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a. one that has 5 electrons in the third energy level b. one with an electron configuration that ends in 4s24p5 c. the Group 6A element in period 4 5. Identify the elements that have electron configurations that end as follows. a. 2s22p4 b. 4s2 c. 3d104s2 6. What is the common characteristic of the electron configurations of the elements Ne and Ar?

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