THERMODYNAMIC QUANTITIES FOR SELECTED SUBSTANCES AT 298.15 K (25 °C)

$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Substance	ΔH_f° (kJ/mol)	ΔG_f^s (kJ/mol)	<i>S</i> ∘ (J/mol-K)	Substance	$\Delta oldsymbol{H_f^s}$ (kJ/mol)	ΔG_f^s (kJ/mol)	<i>S</i> ° (J/mol-K)
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		($C_2H_4(\sigma)$	52.30	68.11	219.4
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0	0	28 32		-84.68	-32.89	229.5
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$							-23.47	269.9
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$							-15.71	310.0
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$Al_2O_3(s)$	-1009.8	-1376.3	31.00			-15.0	231.0
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	3arium						129.7	269.2
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0	0	63.2			124.5	172.8
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$BaCO_3(s)$	-1216.3	-1137.6	112.1				237.6
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	BaO(s)	-553.5	-525.1	70.42				126.8
Bels) 0 0 9.44 $C_2H_5OH(l)$ -277.7 -174.76 160.7 BeO(s) -608.4 -579.1 13.77 $C_6H_{12}O_6(s)$ -1273.02 -910.4 212.1 Be(OH) ₂ (s) -905.8 -817.9 50.21 $CO(g)$ -110.5 -137.2 197.9 $CO_2(g)$ -393.5 -394.4 213.6 Bromine $CO_2(g)$ -393.5 -394.4 213.6 Br(g) 111.8 82.38 174.9 $CH_5COOH(l)$ -487.0 -392.4 159.8 Br(g) 111.8 82.38 174.9 $CH_5COOH(l)$ -487.0 -392.4 159.8 Br(g) 30.71 3.14 245.3 $Color Color Col$	a 11:				•			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		0	0	9.44				160.7
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$								212.1
Bromine Brownine $CO_2(g) = -393.5 = -394.4 = 213.6$ $CO_2(g) = -20.2 = -20.2$ $CO_2(g) = -20.2$ $CO_$					-			197.9
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Be(O11)2(3)	705.0	01717		-			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Bromine							
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Br(g)					107.10	-,	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Br^{-}(aq)$				Cesium			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$Br_2(g)$							
Calcium Ca(g) 179.3 145.5 154.8 Chlorine Ca(s) 0 0 41.4 Cl(g) 121.7 105.7 165.2 CaCO ₃ (s, calcite) -1207.1 -1128.76 92.88 Cl(aq) -167.2 -131.2 56.5 CaCl ₂ (s) -795.8 -748.1 104.6 Cl ₂ (g) 0 0 0 222.5 CaF ₂ (s) -1219.6 -1167.3 68.87 HCl(aq) -167.2 -131.2 56.5 CaO(s) -635.5 -604.17 39.75 HCl(g) -92.30 -95.27 186.6 Ca(OH) ₂ (s) -986.2 -898.5 83.4 CaSO ₄ (s) -1434.0 -1321.8 106.7 Cr(g) 397.5 352.6 174 Carbon Cr(s) 0 0 0 23. C(g) 718.4 672.9 158.0 Cr ₂ O ₃ (s) -1139.7 -1058.1 81. C(s, diamond) 1.88 2.84 2.43 C(s, graphite) 0 0 0 5.69 Co(g) 439 393 179 CCl ₄ (g) -106.7 -64.0 309.4 Co(s) 0 0 28 CCl ₄ (g) -679.9 -635.1 262.3 Copper CH ₄ (g) -74.8 -50.8 186.3 Cu(g) 338.4 298.6 166	$Br_2(l)$				Cs(l)			
Calcium $Ca(g)$ 179.3 145.5 154.8 Chlorine $Ca(s)$ 0 0 41.4 $Cl(g)$ 121.7 105.7 165.2 $CaCO_3(s, calcite)$ -1207.1 -1128.76 92.88 $Cl(aq)$ -167.2 -131.2 56.5 $CaCl_2(s)$ -795.8 -748.1 104.6 $Cl_2(g)$ 0 0 0 222.5 $CaF_2(s)$ -1219.6 -1167.3 68.87 $CaCO_3(s)$ -635.5 -604.17 39.75 $CaCO_3(s)$ -92.30 -95.27 186.6 $CaCO_3(s)$ -986.2 -898.5 83.4 $CaCO_3(s)$ -1434.0 -1321.8 106.7 $CaFO_3(s)$ -1434.0 -1321.8 106.7 $CaCO_3(s)$ -1219.6 $CoCO_3(s)$ -1219.6 $CoCO_3(s)$ -1221.8 106.7 $CoCO_3(s)$ -1221.8 106.8 $CoCOO_3(s)$ -1221.8 10	HBr(g)	-36.23	-53.22	198.49	Cs(s)			85.15
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Calcium				CsCl(s)	-442.8	-414.4	101.2
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		179.3	145.5	154.8	Chlorine			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						121.7	105.7	165.2
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				92.88				56.5
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	=				,			222.96
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$								56.5
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-							186.69
CasO ₄ (s) -1434.0 -1321.8 106.7 Chromium $Cr(g)$ 397.5 352.6 174.2 $Carbon$ $Cr(s)$ 0 0 23.2 $C(g)$ 718.4 672.9 158.0 $Cr2O3(s)$ -1139.7 -1058.1 81.2 $C(s, diamond)$ 1.88 2.84 2.43 $Cobalt$ $C(s, graphite)$ 0 0 5.69 $Co(g)$ 439 393 179 $CCl4(g)$ -106.7 -64.0 309.4 $Co(s)$ 0 0 28 $CCl4(l)$ -139.3 -68.6 214.4 $CF4(g)$ -679.9 -635.1 262.3 $Copper$ $CH4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166					_			
Carbon Cr(g) 397.5 352.6 174.1 $Cr(g)$ 397.5 352.6 174.2 $Cr(g)$ 718.4 672.9 158.0 $Cr_2O_3(s)$ -1139.7 -1058.1 81. $C(s, diamond)$ 1.88 2.84 2.43 $Cobalt$ $C(s, graphite)$ 0 0 5.69 $Col_4(g)$ -106.7 -64.0 309.4 $Co(g)$ 439 393 179 $Col_4(l)$ -139.3 -68.6 214.4 $Cr_4(g)$ -679.9 -635.1 262.3 $Copper$ $CH_4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166	-							
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		1 10 110						174.2
$C(s, diamond)$ 1.88 2.84 2.43 Cobalt $C(s, graphite)$ 0 0 5.69 $Co(g)$ 439 393 179 $CCl_4(g)$ -106.7 -64.0 309.4 $Co(s)$ 0 0 0 28 $CCl_4(l)$ -139.3 -68.6 214.4 $Co(s)$ 0 0 28 $CF_4(g)$ -679.9 -635.1 262.3 Copper $CH_4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166								23.6
$C(s, graphite)$ 0 0 5.69 Cobalt Col ₄ (g) Co(g) 439 393 179 $CCl_4(g)$ -106.7 -64.0 309.4 Co(s) 0 0 0 28 $CCl_4(l)$ -139.3 -68.6 214.4 Co(s) 0 0 28 $CF_4(g)$ -679.9 -635.1 262.3 Copper $CH_4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166	-				$Cr_2O_3(s)$	-1139.7	-1058.1	81.2
$C(s, graphite)$ 0 0 5.69 $Co(g)$ 439 393 179 $CCl_4(g)$ -106.7 -64.0 309.4 $Co(s)$ 0 0 0 28 $CCl_4(l)$ -139.3 -68.6 214.4 $Co(s)$ 0 0 28 $CF_4(g)$ -679.9 -635.1 262.3 Copper $CH_4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166					Cobalt			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						439	393	179
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CCl_4(g)$	-106.7			-			28.4
$CH_4(g)$ -74.8 -50.8 186.3 $Cu(g)$ 338.4 298.6 166	$CCl_4(l)$	-139.3				•	v	201
10)	$CF_4(g)$	-679.9	-635.1					
$C_2H_2(g)$ 226.77 209.2 200.8 $Cu(s)$ 0 0 33	$CH_4(g)$	-74.8	-50.8		-		298.6	166.3
	$C_2H_2(g)$	226.77	209.2	200.8	Cu(s)	0	0	33

Substance	∆ <i>H</i> _f (kJ/mol)	∆G _f (kJ/mol)	S° (J/mol-K)	Substance	∆ <i>H</i> _f (kJ/mol)	$\Delta \mathbf{G}_{f}^{\circ}$ (kJ/mol)	S° (J/mol-K)
CuCl ₂ (s)	-205.9	-161.7	108.1	MgO(s)	-601.8	-569.6	26.8
CuO(s)	-156.1	-128.3	42.59	$Mg(OH)_2(s)$	-924.7	-833.7	63.24
$Cu_2O(s)$	-170.7	-147.9	92.36	Manganasa			
Fluorine				Manganese $Mn(g)$	280.7	238.5	173.6
F(g)	80.0	61.9	158.7	Mn(s)	0	0	32.0
F(g) F(aq)	-332.6	-278.8	-13.8	MnO(s)	-385.2	-362.9	59.7
F(uq) $F_2(g)$	-552.6 0	-2/8.8 0	202.7	$MnO_2(s)$	-519.6	-464.8	53.14
HF(g)	-268.61	-270.70	173.51	$MnO_4^-(aq)$	-541.4	-447.2	191.2
111-(8)	-200.01	-270.70	1/3.31		2.2		
Hydrogen				Mercury		21.76	174.00
H(g)	217.94	203.26	114.60	Hg(g)	60.83	31.76	174.89
$H^+(aq)$	0	0	0	Hg(l)	0	0	77.40
$H^+(g)$	1536.2	1517.0	108.9	$HgCl_2(s)$	-230.1	-184.0	144.5
$H_2(g)$	0	0	130.58	$Hg_2Cl_2(s)$	-264.9	-210.5	192.5
Iodine				Nickel			
I(g)	106.60	70.16	180.66	Ni(g)	429.7	384.5	182.1
I ⁻ (aq)	-55.19	-51.57	111.3	Ni(s)	0	0	29.9
$I_2(g)$	62.25	19.37	260.57	$NiCl_2(s)$	-305.3	-259.0	97.65
$I_2(s)$	0	0	116.73	NiO(s)	-239.7	-211.7	37.99
HI(g)	25.94	1.30	206.3	Nitrogen			
				N(g)	472.7	455.5	153.3
Iron	415.5	369.8	180.5	$N_2(g)$	0	0	191.50
Fe(g)	415.5	0	27.15	$N_2(g)$ $NH_3(aq)$	-80.29	-26.50	111.3
Fe(s)	0		113.4	$NH_3(g)$	-46.19	-16.66	192.5
$Fe^{2+}(aq)$	-87.86	-84.93 -10.54	293.3	$NH_4^+(aq)$	-132.5	-79.31	113.4
$Fe^{3+}(aq)$	-47.69		117.9	$N_2H_4(g)$	95.40	159.4	238.5
$FeCl_2(s)$	-341.8	-302.3	142.3	$NH_4CN(s)$	0.0		
$FeCl_3(s)$	-400	-334	60.75	$NH_4Cl(s)$	-314.4	-203.0	94.6
FeO(s)	-271.9	-255.2 -740.98	89.96	$NH_4NO_3(s)$	-365.6	-184.0	151
$Fe_2O_3(s)$	-822.16		146.4	NO(g)	90.37	86.71	210.62
$Fe_3O_4(s)$	-1117.1	-1014.2 -160.1	52.92	$NO_2(g)$	33.84	51.84	240.45
$FeS_2(s)$	-171.5	-160.1	32.72	$N_2O(g)$	81.6	103.59	220.0
Lead				$N_2O_4(g)$	9.66	98.28	304.3
Pb(s)	0	0	68.85	NOCl(g)	52.6	66.3	264
$PbBr_2(s)$	-277.4	-260.7	161	$HNO_3(aq)$	-206.6	-110.5	146
$PbCO_3(s)$	-699.1	-625.5	131.0	$HNO_3(uq)$ $HNO_3(g)$	-134.3	-73.94	266.4
$Pb(NO_3)_2(aq)$	-421.3	-246.9	303.3	111103(g)	25		
$Pb(NO_3)_2(s)$	-451.9	_		Oxygen			161.0
PbO(s)	-217.3	-187.9	68.70	O(g)	247.5	230.1	161.0
				$O_2(g)$	0	0	205.0
Lithium	159.3	126.6	138.8	$O_3(g)$	142.3	163.4	237.6
Li(g)	0	0	29.09	$OH^-(aq)$	-230.0	-157.3	-10.7
Li(s)	-278.5	-273.4	12.2	$H_2O(g)$	-241.82	-228.57	188.83 69.91
Li ⁺ (aq)	685.7	648.5	133.0	$H_2O(l)$	-285.83	-237.13	
$\operatorname{Li}^+(g)$	-408.3	-384.0	59.30	$H_2O_2(g)$	-136.10	-105.48	232.9
LiCl(s)	- 400.3			$H_2O_2(l)$	-187.8	-120.4	109.6
Magnesium			148.6	Phosphorus			
Mg(g)	147.1	112.5		P(g)	316.4	280.0	163.2
Mg(s)	0	0	32.51 89.6	$P_2(g)$	144.3	103.7	218.1
$MgCl_2(s)$	-641.6	-592.1	07.0	12/8/			

Substance	∆ <i>H</i> _f (kJ/mol)	ΔG_f^s (kJ/mol)	<i>S</i> ∘ (J/mol₋K)	Substance	ΔH_f° (kJ/mol)	ΔG_f^s (kJ/mol)	<i>S</i> ° (J/mol₋K)
$P_4(g)$	58.9	24.4	280	$AgNO_3(s)$	-124.4	-33.41	140.9
$P_4(s, red)$	-17.46	-12.03	22.85	Sodium			
$p_4(s, white)$	0	0	41.08		1055		152.7
$PCl_3(g)$	-288.07	-269.6	311.7	Na(g)	107.7	77.3	153.7
$PCl_3(l)$	-319.6	-272.4	217	Na(s)	0	0	51.45
$PF_5(g)$	-1594.4	-1520.7	300.8	$Na^+(aq)$	-240.1	-261.9	59.0
$PH_3(g)$	5.4	13.4	210.2	$Na^+(g)$	609.3	574.3	148.0
$P_4O_6(s)$	-1640.1			NaBr(aq)	-360.6	-364.7	141.00
$P_4O_{10}(s)$	-2940.1	-2675.2	228.9	NaBr(s)	-361.4	-349.3	86.82
$POCl_3(g)$	-542.2	-502.5	325	$Na_2CO_3(s)$	-1130.9	-1047.7	136.0
$POCl_3(l)$	-597.0	-520.9	222	NaCl(aq)	-407.1	-393.0	115.5
$H_3PO_4(aq)$	-1288.3	-1142.6	158.2	NaCl(g)	-181.4	-201.3	229.8
				NaCl(s)	-410.9	-384.0	72.33
Potassium	20.00	<		$NaHCO_3(s)$	-947.7	-851.8	102.1
K(g)	89.99	61.17	160.2	$NaNO_3(aq)$	-446.2	-372.4	207
K(s)	0	0	64.67	$NaNO_3(s)$	-467.9	-367.0	116.5
KCl(s)	-435.9	-408.3	82.7	NaOH(aq)	-469.6	-419.2	49.8
$KClO_3(s)$	-391.2	-289.9	143.0	NaOH(s)	-425.6	-379.5	64.46
$KClO_3(aq)$	-349.5	-284.9	265.7	$Na_2SO_4(s)$	-1387.1	-1270.2	149.6
$K_2CO_3(s)$	-1150.18	-1064.58	155.44	Strontium			
$KNO_3(s)$	-492.70	-393.13	132.9	SrO(s)	-592.0	-561.9	54.9
$K_2O(s)$	-363.2	-322.1	94.14	Sr(g)	164.4	110.0	164.6
$KO_2(s)$	-284.5	-240.6	122.5				
$K_2O_2(s)$	-495.8	-429.8	113.0	Sulfur			
KOH(s)	-424.7	-378.9	78.91	S(s, rhombic)	0	0	31.88
KOH(aq)	-482.4	-440.5	91.6	$S_8(g)$	102.3	49.7	430.9
Rubidium				$SO_2(g)$	-296.9	-300.4	248.5
Rb(g)	85.8	55.8	170.0	$SO_3(g)$	-395.2	-370.4	256.2
Rb(s)	0	0	76.78	$SO_4^{2-}(aq)$	-909.3	-744.5	20.1
RbCl(s)	-430.5	-412.0	92	$SOCl_2(l)$	-245.6	_	
$RbClO_3(s)$	-392.4	-292.0	152	$H_2S(g)$	-20.17	-33.01	205.6
				$H_2SO_4(aq)$	-909.3	-744.5	20.1
Scandium	255 0	226.1	174.7	$H_2SO_4(l)$	-814.0	-689.9	156.1
Sc(g)	377.8	336.1	34.6	Titanium			
Sc(s)	0	0	34.0	Ti(g)	468	422	180.3
Selenium				Ti(s)	0	0	30.76
$H_2Se(g)$	29.7	15.9	219.0	$TiCl_4(g)$	-763.2	-726.8	354.9
Silicon				$TiCl_4(l)$	-804.2	-728.1	221.9
	368.2	323.9	167.8	$TiO_2(s)$	-944.7	-889.4	50.29
Si(g)	0	0	18.7	$11O_2(3)$	711.7	00777	
Si(s)	-73.22	−70.85	16.61	Vanadium			
SiC(s)		-572.8	239.3	V(g)	514.2	453.1	182.2
$SiCl_4(l)$ $SiO_2(s, quartz)$	-640.1 -910.9	-856.5	41.84	V(s)	0	0	28.9
Silver				Zinc			160.0
Ag(s)	0	0	42.55	Zn(g)	130.7	95.2	160.9
$Ag^+(aq)$	105.90	77.11	73.93	Zn(s)	0	0	41.63
	-127.0	-109.70	96.11	$ZnCl_2(s)$	-415.1	-369.4	111.5
AgCI(s)	-12/11	107.70			-348.0	-318.2	43.9