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4. Name the element that matches the following description. a. one that has 5 electrons in the third energy level b. one with an electron configuration that ends in  $4s^2 4p^5$  c. the Group 6A element in period 4 5. Identify the elements that have electron configurations that end as follows. a.  $2s^2 2p^4$  b.  $4s^2$  c.  $3d^1 4s^2$  6.

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