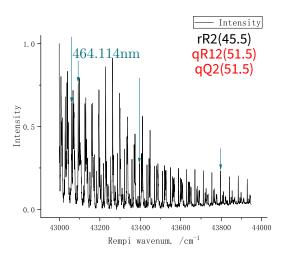


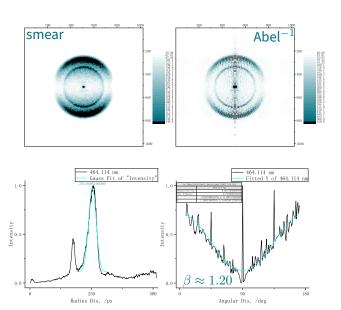


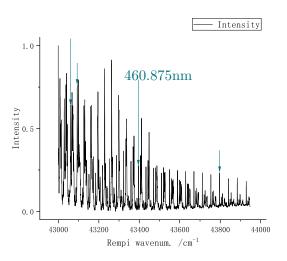


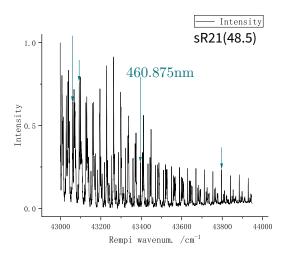


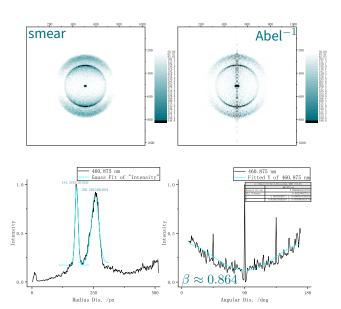


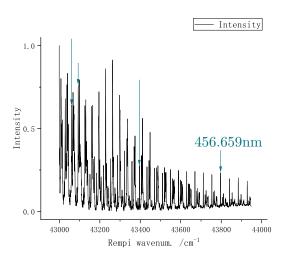


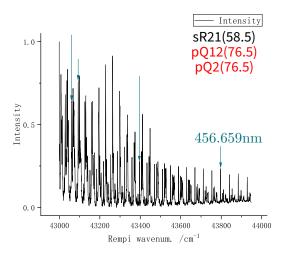


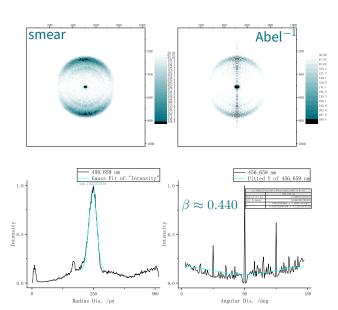












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# Peak assignments

$464.484$ nm $\approx 43058.49$ cm <sup>-1</sup>	$464.114$ nm $\approx 43092.81$ cm <sup>-1</sup>	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	$456.659$ nm $\approx 43796.34$ cm <sup>-1</sup>
px = 253.162	px = 253.655	px = <b>181.319</b> & 256.240	px = 246.776
rR2 (44.5) qR12 (51.5) qQ2 (51.5)	rR2 (45.5) qR12 (51.5) qQ2 (51.5)	sR21  (48.5)	sR21 (58.5) pQ12 (76.5) pP2 (76.5)

# Peak assignments

$464.484$ nm $\approx 43058.49$ cm <sup>-1</sup>	$464.114$ nm $\approx 43092.81$ cm <sup>-1</sup>	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	$456.659$ nm $\approx 43796.34$ cm <sup>-1</sup>
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rR2 (44.5) $qR12 (51.5)$ $qQ2 (51.5)$	rR2 (45.5) qR12 (51.5) qQ2 (51.5)	sR21  (48.5)	sR21 (58.5) pQ12 (76.5) pP2 (76.5)

#### **Notice**

Colored assignments are mismatched, and will not be used to calculate.

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## Trans. energy of NO

	$E_{total}$	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^1$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49$ cm $^{-1}$		$\Delta E_v(1 \to 0) + E(J = 44)$
Peak 2 464.114nm	43 092.81cm <sup>-1</sup>		$\Delta E_v(1 \to 0) + E(J = 45)$
Peak 3 460.875nm	43 395.69cm <sup>-1</sup>	25 128.57cm <sup>-1</sup>	$\Delta E_v(1 \to 0) + E(J = 48)$
Peak 4 456.659nm	43 796.34cm <sup>-1</sup>		$\Delta E_v(1 \to 0) + E(J = 58)$

<sup>&</sup>lt;sup>1</sup>Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

# Trans. energy of NO

	$E_{total}$	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^2$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49\mathrm{cm}^{-1}$		$2341.9327750 \text{cm}^{-1} + E(J = 44)$
Peak 2 464.114nm	43 092.81cm <sup>-1</sup>		$2341.9327750 \text{cm}^{-1} + E(J = 45)$
Peak 3 460.875nm	43 395.69cm <sup>-1</sup>	25 128.57cm <sup>-1</sup>	$2341.9327750 \text{cm}^{-1} + E(J = 48)$
Peak 4 456.659nm	43 796.34cm <sup>-1</sup>	-	$2341.9327750 \text{cm}^{-1} + E(J = 58)$

<sup>&</sup>lt;sup>2</sup>Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

#### Trans. energy of NO

	$E_{total}$	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^2$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49\mathrm{cm}^{-1}$		$2341.9327750 \text{cm}^{-1} + E(J = 44)$
Peak 2 464.114nm	43 092.81cm <sup>-1</sup>		$2341.9327750 \text{cm}^{-1} + E(J = 45)$
Peak 3 460.875nm	43 395.69cm <sup>-1</sup>	25 128.57cm <sup>-1</sup>	$2341.9327750 \text{cm}^{-1} + E(J = 48)$
Peak 4 456.659nm	43796.34cm <sup>-1</sup>		$2341.9327750 \text{cm}^{-1} + E(J = 58)$

#### Vib. energy level

$$E_v = \omega_e \left( v + \frac{1}{2} \right) - \omega_e x_e \left( v + \frac{1}{2} \right)^2 + \omega_e y_e \left( v + \frac{1}{2} \right)^3.$$

<sup>&</sup>lt;sup>2</sup>Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

### Trans. energy of NO

	$E_{total}$	$E_{bond}(\mathrm{O}\mathrm{-NO})^3$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49$ cm $^{-1}$		5814.033cm <sup>-1</sup>
Peak 2 464.114nm	43 092.81cm <sup>-1</sup>		5965.969cm <sup>-1</sup>
Peak 3 460.875nm	43 395.69cm <sup>-1</sup>	25 128.57cm <sup>-1</sup>	6239.696cm <sup>-1</sup>
Peak 4 456.659nm	$43796.34$ cm $^{-1}$	•	8004.278cm <sup>-1</sup>

<sup>&</sup>lt;sup>3</sup>Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

<sup>&</sup>lt;sup>4</sup>Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

#### Trans. energy of NO

	$E_{total}$	$E_{bond}(\mathrm{O}\mathrm{-NO})^3$	$E_{int.}(NO)$
Peak 1 464.484nm	43 058.49cm <sup>-1</sup>		$5814.033$ cm $^{-1}$
Peak 2 464.114nm	43 092.81cm <sup>-1</sup>		5965.969cm <sup>-1</sup>
Peak 3 460.875nm	43 395.69cm <sup>-1</sup>	25 128.57cm <sup>-1</sup>	6239.696cm <sup>-1</sup>
Peak 4 456.659nm	43 796.34cm <sup>-1</sup>		8004.278cm <sup>-1</sup>

#### Rot. energy level

### Simulated data generated by PGOPHER<sup>4</sup>.

<sup>&</sup>lt;sup>3</sup>Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

<sup>&</sup>lt;sup>4</sup>Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

### Trans. energy of NO

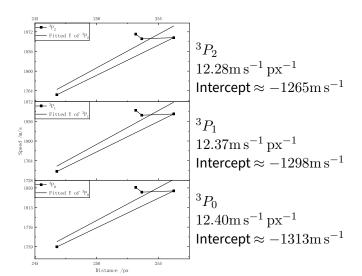
E <sub>int.</sub> (O)	$\begin{split} E_{\rm trans}({\rm total}) &\approx 2.875464 E_{\rm trans}({\rm NO}) \\ &= E_{\rm total} - E_{\rm bond}({\rm O-NO}) - E_{\rm int.}({\rm O}) - E_{\rm int.}({\rm NO}) \end{split}$	$E_{trans}(NO) \\ = \frac{1}{2}m(NO)v^2(NO)$
$^{3}P_{2}$ $(0 \text{cm}^{-1})$	$11081.356 \mathrm{cm}^{-1}$ $10964.609 \mathrm{cm}^{-1}$ $10794.143 \mathrm{cm}^{-1}$ $9398.766 \mathrm{cm}^{-1}$	4375.588cm <sup>-1</sup> 4334.685cm <sup>-1</sup> 4344.824cm <sup>-1</sup> 3870.489cm <sup>-1</sup>
$^{3}P_{1}$ (158.625cm <sup>-1</sup> )	$10922.731 \mathrm{cm}^{-1} \\ 10805.984 \mathrm{cm}^{-1} \\ 10635.518 \mathrm{cm}^{-1} \\ 9240.141 \mathrm{cm}^{-1}$	$4320.423 \text{cm}^{-1}$ $4279.520 \text{cm}^{-1}$ $4289.659 \text{cm}^{-1}$ $3815.324 \text{cm}^{-1}$
$^{3}P_{0}$ (226.977cm <sup>-1</sup> )	$10854.379 \mathrm{cm}^{-1} \ 10737.632 \mathrm{cm}^{-1} \ 10567.166 \mathrm{cm}^{-1} \ 9171.789 \mathrm{cm}^{-1}$	$4296.653 \text{cm}^{-1}$ $4255.749 \text{cm}^{-1}$ $4265.888 \text{cm}^{-1}$ $3791.553 \text{cm}^{-1}$

<sup>&</sup>lt;sup>5</sup>Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.

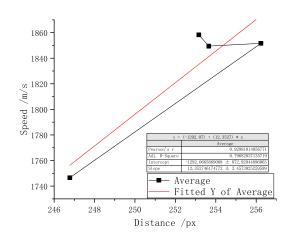
# Trans. speed of NO

E <sub>int.</sub> (O)	$v(NO) = \sqrt{\frac{2E_{trans}(NO)}{m(NO)}}$	$\Delta y$
$^{3}P_{2}$ (0cm <sup>-1</sup> )	$\begin{array}{c} 1867.845\mathrm{ms^{-1}} \\ 1859.094\mathrm{ms^{-1}} \\ 1861.267\mathrm{ms^{-1}} \\ 1756.732\mathrm{ms^{-1}} \end{array}$	253.177 253.650 256.147 246.776
$^{3}P_{1}$ (158.625cm <sup>-1</sup> )	$\begin{array}{c} 1856.033\mathrm{ms^{-1}}\\ 1847.226\mathrm{ms^{-1}}\\ 1849.413\mathrm{ms^{-1}}\\ 1744.168\mathrm{ms^{-1}} \end{array}$	253.177 253.650 256.148 246.776
$^{3}P_{0}$ (226.977cm <sup>-1</sup> )	$\begin{array}{c} 1850.920 \mathrm{m  s}^{-1} \\ 1842.089 \mathrm{m  s}^{-1} \\ 1844.282 \mathrm{m  s}^{-1} \\ 1738.726 \mathrm{m  s}^{-1} \end{array}$	253.177 253.650 256.147 246.776

### Trans. speed of NO



### Trans. speed of NO



# Average $12.35 \rm m\,s^{-1}\,px^{-1}$ Intercept $\approx -1292 \rm m\,s^{-1}$

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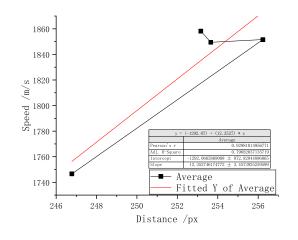
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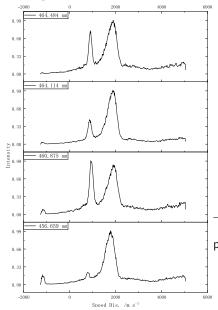
Peak assignments

Speed correction

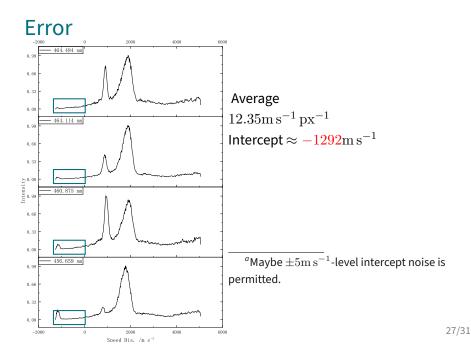


Average  $12.35 \mathrm{m\,s^{-1}\,px^{-1}}$  Intercept  $\approx -1292 \mathrm{m\,s^{-1}}$ 

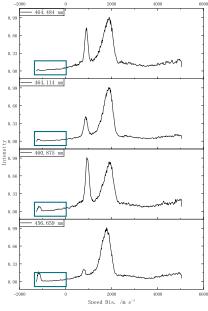




 $^{\sigma}\mathrm{Maybe}\pm5\mathrm{m\,s}^{-1}\text{-level}$  intercept noise is permitted.







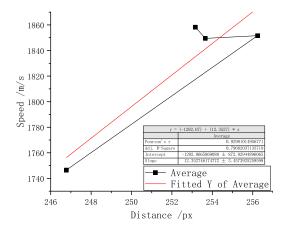
#### Average

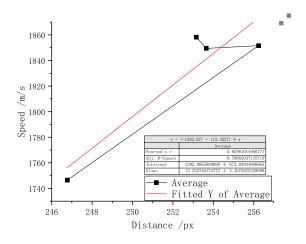
 $12.35\mathrm{m\,s^{-1}\,px^{-1}}$  Intercept  $\approx -1292\mathrm{m\,s^{-1}}$ 

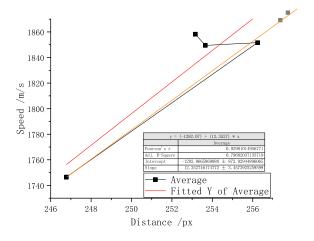
#### **Notice**

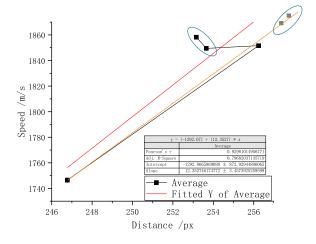
What we are calculating here are actually  $|\mathbf{v}_{\mathrm{NO}}|$ , which are not supposed to be minus  $^a$ .

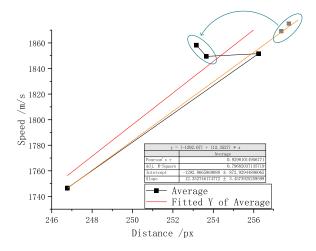
 $<sup>^{</sup>a}$ Maybe  $\pm 5 \mathrm{m \ s^{-1}}$ -level intercept noise is permitted.

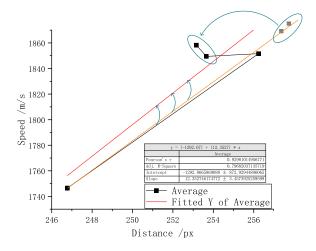


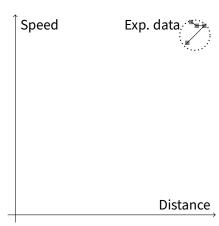


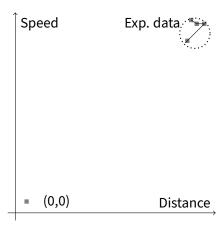


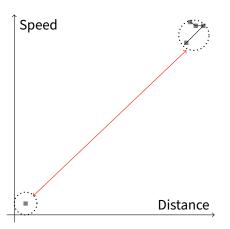






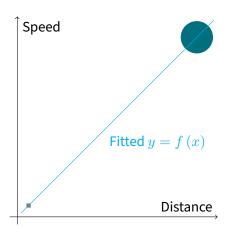






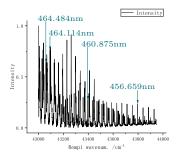
If we assume a virtual zero point: The fake data obtains a huge weight! Statistics tools always treat all data as proper indications.



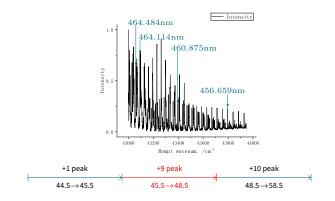


If we assume a virtual zero point: The fake data obtains a huge weight! Statistics tools always treat all data as proper indications.

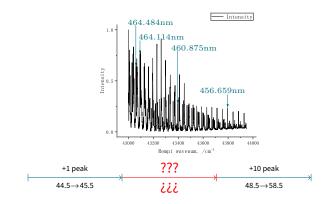
⇒ After assignments, which are the points we should use?



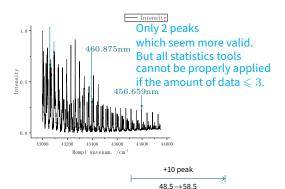
$464.484$ nm $\approx 43058.49$ cm <sup>-1</sup>	$464.114$ nm $\approx 43092.81$ cm <sup>-1</sup>	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	$456.659$ nm $\approx 43796.34$ cm <sup>-1</sup>
px = 253.162	px = 253.655	px = 256.240	px = 246.776
rR2 (44.5)	rR2~(45.5)	sR21 (48.5)	sR21 (58.5)



$464.484$ nm $\approx 43058.49$ cm <sup>-1</sup>	$464.114$ nm $\approx 43092.81$ cm <sup>-1</sup>	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	$456.659$ nm $\approx 43796.34$ cm <sup>-1</sup>
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460.875 nm $\approx 43395.69 \text{cm}^{-1}$	$456.659$ nm $\approx 43796.34$ cm <sup>-1</sup>
px = 256.240	px = 246.776
sR21 (48.5)	sR21 (58.5)

#### Reference

- [1] Rémy Jost et al. The Journal of Chemical Physics 105.3 (July 1996).
- [2] Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.
- [3] Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer 186 (2017), pp. 221–242.