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Review Module / Chapters 1-4 7 Objectives • Define chemistry and differentiate among its traditional divisions • List several reasons to study chemistry Key Terms Part A Completion Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term ...

1.1 Chemistry Section Review - LPS

62 Review Module / Chapters 13-16 ____ 11. The electron configuration of a fluoride ion, F^- , is: a. $1s^2 2s^2 2p^5$. b. the same as that of the neon atom. c. $1s^2 2s^2 2p^6 3s^1$. d. the same as that of a potassium ion. ____ 12. Metals are good conductors of electricity because they: a. form crystal lattices. b. contain positive ions. c. contain mobile valence electrons. d. form ionic bonds.

15 Ionic Bonding and Ionic Compounds Chapter Test A - LPS

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11.3 SECTION REVIEW - Weebly

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