練習実験報告

肖宇笑 May 27, 2024

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Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

Galvano Sepctrum

REMPI scan

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Peak assignments

Galvano Sepctrum

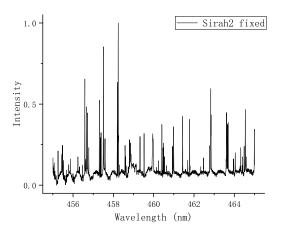


Fig. 1: Wavelen. correction

Galvano Sepctrum

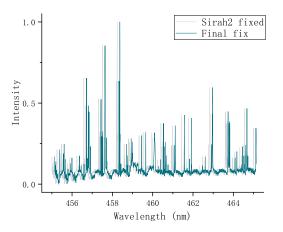


Fig. 1: Wavelen. correction

Galvano Sepctrum

Calibration



Fig. 2: Correction function

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

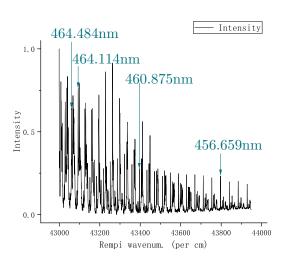
Galvano Sepctrum

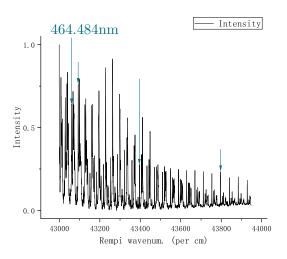
REMPI scan

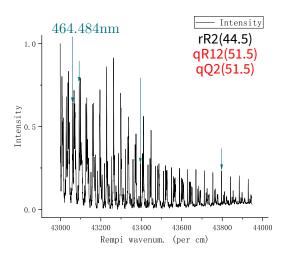
Selected peaks

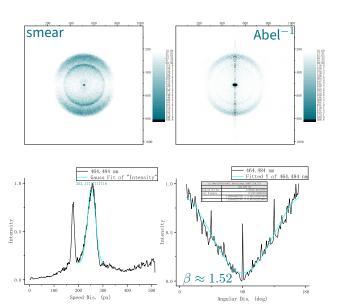
Peak assignments

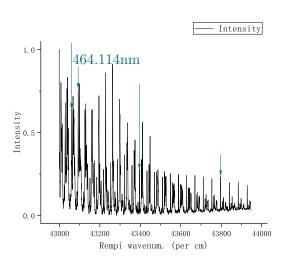
Selected peaks

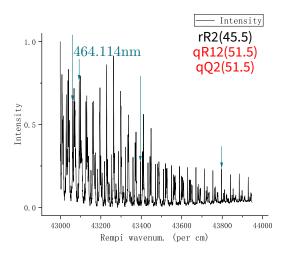


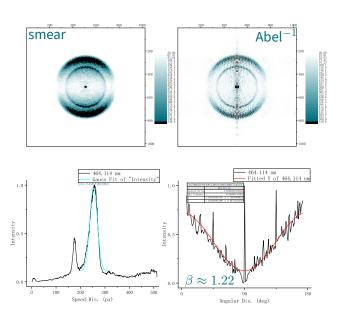


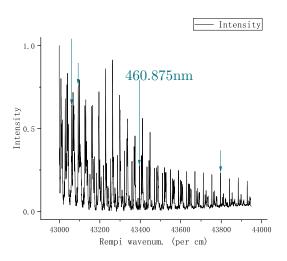


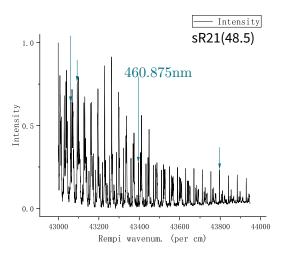


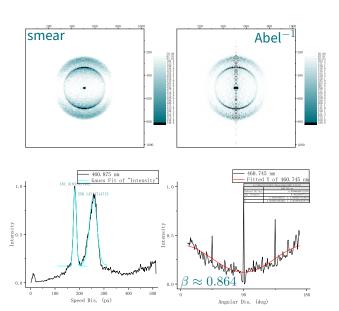


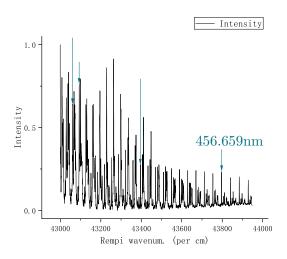


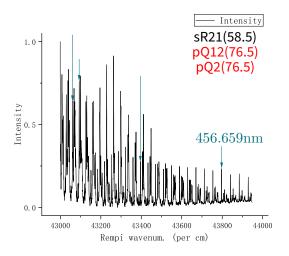


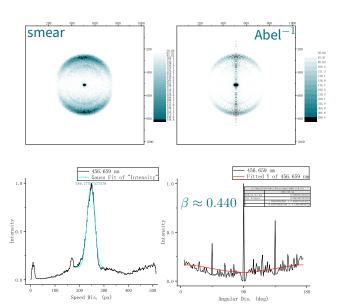












Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

Peak assignments

| 464.484 nm ≈ 43058.49 cm ⁻¹ | 464.114 nm ≈ 43092.81 cm ⁻¹ | 460.875 nm ≈ 43395.69 cm ⁻¹ | 456.659 nm ≈ 43796.34 cm ⁻¹ |
|--|--|--|--|
| px = 253.177 | px = 253.650 | px = 181.319 & 256.147 | px = 246.776 |
| rR2 (44.5) $qR12 (51.5)$ $qQ2 (51.5)$ | rR2 (45.5) qR12 (51.5) qQ2 (51.5) | sR21 (48.5) | sR21 (58.5) pQ12 (76.5) pP2 (76.5) |

Notice

Colored assignments are mismatched, and will not be used to calculate.

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

\boldsymbol{y} trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^1$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|--|-----------------------------------|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | $\Delta E_v(1 \to 0) + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | $\Delta E_v(1 \to 0) + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $\Delta E_v(1 \to 0) + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | - | $\Delta E_v(1 \to 0) + E(J = 58)$ |

¹Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

\boldsymbol{y} trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^2$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|--|---|
| Peak 1 464.484nm | 43058.49 cm $^{-1}$ | | $2341.9327750 \text{cm}^{-1} + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $2341.9327750 \text{cm}^{-1} + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 58)$ |

²Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

³J. Danielak et al. *Journal of Molecular Spectroscopy* **181**.2 (1997), pp. 394–402.

y trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O-NO})^2$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|-----------------------------|---|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | _ | $2341.9327750 \text{cm}^{-1} + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $2341.9327750 \text{cm}^{-1} + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 58)$ |

Vib. energy level³

$$E_v = \omega_e \left(v + \frac{1}{2} \right) - \omega_e x_e \left(v + \frac{1}{2} \right)^2 + \omega_e y_e \left(v + \frac{1}{2} \right)^3.$$

²Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

³J. Danielak et al. Journal of Molecular Spectroscopy **181**.2 (1997), pp. 394–402.

\boldsymbol{y} trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^4$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|--|--------------------------|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | 5814.033cm ⁻¹ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | 5965.969cm ⁻¹ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | 6239.696cm ⁻¹ |
| Peak 4 456.659nm | 43796.34 cm $^{-1}$ | | 8004.278cm ⁻¹ |

⁴Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

⁵Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

y trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O-NO})^4$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|-----------------------------|--------------------------|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | 5814.033cm ⁻¹ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | 5965.969cm ⁻¹ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | 6239.696cm ⁻¹ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | 8004.278cm ⁻¹ |

Rot. energy level

Simulated data generated by PGOPHER⁵.

⁴Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

⁵Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

\boldsymbol{y} trans. energy of NO

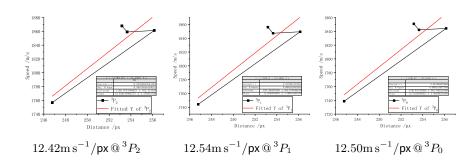
| E _{int.} (O) | $\begin{split} E_{\rm trans}({\rm total}) &\approx 2.875464 E_{\rm trans}({\rm NO}) \\ &= E_{\rm total} - E_{\rm bond}({\rm O-NO}) - E_{\rm int.}({\rm O}) - E_{\rm int.}({\rm NO}) \end{split}$ | $E_{\text{trans}}(\text{NO})$ $= \frac{1}{2}m(\text{NO})v^2(\text{NO})$ |
|--|--|--|
| $^{3}P_{2}$ (0cm^{-1}) | $11081.356 \mathrm{cm}^{-1}$ $10964.609 \mathrm{cm}^{-1}$ $10794.143 \mathrm{cm}^{-1}$ $9398.766 \mathrm{cm}^{-1}$ | 4375.588cm ⁻¹ 4334.685cm ⁻¹ 4344.824cm ⁻¹ 3870.489cm ⁻¹ |
| ³ P ₁ (158.625cm ⁻¹) | $10922.731 \mathrm{cm}^{-1}$ $10805.984 \mathrm{cm}^{-1}$ $10635.518 \mathrm{cm}^{-1}$ $9240.141 \mathrm{cm}^{-1}$ | 4320.423cm^{-1} 4279.520cm^{-1} 4289.659cm^{-1} 3815.324cm^{-1} |
| $^{3}P_{0}$ (226.977cm ⁻¹) | $10854.379 \mathrm{cm}^{-1}$ $10737.632 \mathrm{cm}^{-1}$ $10567.166 \mathrm{cm}^{-1}$ $9171.789 \mathrm{cm}^{-1}$ | 4296.653cm ⁻¹ 4255.749cm ⁻¹ 4265.888cm ⁻¹ 3791.553cm ⁻¹ |

⁶Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.

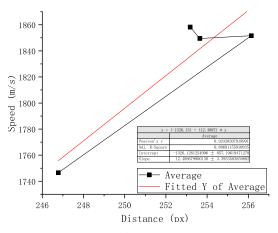
y Trans. speed of NO

| E _{int.} (O) | $v(\text{NO}) = \sqrt{\frac{2E_{\text{trans}}(\text{NO})}{m(\text{NO})}}$ | Δy |
|--|--|--|
| $^{3}P_{2}$ (0cm ⁻¹) | $1867.845 \mathrm{m s}^{-1} \\ 1859.094 \mathrm{m s}^{-1} \\ 1861.267 \mathrm{m s}^{-1} \\ 1756.732 \mathrm{m s}^{-1}$ | 253.177 253.650 256.147 246.776 |
| $^{3}P_{1}$ (158.625cm ⁻¹) | $1856.033 \mathrm{m s^{-1}} \\ 1847.226 \mathrm{m s^{-1}} \\ 1849.413 \mathrm{m s^{-1}} \\ 1744.168 \mathrm{m s^{-1}}$ | 253.177 253.650 256.148 246.776 |
| $^{3}P_{0}$ (226.977cm ⁻¹) | $1850.920 \mathrm{m s}^{-1} \\ 1842.089 \mathrm{m s}^{-1} \\ 1844.282 \mathrm{m s}^{-1} \\ 1738.726 \mathrm{m s}^{-1}$ | 253.177 253.650 256.147 246.776 |

y Trans. speed of NO



y Trans. speed of NO



 $12.49 \mathrm{m\,s^{-1}/px}$ @ Average

Reference

- J. Danielak et al. Journal of Molecular Spectroscopy 181.2 (1997), pp. 394–402.
- [2] Rémy Jost et al. The Journal of Chemical Physics 105.3 (July 1996).
- [3] Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.
- [4] Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer 186 (2017), pp. 221–242.