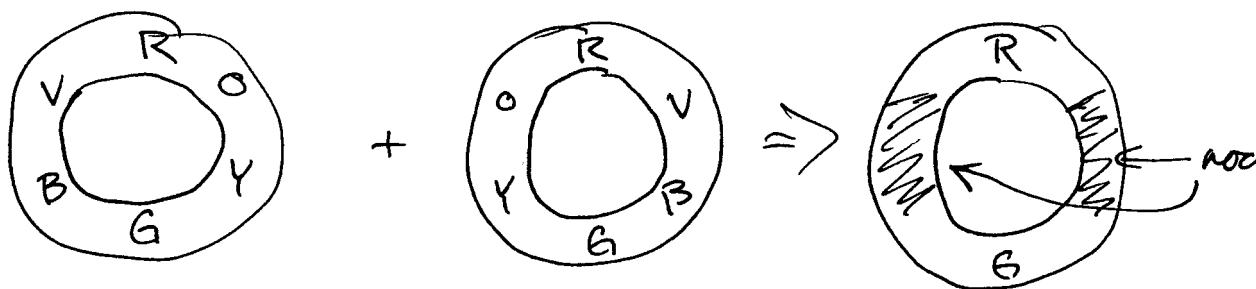


# Physics 131 - Homework XII - Solutions

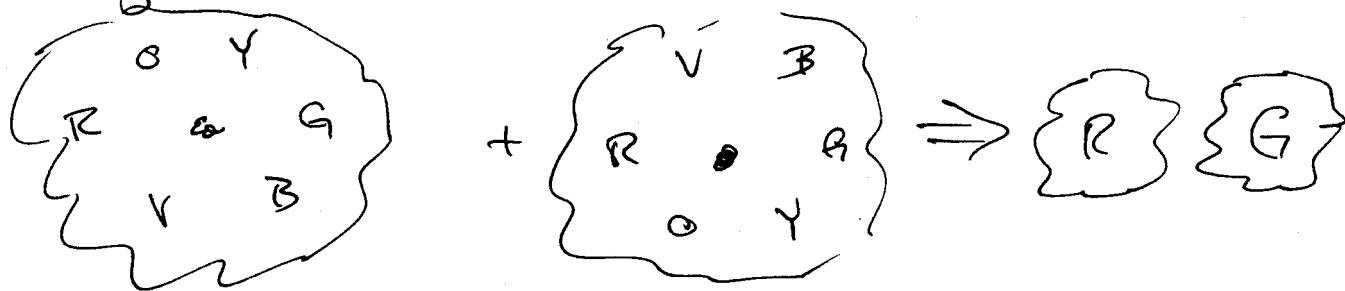
1.



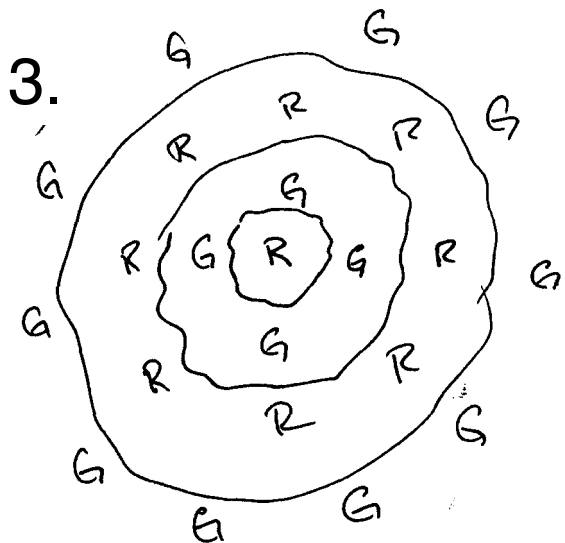
This is like the standing wave we get superposing

$$\begin{array}{c}
 \dots R \ O \ Y \ G \ B \ V \ R \ O \ Y \ G \dots \\
 \dots R \ V \ B \ G \ Y \ O \ R \ V \ B \ G \dots
 \end{array}
 +
 \Rightarrow
 \begin{array}{c}
 R \ \text{m} \ G \ \text{m} \ R \ \text{m} \ G
 \end{array}$$

2. The result is just like above, except not confined to a ~~hoop~~ hoop. So, instead of a bright patch on the hoop, we get bright blobs:



Which looks very like  $n_r=0$ ,  $n_\theta=1$ ,  $n_\phi=0$ , only turned  $90^\circ$  sideways (in fact - it is exactly that!)



This has 3 radial nodes, so  $n_r = 3$ .

This has no angular dependence,

So  ~~$n_\theta = n_\phi = 0$~~ .

$n_\theta = 0$  and

$n_\phi = 0$ ,

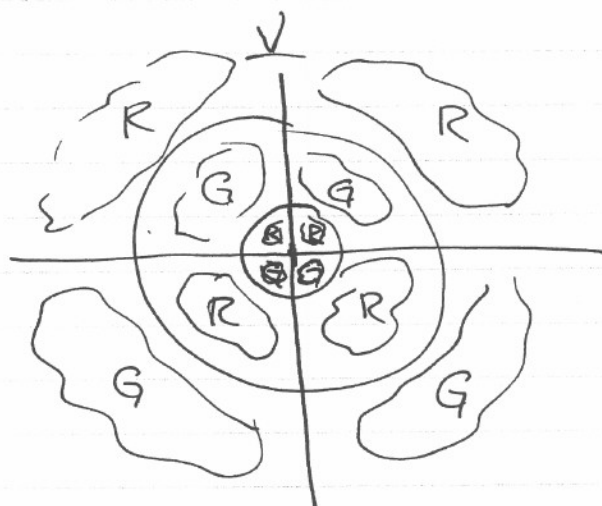
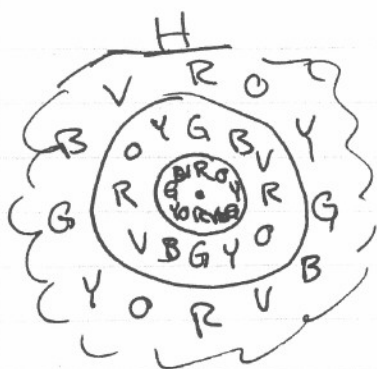
4. 2 radial nodes  $\Rightarrow n_r = 2$   
 3 angular nodes  $\Rightarrow l = 3$   
 2  $\phi$  nodes  $\Rightarrow m_l = 2$  (or  $-2$ )

$$n = n_r + n_\theta + n_\phi + 1 = n_r + l + 1 = \boxed{6}$$

$$\boxed{\begin{array}{l} l = 3 \\ m_l = 2 \text{ (or } -2) \end{array}}$$

$$\boxed{L^2 = l(l+1)\hbar^2 = 12\hbar^2}$$

5. First - draw nodal surfaces - 2 radial, 1  $\theta$   
 Then fill in colors on  $\phi$  cut to make 2  $\phi$  "nodes"  
 & alternate colors on  $V$  cut:



6. First - fill  $n = 1$  - 0 nodes -  $n_r = 0, n_\theta = 0, n_\phi = 0, 2 \text{ spins} \Rightarrow 2 \text{ electrons}$   
 Next - fill  $n = 2$  - 1 node -  $n_r = 1, \text{ rest zero} \Rightarrow 2 \text{ electrons}$   
 $n_\theta = 1, \text{ rest zero} \Rightarrow 2 \text{ electrons}$   
 $n_\phi = +1, \text{ rest zero} \Rightarrow 2 \text{ electrons}$   
 $n_\phi = -1, \text{ rest zero} \Rightarrow 2 \text{ electrons}$   
 } 8 more, for a total of 10  
 Next - fill  $n = 3$  - 2 nodes, but avoid ang nodes, so  $n_r = 2$  for 1st electron  
 $n_\theta = n_\phi = 0$  or  $n = 3, l = 0, m_l = 0$