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chapter 13

13.1) setting the intracellular & extracellular concentrations as:

C_{Int} (Na[†]) = 10 m/h - C_K(K[†]) = 147.5 m/h - C_K(Cl) = 10 m/h - C_K(Na[†]) = 145 m/h - C_K(K[†]) = 4.5 m/h - C_K(Cl) = 715 m/h

The transmembrane potential can be found from:

AV = V2 - V, = KBT ln C1 = (25) mv x ln C1 C2

 $Na^{+} \stackrel{Z=+1}{>} \Delta V = .66.85 \text{ mV}$ $K^{+} \stackrel{Z=+1}{>} \Delta V = .87.24 \text{ mV}$

b) C (Not) = 470 mm C (K+) = 10 mm C (CL) = 550 mm

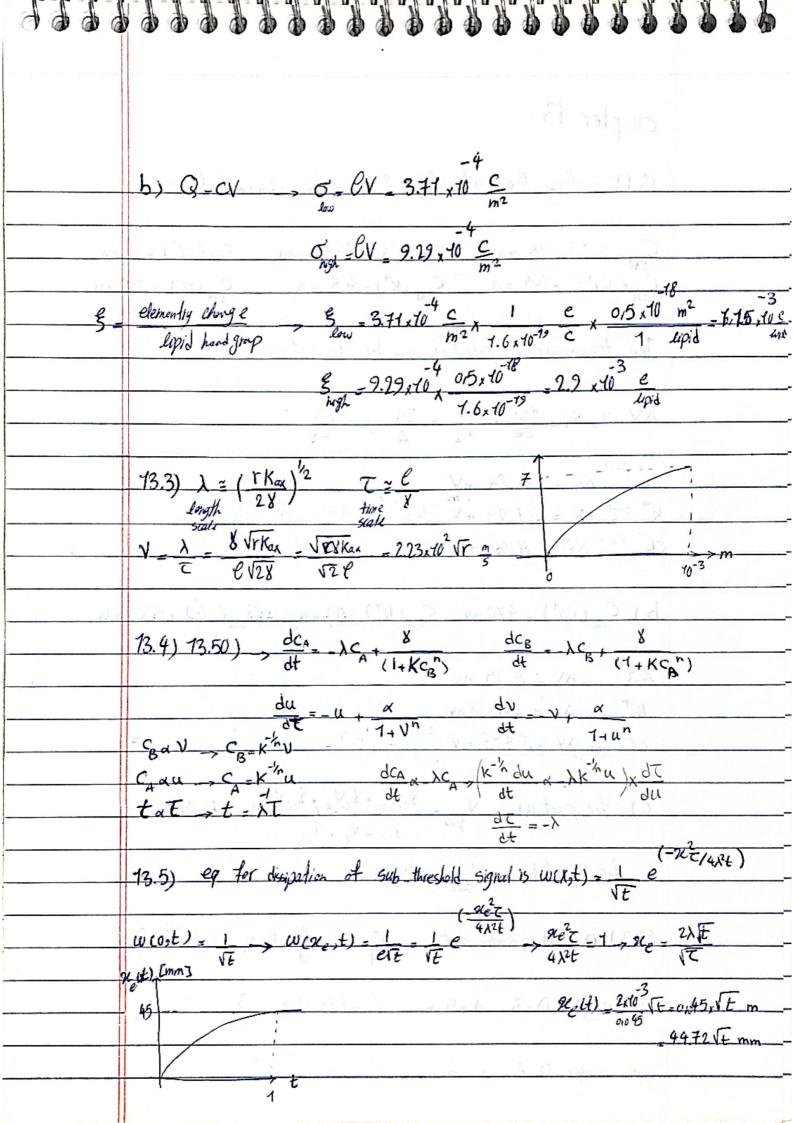
 $Na^{+} \rightarrow \Delta V = 96.25 \text{ mV}$ $K^{+} \rightarrow \Delta V = -67.28 \text{ mV}$ $Cl^{-} \rightarrow \Delta V = -100.18 \text{ mV}$

for part b) -> V953 = 79.13 inv

13.2) D = Em (dickedic const.) -> C = C = A DxE = 8.85 x10 x D

lower_range: D=3, d=5mm -> C=8.85 x 10 F x 3 = 5.31 x 10 F

upper_range: D=6, d=4 nm -, C=8.85×10 x 6 - 43.28 x 10 F



13.6) 0 11 = 5 pm C= CA=10 x (TL(5,406)2) = PF The change passing through the axon during a 100mx potential change is DQ=CN=3.14x10 x 10 = 0,314 x10 c = 7.96x40 e = 0,49x10 e b) The current moves outward at membrane V2: C) ion flow $\frac{0.49}{x70}$ $\frac{0.49 \times 40}{x}$ $\frac{0.49 \times 40}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ $\frac{1.03}{x}$ 13.7) to substitude w(x,t) = w(x,t) exp (+t/2) into 13.26 we $(\frac{\lambda^2}{dx^2} \frac{dv}{dt} - \frac{dv}{dt} - \frac{dv}{dt} - \frac{dv}{dt} - \frac{dv}{dt} + \frac{dv}{dt} - \frac{$ 12 ed du2 - T (du e + V e) - 12 e du2 - 7 d (v(21) + e =) = 0 - a - 1 - 12 d2 (ve) - 7 d (ve) =0 where v(21,t)e = w(21,t) b) substitute west = 1 exp(22/42+) in the eq. we he $\frac{\lambda^{2} \frac{d^{2}u}{dx^{2}} - \frac{\lambda^{2}}{\sqrt{t}} \frac{d}{dx} \left(-\frac{2\alpha T}{4\lambda^{2}t} e^{\frac{-2\lambda^{2}t}{4\lambda^{2}t}}\right) = \frac{\lambda^{2}}{\sqrt{t}} \times \frac{-2T}{4\lambda^{2}t} e^{\frac{-2\lambda^{2}t}{4\lambda^{2}t}} \left(1 - \frac{2\alpha T}{4\lambda^{2}t}\right)$ $= \frac{\lambda^{2} \frac{du^{2}}{4\lambda^{2}t}}{\sqrt{t}} \frac{-\frac{2\lambda^{2}}{4\lambda^{2}t}}{\sqrt{t}} \left(-\frac{1}{2} \frac{t^{2}}{t^{2}} + \frac{1}{\sqrt{t}} \frac{\alpha^{2}t}{4\lambda^{2}t^{2}}\right)$ $= \frac{\lambda^{2} \frac{du^{2}}{4\lambda^{2}t}}{\sqrt{t}} \frac{-\frac{2\lambda^{2}t}{4\lambda^{2}t}}{\sqrt{t}} \left(-\frac{1}{2} \frac{t^{2}}{t^{2}} + \frac{1}{\sqrt{t}} \frac{\alpha^{2}t}{4\lambda^{2}t^{2}}\right)$

A į	73.8) the four steady state solutions of eg 43.31 are:
a yela m	Case 2) V_{55} is small: $U_{55} = \alpha$ $V_{55} = \alpha$ Case 2) V_{55} is small: $U_{55} = \alpha$ $V_{55} = \alpha$
	COISE 3) Us and Vy are large: (0) Us - Vs = 0 T+n
	Cers 4) Uss ore smell (acct) us = Vss = a
	for case 1) -> g(21) - \alpha \frac{d}{dx} > g'(21) - n\alpha 22 \\ (1+22^n)^2
ξ (μ _{i)} μ;)=g(ey) g(ey) = (nd) 2 2; 2 (1,21; 2)2 (1,21; 2)2
j sel	$\frac{(4+2i)^{2}}{(4+2i)^{2}} = \frac{(4+2i)^{2}}{(4+2i)^{2}} = \frac{n^{2}}{(1+2i)^{2}} = \frac{n^{2}}{($
	$\frac{(1+\alpha)^2}{(1+\alpha)^2} \frac{(1+\alpha)^2}{(1+\alpha)^2} \frac{(1+\alpha)^2(1+\alpha^n)^2}{(1+\alpha)^2(1+\alpha^n)^2}$
	for axy = \(\(\(\lambda \) \(\lambda \) \(\alpha \)
	for case 2) $\frac{5(u_{55}, v_{55}) - n^2}{(1+a)^2(1+a^2)^2}$ some as case 1, stable /
	for case 3) \$ (U55, V5) = (nx)2 a -(1+n)(n-1) n32 a -(n2-1)
	of contract of undited
	73.9) for steady state > du du -o
	$\frac{u-\alpha}{55} \frac{v-\alpha}{(1+v_{ss}^n)} \xrightarrow{55} \frac{u(1+\alpha)}{(1+u_{ss}^n)} \xrightarrow{55} \frac{u(1+\alpha)}{1+u_{ss}} = \alpha$
	$u_{ss}(1+u_{ss}^{n}+\alpha)=\alpha+\alpha u_{ss}^{n} \qquad u_{ss}+u_{ss}^{n+1}+\alpha u_{ss}-\alpha-\alpha u_{ss}^{n}$
	$\alpha(u_s - u_s^n + 1) + u_s (4 + u_s^n - 1)$