

Q: Detailed explanation of the “A glimpse into the solution process using the Schrödinger Equation, including separation of variables and application of boundary conditions. “

A: Solving the Infinite Quantum Well with Separation of Variables and Boundary Conditions

The glimpse you saw likely involved a step in solving the Schrödinger Equation (SE) for the infinite quantum well with specific boundary conditions. Here’s a breakdown of the general approach:

The Mighty Schrödinger Equation (SE)

The SE is the fundamental equation in quantum mechanics that relates the total energy (E) of a particle to its wave function (ψ). In the case of the infinite quantum well, the SE takes the following form:

$$-\frac{\hbar^2}{2m} \left(\frac{d^2\psi}{dx^2} \right) + V(x) \cdot \psi(x) = E \cdot \psi(x)$$

where:

- \hbar (pronounced “h-bar”) is the reduced Planck constant
- m is the mass of the particle
- $\frac{d^2\psi}{dx^2}$ is the second derivative of the wave function with respect to position (x)
- $V(x)$ is the potential energy function (zero inside the well, infinite outside)
- E is the total energy of the particle

Separation of Variables: A Clever Trick

To solve this equation, we often use a technique called separation of variables. This assumes the wave function can be written as a product of two separate functions:

$$\psi(x) = X(x) \cdot T(t)$$

In the context of the infinite well, we’re usually interested in the time-independent solutions, so $T(t)$ becomes a constant. This leaves us with:

$$\psi(x) = X(x)$$

We plug this separated form back into the SE and manipulate the equation to obtain two ordinary differential equations (ODEs):

- One ODE for the spatial part ($X(x)$) that depends on the position (x) and the particle’s energy (E).
- Another ODE for the time-dependent part ($T(t)$) in case we’re considering a time-dependent SE.

Solving for the Spatial Part ($X(x)$)

In the infinite quantum well, since the potential energy ($V(x)$) is zero inside the well, the ODE for the spatial part becomes:

$$-\frac{\hbar^2}{2m} \left(\frac{d^2X}{dx^2} \right) = E \cdot X$$

This equation has solutions of the form:

$$X(x) = A \cdot \sin(kx) + B \cdot \cos(kx)$$

where:

(continued)

- A and B are constants determined by the boundary conditions
- k is the wave number (related to the wavelength λ by $k = \frac{2\pi}{\lambda}$)

Applying the Boundary Conditions: The Filter

The key aspect you saw a glimpse of is applying the boundary conditions to this solution. These conditions act as filters, selecting the allowed wave functions and consequently, the energy levels.

For the infinite quantum well, the common boundary conditions are:

- $\psi(0) = 0$: The wave function must be zero at the left wall ($x = 0$).
- $\psi(a) = 0$: The wave function must be zero at the right wall ($x = a$, where a is the width of the well).

Applying these conditions to the solution for $X(x)$ leads to:

- $\psi(0) = A \cdot \sin(0) + B \cdot \cos(0) = 0$: This implies $B = 0$ (since $\cos(0) = 1$)
- $\psi(a) = A \cdot \sin(ka) = 0$: Since A cannot be zero (otherwise, the entire wave function would be zero), this implies $\sin(ka) = 0$.

Quantization of Energy Levels: A Consequence

For $\sin(ka)$ to be zero, ka must be an integer multiple of π (since $\sin(n\pi) = 0$ for any integer n). This means:

$$ka = n\pi$$

Since k is related to the wavelength (λ) through $k = \frac{2\pi}{\lambda}$, we can rewrite this as:

$$n \cdot \lambda = 2a$$

This equation tells us that only specific wavelengths (and their corresponding wave functions) are allowed within the well. The allowed wavelengths must be integer multiples of twice the well width ($2a$). This results in the **quantization of energy levels**. The particle can only exist in specific energy states determined by the allowed wavelengths.

The glimpse you saw likely involved manipulating the equations related to applying the boundary conditions and showing how this leads to the quantization of energy levels.

- In reality, solving