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Make a chart on the board with the necessary information to guide the students. ... [NOTE: The negative key is just below the 3 and is a negative sign ...

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The Combined Gas Law. Expresses the relationship between pressure, volume, and temperature of a fixed amount of gas. $PV/T = k$. or . $P_1 V_1 / T_1 = P_2 V_2 / T_2$ Charles' Law. Episode 903. Ex: A sample of gas has a volume of 201 L when its temperature is 293 K and its pressure is 224 mm Hg. What volume will the gas occupy at STP? Given:

The Combined Gas Law

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Kinetic Theory and Boyle's Law. Pressure of a gas is caused by the molecules of the gas striking the walls of the container.. If the gas is . compressed. to . half. the volume it had, twice. as many . molecules. are present in any . given volume

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