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4th Quarter - scroll down for most recent homework Monday, March 25, 2019 1. Finish notes from class - if need on NB p. 235 using TB pp. 536-543 and NB p. 236 using TB pp. 544-554

### **Homework - Mrs. Chaillou-Hall - Honors Chemistry**

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### **Chapter 11 A Gas Law Practice Problems ( Use notebook ...**

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### **UC Berkeley School of Law - Wikipedia**

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### **9.2 Naming and Writing Formulas for Ionic Compounds**

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Equilibrium chemistry is concerned with systems in chemical equilibrium. The unifying principle is that the free energy of a system at equilibrium is the minimum possible, so that the slope of the free energy with respect to the reaction coordinate is zero. This principle, applied to mixtures at equilibrium provides a definition of an equilibrium constant.

### **Equilibrium chemistry - Wikipedia**

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### **9.5 The Laws Governing Formulas and - Chemistry**

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In the last chapter we learned that deep neural networks are often much harder to train than shallow neural networks. That's unfortunate, since we have good reason to believe that if we could train deep nets they'd be much more powerful than shallow nets. But while the news from the last chapter is discouraging, we won't let it stop us.

### **Neural networks and deep learning**

(Third edition) by Stuart Russell and Peter Norvig. The leading textbook in Artificial Intelligence. Used in over 1400 universities in over 125 countries. The 22nd most cited. computer science publication on Citeseer (and 4th most cited publication of this century).

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