High-temperature superconductivity of ternary $Ca_4B_xC_{23-x}$ clathrates at moderate pressure

Peng Chen ,¹ Zepeng Wu ,¹ Yang Sun ,¹ Tie-Yu Lü,¹ Xinrui Cao ,^{1,2} Zi-zhong Zhu ,^{1,2} and Shunqing Wu ,^{1,*}

¹ Department of Physics, OSED, Key Laboratory of Low Dimensional Condensed Matter Physics

(Department of Education of Fujian Province), Xiamen University, Xiamen 361005, China

² Fujian Provincial Key Laboratory of Theoretical and Computational Chemistry, Xiamen University, Xiamen 361005, China



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The synthesis of SrB_3C_3 and the research on its analogs demonstrate the high-temperature superconductivity of ternary metal-boron-carbon clathrates at ambient or moderate pressure. Here, we explore the stability and superconductivity of calcium-doped and boron-substituted type-VIII carbon clathrates $Ca_4B_xC_{23-x}$ (x=0-23) at 25–100 GPa. Based on the structural features and density functional theory (DFT) calculations, we screened 8 388 608 possible B-decorated structures and ultimately identified five thermodynamically metastable superconductors. Among them, $Ca_4B_{10}C_{13}$ (space group $I\bar{4}3m$) exhibited the highest T_c of \sim 101–109 K, between 25 and 100 GPa. The high T_c of $Ca_4B_{10}C_{13}$ originates from strong electron-phonon coupling caused by phonon softening and high density of states at the Fermi level. Our findings suggest that metal-doped and boron-substituted carbon clathrates are promising high-temperature superconductors at moderate pressure.

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I. INTRODUCTION

In recent years, a series of high-temperature hydrogenrich superconductors have been theoretically predicted and successfully synthesized, such as CaH₆ (215 K, 172 GPa [1,2]), YH₉ (243 K, 201 GPa [3,4]), LaH₁₀ (~250 K, 170 GPa [5,6]). The common structural feature of these hydrogenrich high-temperature superconductors is a cagelike structure formed by hydrogen atoms. However, the extremely high stabilizing pressure required for hydrogen-rich superconductors limits their practical application. Consequently, researchers have been working to explore high-temperature superconductors at lower pressure. For example, a series of compounds, represented by Mg₂IrH₆, have been predicted to be high- T_c superconductors at ambient pressure very recently [7–9]. Additionally, a series of few-hydrogen compounds have also been expected to be high-temperature superconductors at ambient pressure, such as Al₄H (\sim 54 K [10]), (Be₄)₂H $(\sim 72-84 \text{ K} [11])$, and MgHCu₃ ($\sim 42 \text{ K} [12]$).

Another effective approach is to replace H with other light elements, such as B, C, and N. Some binary carbides were predicted to have a high T_c at ambient pressure. For instance, NaC₆ in the sodalite structure has been predicted as a potential superconductor with $T_c \sim 116$ K [13]. MC₆ (where M represents other metals replacing Na) with the same structure has also been theoretically studied [14,15]. It was found that the compounds with a carbon-cage-network structure have higher T_c values than those carbon-based materials without a cage. Nonetheless, the high formation energy characteristic poses a significant challenge to synthesis. Zeng et al. theoretically studied Li-filled, B-substituted carbon clathrates and found that the strategy of boron substitution could improve

the thermodynamic stability of carbon clathrate frameworks [16]. Recently, SrB_3C_3 (Pm-3n) was successfully synthesized at 50 GPa and was recovered to ambient pressure under inert conditions with T_c of \sim 20 K (27–43 K) at 40 GPa by experiment (theory) [17,18]. SrB_3C_3 consists of a sp^3 -bonded framework with equal amounts of boron and carbon in the sodalite structure. The substitution of half of the C atoms with B provides p-type doping which maintains its dynamic stability at 0–200 GPa and thermodynamic stability at 50–200 GPa [17].

Due to the host-guest structure of boron-carbon clathrates, it is a promising path to enhance the T_c by tuning guest atoms or host framework of clathrates. For example, CaB₃C₃ and BaB3C3 with similar structures to SrB3C3 were predicted to be dynamically stable at ambient pressure with T_c of 48 K [19] and 43 K [20], respectively. Additionally, Rb-doped SrB_3C_3 was predicted to have a high T_c above 70 K [21,22]. Other boron-carbon clathrates as the analogs of SrB₃C₃ have also been predicted to exist high values of T_c at ambient pressure, such as KPbB₆C₆ ($T_c \sim 88$ K [23]), RbYbB₆C₆ $(T_c \sim 67 \text{ K [24]})$, and RbBaB₆C₆ $(T_c \sim 68 \text{ K [24]})$. Very recently, SrNH₄B₆C₆ was designed by doping SrB₃C₃ with small hydride units NH₄. This modification effectively improved T_c to 73 K at 1 atm. Its analog PbNH₄B₆C₆ was predicted with a T_c of ~86 K at 1 atm [25]. On the other hand, LaH₁₀-like host frameworks have been used to construct boron-carbon clathrates of the type $Fm\bar{3}m-XB_2C_8$ (where X is metal elements) [26,27], which possess T_c around or above liquid nitrogen temperature at ambient pressure. Additionally, some boron-nitrogen and boron-silicon clathrates have been predicted to be superconductors at ambient pressure, such as Al(BN)₆ ($T_c \sim 47$ K [28]), Al₂(BN)₆ ($T_c \sim 72$ K [29]), RbB₃Si₃ ($T_c \sim 14 \text{ K } [30]$).

These studies suggest that B-C clathrates have great potential for exploration in the search for high-temperature

^{*}Contact author: wsq@xmu.edu.cn

superconductors at ambient or moderate pressure. Many metastable carbon clathrate frameworks have been theoretically investigated [31–39], such as the sp^3 -bonded type-VIII carbon clathrate derived from Ba₈Ga₁₆Sn₃₀ [40]. This raises the question of whether B-C clathrate superconductors could be designed using a similar strategy based on other sp^3 bonded carbon clathrates. We also noted that a recent study predicted a thermodynamically stable Ca₈B₁₆C₃₀ at 50 GPa [41]. However, it is a semiconductor with a band gap of \sim 2.5 eV. In other studies of Ca-B-C systems, while some phases are predicted to be superconductors, their T_c are below 40 K at ambient pressure, such as CaB_2C_4 ($T_c \sim 1.7$ K), CaB_4C_2 ($T_c \sim 15$ K) [42], Ca_2BC_{11} ($T_c \sim 8.9$ K), $CaBC_5$ $(T_c \sim 5.2 \text{ K})$, and CaB₃C₃ $(T_c \sim 26.1 \text{ K})$ [43]. Thus, we look forward to further exploring clathrates with high T_c in the Ca-B-C ternary system with a type-VIII carbon clathrate framework.

In this paper, based on condition screening and density functional theory (DFT) calculations, we explored the stability and superconductivity of ternary $Ca_4B_xC_{23-x}$ (x=0-23) clathrates at 25–100 GPa. The structure motif used is the type-VIII carbon clathrates from Refs. [31,40]. Boron atoms were introduced by partially replacing carbon sites, and all possible replacement sites were considered for $Ca_4B_xC_{23-x}$ with varying boron contents. After screening the structural features and performing DFT calculations, we identified five thermodynamically metastable, dynamically stable, and metallic phases at 25–100 GPa. Their superconductivity was assessed using DFPT (density functional perturbation theory) calculations, such as $Ca_4B_{10}C_{13}$ with T_c above 100 K at 25–100 GPa.

II. COMPUTATIONAL DETAILS

Structure relaxations and electronic properties were carried out using the Perdew-Burke-Ernzerhof (PBE) generalized gradient approximation (GGA) [44] in the framework of the projector augmented wave (PAW) [45] method as implemented in the VASP code [46]. The valence electrons of $3s^2 3p^6 4s^2$, $2s^2 2p^1$, and $2s^2 2p^2$ for Ca, B, and C were used in the PAW method, respectively. The plane-wave basis set with an energy cutoff of 600 eV and a Monkhorst-Pack Brillouin zone (BZ) sampling density of $2\pi \times 0.015 \text{ Å}^{-1}$ were employed in the structure relaxations and self-consistent calculations. The structural optimization was performed using the conjugate gradient algorithm with variable cell shapes, and the convergence threshold is -0.01 eV/Å for the atomic force. Phonon calculations were carried out in the finite displacement method using the VASP package coupled to the PHONOPY code [47]. For these calculations, $2 \times 2 \times 2$ supercells with ~216 atoms and a Monkhorst-Pack BZ sampling density of $2\pi \times 0.025 \text{ Å}^{-1}$ were employed.

Within the framework of DFPT [48], electron-phonon coupling (EPC) was calculated by ultrasoft pseudopotentials (USPP) [49] with PBE functional implemented in the QUANTUM ESPRESSO package (QE) [50,51]. The same configurations of valence electrons as used in PAW were applied in USPP. The kinetic energy cutoffs were 70 Ry for wave functions and 560 Ry for charge density. The calculating electronic structures of Ca₄B₁₀C₁₃ under different pressures performed pretty consistent results between VASP and QE in

Supplemental Material Note S1 [52]. Q-point meshes with $3 \times 3 \times 3$ were used for DFPT calculation. Self-consistent electron density and EPC were calculated by employing a $6 \times 6 \times 6$ k-point mesh, and a dense $12 \times 12 \times 12$ grid was used for evaluating an accurate electron-phonon interaction matrix. We also tested q-point meshes with $4 \times 4 \times 4$, and the results of EPC were similar to those obtained with the $3 \times 3 \times 3$ meshes.

The calculations of T_c are based on the Eliashberg spectrum function defined as [53–55]

$$\alpha^2 F(\omega) = \frac{1}{2\pi N(E_F)} \sum_{qv} \frac{\gamma_{qv}}{\hbar \omega_{qv}} \delta(\omega - \omega_{qv}), \tag{1}$$

where $N(E_F)$ is the density of states (DOS) at the Fermi level, ω_{qv} represents the phonon frequency of the mode v with wave vector q, and γ_{qv} is the phonon linewidth defined as

$$\gamma_{qv} = \frac{2\pi\omega_{qv}}{\Omega_{\rm BZ}} \sum_{i,j} \int d^3k \left| g_{k,qv}^{i,j} \right|^2 \delta(\varepsilon_{i,q} - E_F) \delta(\varepsilon_{j,k+q} - E_F).$$
(2

Here, $g_{k,qv}^{i,j}$ is the EPC matrix element, and $\Omega_{\rm BZ}$ is the volume of the Brillouin zone. The EPC constant can be determined as

$$\lambda = \sum_{qv} \frac{\gamma_{qv}}{\pi \hbar N(E_F)\omega_{qv}^2} = 2 \int \frac{\alpha^2 F(\omega)}{\omega} d\omega.$$
 (3)

For EPC constant $\lambda < 1.5$, T_c was firstly estimated from the McMillan-Allen-Dynes (MAD) formula [53,54]:

$$T_c = \frac{\omega_{\log}}{1.2} \exp\left[-\frac{1.04(1+\lambda)}{\lambda - \mu^*(1+0.62\lambda)}\right].$$
 (4)

For the strong EPC system with $\lambda > 1.5$, T_c was corrected by the Allen-Dynes–corrected McMillan formula [55]:

$$T_c = f_1 f_2 \frac{\omega_{\log}}{1.2} \exp\left[-\frac{1.04(1+\lambda)}{\lambda - \mu^*(1+0.62\lambda)}\right],$$
 (5)

where the correction factors f_1 , f_2 were defined by λ , μ^* , ω_{\log} , and mean-square frequency ω_2 . Here, the renormalized Coulomb potential $\mu^* = 0.10$, which is a conventional setting for B-C clathrates [19,23], was used. The logarithmic average frequency ω_{\log} is computed as

$$\omega_{\log} = \exp\left(\frac{2}{\lambda} \int \frac{\alpha^2 F(\omega)}{\omega} \ln \omega d\omega\right).$$
 (6)

We also estimated the superconducting gap and T_c using the Migdal-Eliashberg (ME) approach by numerically solving isotropic Eliashberg equations [56,57]. For the details of the computational method, refer to Refs. [58–60].

III. RESULTS AND DISCUSSION

A. Structure screening

The space group of type-VIII clathrate is $I\bar{4}3m$, and the carbon atoms occupy the 2a (0, 0, 0), 8c (x_c, x_c, x_c), 12d (1/4, 1/2, 0), and 24g (x_g, x_g, z_g) Wyckoff positions, where $x_c \approx 0.137$, $x_g \approx 0.586$, and $z_g \approx 0.856$. As shown in Fig. 1(a), there are eight [3³ 4³ 5⁹] cages (pentadecahedron) and three small [3⁴ 4⁴] (octahedron) voids in the conventional cell of type-VIII clathrate with 46 carbon atoms. In

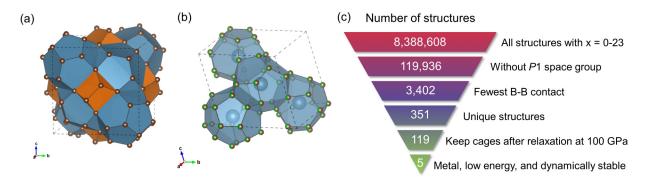


FIG. 1. (a) Crystal structure of type-VIII clathrate from Refs. [31,40], where the C atoms are shown as brown spheres. The blue polyhedrons represent [3 3 4 3 5 9] cages, and the orange ones represent small [3 4 4 4] voids. The cage representation was the same as Ref. [31], e.g., [3 3 4 3 5 9] means a cage with 3 three-membered rings, 3 four-membered rings, and 9 five-membered rings. (b) Crystal structure of Ca₄B_xC_{23-x} (x = 0 - 23), where the Ca and B/C atoms are shown as blue and green/brown spheres, respectively. (c) Number of structures in Ca₄B_xC_{23-x} with possible B substitution sites and the number of structures remaining after screening under different conditions.

the following study, the primitive cell with 23 atoms of type-VIII clathrate was used to construct B-C clathrates with four $[3^3 4^3 5^9]$ cages. The shortest distance between the center point and the vertices of the $[3^4 4^4]$ carbon cage is only 1.25 Å at 0 GPa, which is shorter than the other carbon cages (1.63 Å–2.55 Å) [31]. Considering the smaller capacity of $[3^4 4^4]$ voids, we constructed B-C clathrates of $Ca_4B_xC_{23-x}$ by doping calcium atoms only in $[3^3 4^3 5^9]$ cages and boron substitution as shown in Fig. 1(b).

At a definite x value, the number of total possible structures of $Ca_4B_xC_{23-x}$, N_x , is given by

$$N_x = C_{23}^x = \frac{23!}{(23-x)!x!}. (7)$$

As shown in Fig. 1(c), the sum of N_x from x = 0 to x = 23is 8 388 608, which is too immense to direct DFT calculations. In previous studies on boron-substituted carbon clathrates, it was found that B-B contact was energetically disadvantageous [16,41], and those with P1 space-group symmetry were rarely observed as ground states [16,41,61,62]. Thus, to decrease N_r , we set up filters to remove structures with P1 space group symmetry and only reserve those structures with the fewest B-B contact. To test the rationality of such filters, we computed the enthalpies of Ca₄B₁₀C₁₃ with P1 space-group symmetry at 100 GPa (see Supplemental Note S2 [52]). We found that the B-B contacts are energetically unfavorable. Here, the structures of $Ca_4B_xC_{23-x}$ without B-B contact can be found when $x \leq 10$, but they will inevitably have B-B contact when $x \ge 11$. As shown in Fig. 1(c), the filters effectively reduce the number of possible structures to 3402, and it can be further diminished to 351 after removing the symmetry equivalence structures. The detailed number of remaining structures can be found in Table S1 [52].

The full relaxation of these structures was performed at 25, 50, and 100 GPa for these 351 structures. We found that some of these structures failed to maintain the cagelike structure or had local defects, which reflected the structural instability caused by the corresponding boron substitution scheme. However, we were only interested in those cagelike structures, so we set up a criterion with C-C bond length < 2.1 Å, B-C bond length < 2.2 Å, and B-B bond length < 2.3 Å to identify the cagelike structures. Under the screening of this criterion, 70,

85, and 119 cage structures were left at 25, 50, and 100 GPa, respectively.

B. Phase stability

To evaluate the thermodynamic stability of the selected structures after structure screening, we first calculated the formation enthalpy ΔH at 25–100 GPa, which is defined as

$$\Delta H = \frac{H(\text{Ca}_4\text{B}_x\text{C}_{23-x}) - 4H(\text{Ca}) - xH(\text{B}) - (23-x)H(\text{C})}{27},$$
(8)

where $H(\text{Ca}_4\text{B}_x\text{C}_{23-x})$ is the total enthalpy of the $\text{Ca}_4\text{B}_x\text{C}_{23-x}$ compounds. Here, H(Ca), H(B), and H(C) are the enthalpies of the ground states of Ca, B, and C, respectively, at corresponding pressure (see Table S2 [52]). As shown in Fig. 2(a), when $x \le 7$ and $x \ge 11$, there are a lot of relaxed structures that could not keep cage characters, which are labeled with a cross. Besides, focusing on those cage structures, we found that ΔH shows a decreasing trend with x from 0 to 8 and turns to be increasing when x is from 8 to 23 at different pressures. Besides, $\text{Ca}_4\text{B}_x\text{C}_{23-x}$ has lower values of ΔH with pressure increasing, and the lowest ΔH reaches -0.524 eV/atom at 100 GPa in x = 8.

Next, we further verified the thermodynamic stability for the cage structures by calculating the convex hull of the Ca-B-C system at 25, 50, and 100 GPa, as shown in Fig. S5 [52], and the results of calculated enthalpy above the convex hull (E_d) were summarized in Fig. 2(b). A similar V-shaped trend of first decreasing and then increasing with growing x can also be observed in the E_d of $Ca_4B_xC_{23-x}$, and the low point of E_d is around x = 8 at different pressures. This indicates that those structures with a boron content close to Ca₄B₈C₁₅ (insulators, see Fig. S6 [52]) are energy favorable. And the pressure decreases E_d of $Ca_4B_xC_{23-x}$, which is consistent with other studies about B-C clathrates [16,17,41,63]. Here, we are more interested in those structures that have lower E_d under pressure, which are more likely to be synthesized at high pressure [17,63]. As discussed in Refs. [64,65], many metastable inorganic materials can be synthesized within a specific energy range. There are also some metastable M-B, M-C, and M-B-C (M means metal element) compounds

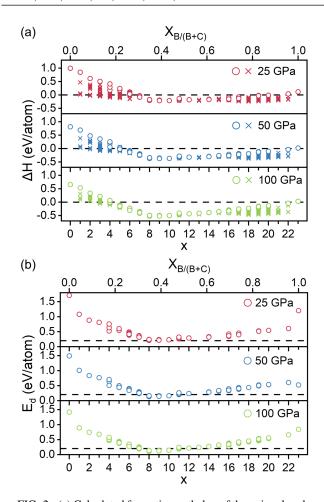


FIG. 2. (a) Calculated formation enthalpy of the uniquely relaxed phases without P1 space-group symmetry and with the fewest B-B contacts at 25, 50, and 100 GPa. The relaxed structures with (without) keeping intact cage characters are labeled with a circle (cross). (b) Calculated enthalpy above convex hull (E_d) of the uniquely relaxed phases with cagelike structures (cage structure) at 25, 50, and 100 GPa. In (a) and (b), the lower x axis is the value of x in $Ca_4B_xC_{23-x}$, and the upper x axis is the ratio of B to (B + C) in $Ca_4B_xC_{23-x}$ (labeled as $X_{B/(B+C)}$). The dashed line in (b) indicates the 0.2 eV/atom threshold to identify the phases with low enthalpy. Note that we excluded the data points of structures that did not maintain the cage structure after relaxation.

that have been experimentally synthesized, but we confirmed that they are theoretically thermodynamic metastable. For example, $I4_1/acd \text{ Na}_2\text{C}_2$ [66], $Fm\bar{3}m \text{ ZrB}$ [67], and P4m2 Li₂B₂C [68] have been experimentally synthesized under certain conditions. And our calculations indicate that $I4_1/acd \text{ Na}_2\text{C}_2 \ (E_d = 0.224 \text{ eV/atom}), \ Fm\bar{3}m \ \text{ZrB} \ (E_d = 0.224 \text{ eV/atom})$ 0.378 eV/atom), and $P\bar{4}m2 \text{ Li}_2B_2C$ ($E_d = 0.243 \text{ eV/atom}$) are metastable compounds. Given these examples, we may not be able to rule out the synthesizability of those structures with $E_{\rm d} > 0.2$ eV/atom. However, to screen out the more synthesizable structures efficiently, we used a criterion of $E_d < 0.2 \,\mathrm{eV/atom}$ at 100 GPa to screen superconducting candidates. As shown in Fig. 2(b), 2, 2, 1, and 2 structures meet the criterion of E_d when x = 7, 9, 10, and 11, respectively. The phases with x = 8 were not considered because they were insulators as shown in Fig. S6 [52]. So, we get seven metallic structures with $E_d < 0.2 \,\mathrm{eV/atom}$ at 100 GPa, which were $\mathrm{Ca_4B_7C_{16}}$ ($I\bar{4}3m$), $\mathrm{Ca_4B_7C_{16}}$ (R3m), $\mathrm{Ca_4B_9C_{14}}$ (R3m), $\mathrm{Ca_4B_9C_{14}}$ ($I\bar{4}$), $\mathrm{Ca_4B_{10}C_{13}}$ ($I\bar{4}3m$), $\mathrm{Ca_4B_{11}C_{12}}$ (Cm), and $\mathrm{Ca_4B_{11}C_{12}}$ (C2), and their detailed structure information was shown as in Table S5 [52].

Then we evaluated the dynamic stability of seven structures, and their phonon spectra at 25, 50, and 100 GPa are shown in Figs. S7–S9 [52]. The results showed that five phases of R3m Ca₄B₉C₁₄, $I\bar{4}$ Ca₄B₉C₁₄, $I\bar{4}3m$ Ca₄B₁₀C₁₃, Cm Ca₄B₁₁C₁₂, and C2 Ca₄B₁₁C₁₂ were harmonic dynamically stable from 25 to 100 GPa, but the other two phases of Ca₄B₇C₁₆ were dynamically unstable at the harmonic level from 25 to 100 GPa for the imaginary phonon frequency. Considering the anharmonic effect in the light element system [69–71], these two Ca₄B₇C₁₆ phases might be dynamically stable when the anharmonic effect is taken into account. However, it was beyond the scope of this study. So, in the following study, we excluded two Ca₄B₇C₁₆ phases and only considered the other five phases as shown in Fig. 3.

C. Electron-phonon coupling and superconductivity

For these five metastable phases, we first performed the fast frozen-phonon calculation of the zone-center EPC strength to evaluate the superconductivity [72]. This method has been successfully applied in ternary metal diborides and hydrogenrich compositions [60,73]. As shown in Table I, three phases of R3m Ca₄B₉C₁₄, $I\bar{4}$ Ca₄B₉C₁₄, and $I\bar{4}3m$ Ca₄B₁₀C₁₃ had high λ_{Γ} from 0.40 to 0.49 at 50 GPa, which indicated their potential strong EPC strength compared to the $\lambda_{\Gamma} = 0.44$ of MgB₂ [72]. We then performed the DFPT calculations to further demonstrate their superconductivity and evaluated the T_c by MAD formula and Migdal-Eliashberg approach as shown in Table I. Their superconducting energy gap is shown in Fig. S10 [52]. As is expected, the full Brillouin zone EPC λ of 0.62 for R3m Ca₄B₉C₁₄ and 0.70 for $I\bar{4}$ Ca₄B₉C₁₄ are comparable to MgB₂ ($\lambda \sim 0.7$) [74] at 0 GPa and SrB₃C₃ ($\lambda \sim 0.7$) [18] at 50 GPa. However, Cm Ca₄B₁₁C₁₂ and C2 Ca₄B₁₁C₁₂ had relatively small values of λ_{Γ} , but they still performed considerable λ of 0.68 and 1.00 at 50 GPa, respectively, which indicates those phonon modes away from the zone center play an important role in EPC for these two phases. Remarkably, Ca₄B₁₀C₁₃ possesses strong EPC and high superconduct temperature at 50 GPa and has the highest λ of 1.30 and T_c of 101 K above liquid nitrogen temperature. Compared with such a high T_c of Ca₄B₁₀C₁₃, the T_c difference between the two $Ca_4B_9C_{14}$ phases and between the two $Ca_4B_{11}C_{12}$ phases were all smaller, indicating that the boron contents played a more important role in T_c than the distribution of boron.

To further understand the origin of such high λ and T_c values of Ca₄B₁₀C₁₃, we plotted its phonon spectrum, projected phonon density of states (PHDOS), Eliashberg spectral function $\alpha^2 F(\omega)$, EPC integral $\lambda(\omega)$, fat electron band structure, projected density of states (PDOS), and total density of states (TDOS) at 50 GPa in Fig. 4. For comparison, the results of ground states of R3m Ca₄B₉C₁₄, and Cm Ca₄B₁₁C₁₂ at 50 GPa were also plotted in Fig. 4, and the results of I4 Ca₄B₉C₁₄ and C2 Ca₄B₁₁C₁₂ can be found in Figs. S11–S13 [52]. Firstly, we observed the phonon modes dominated by the B-C covalent framework strongly coupled with the high-density B-C

TABLE I. Space group, DOS at the Fermi level $[N(E_F)]$, zone-center EPC λ_{Γ} , full Brillouin zone EPC λ , logarithmic average frequency ω_{\log} , and T_c of R3m Ca₄B₉C₁₄, $I\bar{4}$ Ca₄B₉C₁₄, $I\bar{4}3m$ Ca₄B₁₀C₁₃, Cm Ca₄B₁₁C₁₂, and C2 Ca₄B₁₁C₁₂ at 50 GPa. T_c is estimated using the MAD formula $T_{c,MAD}$ and Migdal-Eliashberg approach $T_{c,ME}$ by setting the screened Coulomb potential $\mu^* = 0.10$.

Structure	Space group	$N(E_F)$ states/(eV f.u.)	λ_{Γ}	λ	$\omega_{\log}(K)$	$T_{c_{ m MAD}}$ (K)	$T_{c_{-}\mathrm{ME}}\left(\mathrm{K}\right)$
$Ca_4B_9C_{14}$	R3m	8.53	0.46	0.62	964	24	30
$Ca_4B_9C_{14}$	$I\bar{4}$	8.75	0.40	0.70	985	34	41
$Ca_4B_{10}C_{13}$	$I\bar{4}3m$	9.12	0.49	1.30	870	85	101
$Ca_{4}B_{11}C_{12}$	Cm	3.01	0.12	0.76	642	27	32
$Ca_4B_{11}C_{12}$	C2	5.28	0.24	1.00	429	30	35

covalent electrons at the Fermi level in all three phases as shown in Fig. 4, which brings the main contribution to their EPC and is the same with other B-C clathrate superconductors [18-23]. Among them, the phonon modes that played a major role in EPC are mainly concentrated in middle- and high-frequency regions, such as the breathing modes at Γ point as shown in Fig. S14 [52]. Especially, we noted that Ca₄B₁₀C₁₃ has significantly softened phonon modes accompanied by large phonon linewidth in the medium frequency region compared with R3m Ca₄B₉C₁₄, as shown in Figs. 4(a) and 4(b). Note that the softened phonons here mean the overall frequency descent behavior of many phonon modes in middlefrequency regions, as discussed in Supplemental Material Note S3 [52]. Because of the large phonon linewidth of the phonon modes dominated by B and C in the middle-frequency regions, the Eliashberg spectral functions $\alpha^2 F(\omega)$ of R3m Ca₄B₉C₁₄ and Ca₄B₁₀C₁₃ all show a peaklike distribution in middle-frequency regions. If we integrate Eq. (3) in the range of $0-600\,\mathrm{cm}^{-1}$, we find that the contribution to λ of R3m $Ca_4B_9C_{14}$ is 0.124 in this range (see Table S3 [52]). In contrast, $Ca_4B_{10}C_{13}$ has a high proportion of integral λ of 0.559 in the range of $0-600\,\mathrm{cm}^{-1}$ due to the softened phonon modes. Such phonon softening leads to a higher total EPC constant $\lambda = 1.30$ and $T_c = 101$ K of Ca₄B₁₀C₁₃ than R3m Ca₄B₉C₁₄ at 50 GPa. This analysis illustrates that Ca₄B₁₀C₁₃ with higher B content leads to significant phonon softening compared to R3m Ca₄B₉C₁₄, which enhances EPC and improves T_c .

In addition, the high $N(E_F)$ is also an important factor leading to the high T_c of Ca₄B₁₀C₁₃. As shown in Figs. 4(d)–4(f), the increase of boron content has the effect of hole doping, which significantly regulates the electronic band structure of these three phases. The Van Hove singularity (VHS) lies on the Fermi level at the F point for R3m Ca₄B₉C₁₄, and the multiple bands cross the Fermi level for Ca₄B₁₀C₁₃, leading to their high $N(E_F)$. As shown in Eq. (3), λ appears to have a negative correlation with $N(E_F)$. However, there is a positive correlation between γ_{qv} and the square of $N(E_F)$; as shown in Eq. (2), the sum contains the product of two δ functions. So, it actually results in a positive correlation between λ and $N(E_F)$, and λ can be written in the form of $\lambda = \frac{N(E_F)\langle I^2 \rangle}{M\langle \omega^2 \rangle}$, as described in Refs. [53,55]. Therefore, the high $N(E_F)$ favor large λ and T_c . As shown in Figs. 4(c) and 4(f), although Cm Ca₄B₁₁C₁₂ has more softened phonons, their contributions to EPC were greatly hindered due to the low $N(E_F)$. This is also the key to why the T_c of Cm $Ca_4B_{11}C_{12}$ is much lower than that of Ca₄B₁₀C₁₃ at 50 GPa. Given the above, it can be concluded

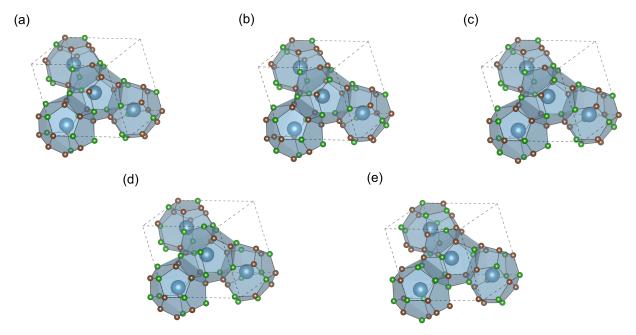


FIG. 3. Crystal structures of (a) R3m Ca₄B₉C₁₄, (b) $I\bar{4}$ Ca₄B₉C₁₄, (c) $I\bar{4}3m$ Ca₄B₁₀C₁₃, (d) Cm Ca₄B₁₁C₁₂, and (e) C2 Ca₄B₁₁C₁₂ at 50 GPa, where the Ca, B, and C atoms are shown as blue, green, and brown spheres, respectively.

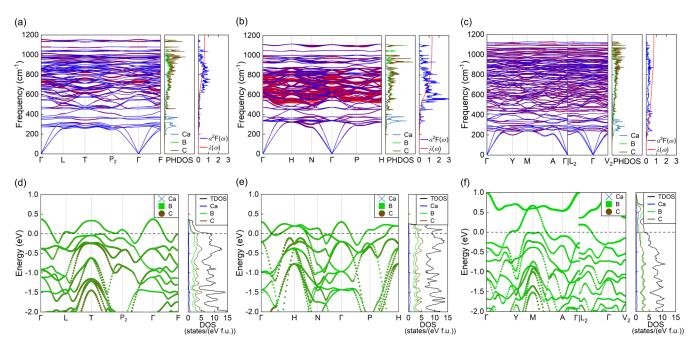


FIG. 4. Phonon dispersion curves, projected phonon density of states (PHDOS), Eliashberg spectral function $\alpha^2 F(\omega)$ and EPC integral $\lambda(\omega)$ of (a) R3m Ca₄B₉C₁₄, (b) $I\bar{4}3m$ Ca₄B₁₀C₁₃, and (c) Cm Ca₄B₁₁C₁₂ at 50 GPa. Phonon dispersion is represented by blue thin lines, decorated with the respective phonon linewidth $\gamma_{\nu q}$ in red circles. Fat electron band structure, projected density of states (PDOS), and total density of states (TDOS) of (d) R3m Ca₄B₉C₁₄, (e) $I\bar{4}3m$ Ca₄B₁₀C₁₃, and (f) Cm Ca₄B₁₁C₁₂ at 50 GPa.

that the T_c of Ca₄B_xC_{23-x} (x = 9-11) is sensitive to boron content, which can influence the phonon softening and $N(E_F)$ of phases.

Given that B-C clathrates may be synthesized at high pressure and depressurized to moderate even atmospheric pressure [17,18,63], we also calculated the superconductivity of five $Ca_4B_xC_{23-x}$ phases under different pressure conditions as shown in Table S4 [52] and summarized in Fig. 5. As a result,

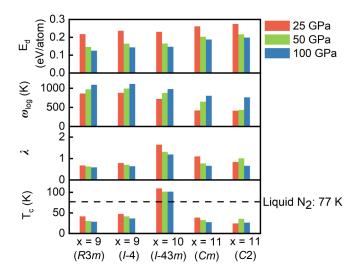


FIG. 5. Calculated enthalpy above convex hull (E_d) , logarithmic average frequency ω_{\log} , EPC constant λ , and T_c of R3m Ca₄B₉C₁₄, $I\bar{4}$ Ca₄B₉C₁₄, $I\bar{4}3m$ Ca₄B₁₀C₁₃, Cm Ca₄B₁₁C₁₂, and C2 Ca₄B₁₁C₁₂ at different pressure conditions, respectively. Here, the values of T_c are estimated using the Migdal-Eliashberg approach by setting the screened Coulomb potential $\mu^* = 0.10$.

these five phases can keep cage structures and are dynamically stable from 25 to 100 GPa (see Figs. S7–S9 [52]), and their E_d show a decreasing trend as the pressure increases in this pressure range as shown in Fig. 5. As expected, with pressure increases the length of B-C bond decreases, significantly hardening the phonon modes as shown in Fig. 4 and Figs. S11–S13 [52]. Therefore, ω_{\log} increases and λ decreases, and in the competition between ω_{\log} and λ , λ plays the dominant role, which leads to the T_c s having a consistent trend with pressure as λ . However, $C2 Ca_4B_{11}C_{12}$ is an exception, as seen in Fig. 5; λ increased at 25–50 GPa, leading the value of T_c at 50 GPa to be larger than that at 25 GPa.

A further question is, can any of these superconducting phases be quenched to ambient pressure? By calculating the phonon spectra of these superconducting phases, we found that three phases (R3m Ca₄B₉C₁₄, $I\bar{4}$ Ca₄B₉C₁₄, and $I\bar{4}3m$ Ca₄B₁₀C₁₃) maintain dynamical stability at ambient pressure. In comparison, the other two phases (Cm Ca₄B₁₁C₁₂ and C2 Ca₄B₁₁C₁₂) are dynamically unstable at ambient pressure. It indicates that the three dynamically stable superconducting phases have the potential to quench to ambient pressure. By solving the isotropic Eliashberg equation, we estimated the values of T_c for R3m Ca₄B₉C₁₄ (63 K), $I\bar{4}$ Ca₄B₉C₁₄ (79 K), and $I\bar{4}3m$ Ca₄B₁₀C₁₃ (107 K) at ambient pressure.

IV. CONCLUSIONS

In summary, we explored the stability and superconductivity of ternary $Ca_4B_xC_{23-x}$ (x = 0-23) clathrates at 25–100 GPa. A total of 8 388 608 possible structures of $Ca_4B_xC_{23-x}$ were considered. After symmetry screening, B-B bond screening, and structure deduplication, 351 unique

candidate structures remained. Based on the DFT calculations, we identified five metallic, thermodynamically metastable phases that preserved good cage characteristics at 25–100 GPa. Their superconductivity was demonstrated by the Migdal-Eliashberg theory and the DFPT calculations. These include R3m Ca₄B₉C₁₄ with T_c of \sim 28–41 K, $I\bar{4}$ Ca₄B₉C₁₄ with T_c \sim 36–47 K, $I\bar{4}3m$ Ca₄B₁₀C₁₃ with T_c of \sim 101–109 K, Cm Ca₄B₁₁C₁₂ with T_c of \sim 27–38 K, and C2 Ca₄B₁₁C₁₂ with T_c \sim 24–35 K across 25–100 GPa. The high λ and T_c values arise from the strong coupling between the B-C electronic states at the Fermi level and the phonon modes

of the B-C atoms. Our findings suggest that metal doping and boron substitution in carbon clathrates are effective strategies for exploring high-temperature superconductors.

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