VE320 – Summer 2024

Introduction to Semiconductor Devices

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Chapter 9 Metal-Semiconductor Schottky Junction

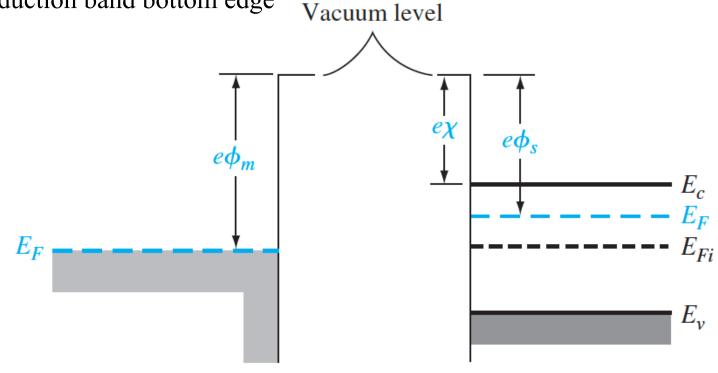
Outline

9.1 The Schottky barrier diode

9.2 Metal-semiconductor Ohmic contacts

Qualitative characteristics

- Work function: energy difference between the vacuum energy level and the Fermi level
- Electron affinity: energy different between the vacuum energy level and conduction band bottom edge



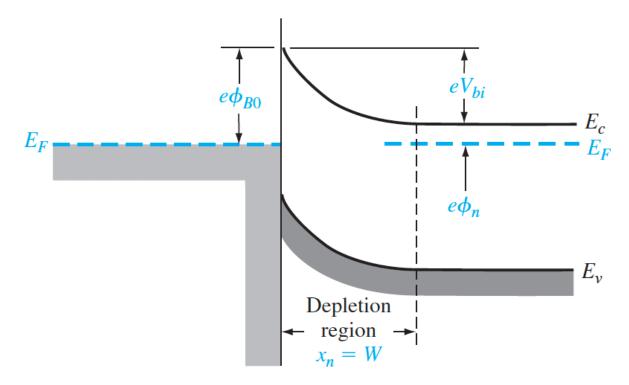


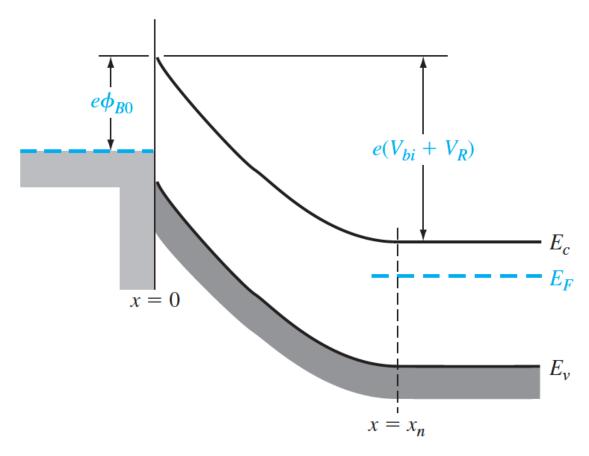
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Element	Work function, $\phi_{\scriptscriptstyle m}$
Ag, silver	4.26
Al, aluminum	4.28
Au, gold	5.1
Cr, chromium	4.5
Mo, molybdenum	4.6
Ni, nickel	5.15
Pd, palladium	5.12
Pt, platinum	5.65
Ti, titanium	4.33
W, tungsten	4.55

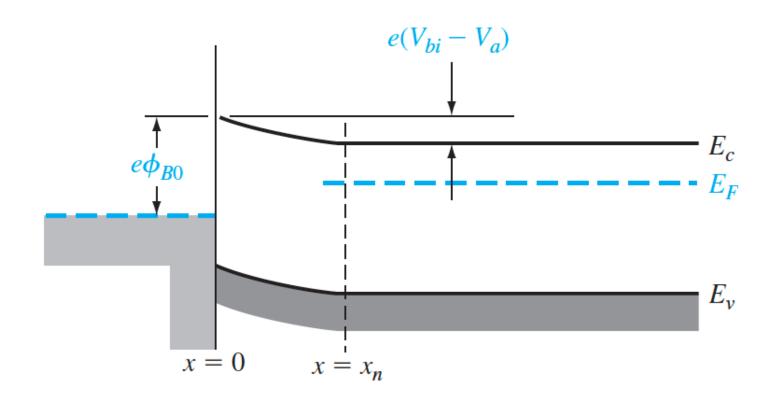
Element	Electron affinity, χ
Ge, germanium	4.13
Si, silicon	4.01
GaAs, gallium arsenide	4.07
AlAs, aluminum arsenide	3.5

- Schottky barrier: $\phi_{B0} = (\phi_m \chi)$
- Built-in potential barrier: $V_{\rm bi} = \phi_{B0} \phi_n$



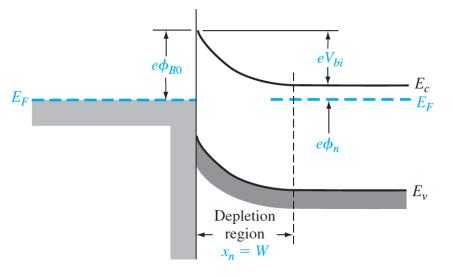


Reverse bias

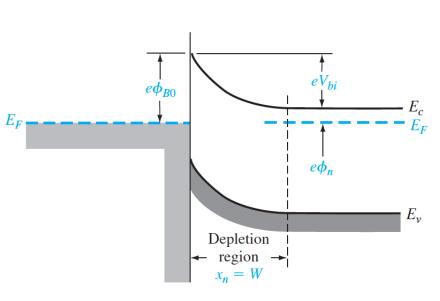


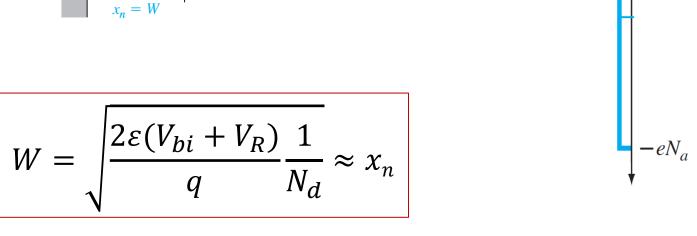
Forward bias

Ideal junction properties (electric field)













 $+x_n$

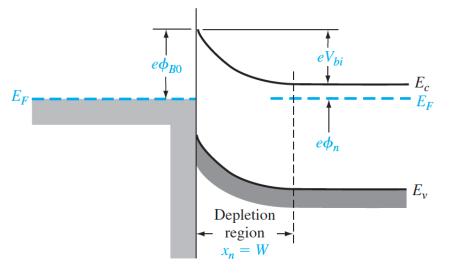
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 $+eN_d$

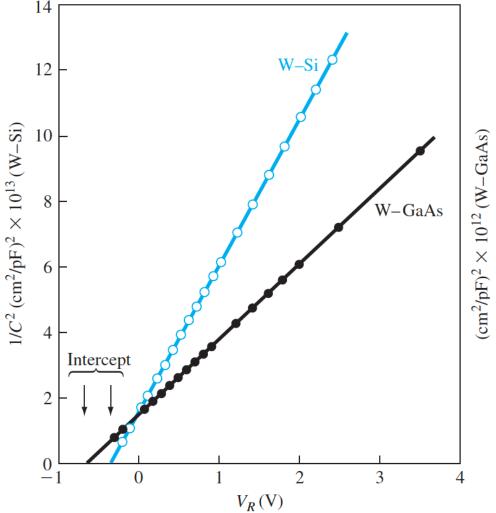
Problem example #1

A metal-semiconductor junction is formed between a metal with a work function of 4.3 eV and p-type silicon with an electron affinity of 4.0 eV. The acceptor doping concentration in the silicon is $N_a = 5 \times 10^{16}$ cm⁻³. Assume T = 300K. (a) Sketch the energy-band diagram. (b) Determine the height of the Schottky barrier. (c) Sketch the energy-band diagram with an applied reverse-biased voltage of $V_R = 3V$. (d) Sketch the energy-band diagram with applied forward-bias voltage of $V_a = 0.25V$.

Ideal junction properties (capacitance)



Ideal junction properties



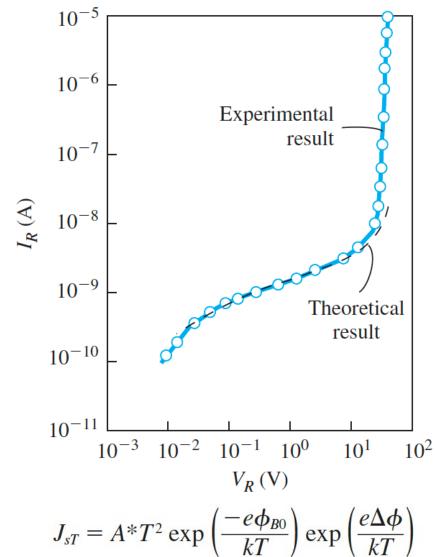
Current-voltage relationship

$$J = J_{sT} \left[\exp\left(\frac{eV_a}{kT}\right) - 1 \right]$$

$$J_{sT} = A * T^2 \exp\left(\frac{-e\phi_{Bn}}{kT}\right)$$

$$A^* \equiv \frac{4\pi e m_n^* k^2}{h^3}$$

Richardson constant



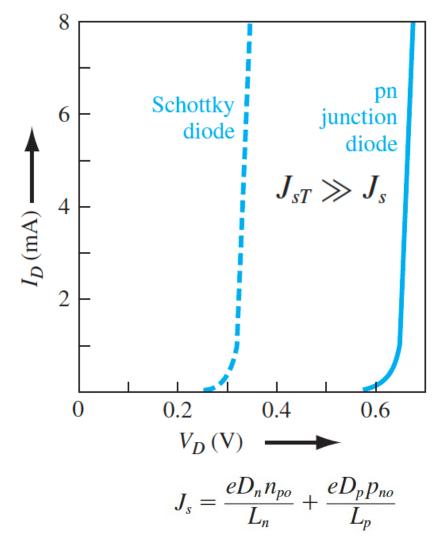
Compare Schottky diode and PN junction

$$J = J_{sT} \left[\exp\left(\frac{eV_a}{kT}\right) - 1 \right]$$

$$J_{sT} = A * T^2 \exp\left(\frac{-e\phi_{Bn}}{kT}\right)$$

$$A^* \equiv \frac{4\pi e m_n^* k^2}{h^3}$$

Richardson constant



Problem example #2

Consider a tungsten barrier on silicon with a measured barrier height of ϕ_{Bn} = 0.67eV. The effective Richardson constant is A* = 114 A/K²cm². T = 300K.

Problem example #3

Control of the Schottky Barrier Height in Monolayer WS₂ FETs using Molecular Doping

Siyuan Zhang, Hsun-Jen Chuang, Son T. Le, Curt A. Richter, Kathleen M. McCreary, Berend T. Jonker, Angela R. Hight Walker, Christina A. Hacker*

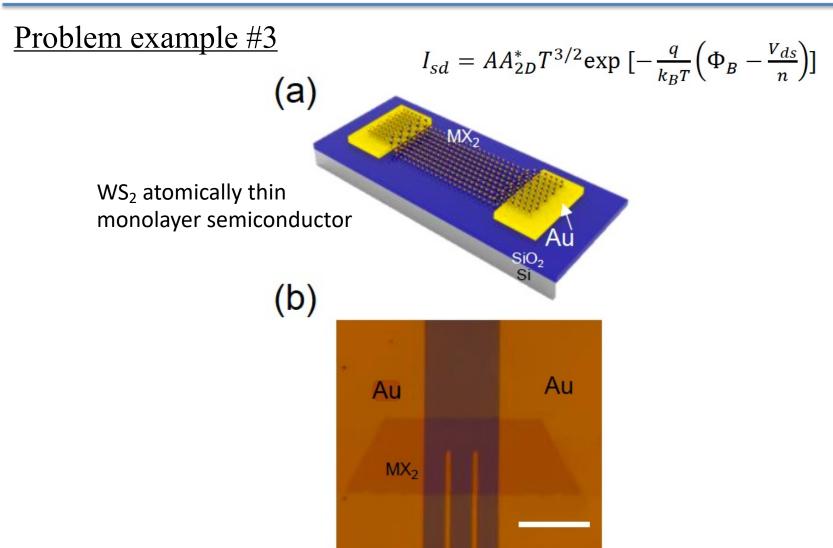
Dr. S. Zhang, Dr. S. T. Le Theiss Research, La Jolla, CA 92037, USA

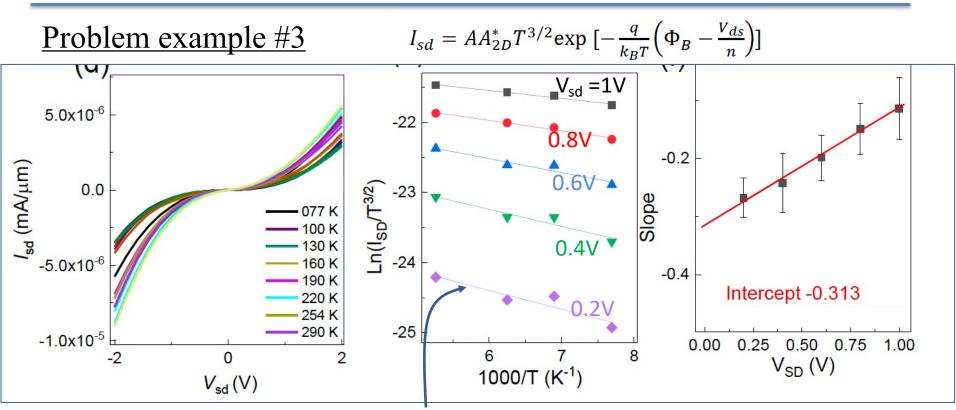
Dr. H.- J. Chuang, Nova Research Inc Washington DC 20375, USA Under review in AIP Advances

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1) Write the analytical expression of Line 1 if we take 1000/T as x and ln(ISD/T2/3) as y?

Line 1

- 2) Write the expression of Slope in the right figure.
- 3) Find Schottky barrier height Φ_{B}

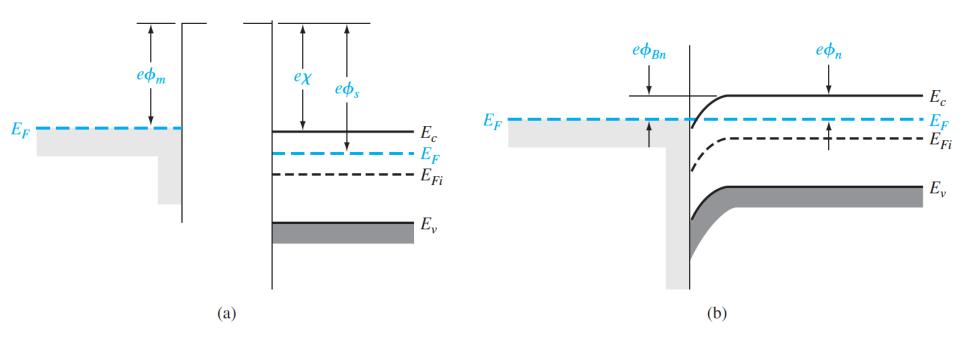
Problem example #3

Outline

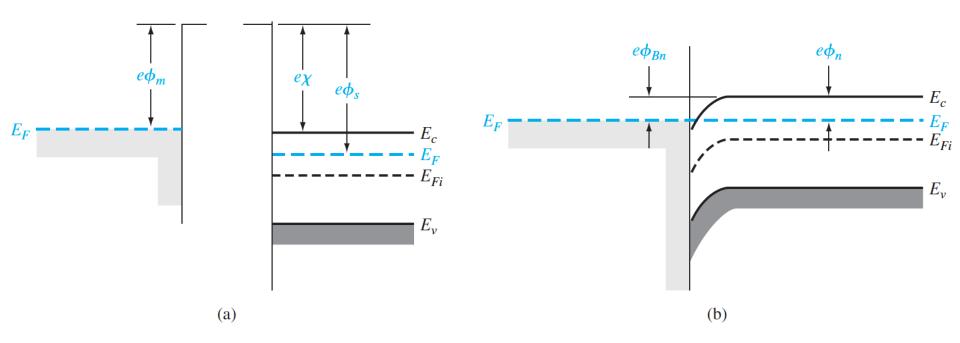
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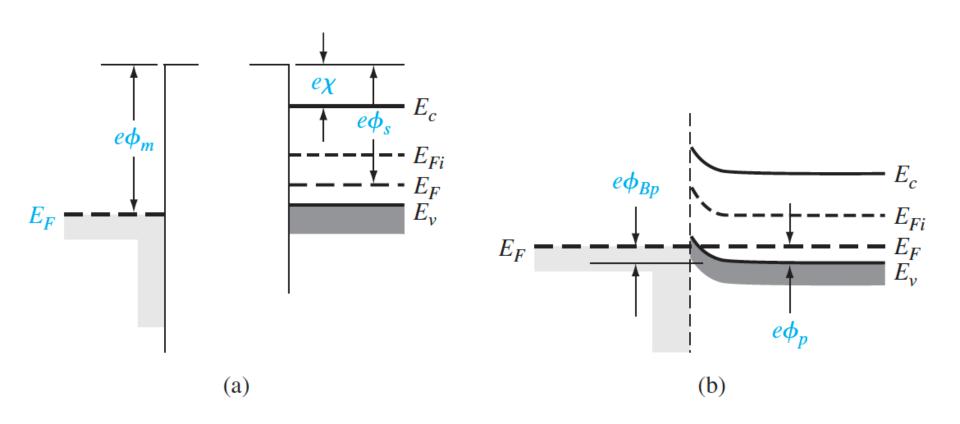
Ideal Nonrectifying Barrier

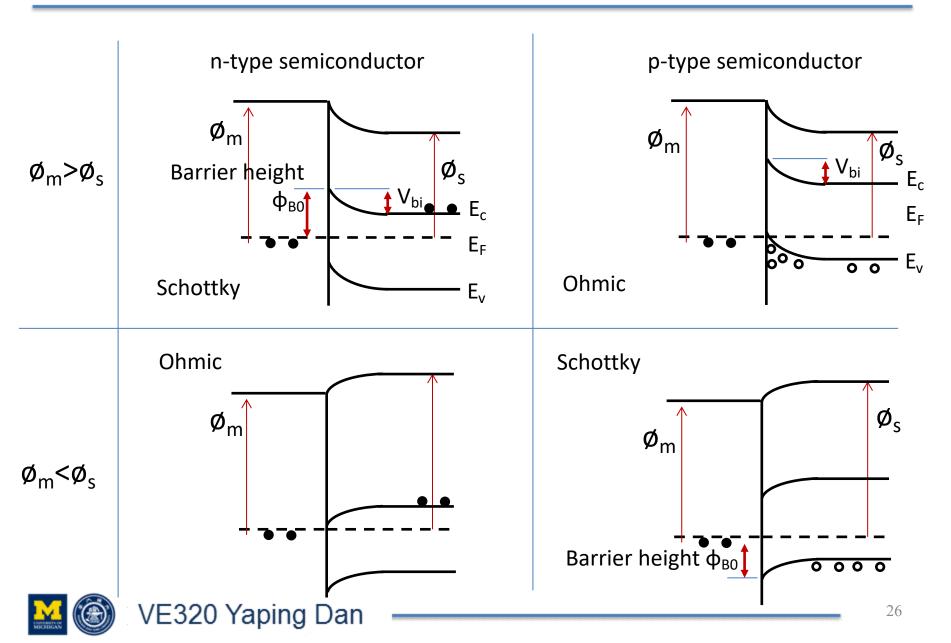


Ideal Nonrectifying Barrier



<u>Ideal Nonrectifying Barrier</u>





Problem example #4

For Si, if it is doped with phosphorus at a concentration of 10¹⁵ cm⁻³, what metal you can choose from the list for Ohmic contact.

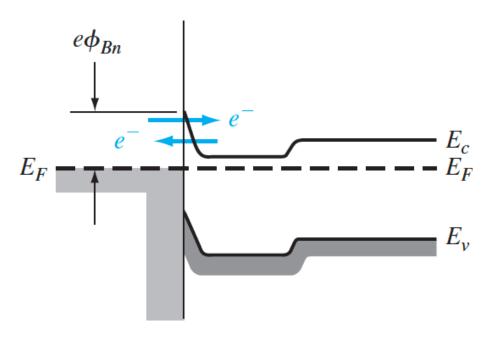
Repeat the question above for p-type Si doping at the concentration of 10^{17} cm⁻³. Si has an electron affinity of 4.01 eV and a bandgap of 1.12eV.

Table 9.1 | Work functions of some elements

Element	Work function, $\phi_{\scriptscriptstyle m}$
Ag, silver	4.26
Al, aluminum	4.28
Au, gold	5.1
Cr, chromium	4.5
Mo, molybdenum	4.6
Ni, nickel	5.15
Pd, palladium	5.12
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Ti, titanium	4.33
W, tungsten	4.55

Problem example #4

1.Tunneling Barrier



The tunneling current has the form

$$J_t \propto \exp\left(\frac{-e\phi_{Bn}}{E_{oo}}\right)$$

where

$$E_{oo} = \frac{e\hbar}{2} \sqrt{\frac{N_d}{\epsilon_s m_n^*}}$$

The tunneling current increases exponentially with doping concentration.





2.Silicide alloy

Nickel silicide, NiSi

<u>Titanium silicide</u>, TiSi₂

