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Atoms and Bonding Chapter Test A Atoms and Bonding Using Science Skills Use the portion of the periodic table below to answer the following questions. 26. Classifying A new period of very heavy elements from outer space has been discovered by scientists. In which group does each element belong? Write the element in the appropriate square of ...

Atoms and Bonding Chapter Test A Atoms and Bonding

Atoms and Bonding Chapter Test Using Science Skills Use the electron dot diagrams to answer the following. 21. Applying Concepts Predict the formula for the compound, if any, that would form from the elements listed below. If a compound is unlikely, explain the reason why. a. magnesium and chlorine

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What kind of bonds do these compounds have? a. ionic bonds b. covalent bonds c. metallic bonds ____ 24. Applying Concepts Which compound has a polar bond? a. HF b. F 2 c. O 2 ____ 25. Applying Concepts Which compound has a triple bond? a. O 2 b. F 2 c. N 2 H (+) (-) F O O N N F F Atoms and Bonding

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Atoms and Bonding Chapter Test A Multiple Choice Write the letter of the correct answer on the line at the left. ____ 1. Which is a property shared by most molecular compounds? a. high boiling point b. high melting point c. low melting point d. nonpolar bonds ____ 2. When an atom loses an electron, it becomes a a. positive ion. b. negative ion.

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IONIC BONDING AND IONIC COMPOUNDS CHAPTER TEST A 15 ... Atoms in compounds tend to have the electron configuration of a noble gas. Column A electron dot structure valence electron ionic bond ... 15 Ionic Bonding and Ionic Compounds Chapter Test A Author: Prentice Hall Created Date:

15 Ionic Bonding and Ionic Compounds Chapter Test A

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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