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A B; Why are atoms invisible to the naked eye? to small to reflect the light waves that we can see: the three examples of subatomic particles? proton, neutron, electron

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Unit 4: Waves, Sound, and Light: Chapter 15. Waves

A B; The most significant source of nuclear energy is the ____ natural background radiation: Radioactivity can be defined as: Disintegration of the Atomic Nucleus

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Physical Science Test - Final Exam 14 38. (a) (1 point) What is the name of another element in the Periodic Table that is expected to have chemical properties that are similar to the chemical properties of Argon?

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