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$\text{Zn(s)} + 2\text{HCl(aq)} \rightarrow \text{ZnCl}_2\text{(aq)} + \text{H}_2\text{(g)}$ The rate of the reaction is followed by measuring the loss in mass of the flask and its contents at regular time intervals.

NATIONAL SENIOR CERTIFICATE GRADE 12

Physical Science DoE/Exemplar 2008 NSS - Memorandum 4 QUESTION 3 / VRAAG 3 3.1 True / Waar [12.2.1] (2) 3.2 False / Onwaar Concentrations of products and reactants are constant. Konsentrasies van produkte en reaktante is konstant. OR/OF Rate of forward reaction = rate of reverse reaction

PHYSICAL SCIENCE P2 MEMO - Primex

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