

Lecture 2

Read Hund's rule (Ashcroft & Mermin : pg. 650 - 652)

Lande's g factor :

$$g = 1 + \frac{J(J+1) + S(S+1) - L(L+1)}{2J(J+1)} = \frac{3}{2} + \frac{1}{2} \left[\frac{S(S+1) - L(L+1)}{J(J+1)} \right]$$

HW: Use eq. (31.50) to calculate for effective Bohr magneton number for all elements in the periodic table

$$\text{E.G.: Ce} \rightarrow 4f^1 \quad \begin{array}{ccccccc} \downarrow & - & - & - & - & - & - \\ +3 & +2 & +1 & 0 & -1 & -2 & -3 \end{array}$$

$$S = \frac{1}{2}, L = 3, J = \frac{5}{2}$$

$$g = \frac{3}{2} + \frac{1}{2} \left(\frac{\frac{1}{2}(\frac{3}{2}) + 3(3+1)}{\frac{5}{2}(\frac{7}{2})} \right) = \frac{6}{7}$$

$$p = \frac{6}{7} \left[\frac{5}{2} \left(\frac{7}{2} \right) \right]^{1/2} = \frac{6}{7} \cdot \frac{\sqrt{35}}{2} \approx \frac{36}{14} \approx 2.57$$