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Water beads up into small, nearly spherical drops on a paper towel. c. Water forms nearly spherical drops at the end of an eyedropper. d. An insect called a water strider is able to "walk" on water. 7. Using Figure 15.4, explain why a water drop has surface tension. 8.

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CHAPTER 15, Water and Aqueous Systems(continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass. b. Water beads up into small, nearly spherical drops on a paper towel.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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Chapter 15- Water and Aqueous Systems. because of water's hydrogen bonds, the molecules inside the water are all connected by hydrogen bonds and surround each other with hydrogen bonds, where the attractive forces are all balanced, but at the surface, they're unbalanced, so they are drawn in towards the rest of the water molecules, holding them in surface tension.

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Chapter 15 - Water and Aqueous Systems - 15.2 Homogeneous Aqueous Systems - 15.2 Lesson Check - Page 501: 17 Answer CH4 doesn't dissolve in water because it is a molecular compound with no net dipole KCl is soluble because of its ionic bonds.

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Water And Aqueous Systems Chapter 15 Chemistry. In a full glass of water, the water surface seems to bulge over the rim of the glass. Water forms nearly spherical drops at the end of an eyedropper. An insect called a water strider is able to "walk" on water.

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