練習実験報告

肖宇笑

May 22, 2024

Galvano Sepctrum

REMPI scan

Selected peaks

Radius and angular distributions

Peak assignments

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Galvano Sepctrum

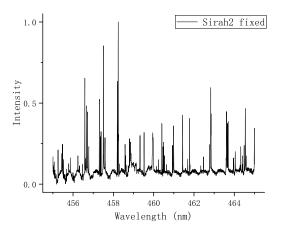


Fig. 1: Wavelen. correction

Galvano Sepctrum



Fig. 1: Wavelen. correction

Galvano Sepctrum

Correction



Fig. 2: Correction function

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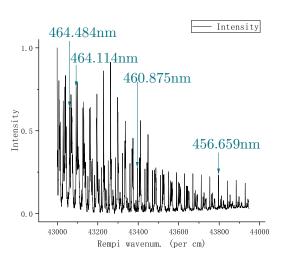
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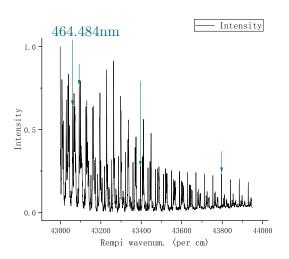
Galvano Sepctrum

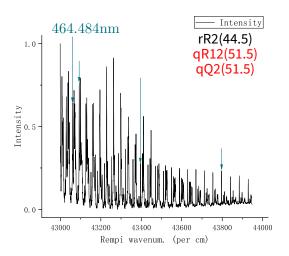
REMPI scan

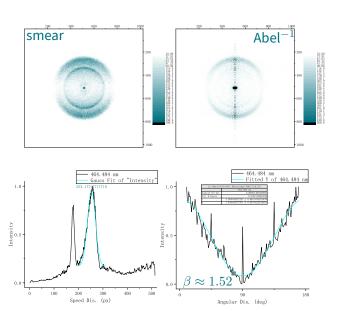
Selected peaks

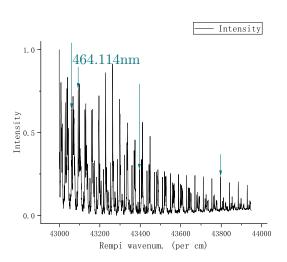
Radius and angular distributions

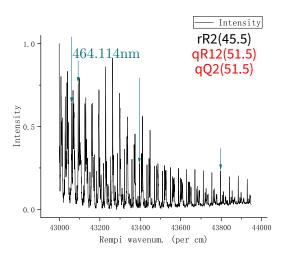
Peak assignments

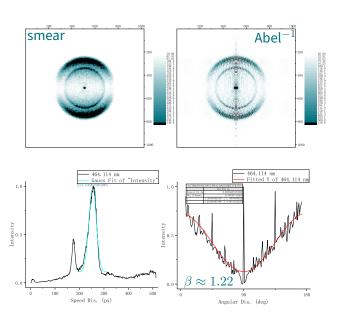


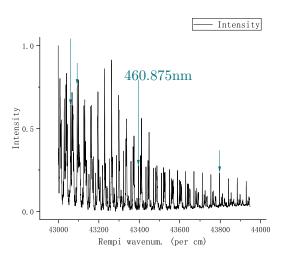


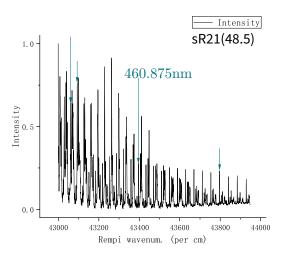


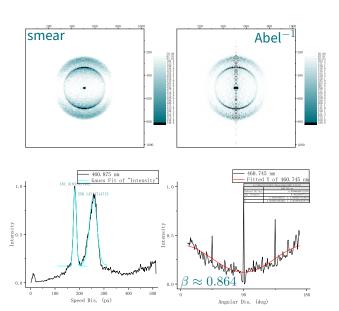


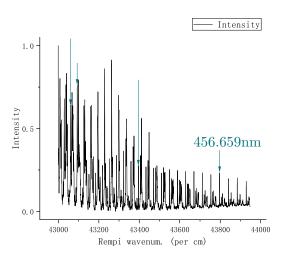


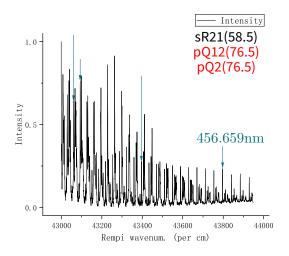


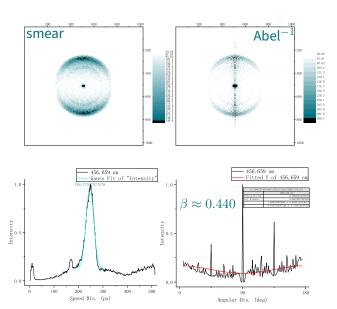












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| 464.484 nm ≈ 43058.49 cm ⁻¹ | 464.114 nm ≈ 43092.81 cm ⁻¹ | 460.875 nm ≈ 43395.69 cm ⁻¹ | 456.659 nm ≈ 43796.34 cm |
|--|--|--|---|
| px = 253.177 | px = 253.650 | px = 181.319 & 256.147 | px = 246.776 |
| $rR2\ 44.5$ $qR12\ 51.5$ $qQ2\ 51.5$ | $rR2\ 45.5$ $qR12\ 51.5$ $qQ2\ 51.5$ | sR21~48.5 | $sR21\ 58.5$ $pQ12\ 76.5$ $pP2\ 76.5$ |

Notice

Colored assignments are mismatched, and will not be used to calculate.

Galvano Sepctrum

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\boldsymbol{y} trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O}\!-\!\mathrm{NO})^1$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|--|-----------------------------------|
| Peak 1 464.484nm | 43058.49 cm $^{-1}$ | | $\Delta E_v(1 \to 0) + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | - | $\Delta E_v(1 \to 0) + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $\Delta E_v(1 \to 0) + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | - | $\Delta E_v(1 \to 0) + E(J = 58)$ |

¹Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

\boldsymbol{y} trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O-NO})^2$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|-----------------------------|---|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $2341.9327750 \text{cm}^{-1} + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 58)$ |

²Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

³J. Danielak et al. *Journal of Molecular Spectroscopy* **181**.2 (1997), pp. 394–402.

y trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O-NO})^2$ | $E_{int.}(\mathrm{NO})$ |
|---------------------|---------------------------|-----------------------------|---|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 44)$ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | _ | $2341.9327750 \text{cm}^{-1} + E(J = 45)$ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | $2341.9327750 \text{cm}^{-1} + E(J = 48)$ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | $2341.9327750 \text{cm}^{-1} + E(J = 58)$ |

Vib. energy level³

$$E_v = \omega_e \left(v + \frac{1}{2} \right) - \omega_e x_e \left(v + \frac{1}{2} \right)^2 + \omega_e y_e \left(v + \frac{1}{2} \right)^3.$$

²Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

³J. Danielak et al. Journal of Molecular Spectroscopy **181**.2 (1997), pp. 394–402.

y trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O}\mathrm{-NO})^4$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|--------------------------------------|--------------------------|
| Peak 1 464.484nm | 43058.49 cm $^{-1}$ | | 5814.033cm ⁻¹ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | 5965.969cm ⁻¹ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | 6239.696cm ⁻¹ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | 8004.278cm ⁻¹ |

(2017), pp. 221–242.

⁴Rémy Jost et al. *The Journal of Chemical Physics* **105**.3 (July 1996).

⁵J. Danielak et al. *Journal of Molecular Spectroscopy* **181**.2 (1997), pp. 394–402.

⁶Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer **186**

y trans. energy of NO

| | E_{total} | $E_{bond}(\mathrm{O-NO})^4$ | $E_{int.}(NO)$ |
|---------------------|---------------------------|-----------------------------|--------------------------|
| Peak 1 464.484nm | 43 058.49cm ⁻¹ | | 5814.033cm ⁻¹ |
| Peak 2 464.114nm | 43 092.81cm ⁻¹ | | 5965.969cm ⁻¹ |
| Peak 3 460.875nm | 43 395.69cm ⁻¹ | 25 128.57cm ⁻¹ | 6239.696cm ⁻¹ |
| Peak 4 456.659nm | 43 796.34cm ⁻¹ | | 8004.278cm ⁻¹ |

Rot. energy level⁵

Simulated data generated by PGOPHER⁶.

⁶Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer **186**

(2017), pp. 221–242.

⁴Rémy Jost et al. The Journal of Chemical Physics **105**.3 (July 1996).

⁵J. Danielak et al. Journal of Molecular Spectroscopy **181**.2 (1997), pp. 394–402.

y trans. energy of NO

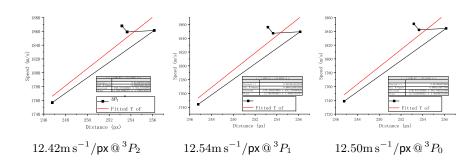
| F _{int.} (O) | $\begin{split} E_{trans}(total) &\approx 2.875464 E_{trans}(NO) \\ &= E_{total} - E_{bond}(O-NO) - E_{int.}(O) - E_{int.}(NO) \end{split}$ | $E_{trans}(NO) \\ = \frac{1}{2}m(NO)v^2(NO)$ |
|--|--|--|
| $^{3}P_{2}$ (0cm^{-1}) | $11081.356 \mathrm{cm}^{-1}$ $10964.609 \mathrm{cm}^{-1}$ $10794.143 \mathrm{cm}^{-1}$ $9398.766 \mathrm{cm}^{-1}$ | 4375.588cm ⁻¹ 4334.685cm ⁻¹ 4344.824cm ⁻¹ 3870.489cm ⁻¹ |
| ³ P ₁ (158.625cm ⁻¹) | $10922.731 \mathrm{cm}^{-1}$ $10805.984 \mathrm{cm}^{-1}$ $10635.518 \mathrm{cm}^{-1}$ $9240.141 \mathrm{cm}^{-1}$ | 4320.423cm^{-1} 4279.520cm^{-1} 4289.659cm^{-1} 3815.324cm^{-1} |
| $^{3}P_{0}$ (226.977cm ⁻¹) | $10854.379 \mathrm{cm}^{-1} \ 10737.632 \mathrm{cm}^{-1} \ 10567.166 \mathrm{cm}^{-1} \ 9171.789 \mathrm{cm}^{-1}$ | 4296.653cm^{-1} 4255.749cm^{-1} 4265.888cm^{-1} 3791.553cm^{-1} |

⁷Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.

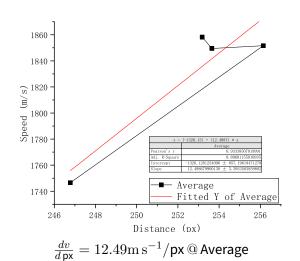
y Trans. speed of NO

| E _{int.} (O) | $v(NO) = \sqrt{\frac{2E_{trans}(NO)}{m(NO)}}$ | Δy |
|--|--|--|
| $^{3}P_{2}$ (0cm ⁻¹) | $1867.845 \mathrm{m s}^{-1} \\ 1859.094 \mathrm{m s}^{-1} \\ 1861.267 \mathrm{m s}^{-1} \\ 1756.732 \mathrm{m s}^{-1}$ | 253.177 253.650 256.147 246.776 |
| $^{3}P_{1}$ (158.625cm ⁻¹) | $1856.033 \mathrm{m s^{-1}} \\ 1847.226 \mathrm{m s^{-1}} \\ 1849.413 \mathrm{m s^{-1}} \\ 1744.168 \mathrm{m s^{-1}}$ | 253.177 253.650 256.148 246.776 |
| $^{3}P_{0}$ (226.977cm ⁻¹) | $1850.920 \mathrm{m s}^{-1} \\ 1842.089 \mathrm{m s}^{-1} \\ 1844.282 \mathrm{m s}^{-1} \\ 1738.726 \mathrm{m s}^{-1}$ | 253.177 253.650 256.147 246.776 |

y Trans. speed of NO



y Trans. speed of NO



Reference

- J. Danielak et al. Journal of Molecular Spectroscopy 181.2 (1997), pp. 394–402.
- [2] Rémy Jost et al. The Journal of Chemical Physics 105.3 (July 1996).
- [3] Charlotte Emma Moore and Jean W. Gallagher. "Tables of spectra of hydrogen, carbon, nitrogen, and oxygen atoms and ions". 1993.
- [4] Colin M. Western. Journal of Quantitative Spectroscopy and Radiative Transfer 186 (2017), pp. 221–242.

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