

## Chemistry Tutorial



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### ***Reviews***

*It is an awesome book that we have possibly go through. It is actually writter in straightforward words and phrases and not confusing. It is extremely difficult to leave it before concluding, once you begin to read the book.*

***(Tierra Kunde)***

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paperback. Book Condition: New. Ship out in 2 business day, And Fast shipping, Free Tracking number will be provided after the shipment. Pages Number: 296 Publisher: Dalian University of Technology Press Pub. Date :2009-02. Chemistry-based tutorial to Inorganic Chemistry Teaching Higher Technical School basic requirements and engineering the basic content of general chemistry. as the basis for many years engaged in inorganic chemistry with editors and the general college chemistry teaching. and writing several chemical materials experience. Inorganic Chemistry Essentials (Mu Wensheng. Yong-fresh. weeks Boron series. Dalian University of Technology Press) written in the basis. The book is divided into eleven chapters. describes the chemical reaction of the energy and direction. precipitation of dissolved acid-base balance and equilibrium. and electrochemical redox reactions based on atomic structure and periodic law of elements. coordination compounds and so on. The teaching materials of the narrative to experimental facts. we could achieve a simple and concise language. easy to understand. easy to self-study. Contents: Chapter 1 1.1 gas and dilute solution of the ideal gas law gas equation of state 1.1.1 1.1.2 1.2 Gas Law partial pressure according to the number of dilute solution of the concentration of solution 1.2.1 1.2.2 Exercise dilute solution in accordance with Chapter 2 of the energy and direction of chemical reaction chemical reaction 2.1 the quality of relationships in chemical reactions of metered-2.1.1 2.1.2 2.2 chemical reaction of reaction 2.2.1 Thermodynamics of energy relations in the common terminology and basic concepts first law of thermodynamics 2.2.2 2.2.3 2.2.4 thermochemical heat chemical reaction chemical reaction equation 2.2.5 Enthalpy 2.3 Entropy and entropy changes in 2.3.1 2.3.2 spontaneous chemical reaction with the degree of disorder and response entropy 2.4 gibbs function of the direction of 2.4.1 gibbs function 2.4.2 gibbs function change and response exercises in Chapter 3 in the direction...



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