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Chapter 5

Chapter 5 Electrons in Atoms43 SECTION 5.1 MODELS OF THE ATOM (pages 127–132) This section summarizes the development of atomic theory. It also explains the significance of quantized energies of electrons as they relate to the quantum mechanical model of the atom. The Development of Atomic Models (pages 127–128) 1.

SECTION 5.1 MODELS OF THE ATOM (pages 127–132)

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atoms. • electron configuration: the arrangement of various orbitals around the nuclei of atoms

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National Science Education Standards are covered in this chapter. Refer to page 4T of the Teacher Guide for an explanation of the correlations. Systems, order, and organization (UCP.1) Evidence, models, and explanation (UCP.2) ... LS. Chapter 5 Electrons in Atoms electrons. ELL. 1 s The .

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106 Core Teaching Resources ____ 11. The number of sublevels in an energy level is equal to the square of the principal quantum number of that energy level. ____ 12. The maximum number of electrons that can occupy the fourth

05 CTR ch05 7/9/04 3:26 PM Page 105 MODELS OF THE ATOM 5

Chapter 5 "Electrons in Atoms" ... • The next electrons go into the 2p orbital • only 5 more... I n c r e a s i n g e n e r g y 1s 2s 3s 4s 5s 6s 7s 2p 4p 5p 6p 3d 4d 5d 7p 6d 4f 5f. 6 • The next electrons go into the 3s orbital • only 3 more... I n c r e a s i n g e n e r g y 1s 2s 3s 4s 5s 6s 7s 2p 3p 4p 5p 6p 3d 4d 5d

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Chapter 6 The Periodic Table 145 ____ 23. Atomic size generally a. increases from left to right across a period. b. decreases from top to bottom within a group. c. remains constant within a period. d. decreases from left to right across a period. ____ 24. Of the following atoms, which one has the smallest first ionization energy?

Chapter Test A - Campbell County Schools

Name ____ Date ____ Class ____ Chapter 5 Atoms and Bonding Section 1 Summary Atoms, Bonding, and the Periodic Table Key Concepts How is the reactivity of elements related to valence electrons in atoms? What does the periodic table tell you about the atoms of elements?

Chapter 5 Atoms and Bonding Section 1 Summary | Kasim ...

valence electrons. The Alkali Metals (page 140) 3. The reactivity of alkali metals from the top of Group 1A to the bottom. 4. Sodium is stored under oil because it . The Alkaline Earth Metals (page 141) 5. Differences in reactivity among alkaline earth metals are shown by the way they react with . Find and match two properties to each element ...

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