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a. one that has 5 electrons in the third energy level b. one with an electron configuration that ends in 4s²4p⁵ c. the Group 6A element in period 4 5. Identify the elements that have electron configurations that end as follows. a. 2s²2p⁴ b. 4s² c. 3d¹⁰4s² 6. What is the common characteristic of the electron configurations of the elements Ne and Ar?

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