

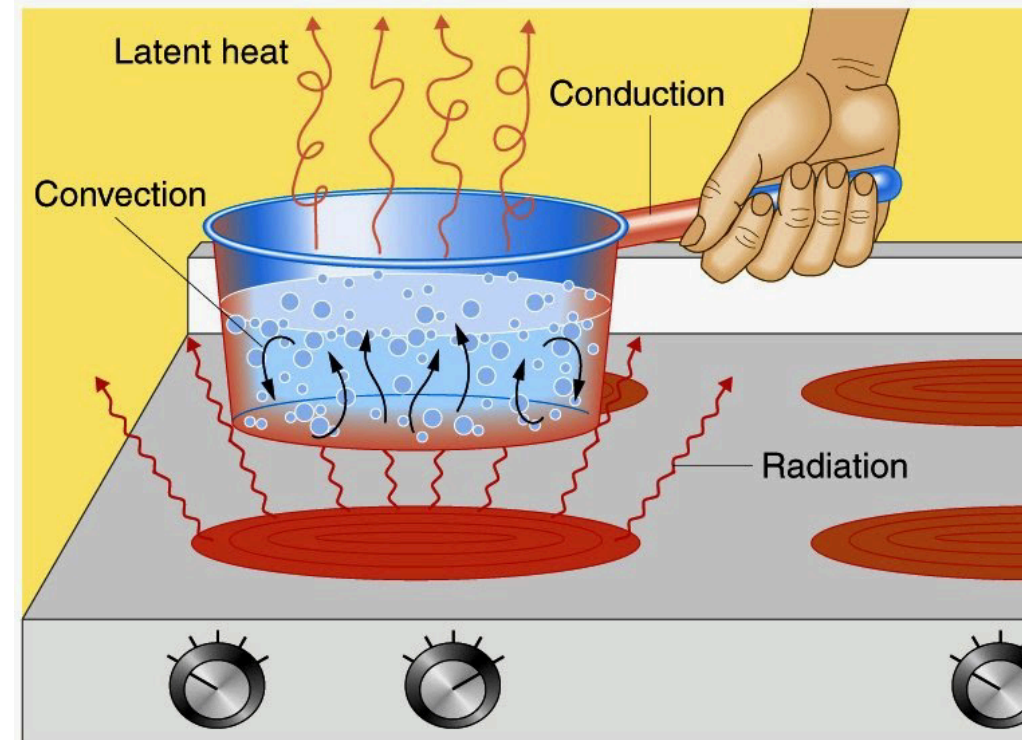
Global Warming

Lecture 3.1

Electromagnetic Radiation

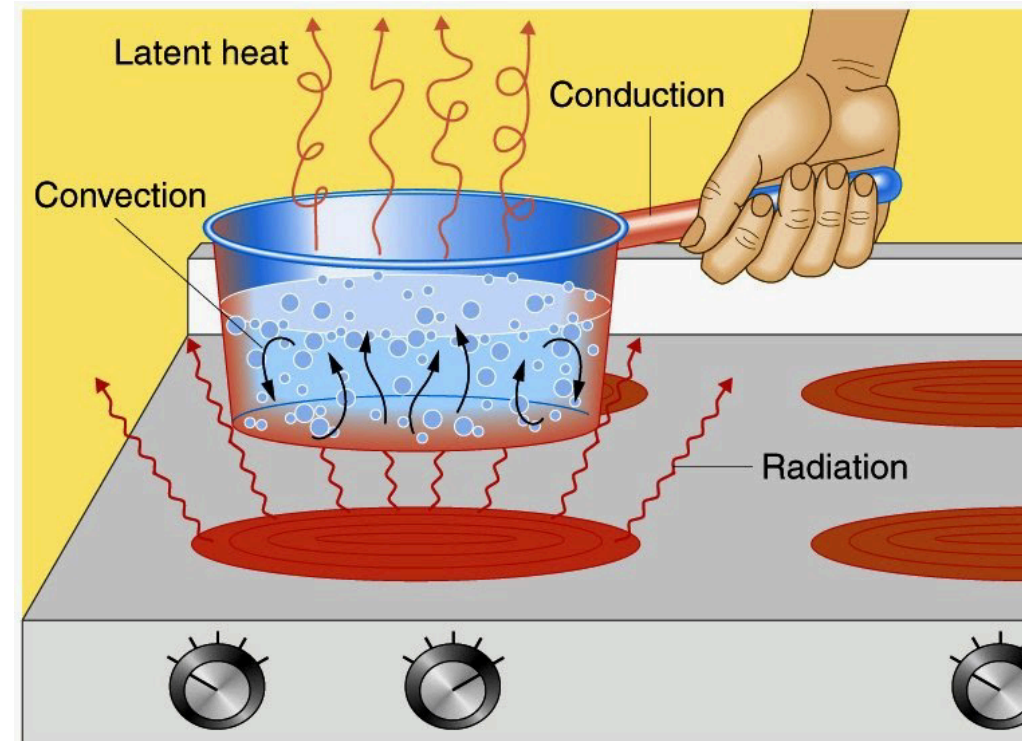
Ways Energy can be moved

- **1. Convection**
 - Convection is when a fluid (either a liquid like water or a gas like the atmosphere) moves to carry heat from one place to another.



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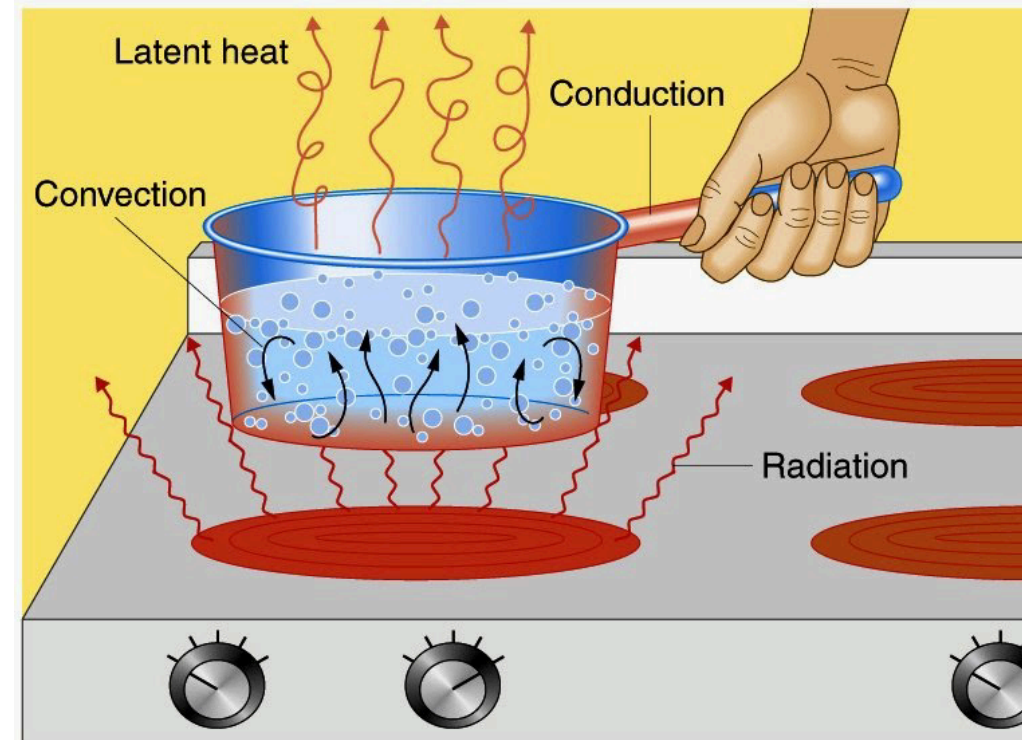
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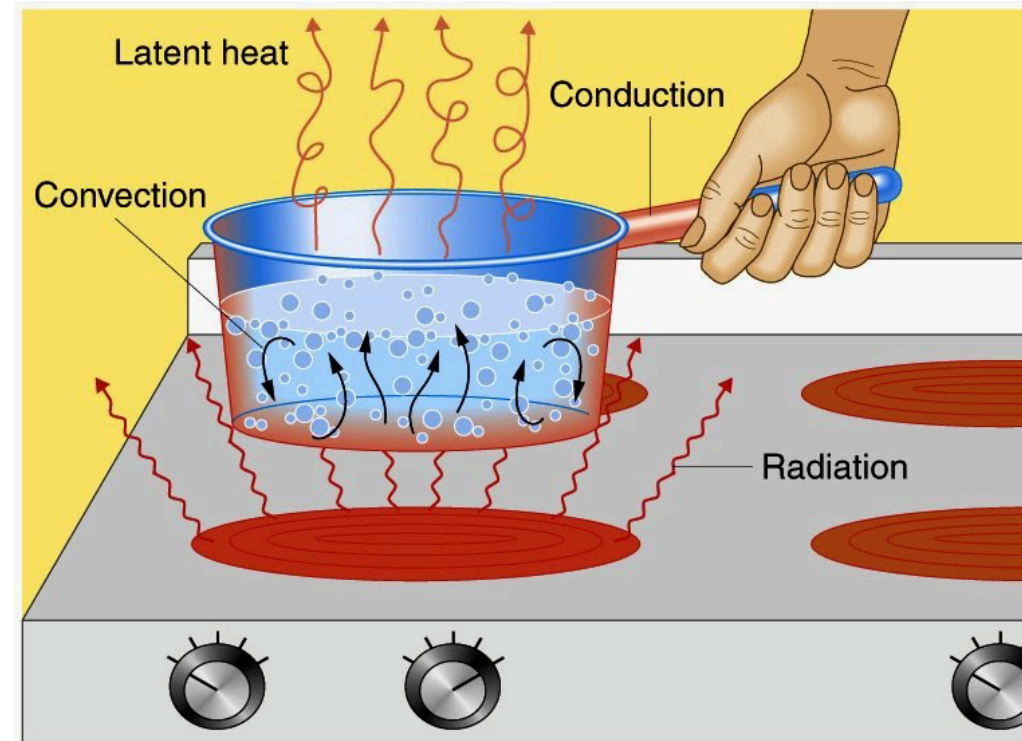


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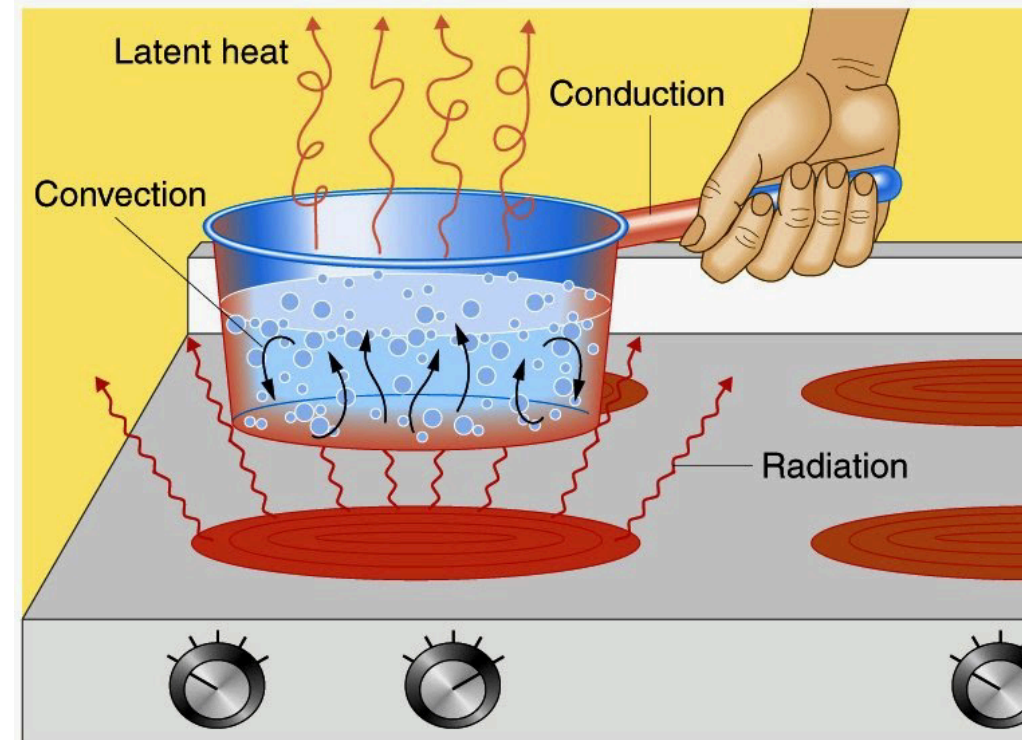
Difference between them?

Convection: both medium and heat “move” around on a human scale
Conduction: only heat “moves” around



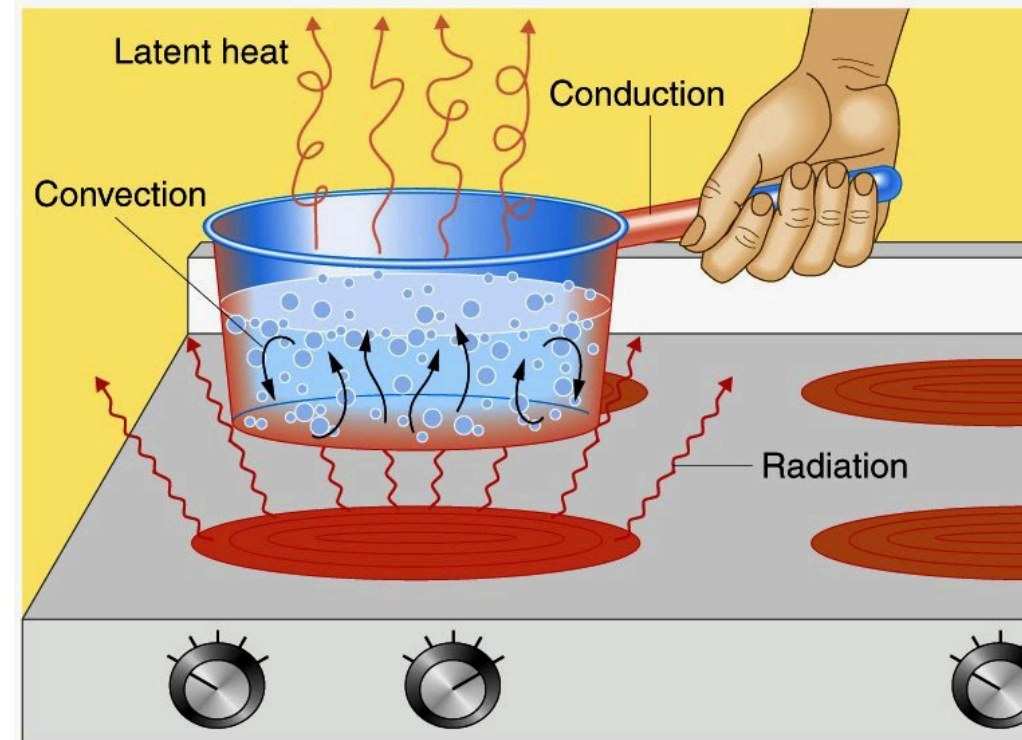
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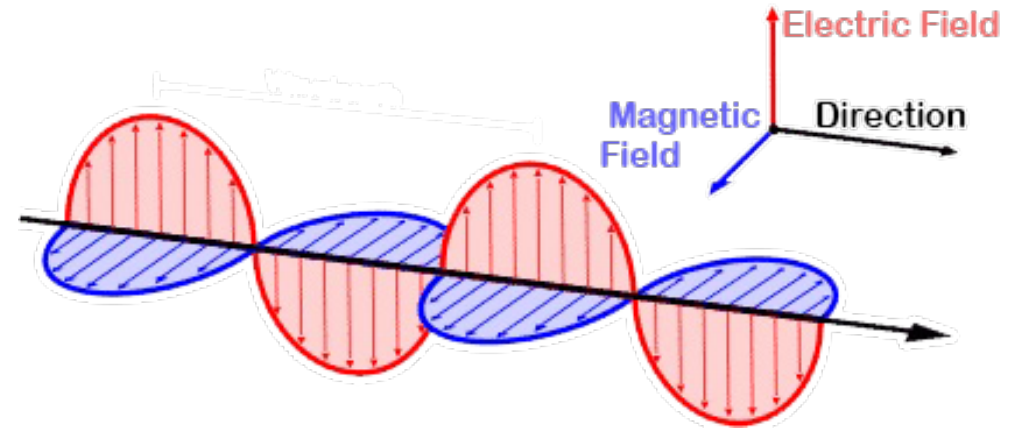
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- **4. Electromagnetic radiation**



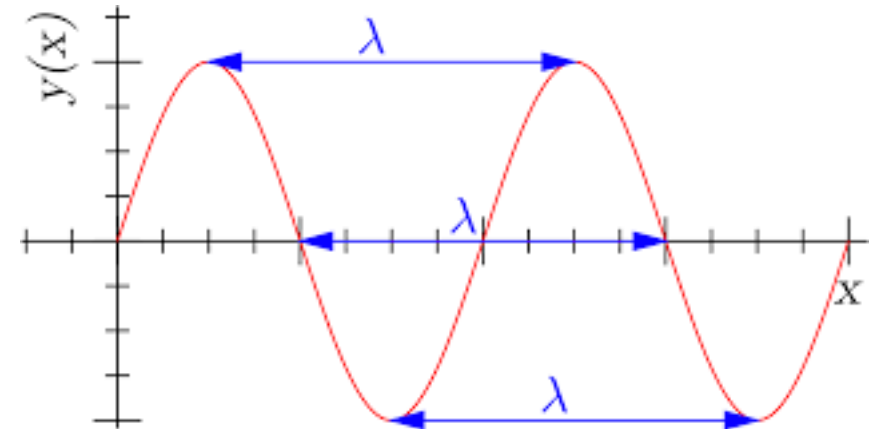
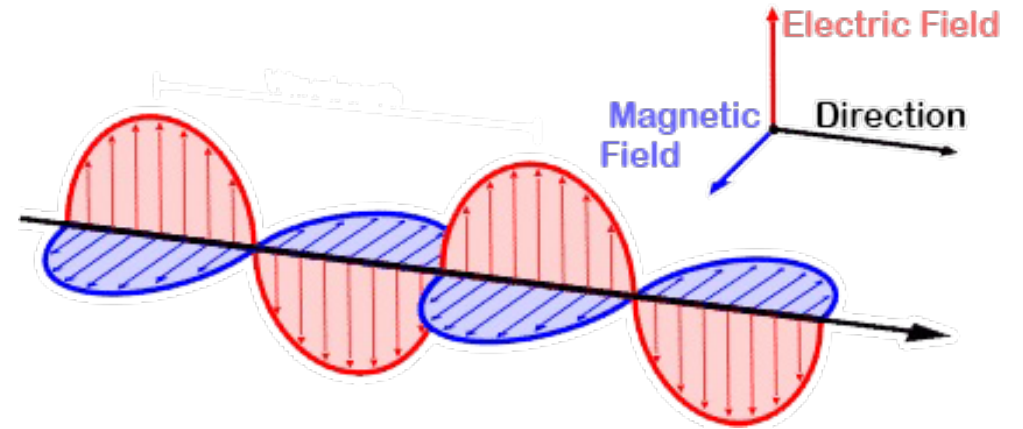
Electromagnetic radiation

- Propagating oscillations (or waves) in electric and magnetic fields. (Waves carrying energy propagates)



Electromagnetic radiation

- Propagating oscillations (or waves) in electric and magnetic fields.
- Electromagnetic radiation has a frequency (ν) and a wavelength (λ):
 - Wavelength (λ): The distance between two wave crests (or troughs)
 - Frequency (ν) is number of wave crests that pass over the origin every second

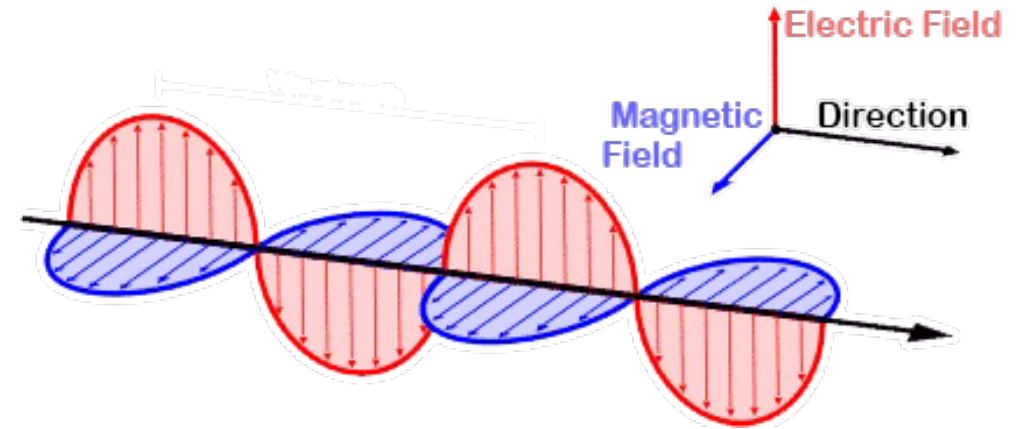


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$$c = \lambda \nu$$

$$\nu = c / \lambda$$



$$c = \lambda \nu$$

speed of light
 $2.998 \times 10^8 \text{ ms}^{-1}$
 $\sim 3 \times 10^8 \text{ m/s}$

wavelength
(in m)

frequency
(in s^{-1} or Hz)

constant for all wavelengths in vacuum!

Units!

Example:

Speed of light in a vacuum:

$$c = 3.0 \times 10^8 \text{ m s}^{-1} \text{ constant for all wavelengths!}$$

$$\nu \sim \text{s}^{-1}$$

$$\lambda = \frac{c}{\nu}$$

Blue light has a frequency of $6.67 \times 10^{14} \text{ s}^{-1}$.

What wavelength does this correspond to (in nm)?

= **nm**

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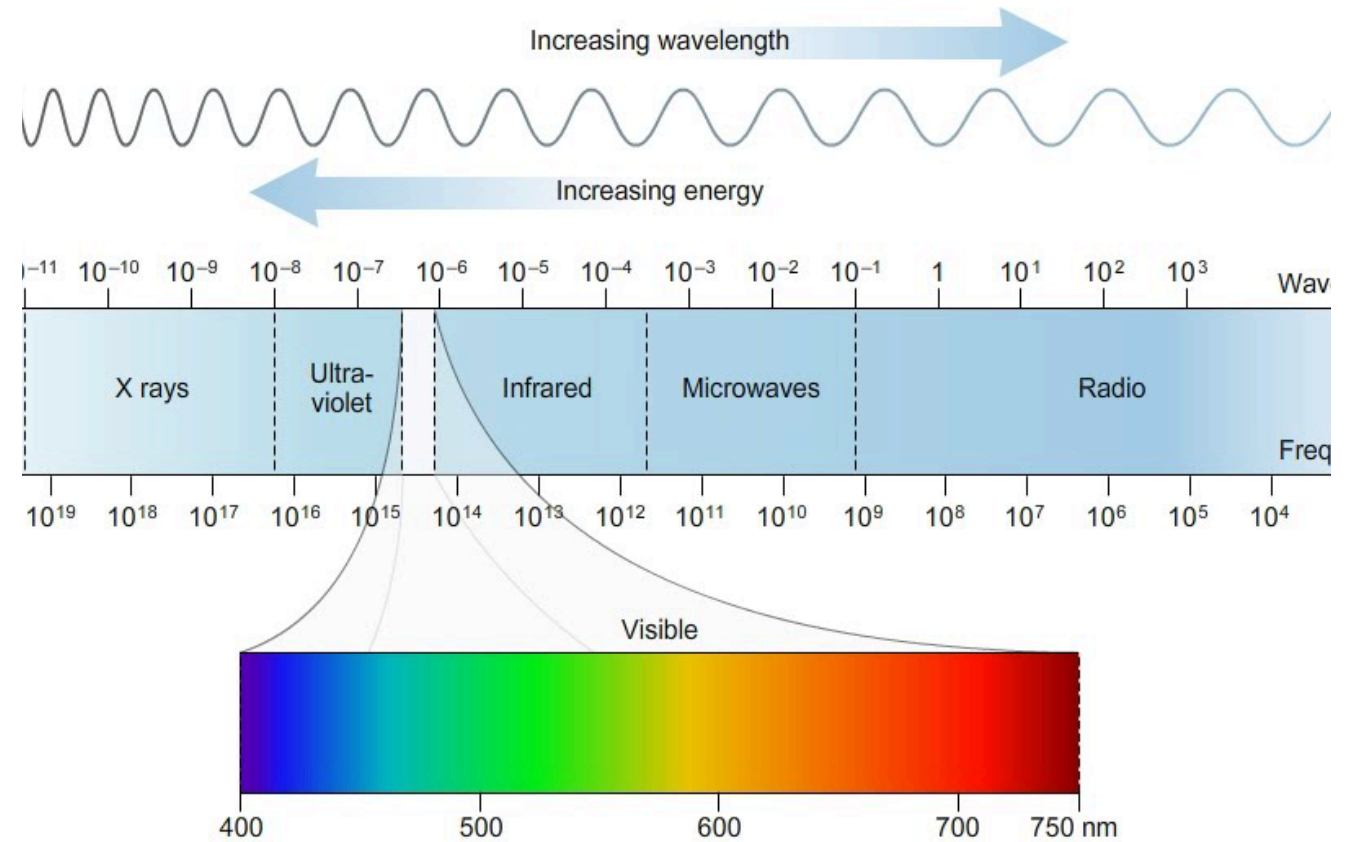
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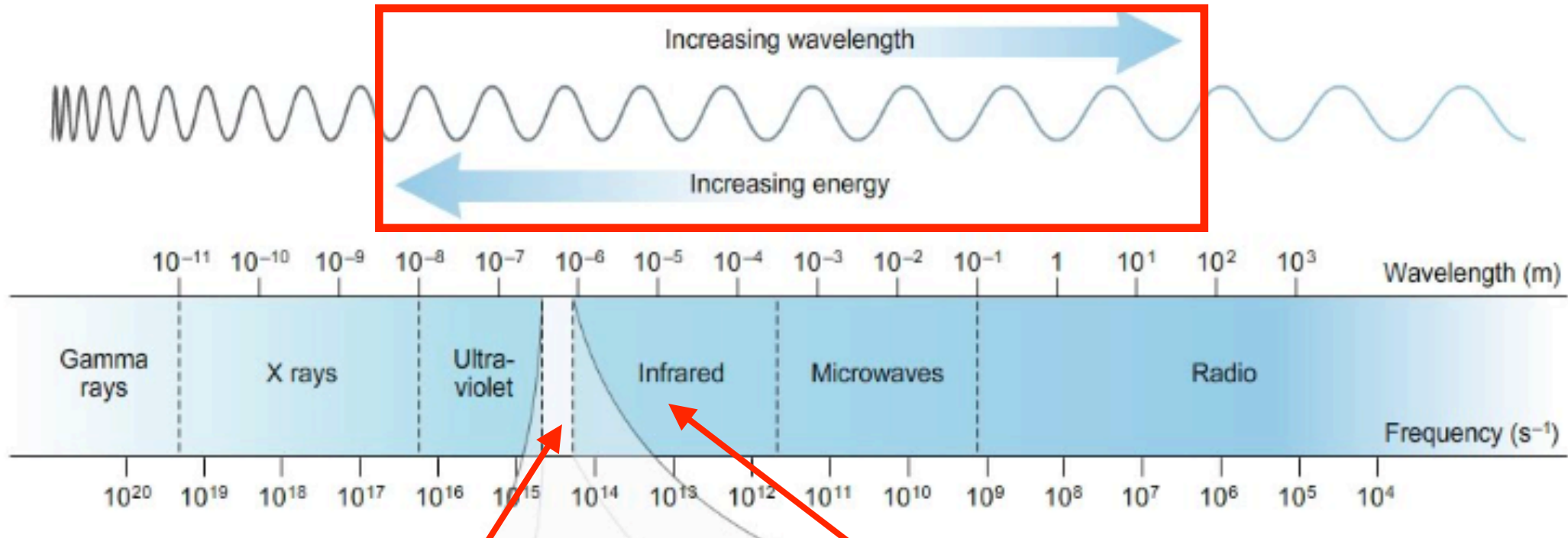
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Types of electromagnetic radiation

- Visible Light:
 - Wavelength: 400- 750nm
- Ultraviolet Radiation
 - Wavelength: 0 nm to 400 nm
- Infrared radiation:
 - Wavelength: 700 (nm) to 1 (mm).



Longwave and shortwave radiation



Sun: shortwave radiation
visible and near infrared

Earth: longwave radiation
infrared

Radiation lab:

1. Human bodies

2. Classroom, computers

3. Glass