練習実験報告

肖宇笑 2024年6月18日

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

Speed correction

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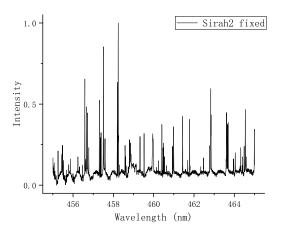


Fig. 1: Wavelen. correction

Galvano Sepctrum

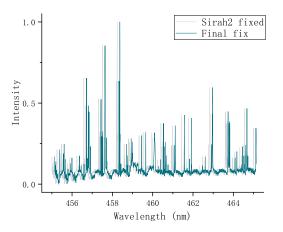


Fig. 1: Wavelen. correction

Galvano Sepctrum

Calibration

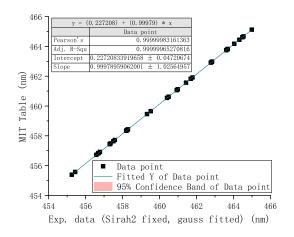


Fig. 2: Correction function

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REMPI scan

Selected peaks

Peak assignments

Speed correction

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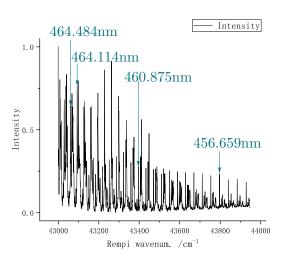
REMPI scan

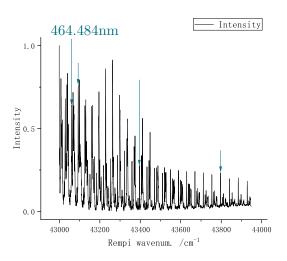
Selected peaks

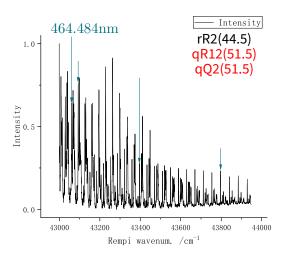
Peak assignments

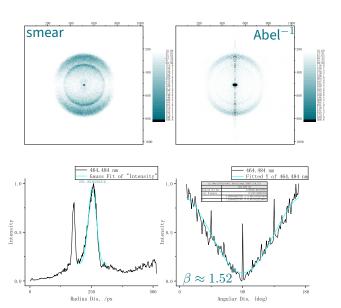
Speed correction

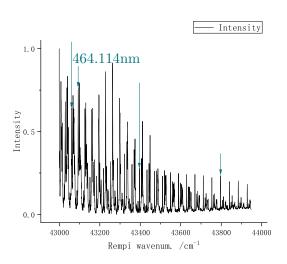
Selected peaks

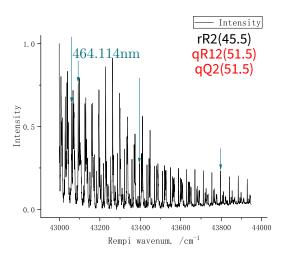


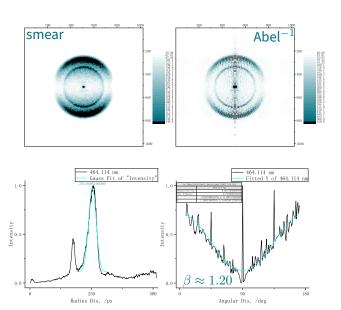


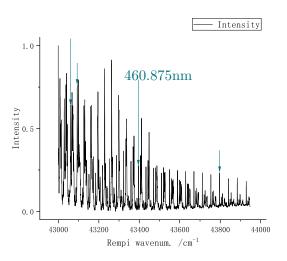


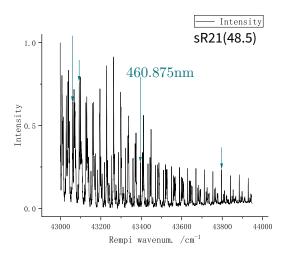


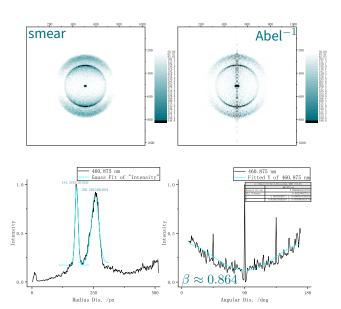


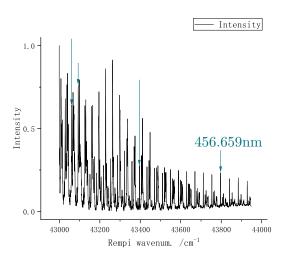


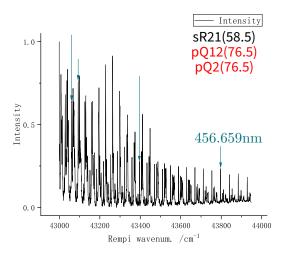


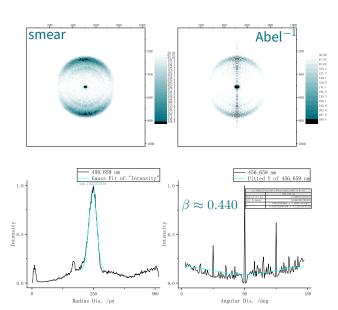












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REMPI scan

Selected peaks

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464.484 nm ≈ 43058.49 cm ⁻¹	464.114 nm ≈ 43092.81 cm ⁻¹	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	456.659 nm ≈ 43796.34 cm ⁻¹
px = 253.162	px = 253.655	px = 181.319 & 256.240	px = 246.776
rR2 (44.5) qR12 (51.5) qQ2 (51.5)	rR2 (45.5) qR12 (51.5) qQ2 (51.5)	sR21 (48.5)	sR21 (58.5) pQ12 (76.5) pP2 (76.5)

Peak assignments

464.484 nm ≈ 43058.49 cm ⁻¹	464.114 nm ≈ 43092.81 cm ⁻¹	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	456.659 nm ≈ 43796.34 cm ⁻¹
px = 253.162	px = 253.655	px = 181.319 & 256.240	px = 246.776
rR2 (44.5) $qR12 (51.5)$ $qQ2 (51.5)$	rR2 (45.5) qR12 (51.5) qQ2 (51.5)	sR21 (48.5)	sR21 (58.5) pQ12 (76.5) pP2 (76.5)

Notice

Colored assignments are mismatched, and will not be used to calculate.

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Trans. energy of NO

	E_{total}	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^1$	$E_{int.}(NO)$
Peak 1 464.484nm	43058.49 cm $^{-1}$		$\Delta E_v(1 \to 0) + E(J = 44)$
Peak 2 464.114nm	43 092.81cm ⁻¹		$\Delta E_v(1 \to 0) + E(J = 45)$
Peak 3 460.875nm	43 395.69cm ⁻¹	25 128.57cm ⁻¹	$\Delta E_v(1 \to 0) + E(J = 48)$
Peak 4 456.659nm	43 796.34cm ⁻¹		$\Delta E_v(1 \to 0) + E(J = 58)$

¹Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

Trans. energy of NO

	E_{total}	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^2$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49\mathrm{cm}^{-1}$		$2341.9327750 \text{cm}^{-1} + E(J = 44)$
Peak 2 464.114nm	43 092.81cm ⁻¹		$2341.9327750 \text{cm}^{-1} + E(J = 45)$
Peak 3 460.875nm	43 395.69cm ⁻¹	25 128.57cm ⁻¹	$2341.9327750 \text{cm}^{-1} + E(J = 48)$
Peak 4 456.659nm	43 796.34cm ⁻¹	-	$2341.9327750 \text{cm}^{-1} + E(J = 58)$

²Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

Trans. energy of NO

	E_{total}	$E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^2$	$E_{int.}(NO)$
Peak 1 464.484nm	$43058.49\mathrm{cm}^{-1}$		$2341.9327750 \text{cm}^{-1} + E(J = 44)$
Peak 2 464.114nm	43 092.81cm ⁻¹		$2341.9327750 \text{cm}^{-1} + E(J = 45)$
Peak 3 460.875nm	43 395.69cm ⁻¹	25 128.57cm ⁻¹	$2341.9327750 \text{cm}^{-1} + E(J = 48)$
Peak 4 456.659nm	43796.34cm ⁻¹		$2341.9327750 \text{cm}^{-1} + E(J = 58)$

Vib. energy level

$$E_v = \omega_e \left(v + \frac{1}{2} \right) - \omega_e x_e \left(v + \frac{1}{2} \right)^2 + \omega_e y_e \left(v + \frac{1}{2} \right)^3.$$

²Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

Trans. energy of NO

	E_{total}	$E_{bond}(\mathrm{O}\mathrm{-NO})^3$	$E_{int.}(NO)$
Peak 1 464.484nm	43058.49 cm $^{-1}$		5814.033cm ⁻¹
Peak 2 464.114nm	43 092.81cm ⁻¹		5965.969cm ⁻¹
Peak 3 460.875nm	43 395.69cm ⁻¹	25 128.57cm ⁻¹	6239.696cm ⁻¹
Peak 4 456.659nm	43796.34 cm $^{-1}$	•	8004.278cm ⁻¹

³Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

⁴Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

Trans. energy of NO

	E_{total}	$E_{bond}(\mathrm{O}\mathrm{-NO})^3$	$E_{int.}(NO)$
Peak 1 464.484nm	43 058.49cm ⁻¹		5814.033 cm $^{-1}$
Peak 2 464.114nm	43 092.81cm ⁻¹		5965.969cm ⁻¹
Peak 3 460.875nm	43 395.69cm ⁻¹	25 128.57cm ⁻¹	6239.696cm ⁻¹
Peak 4 456.659nm	43 796.34cm ⁻¹		8004.278cm ⁻¹

Rot. energy level

Simulated data generated by PGOPHER⁴.

³Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

⁴Colin M. Western. *Journal of Quantitative Spectroscopy and Radiative Transfer* **186** (2017), pp. 221–242.

Trans. energy of NO

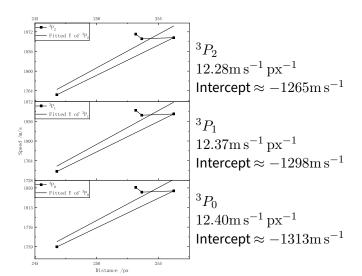
E _{int.} (O)	$\begin{split} E_{\rm trans}({\rm total}) &\approx 2.875464 E_{\rm trans}({\rm NO}) \\ &= E_{\rm total} - E_{\rm bond}({\rm O-NO}) - E_{\rm int.}({\rm O}) - E_{\rm int.}({\rm NO}) \end{split}$	$E_{trans}(NO) \\ = \frac{1}{2}m(NO)v^2(NO)$
$^{3}P_{2}$ (0cm^{-1})	$11081.356 \mathrm{cm}^{-1}$ $10964.609 \mathrm{cm}^{-1}$ $10794.143 \mathrm{cm}^{-1}$ $9398.766 \mathrm{cm}^{-1}$	4375.588cm ⁻¹ 4334.685cm ⁻¹ 4344.824cm ⁻¹ 3870.489cm ⁻¹
$^{3}P_{1}$ (158.625cm ⁻¹)	$10922.731 \mathrm{cm}^{-1} \\ 10805.984 \mathrm{cm}^{-1} \\ 10635.518 \mathrm{cm}^{-1} \\ 9240.141 \mathrm{cm}^{-1}$	4320.423cm^{-1} 4279.520cm^{-1} 4289.659cm^{-1} 3815.324cm^{-1}
$^{3}P_{0}$ (226.977cm ⁻¹)	$10854.379 \mathrm{cm}^{-1} \ 10737.632 \mathrm{cm}^{-1} \ 10567.166 \mathrm{cm}^{-1} \ 9171.789 \mathrm{cm}^{-1}$	4296.653cm^{-1} 4255.749cm^{-1} 4265.888cm^{-1} 3791.553cm^{-1}

⁵Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.

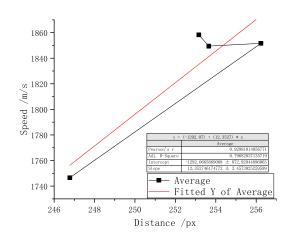
Trans. speed of NO

E _{int.} (O)	$v(NO) = \sqrt{\frac{2E_{trans}(NO)}{m(NO)}}$	Δy
$^{3}P_{2}$ (0cm ⁻¹)	$\begin{array}{c} 1867.845\mathrm{ms^{-1}} \\ 1859.094\mathrm{ms^{-1}} \\ 1861.267\mathrm{ms^{-1}} \\ 1756.732\mathrm{ms^{-1}} \end{array}$	253.177 253.650 256.147 246.776
$^{3}P_{1}$ (158.625cm ⁻¹)	$\begin{array}{c} 1856.033\mathrm{ms^{-1}}\\ 1847.226\mathrm{ms^{-1}}\\ 1849.413\mathrm{ms^{-1}}\\ 1744.168\mathrm{ms^{-1}} \end{array}$	253.177 253.650 256.148 246.776
$^{3}P_{0}$ (226.977cm ⁻¹)	$\begin{array}{c} 1850.920 \mathrm{m s}^{-1} \\ 1842.089 \mathrm{m s}^{-1} \\ 1844.282 \mathrm{m s}^{-1} \\ 1738.726 \mathrm{m s}^{-1} \end{array}$	253.177 253.650 256.147 246.776

Trans. speed of NO



Trans. speed of NO



Average $12.35 \rm m\,s^{-1}\,px^{-1}$ Intercept $\approx -1292 \rm m\,s^{-1}$

Contents

Galvano Sepctrum

REMPI scan

Selected peaks

Peak assignments

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Contents

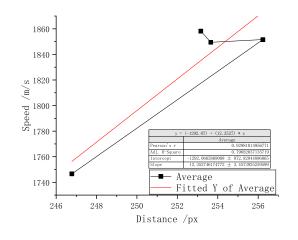
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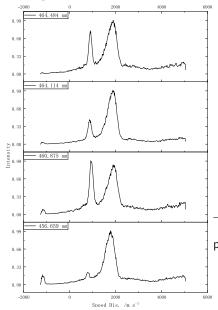
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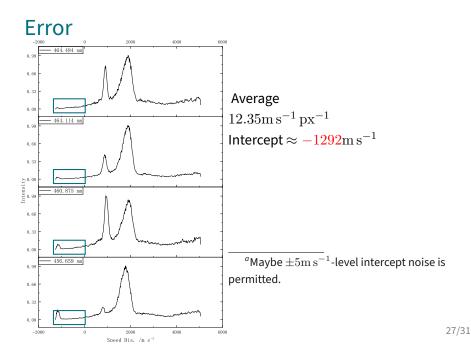


Average $12.35 \mathrm{m\,s^{-1}\,px^{-1}}$ Intercept $\approx -1292 \mathrm{m\,s^{-1}}$

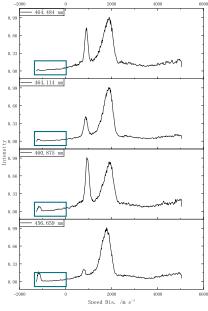




 $^{\sigma}\mathrm{Maybe}\pm5\mathrm{m\,s}^{-1}\text{-level}$ intercept noise is permitted.







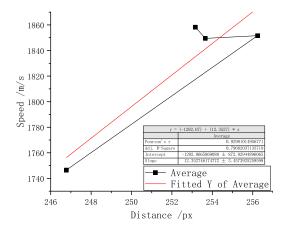
Average

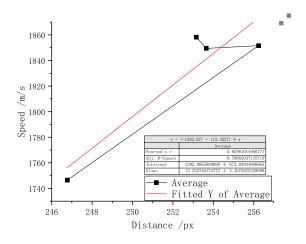
 $12.35\mathrm{m\,s^{-1}\,px^{-1}}$ Intercept $\approx -1292\mathrm{m\,s^{-1}}$

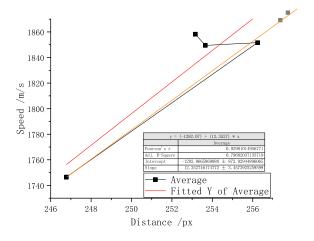
Notice

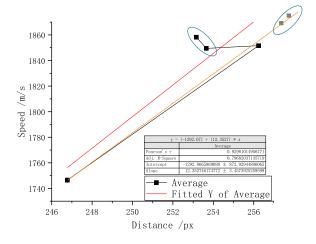
What we are calculating here are actually $|\mathbf{v}_{\mathrm{NO}}|$, which are not supposed to be minus a .

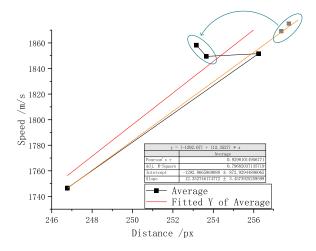
 $^{^{}a}$ Maybe $\pm 5 \mathrm{m \ s^{-1}}$ -level intercept noise is permitted.

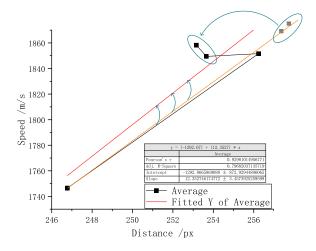


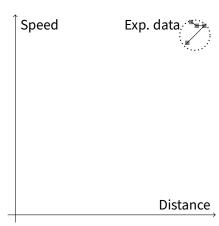


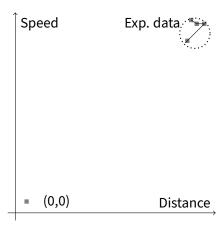


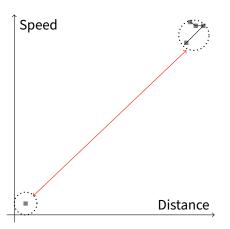






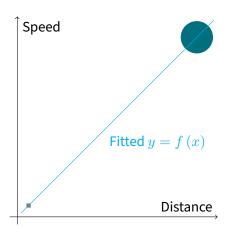






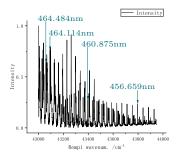
If we assume a virtual zero point: The fake data obtains a huge weight! Statistics tools always treat all data as proper indications.



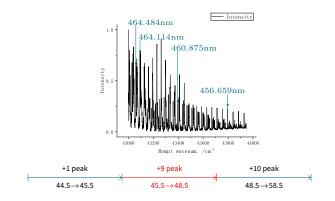


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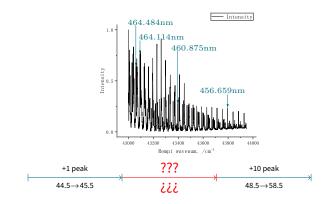
⇒ After assignments, which are the points we should use?



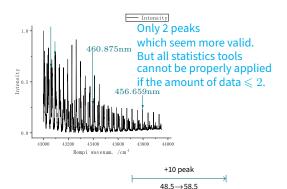
464.484 nm ≈ 43058.49 cm ⁻¹	464.114 nm ≈ 43092.81 cm ⁻¹	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	456.659 nm ≈ 43796.34 cm ⁻¹
px = 253.162	px = 253.655	px = 256.240	px = 246.776
rR2 (44.5)	rR2~(45.5)	sR21 (48.5)	sR21 (58.5)



464.484 nm ≈ 43058.49 cm ⁻¹	464.114 nm ≈ 43092.81 cm ⁻¹	460.875 nm $\approx 43395.69 \text{cm}^{-1}$	456.659 nm ≈ 43796.34 cm ⁻¹
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px = 256.240	px = 246.776
sR21 (48.5)	sR21 (58.5)

Reference

- [1] Rémy Jost et al. The Journal of Chemical Physics 105.3 (July 1996).
- [2] Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.
- [3] Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer 186 (2017), pp. 221–242.