

Designing minimally segregating granular mixtures for gravity-driven surface flows

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Abstract

In dense flowing bidisperse particle mixtures varying in size or density alone, smaller particles sink (percolation-driven) and lighter particles rise (buoyancy-driven). But when particle species differ from each other in both size and density, percolation and buoyancy can either enhance (large/light and small/heavy) or oppose (large/heavy and small/light) each other. In the latter case, a local equilibrium can exist in which the two mechanisms balance and particles remain mixed: this allows the design of minimally segregating mixtures by specifying particle size ratio, density ratio, and mixture concentration. Using DEM simulations, we show that mixtures specified by the design methodology remain relatively well-mixed in heap and tumbler flows. Furthermore, minimally segregating mixtures prepared in a fully segregated state in a tumbler mix over time and eventually reach a nearly uniform concentration. Tumbler experiments with large steel and small glass particles validate the DEM simulations and the potential for designing minimally segregating mixtures.

KEY WORDS

granular flow, heaps, mixing, non-segregating, segregation, tumblers

1 | INTRODUCTION

Dense flows of granular materials tend to spontaneously segregate by particle size,^{1–7} density,^{8–12} shape,^{13–16} friction coefficient,^{17,18} or other physical properties, which can be problematic in many industries due to the deleterious impact of inhomogeneity on product quality.^{19–25} Among the particle properties that drive segregation, size and density are usually the dominant factors.^{26,27} In dense flows of size-disperse equal-density particles (S-system), large particles tend to rise as small particles fall through voids,^{28–30} a segregation mechanism known as percolation. For density-disperse equal-size particle mixtures (D-system), segregation is driven by a buoyant force mechanism in which heavy particles sink and

light particles rise.^{31–33} When particle species differ from each other in both size and density (SD-system), the two segregation mechanisms interact, resulting in more complicated segregation behavior.^{34–39} Though size and density differences can reinforce each other, that is, in mixtures of large light particles and small heavy particles, we are interested here in the opposite situation where the two segregation mechanisms oppose each other, that is, in mixtures of large heavy particles and small light particles, as this case has the potential to reduce species segregation compared with the corresponding pure S- or pure D-system.

Previous studies show that the tendency of spherical particles to sink or rise in a bidisperse mixture can be characterized by the ratios of large to small particle diameter, $R_d = d_l/d_s$ (subscript l for large

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particles and s for small particles), and density, $R_p = \rho_l/\rho_s$, along with the mixture volume concentration c_l (or equivalently c_s , as $c_l + c_s = 1$, assuming the solid volume fraction ϕ is constant).^{40–44} Unlike size or density segregation alone, where which species rises or sinks depends only on R_d or R_p but does not depend on mixture concentration, the segregation direction in an SD-system can be concentration dependent, and a mixed equilibrium state can exist where the effects of particle size and density differences balance.^{26,45,46}

Although granular systems are challenging to study analytically compared with fluid flows because governing equations analogous to the Navier-Stokes equations for fluids are not yet fully established (although progress is being made in this direction^{47–52}), granular materials can offer a conceptual advantage compared with fluids. In many systems of practical interest—particularly tumbling and heap formation—the granular flow occurs only in thin regions of rapid flow,⁵³ even in large-scale industrial processes. Thus, understanding flow and segregation in this thin shear layer provides a key building block for understanding how to scale-up processes of industrial significance.⁵⁴ In fact, we recently developed a segregation model based on the continuum advection-diffusion equation^{24,25,55,56} that predicts the degree to which two non-cohesive particle species differing in both size and density will segregate in thin rapid surface flows.⁵⁷ Specifically, the model predicts a segregation velocity that depends linearly on the local shear rate and quadratically on the species concentration. The segregation velocity is characterized by two empirical coefficients that are functions of R_d and R_p . Streamwise concentration profiles predicted by incorporating this segregation velocity model into a continuum advection-diffusion-segregation transport model match DEM simulation results well for free surface heap flows over a wide range of R_d and R_p . An important feature of the model is the ability to predict the large particle “equilibrium concentration,” $c_{l,eq}$, at which size-related percolation and density-related buoyancy locally offset one another such that the segregation flux of each of the two species is zero.^{57,58} In other words, an initially mixed bidisperse particle mixture does not segregate.

In this article, we use simulation predictions for $c_{l,eq}$ to demonstrate how minimally segregating granular mixtures can be designed based on appropriate choices of R_d , R_p , and c_l . Particle mixtures prepared near the predicted equilibrium conditions are tested in discrete element method (DEM) simulations for free surface flows on bounded heaps and in rotating tumblers to characterize the degree to which the particles remain mixed. We also provide a limited number of experiments confirming the rotating tumbler simulations. Because flows of granular materials are often restricted to thin regions of rapid surface flow even in large-scale systems, this research offers an approach to intentionally design particle systems for industrial processes for which otherwise segregating particles will become or remain relatively well-mixed.

2 | PREDICTING NON-SEGREGATING MIXTURE CONDITIONS

Consider DEM simulations of combined size and density segregation of a bidisperse mixture in a single-sided quasi-2D bounded heap flow.

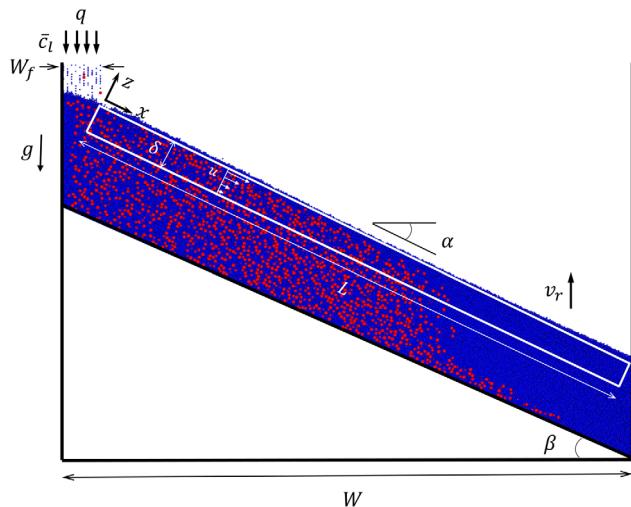


FIGURE 1 Quasi-2D bounded heap setup and segregation example from DEM simulation. For a large-particle feed concentration of $\bar{c}_l = 0.2$, large heavy particles (red, $d_l = 3$ mm, $\rho_l = 4$ g/cm 3) sink, while small light particles (blue, $d_s = 1.5$ mm, $\rho_s = 1$ g/cm 3) rise in the flowing layer producing a large particle enriched upstream region and a nearly pure-small-particle downstream region in the static portion of the heap. $R_d = 2$, $R_p = 4$, $W_f = 3.3$ cm, $W = 50$ cm, $L = 52$ cm, $q = 20$ cm 2 /s, $\delta \approx 1.5$ cm, $v_r = 0.4$ cm/s, $\alpha = 26.3^\circ$.

For the example shown in Figure 1, the domain has width W and thickness T in the y -direction. Particles flow down and to the right in a thin flowing layer of length L and relatively constant thickness δ ⁵⁹ (corresponding approximately to the white rectangle having a depth exaggerated by a factor of about two to make it more visible) and are deposited continuously at a uniform rate onto the static bed. Particles are fed onto the left side of the heap at an effective 2-D flow rate $q = Q/T$ (Q is the volumetric feed rate) and with large particle feed concentration \bar{c}_l , where the bar indicates the feed concentration or the global concentration rather than the local large particle concentration, c_l ; the heap rises with velocity $v_r = q/WT$. The coordinate system is rotated by the repose angle, α , such that the x -axis is in the streamwise direction, the y -axis is in the spanwise direction, and the z -axis is normal to the free surface. The origin is located on the free surface at the downstream edge of the vertical feed region and rises with the free surface of the heap. To reduce computation time, the bottom wall is inclined at an angle $\beta = 28^\circ$, roughly matching the repose angle α . Simulations are performed using our in-house DEM code,^{57,60} which runs on CUDA-enabled GPUs. The DEM code has been previously validated in heap flows by comparing flow fields and concentration profiles from experiments with DEM simulation results.^{61,62} For all simulations, particle-particle and particle-wall contacts use a friction coefficient of $\mu = 0.4$, a binary collision time of $t_c = 0.5$ ms, and a relatively low restitution coefficient of $e = 0.2$ to minimize the downstream flux of bouncing particles. Segregation in dense gravity-driven surface flows is largely insensitive to collision parameters, that is, μ , e , and t_c , with the exception of very low friction coefficients ($\mu < 0.2$) that are atypical of most industrially relevant materials.^{12,39}

In the case shown in Figure 1 for large heavy particles (red) with $\bar{c}_l = 0.2$ and small light particles (blue) with $R_d = 2$ and $R_p = 4$, the conditions are such that the small light particles (blue) rise to the surface of the flowing layer. As a result, the small light (blue) particles flow further down the slope to deposit in a nearly pure blue particle region at the downstream end of the heap, while large heavy particles (red) mixed with the small light particles (blue) deposit on the upstream portion of the heap.

The advantage of considering heap flows over other flow configurations (e.g., plane shear flows or chute flows) is that the local shear rate $\dot{\gamma}$ and the particle species concentration c_i vary throughout the length and depth of the flowing layer but remain constant at a particular location in the flow (when analyzed in a reference frame that rises with the heap surface at rise velocity, v_r). As a result, the time-averaged segregation flux $\phi_{\text{seg},i}$ ($\phi_{\text{seg},i} = w_i c_i$ where w_i is the species-specific velocity in the z-direction) for a wide range of flow conditions ($\dot{\gamma}$ and c_i) can be obtained at different locations in the flowing layer from just one simulation. However, the full range of local concentrations is not usually realized in a single simulation, especially for weakly segregating mixtures, so typically several simulations are conducted with different large particle feed concentrations, \bar{c}_l , to provide data covering the full range of possible local concentrations $0 \leq c_l \leq 1$.^{44,57}

Based on simulations like that shown in Figure 1, the segregation flux dependence on species concentration can be obtained for different values of R_d and R_p for SD-systems. To calculate the segregation flux $\phi_{\text{seg},i}$, the species-specific velocity in the z-direction, w_i , and species concentration, c_i , are first calculated from spatial and temporal averages of local simulation data sampled over the entire flowing layer. The spatial average is determined by a volume-weighted binning method⁵⁹ using right cuboid bins oriented with two faces parallel to the free surface, two faces perpendicular to the free surface and two faces parallel to the sidewalls. Each bin has a streamwise length of 1 cm (3.33 d_l) and a height (normal to the free surface) of 1 mm (0.33 d_l). The concentration and velocity for particles in each bin are sampled at 0.01 s intervals and averaged over 5 s after the system reaches steady state. The local segregation flux is then calculated as the product of the species concentration and the species-specific velocity relative to the local bulk velocity. Further details are provided elsewhere.⁵⁷ Figure 2 shows the mean scaled local segregation flux $\phi_{\text{seg},i}/\dot{\gamma}d_s$ versus local large particle concentration c_l from heap flow simulations for three (R_d , R_p) pairs including the example in Figure 1. For all three cases, the average segregation fluxes (data points) for the small light species (blue) and the large heavy species (red) are always equal and opposite at any particular local value of c_l , as expected for the range of size ratios studied.⁶³ The curves through the data in Figure 2 are best fits of the cubic segregation flux model,⁵⁷

$$\phi_{\text{seg},i} = d_s \dot{\gamma} c_l [A_i + B_i(1 - c_l)](1 - c_l), \quad (1)$$

where A_i and B_i are fitting parameters dependent on both R_d and R_p . For D-system segregation with $R_d = 1$ and $R_p = 4$ in Figure 2A, light particles rise and heavy particles sink except at the extremes of concentration, where the flux is zero because only one species is present.

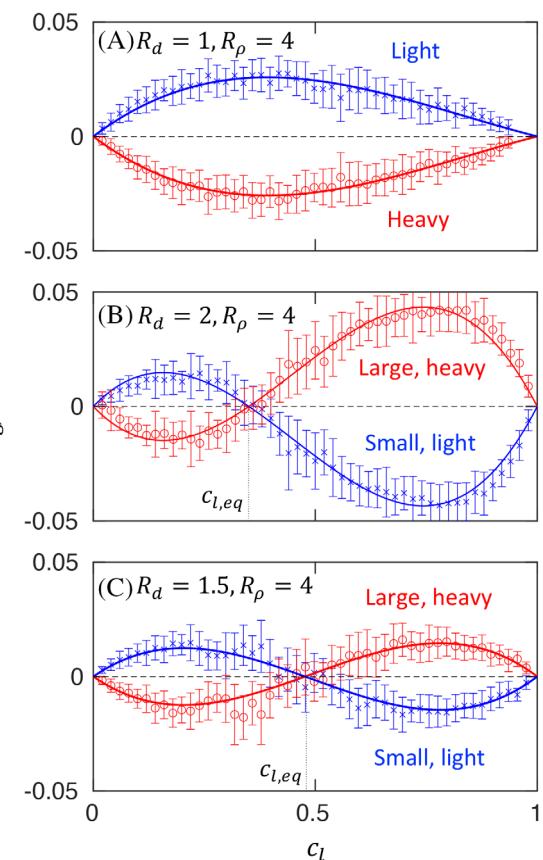


FIGURE 2 Non-dimensionalized segregation flux data, $\phi_{\text{seg},i}/\dot{\gamma}d_s$, for large (O) and small particles (x) averaged over 0.02 wide increments of c_l from heap flow simulations. Error bars represent the standard deviation for each averaging interval of c_l . Solid curves are fits of the segregation velocity model, Equation (1).⁵⁷

The direction of the segregation is reflected by the segregation flux curves that are either positive (upward segregation for light particles) or negative (downward segregation for heavy particles) across the entire bidisperse concentration range. Unlike the ‘unidirectional’ segregation case in Figure 2A, size and density differences compete with each other for $R_d = 2$ and $R_p = 4$ in Figure 2B and for $R_d = 1.5$ and $R_p = 4$ in Figure 2C, and the segregation direction is concentration dependent. For $R_d = 2$ and $R_p = 4$ small, light particles segregate upward for small values of c_l and downward for large values of c_l . The segregation flux direction reverses at $c_{l,\text{eq}} = 0.36$ where $\phi_{\text{seg},i} = 0$. For $R_d = 1.5$ and $R_p = 4$, a similar situation occurs except that $c_{l,\text{eq}}$ increases to 0.47. In addition, the overall concentration averaged segregation flux is smaller for the case in Figure 2C than the case in Figure 2B due to the near balance between the two segregation mechanisms across the full range of concentrations.

Repeating heap flow simulations like that shown in Figure 1 over many (R_d , R_p) combinations results in figures analogous to those in Figure 2 from which $c_{l,\text{eq}}$ can be obtained. Using this approach, the dependence of $c_{l,\text{eq}}$ on R_d and R_p was previously explored for $1 \leq R_d \leq 2$ and $1 \leq R_p \leq 4$.⁵⁷ Here, we extend $c_{l,\text{eq}}$ to $1 \leq R_d \leq 3$ and $1 \leq R_p \leq 5$.

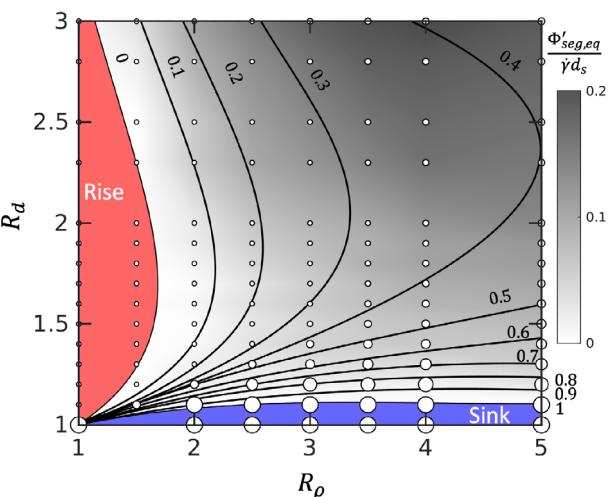


FIGURE 3 Local equilibrium (no segregation) concentration of large particles $c_{l,eq}$ versus particle size and density ratios. Circles denote combinations of R_d and R_p for which heap flow simulations have been performed, and circle diameter is proportional to $c_{l,eq}$ in the range 0 to 1. Iso-concentration curves for $c_{l,eq}$ are interpolated between data points. Particles remain mixed along the curve for $c_l = c_{l,eq}$ for the corresponding R_d and R_p . Segregation is uni-directional in the filled regions for $c_{l,eq} = 0$ (large particles rise, red) and $c_{l,eq} = 1$ (large particles sink, blue). Grayscale is interpolated from the dimensionless large particle segregation flux gradient at the equilibrium concentration, $\phi'_{seg,eq}/\gamma d_s$.

The curves in Figure 3 are cubic fits to the interpolated data for $c_{l,eq}$ forced to pass through $(R_d = 1, R_p = 1)$, where the segregation is necessarily zero. Particles remain mixed along the curve for $c_l = c_{l,eq}$ for the corresponding R_d and R_p . For R_d and R_p combinations to the left of or above the iso-concentration curve for $c_l = c_{l,eq}$, large particles rise, while large particles sink for combinations to the right of or below the iso-concentration curve for $c_l = c_{l,eq}$. Along each axis are regions (colored red or blue) where the segregation is uni-directional, corresponding to cases like that shown in Figure 2A. Size segregation dominates regardless of density in the red region adjacent to the vertical axis where large particles rise even if they are slightly heavier than the small particles, and density segregation dominates in the narrow blue band adjacent to the horizontal axis where large particles sink because they are heavier than the small particles. Finally, for completeness, consider mixtures of large light and small heavy particles (i.e., $1 \leq R_d \leq 3$ and $0 < R_p < 1$, noting that $R_d = d_l/d_s \geq 1$ by definition). In this case, large particles always rise, which would correspond to an all red region to the left of the vertical axis for $0 < R_p < 1$, which is not included in Figure 3.

As previously described,^{57,58} Figure 3 allows an approach for the intentional design of bidisperse particle mixtures that avoid segregation. For instance, in many industrial situations the material of each particle species is specified, thereby fixing the density ratio, and the concentration of each species is fixed based on the product requirements. However, the species sizes can be altered by one of several standard processes such as agglomeration or grinding. Thus, it is possible to specify a size ratio for a given density ratio and relative concentration of species that minimizes segregation. For example,

suppose that a 20:80 species concentration mixture is required and the density ratio of the two species is $R_p = 2.5$. Starting at $R_p = 2.5$ and reading upward to the $c_{l,eq} = 0.2$ contour in Figure 3 indicates that a size ratio of $R_d = 1.6$ should result in a non-segregating mixture. In the remainder of this article, we explore this approach to designing minimally segregating bidisperse particle systems.

The tendency to mix rather than segregate is unaffected by small fluctuations in local concentration around $c_{l,eq}$, mainly due to granular diffusion, as discussed later in the article. Nevertheless, it is helpful to consider how quickly the segregation flux deviates from the non-segregating condition near $c_{l,eq}$. This sensitivity toward segregation can be measured in terms of the derivative of the large-particle segregation flux at the equilibrium concentration, $\phi'_{seg,eq}/\gamma d_s$, where $\phi'_{seg,eq} = d\phi_{seg}/dc_l$ is the slope of the flux curves in Figure 2 at $c_{l,eq}$, shown as gray-scale in Figure 3. Increasing both R_d and R_p (darker areas of Figure 3) increases the sensitivity toward segregation associated with small deviations of c_l from $c_{l,eq}$. Nevertheless, we will show later in this article that the tendency toward mixing near $c_{l,eq}$ is robust.

The equilibrium concentrations corresponding to no segregation in Figure 3 are determined for each combination of R_d and R_p from the local segregation flux, which is based on the local segregation velocity and local species concentration, over a wide range of local shear rates $\dot{\gamma}$ and local particle concentrations c_l that occur in a bounded heap flow. Other flow parameters such as shear rate gradient⁶⁴ and pressure⁶⁵ also affect the segregation flux and are partially responsible for the magnitude of the error bars in Figure 2 (along with randomness due to individual particle collisions). However, on average these effects are implicitly included by considering the segregation flux for conditions throughout the thin flowing surface layer. Since the flow kinematics of a thin surface layer of particles flowing down a slope is similar regardless of the overall flow configuration,⁴⁸ Figure 3 should be generally applicable to thin gravity-driven surface flows in other geometries, as we show later in this article for flow in a rotating tumbler.

Before continuing with the approach of designing minimally segregating particle mixtures by using the appropriate combination of R_d , R_p , and $c_{l,eq}$, we consider briefly an alternative approach that ignores $c_{l,eq}$ and is therefore more general. Returning to Figure 2, it is evident that the magnitude of the segregation flux depends on size and density ratios in addition to the mixture concentration. For instance, the segregation flux across all values of c_l for $R_d = 1.5$ and $R_p = 4$ in Figure 2C is generally less than that for $R_d = 2$ and $R_p = 4$ in Figure 2B. Hence, a system with $R_d = 1.5$ and $R_p = 4$ would tend to segregate less, regardless of mixture concentration, than a system with $R_d = 2$ and $R_p = 4$. Using data across the full range of R_d and R_p considered here, we quantify this average tendency toward segregation at a particular combination of R_d and R_p as the square of the dimensionless segregation flux averaged over all possible concentrations,

$$\overline{\Phi^2} = \int_0^1 \left(\phi_{seg,l} / \gamma d_s \right)^2 dc_l, \quad (2)$$

which is plotted in Figure 4 as a function of size and density ratios. The grayscale reflects the sensitivity of the system to segregation

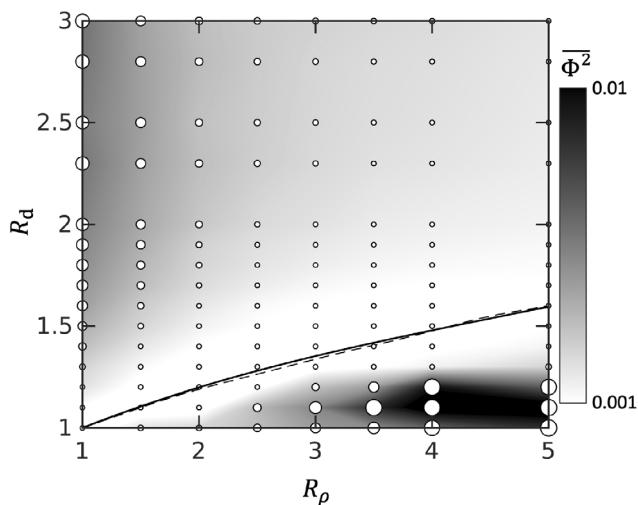


FIGURE 4 Concentration averaged dimensionless segregation flux squared, $\bar{\Phi}^2$, versus particle size and density ratios. Circular symbols indicate (R_d, R_p) pairs for which simulations are performed; circle diameter is proportional to $\bar{\Phi}^2$ in the range 0.001 to 0.01. The minimum value of $\bar{\Phi}^2$ is well-approximated by $R_d = 0.163(R_p - 1) + 1$ (dashed curve), which nearly matches the solid curve for equilibrium concentration $c_{l,\text{eq}} = 0.5$ (reproduced from Figure 3).

across all possible concentrations c_l . Combinations of R_d and R_p in brighter areas of Figure 4 tend to segregate less across all mixture concentrations, while combinations in darker areas tend to segregate more. The least segregation occurs for combinations of R_d and R_p along the dashed curve, which is calculated as $\partial\bar{\Phi}^2/\partial R_d = 0$.^{*} Interestingly, this curve for the least segregation corresponds closely to the $c_{l,\text{eq}} = 0.5$ curve from Figure 3 which is reproduced in Figure 4 as the solid curve. This is because the lowest flux magnitudes tend to occur when $c_{l,\text{eq}}$ is around 0.5 as is evident in comparing Figures 2B, C.

Figure 4 offers an alternative approach to the one based on Figure 3 where \bar{c}_l is set to $c_{l,\text{eq}}$ to prevent segregation. This alternative is simply to ignore the mixture concentration and operate in the light colored region of Figure 4 along the dashed curve for the minimum value of $\bar{\Phi}^2$. In this case, for a given density ratio for the two particle species, the size ratio would be adjusted to fall on the minimum segregation curve in Figure 4. For instance, for the preceding example with a density ratio $R_p = 2.5$, the size ratio should be set to $R_d \approx 1.3$ to minimize segregation across all possible concentrations. While this approach will result in more segregation when $c_l \neq c_{l,\text{eq}}$, it may be easier to implement in some industrial situations. However, since the approach in Figure 3 using $c_{l,\text{eq}}$ is more precise in avoiding segregation than the alternative approach based on Figure 4, we focus on the former in the remainder of this article.

3 | VALIDATING PREDICTIONS FOR NON-SEGREGATING MIXTURES

The combinations of size ratio, density ratio, and large particle concentration leading to no segregation indicated in Figure 3 are based

on the local segregation flux, and the equilibrium concentration $c_{l,\text{eq}}$ is inherently a local variable at each point within the thin flowing layer typical of many gravity-driven granular flows. However, the requirement that $c_l = c_{l,\text{eq}}$ everywhere in the flow to prevent segregation is unlikely to be realized exactly in real systems due to the stochastic nature of granular flows which drives local variations in the mixture concentration, as well as to deviations in the actual velocity profile from that of the bounded heap for which the $c_{l,\text{eq}}$ curves in Figure 3 were obtained. Furthermore, the averaging approach used to obtain Figure 3 and Equation (1) ignores the potential influence of absolute pressure⁶⁵ and pressure variation in combination with the effects of the shear rate gradient⁶⁴ on segregation. Hence, the question is if a uniformly mixed system with a global concentration of $\bar{c}_l = c_{l,\text{eq}}$ remains mixed or locally segregates across different flow geometries and initial conditions. In this section, we validate and demonstrate the potential for designing minimally segregating granular mixtures using the approach of specifying R_d , R_p , and $c_{l,\text{eq}}$ (Figure 3) by quantifying and comparing the segregation of particles near the equilibrium conditions shown in Figure 3 predicted by Equation (1) in bounded heaps and rotating tumblers. The goal is to determine how the *local* propensity for mixing or segregation shown in Figure 3 affects the *global* segregation across the entire flow domain.

3.1 | Bounded heap flow

We consider first the global segregation of particle mixtures prepared at the equilibrium concentration (i.e., $\bar{c}_l = c_{l,\text{eq}}$) and fed onto a bounded heap. Figure 5 shows the segregation patterns resulting from a feed concentration of $\bar{c}_l = 0.1$, $R_d = 1.5$, and three different values of R_p . For $R_p = 1$ large red particles segregate upward to the free surface of the flowing layer and deposit downstream near the endwall, indicating that size-based percolation dominates. Particles remain relatively mixed over most of the heap for $R_p = 2$, as the two segregation mechanisms nearly balance. Segregation reverses for $R_p = 3$ with small light particles (blue) depositing at the downstream end of the heap as buoyancy dominates over percolation.

To quantify these results, Figure 6 plots streamwise c_l profiles in the deposited heap for the three cases in Figure 5. For $R_p = 1$ (red curve), the concentration of large particles is highest near $x/L \approx 1$ as they deposit on the downstream portion of the heap along with a few small particles, consistent with Figure 5A. The situation reverses for $R_p = 3$ (blue curve), where a higher fraction of large particles deposit on the upstream portion of the heap, and pure small particles ($c_l = 0$) deposit on the downstream portion, as shown in Figure 5C. For the intermediate density ratio where size and density effects nearly balance, large particles deposit almost uniformly on the heap and the streamwise variation in c_l (green curve) is the smallest of the three cases.

The degree and direction of segregation along the length of the heap is calculated using the signed and scaled standard deviation of the large particle concentration,

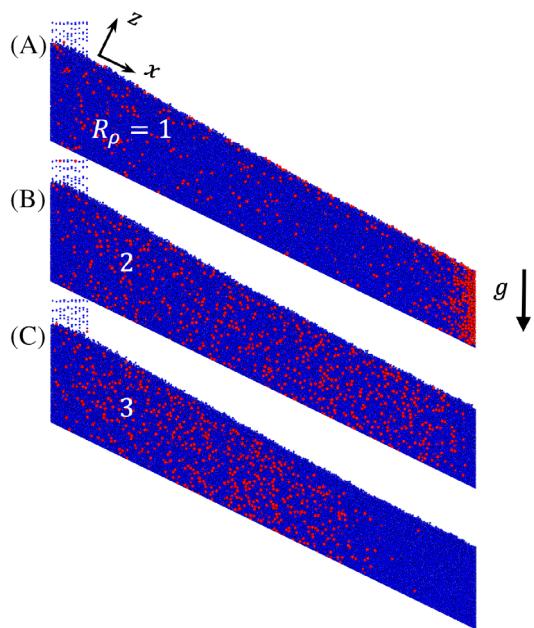


FIGURE 5 Heap flow segregation examples from DEM simulations for feed concentration of large red particles $\bar{c}_l = 0.1$ and size ratio $R_d = 1.5$ showing reversal in large particle segregation direction for different density ratios R_p . Other conditions are as in Figure 1.

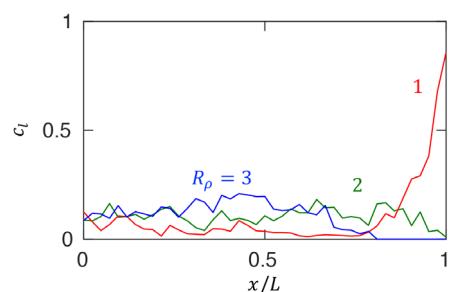


FIGURE 6 Streamwise concentration profiles of large particles, c_l , deposited on the heap for the three different R_p values in Figure 4 with $\bar{c}_l = 0.1$ and $R_d = 1.5$.

$$\hat{\sigma}_l = \text{sgn}(c_{l,N} - \langle c_l \rangle) \frac{\sqrt{\sum_{k=1}^N (c_{l,k} - \langle c_l \rangle)^2 / N}}{\langle c_l \rangle (1 - \langle c_l \rangle)}, \quad (3)$$

where $N = 50$ is the number of uniform width bins for calculating c_l at different streamwise positions, $c_{l,k}$ is the local depth-averaged volume concentration for particles deposited on the heap below the flowing layer in bin k , and $\langle c_l \rangle$ is the average value of c_l for $0 \leq x \leq L$. Note that $\langle c_l \rangle$ differs slightly from \bar{c}_l because it excludes the portion of the heap below the feed zone (see Figure 1). The value $\hat{\sigma}_l$ is essentially the standard deviation of c_l , but with two additional multiplicative terms. The sign function, $\text{sgn}(c_{l,N} - \langle c_l \rangle)$, indicates the large particle segregation direction: a value of -1 indicates downward segregation resulting in $c_l > \langle c_l \rangle$ on the upstream portion of the heap and $c_{l,N} = c_l(x=L) < \langle c_l \rangle$ at

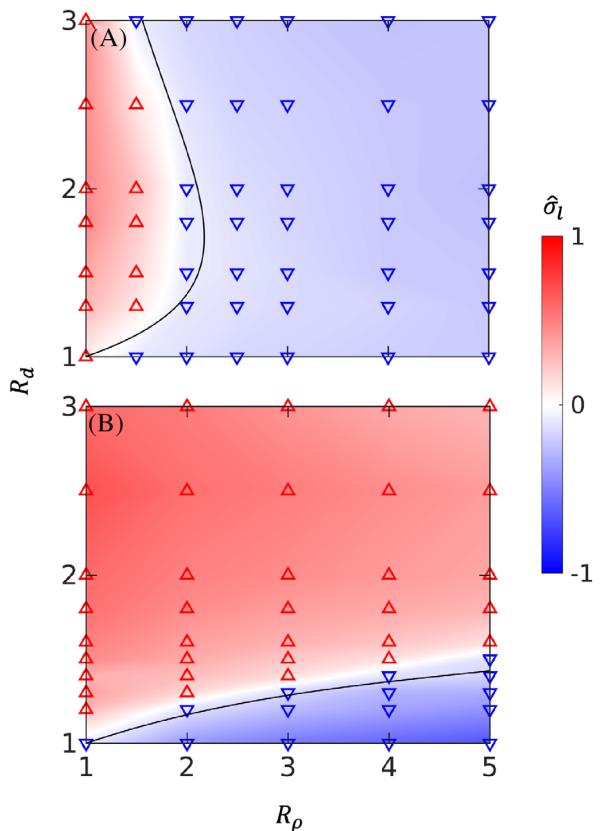


FIGURE 7 Scaled deviation of large particle concentration, $\hat{\sigma}_l$, (color contours) for feed concentration \bar{c}_l of (A) 0.1 and (B) 0.6. Black curves show predicted non-segregating R_d and R_p values for those feed concentrations; white regions correspond to particles remaining mixed in heap simulations. Upward red triangles indicate large particles rise; downward blue triangles indicate large particles sink.

the downstream end; a value of $+1$ corresponds to upward segregation resulting in $c_l < \langle c_l \rangle$ on the upstream portion of the heap and $c_l > \langle c_l \rangle$ at the downstream end. The $\langle c_l \rangle(1 - \langle c_l \rangle)$ term in the denominator normalizes the measured standard deviation by that for perfect segregation. A fully mixed, non-segregating case with large particles depositing uniformly along the surface of the heap has $\hat{\sigma}_l = 0$, while for complete segregation, $\hat{\sigma}_l = -1$ for sinking large particles and $\hat{\sigma}_l = 1$ for rising large particles.

Figure 7 demonstrates how $\hat{\sigma}_l$ captures the rise-sink transition in heap flow simulations (data points) for 49 (R_d , R_p) combinations with $\bar{c}_l = 0.1$ (Figure 7A) and 50 (R_d , R_p) combinations with $\bar{c}_l = 0.6$ (Figure 7B). Red and blue shading based on interpolating these data corresponds to the degree to which $\hat{\sigma}_l$ deviates from the perfectly mixed state value of $\hat{\sigma}_l = 0$ with white corresponding to particles remaining mixed. The predicted equilibrium curve for the local value of c_l from Figure 3, shown by the black curve, corresponds closely to the white region, indicating that particles remain mixed at that feed concentration. In other words, the $c_{l,\text{eq}}$ curve in the R_d , R_p -plane along which particles are predicted to locally remain mixed from Figure 3 (black curve) corresponds closely to (R_d, R_p) pairs along which particles remain globally mixed for that feed concentration under heap flow

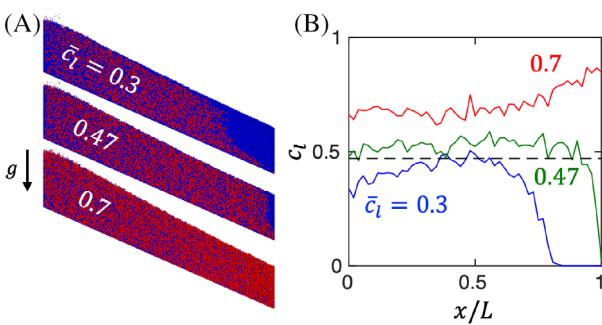


FIGURE 8 (A) DEM simulation images and (B) large particle concentration versus streamwise position for heap flow segregation with $R_d = 1.5$, $R_p = 4$, and different values of large particle feed concentration \bar{c}_l . Large heavy particles (red) sink while small light particles (blue) rise for $\bar{c}_l = 0.3 < c_{l,\text{eq}}$, as buoyancy overcomes percolation. In contrast, for $\bar{c}_l = 0.7 > c_{l,\text{eq}}$ segregation is reversed and percolation dominates over buoyancy. Particles remain relatively mixed for $\bar{c}_l = c_{l,\text{eq}} = 0.47$ (horizontal dashed line).

(white region where $\hat{\sigma}_l = 0$). The small deviation of the black curve from the white region could result from many factors including the resolution of data points in the (R_d, R_p) space and fluctuations in the local concentration similar to those observed for the weakly segregating case (green curve) shown in Figure 6.

Particle mixtures with (R_d, R_p) off the equilibrium curve in Figure 3 segregate, as expected. Conditions below and to the right of the $c_{l,\text{eq}} = \bar{c}_l$ equilibrium curve in the R_d, R_p -plane correspond to smaller size ratios and larger density ratios, indicating that large particles sink for these conditions; conditions above and to the left of the equilibrium curve correspond to larger size ratios and smaller density ratios, indicating that large particles rise.

An equivalent alternate description is that if the feed concentration \bar{c}_l for a particular (R_d, R_p) pair is above the equilibrium value in Figure 3 ($\bar{c}_l > c_{l,\text{eq}}$), then the feed concentration is too high to maintain equilibrium, and the large particles will rise resulting in $\hat{\sigma}_l > 0$. If the feed concentration \bar{c}_l for the (R_d, R_p) pair is below the equilibrium value in Figure 3 ($\bar{c}_l < c_{l,\text{eq}}$), then the feed concentration is too low to maintain equilibrium, and large particles will sink resulting in $\hat{\sigma}_l < 0$. This is evident in the segregation examples with $R_d = 1.5$, $R_p = 4$, and $c_{l,\text{eq}} = 0.47$ shown in Figure 8. Large heavy particles (red) sink for $\bar{c}_l = 0.3 < c_{l,\text{eq}}$, depositing on the upstream portion of the heap, whereas for $\bar{c}_l = 0.7 > c_{l,\text{eq}}$ large heavy particles (red) rise, depositing at a higher concentration on the downstream portion of the heap. For $\bar{c}_l = c_{l,\text{eq}}$ particles remain well-mixed except near the downstream bounding endwall where c_l decreases somewhat, potentially due to a pressure dependence of the equilibrium concentration that is unaccounted for in Equation (1) and the predictions of Figure 3, as discussed shortly with regard to the rotating tumbler results.

The implication of these results for heap flow is that the local propensity for mixing or segregation as predicted in Figure 3 also reflects the global propensity for mixing or segregation, at least to a first order. The consequence is that if granular material with $\bar{c}_l = c_{l,\text{eq}}$ is mixed at the feed, it remains mixed as it deposits on the heap. Note

that small fluctuations in \bar{c}_l have little influence on the segregation results, likely due to granular diffusion. In addition, the facts that particles remain relatively mixed for (R_d, R_p) pairs near the equilibrium curve as shown in Figure 7, and that $\hat{\sigma}_l$ varies smoothly across the (R_d, R_p) space in Figure 7 indicate that slight deviation of \bar{c}_l from $c_{l,\text{eq}}$ or small fluctuations in c_l do not affect the overall propensity for particles to remain mixed in heap flows for appropriate values of R_d and R_p .

3.2 | Rotating tumbler flow

In the previous section, we demonstrated the application of our approach for designing minimally segregating mixtures to heap flows. Since the data for the local equilibrium concentrations in Figure 3 upon which the approach is based come from segregation in heap flows, its effectiveness is, perhaps, not surprising. The question now is if the equilibrium concentrations obtained from heap flows can be used to identify minimally segregating mixtures in a gravity-driven flow where the velocity field and boundary conditions differ. Here, we apply the approach to rotating tumbler flow.^{25,27,34,66}

Tumblers are used to coat, crush, and mix particles. Unlike bounded heaps where segregation takes place during the short period of time before the particles deposit onto the fixed bed, segregation in rotating tumblers is an ongoing process as particles repeatedly flow down the slope, enter solid-body rotation in the downstream half of the flowing layer, and then re-enter the upper half of the flowing layer after solid body rotation. Segregation occurs only in the shear of the flowing surface layer; below the flowing surface layer, particles are in near solid-body rotation with the tumbler (i.e., the quasi-static or fixed bed region) where the segregation is negligible. There is an initial transient as initially mixed particles segregate in the flowing layer and then deposit in a segregated pattern in the first half-rotation, followed by enhanced segregation as the particles repeatedly flow down the surface. A steady segregated pattern is established after only a few tumbler rotations^{43,67,68} in which segregated particles in the fixed bed enter the flowing layer, flow down the surface, and maintain their segregated pattern upon re-entering the fixed bed.

Here we test the same bidisperse particle mixtures as used in the heap flow simulations described in the previous section, noting that \bar{c}_l here refers to the overall large particle concentration in the tumbler rather than the feed concentration, as was the case for the heap. The half-full tumbler is $r_0 = 7.5$ cm in radius ($r_0/d_l = 25$) and 1.5 cm in axial extent with periodic boundaries in the axial direction to avoid endwall effects. The cylindrical tumbler wall is formed from 3 mm particles overlapping by 1.5 mm to reduce slip between the particle bed and the wall. The tumbler rotates at $\Omega = 10$ rev/min. The Froude number is $F_r = \Omega^2 r_0/g = 0.0084$, which corresponds to the flat surface continuous flow regime.⁶⁹

Figure 9 shows examples of the visually steady segregation patterns, which occur after 5–7 tumbler rotations ($t\Omega = 5$ –7) and are more clearly quantified shortly, for the same particle mixtures used in the bounded heap examples in Figure 8 starting from a well-mixed initial condition. The equilibrium concentration of large particles for

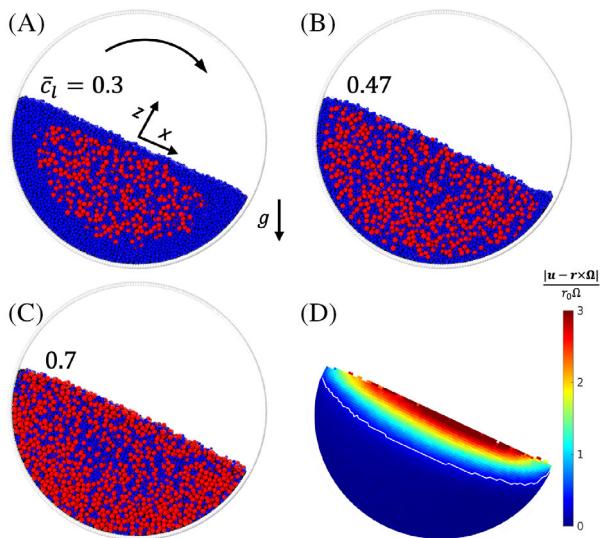


FIGURE 9 Rotating tumbler DEM simulation setup and segregation examples in steady state for $R_d = 1.5$, $R_p = 4$, and global large particle concentration \bar{c}_l of (A) 0.3, (B) 0.47, and (C) 0.7 after seven rotations, or 42 s, starting from a fully mixed initial condition. Large particles are red ($d_l = 3$ mm, $\rho_l = 4$ g/cm³) and small particles are blue ($d_s = 2$ mm, $\rho_s = 1$ g/cm³). (D) Non-dimensional speed in rotating reference frame, $|u - \mathbf{r} \times \boldsymbol{\Omega}| / r_0 \Omega$, for all particles in case (B) averaged over 10 s. Solid curve shows where the streamwise velocity in the lab frame is zero, that is, $u_x(x, z) = 0$, which delineates the flowing layer from the quasi-static fixed bed.

$R_d = 1.5$ and $R_p = 4$ is $c_{l,eq} = 0.47$ according to Figures 2C, 3. Similar to large heavy particles depositing in the upstream portion of the heap for $\bar{c}_l < c_{l,eq}$, large heavy particles sink in the flowing layer and segregate to the central portion of the tumbler bed for $\bar{c}_l = 0.3$ in Figure 9A. Figure 9B shows that for $\bar{c}_l = c_{l,eq} = 0.47$ particles remain relatively mixed, as the two segregation mechanisms nearly balance. Segregation is reversed at $\bar{c}_l = 0.7 > c_{l,eq}$ in Figure 9C as percolation dominates over buoyancy, that is, large heavy particles segregate to the tumbler periphery analogous to their deposition in the downstream portion of the heap.

For reference, Figure 9D shows the non-dimensional particle speed in the rotating reference frame of the tumbler for $\bar{c}_l = 0.47$, that is, $|u - \mathbf{r} \times \boldsymbol{\Omega}| / r_0 \Omega$ where u is measured in the lab frame. The speed of the particles is largest at the midpoint of the free surface of the flowing layer and decreases with depth. Segregation takes place in the flowing surface layer due to shear; below this layer particles are in near solid-body rotation with the tumbler and do not segregate. The elapsed time between particle passes through the flowing layer (i.e., the solid body rotation residence time) is slightly less than half of the tumbler rotation period, and the time a particle typically spends in the flowing layer is one order of magnitude less than the tumbler rotation period.⁷⁰ Similar velocity fields are observed for the other two values of \bar{c}_l .

The core of the tumbler bed is typically thought of as the inner bed region where small (S-system) or heavy (D-system) particles segregate, and which is surrounded by the other particle species (large or

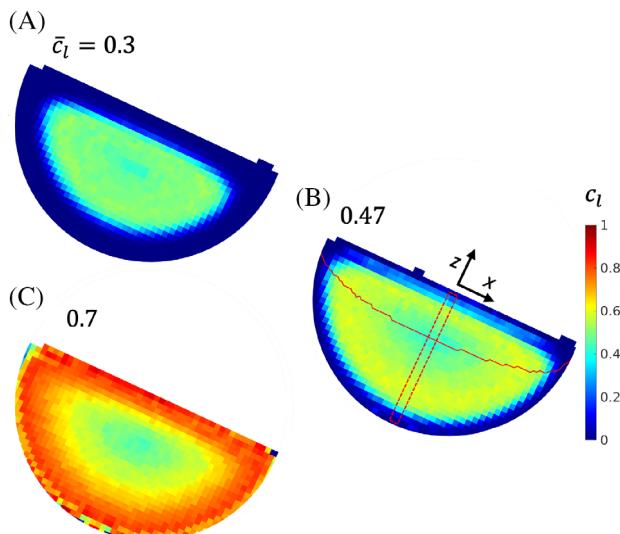


FIGURE 10 Steady-state large-particle concentration c_l averaged over 10 s after seven rotations (42 s) in the particle-filled portion of the tumbler rotating at 10 rpm for $R_d = 1.5$, $R_p = 4$, and \bar{c}_l equal to (A) 0.3, (B) 0.47 and (C) 0.7. Solid curve in (B) shows the bottom of the flowing layer where the streamwise velocity in the lab frame is zero, that is, $u_x(x, z) = 0$. Rectangular box in (B) is the region used for c_l profiles.

light particles, respectively). For SD-systems, where segregation is concentration dependent, we consider the core to be the region of the inner bed where the concentration is nearly constant but typically not fully segregated as in S- or D-system. The core includes particles that are both in the flowing layer and in the fixed bed, that is, in solid body rotation. This is evident comparing the lower bound of the flowing layer (white contour in Figure 9D) with the extent of the large particle enriched core in Figure 9A. The core spans the lower portion of the flowing layer and the inner portion of the solid body rotation region. This is most easily understood in terms of steady-state pathlines, which are, by definition, circular arcs in the portion of the bed in solid body rotation and generally in the streamwise direction, though slightly curved, in the flowing layer.^{8,27} For steady-state segregation, there is a one-to-one correspondence between the particle species distribution in the fixed bed and that in the flowing layer. Hence, the core extends into both the flowing layer and the fixed bed, with the core center (rotation center of the particles) at the boundary between them along a radial line at $x = 0$. Here, we define the approximate center of the tumbler core as the point where the velocity field is zero (white contour in Figure 9D) at the midpoint of the flowing layer ($x = 0$). For all cases studied here, this location is well approximated by a constant value of $z/r_0 = -0.3$.

The local concentration of large particles in the three rotating tumbler cases in Figure 9 can be calculated from spatial and temporal averages of simulation data after the system reaches steady-state. Figure 10 shows the steady state concentration of large particles. In the core of the tumbler bed, away from the flowing surface layer and cylindrical tumbler wall, c_l is nearly the same for all three cases, while

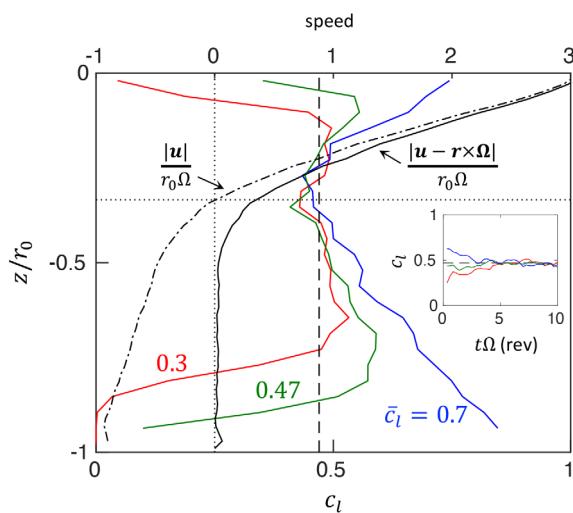


FIGURE 11 Depth profile of scaled particle speed in the lab frame (dash-dot black curve) and in the rotating tumbler frame (solid black curve) along a radius normal to the free surface and averaged across the region indicated by the dashed rectangle in Figure 10B. Dotted horizontal line at $z/r_0 = -0.3$ indicates the approximate bottom of the flowing layer where $u_x(x = 0, z = -\delta) \approx 0$ and which also corresponds to the core center. Colored curves are radial profiles of average large particle concentration normal to the free surface, as indicated by the dashed rectangle in Figure 10B, in steady state for three different \bar{c}_l values. At the core's approximate center ($z/r_0 = -0.3$, dotted horizontal line), $c_l \approx c_{l,\text{eq}}$ (dashed vertical line) even when $\bar{c}_l \neq c_{l,\text{eq}}$. Inset: c_l at the core center ($z/r_0 = -0.3$) tends toward the equilibrium concentration, $c_{l,\text{eq}}$ with increasing number of rotations regardless of the overall concentration, \bar{c}_l .

the concentration surrounding the core and outward to the periphery differs significantly between the three cases. For $\bar{c}_l = 0.3 < c_{l,\text{eq}}$, c_l at the tumbler periphery is effectively zero, consistent with Figure 9A where small light particles (blue) segregate to the tumbler periphery. In contrast, for $\bar{c}_l = 0.7 > c_{l,\text{eq}}$ the segregation direction is reversed, and c_l increases approaching the tumbler periphery, consistent with an excess of large heavy particles (red) at the periphery in Figure 9C.

The relationship between the mixture concentration and the flowing layer depth is depicted on a more quantitative basis in Figure 11. This figure shows the depth profile of the large particle concentration, c_l , along a radius normal to the free surface and averaged across the narrow region within the rectangle in Figure 10B (indicated along the lower horizontal axis), as well as the scaled particle speed in the lab frame and in the rotating tumbler frame (indicated along the upper horizontal axis). Concentration data are excluded for locations where the volume fraction is below 0.3 [corresponding to near the free surface ($z/r_0 \approx 0$) and near the tumbler wall ($z/r_0 \approx -1$) due to the overlap of the rectangular averaging regions with the circular boundary of the tumbler].

The c_l concentration profiles for the three cases in Figure 10 have a concentration in the core, corresponding to $-0.5 \leq z/r_0 \leq -0.2$, that is approximately equal to the equilibrium concentration, $c_{l,\text{eq}}$, as shown in Figure 11. In particular, the profiles for all three cases overlap and have a minimum value of $c_l \approx c_{l,\text{eq}}$ (dashed vertical line) at $z/r_0 \approx -0.3$ (horizontal dotted line), the approximate radial position of the core center. Thus, the center of the core and the region around it is mixed at the equilibrium concentration, $c_{l,\text{eq}}$, while the particle concentration in the periphery adjusts to a concentration necessary to accommodate the remaining particles, whether they are small particles (Figure 10A) or large particles (Figure 10C).

Keeping in mind that the particles are initially fully mixed, the question arises about how long it takes for the core to reach the equilibrium concentration. To answer this, the inset in Figure 11 shows time series of local concentration at the core center (i.e., c_l at $z/r_0 = -0.3$). The concentration c_l ($z/r_0 = -0.3$) starts at approximately the initial concentration \bar{c}_l in each case, and the three curves converge to the equilibrium concentration within five rotations ($t\Omega = 5$, or 30 s), which is the point after which we consider the segregation to be steady-state. This is also about the time necessary for the global segregation pattern to visually reach steady state.

From the concentration profile for $\bar{c}_l = c_{l,\text{eq}} = 0.47$ in Figure 11, it is evident that the mixing is imperfect at the equilibrium condition. Nevertheless, c_l is within a relatively narrow range $0.44 \leq c_l \leq 0.58$ except near the free surface, $z/r_0 \lesssim -0.1$, and near the tumbler wall, $z/r_0 \gtrsim -0.9$, where it decreases. Note that to a first approximation, the concentration profile in the solid body rotation portion of the particles is simply a stretched reflection of the concentration profile in the flowing layer. This occurs because particles deposit on the steady-state pathlines described earlier. Hence, these decreased concentrations at the surface and the tumbler wall are essentially mirror images of one another. The reduced large particle concentration near the tumbler wall has been noted previously^{66,71} but not explained, to our knowledge. The deviation of c_l from $c_{l,\text{eq}} = 0.47$ at the top of the flowing layer and near the cylindrical wall may be due to the effect of depth-varying lithostatic pressure, which is unaccounted for in the equilibrium concentration results in Figure 3. An extended discussion of this deviation in c_l is included in Sec. 6.

Two significant results are evident in Figures 9–11. First, segregation in a rotating tumbler is significantly reduced for $\bar{c}_l = c_{l,\text{eq}}$, much like the earlier case of bounded heap flows; second, the particle concentration in the core of the rotating tumbler bed saturates at approximately the equilibrium concentration regardless of \bar{c}_l . Similar results are also observed for the same size and density ratios ($R_d = 1.5$, $R_p = 4$) but with smaller absolute particle diameters (i.e., both d_l and d_s are scaled by a factor of two such that $r_0/d_l = 50$), shown in Figure 12. Particles remain mixed at steady state throughout most of the particle bed, indicating that the relative size of the tumbler compared with the particles is not crucial to the particles remaining mixed at $c_l = c_{l,\text{eq}}$. What is more interesting in this example is that the particles remain reasonably well-mixed not only for $\bar{c}_l = c_{l,\text{eq}}$ in Figures 12C,D, but also for $\bar{c}_l = c_{l,\text{eq}} - 0.03$ in Figures 12A,B and $\bar{c}_l = c_{l,\text{eq}} + 0.03$ in Figures 12E,F. Radial concentration profiles for these three cases are shown in Figure 13. The nearly identical concentration profiles indicate that small deviations of \bar{c}_l from $c_{l,\text{eq}}$ have little influence on the overall mixing of the particles in rotating tumblers. Moreover, the concentration profile for $\bar{c}_l = 0.47$ in Figure 11 for $r_0/d_l = 25$ is nearly identical to those in Figure 13 for $r_0/d_l = 50$. This is

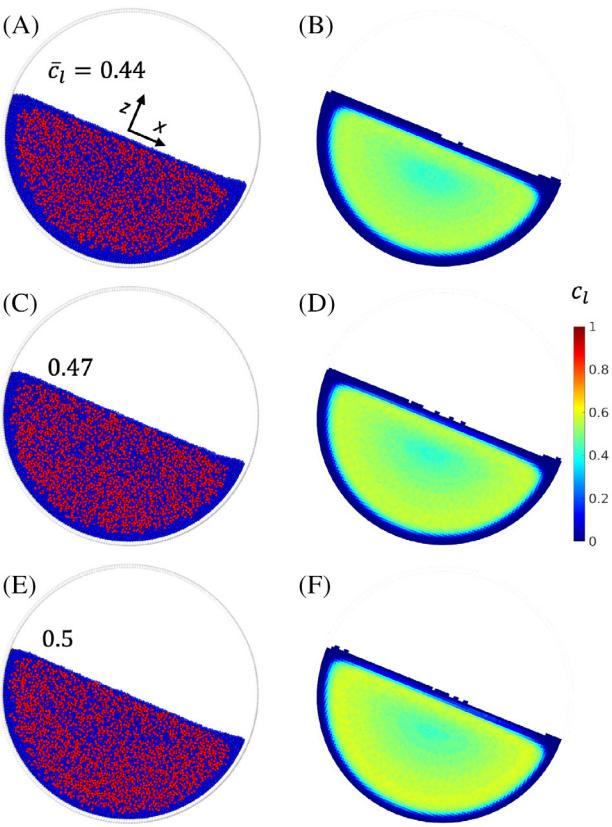


FIGURE 12 Segregation at steady state for $R_d = 1.5$, $R_p = 4$, and \bar{c}_l of (A) 0.44, (B) 0.47, and (C) 0.5 using finer particles with $r_0/d_l = 50$. Other parameters are the same as in Figure 7B. (B,D,F) Time-averaged large particle concentration, c_l .

because the particle concentration fields at steady state result from the balance between diffusion flux and segregation flux in the flowing layer. Since both fluxes are proportional to particle diameter,⁴² the effects of particle diameter cancel, and the concentration profiles are independent of r_0/d_l , resulting in the green curve in Figure 13 matching the green curve in Figure 11 except near the surface and the corresponding region on the tumbler wall. One difference between $r_0/d_l = 25$ and $r_0/d_l = 50$ is that the layers of nearly pure small particles (small c_l) at the flowing layer surface and near the tumbler wall are thinner for $r_0/d_l = 50$ suggesting that the phenomenon driving this effect is related to the size of the particles compared with that of the tumbler.

Returning now to the result in Figures 10 and 12 showing that $c_l \approx c_{l,\text{eq}}$ in the core of the tumbler bed regardless of the global concentration of large particles, we consider a different scenario in which the global concentration of large particles is held constant and R_p is varied. Figure 14 compares the radial concentration profiles and time evolution of c_l at the core center with $R_d = 1.5$ and $\bar{c}_l = 0.3$, for different values of R_p . In steady state c_l is close to $c_{l,\text{eq}}$ (dashed lines) at the core center ($z/r_0 \approx -0.3$) for the corresponding (R_d, R_p) for all four values of R_p , as shown in Figure 14A. This occurs regardless of whether c_l is increasing relative to \bar{c}_l to reach $c_{l,\text{eq}}$, as is the case for $R_p = 2$ in Figure 14B, or decreasing from \bar{c}_l to $c_{l,\text{eq}}$, as is the case for $R_p = 3$,

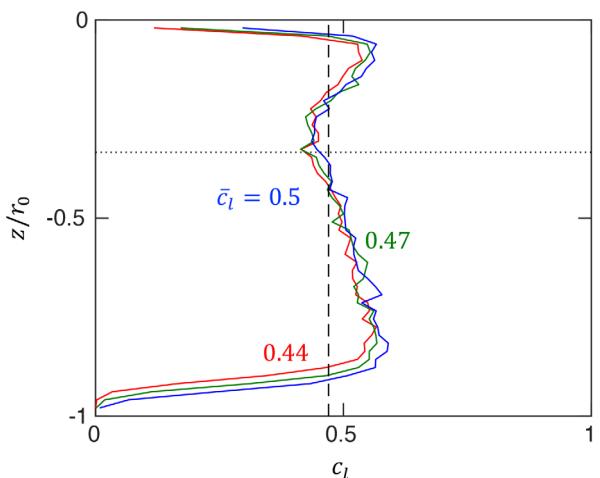


FIGURE 13 Large-particle concentration c_l along a radial slice normal to the free surface in steady state for smaller particles with $r_0/d_l = 50$ at three different values of \bar{c}_l . At the core center ($z/r_0 = -0.3$, dotted horizontal line), $c_l \approx c_{l,\text{eq}}$ (dashed vertical line).

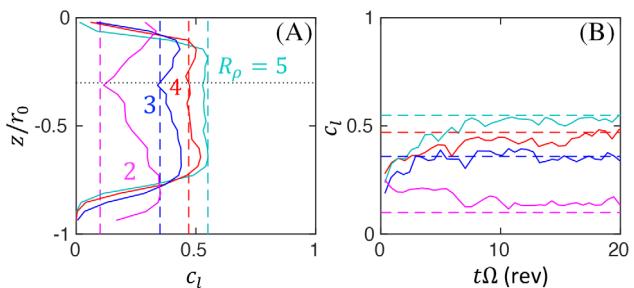


FIGURE 14 For differing R_p and fixed global large-particle concentration $\bar{c}_l = 0.3$, the local large-particle concentration, c_l , at the center of the tumbler core (i.e., $-0.32 < z/r_0 < -0.28$) is similar to the corresponding local equilibrium concentration $c_{l,\text{eq}} = 0.10, 0.35, 0.47$, and 0.55 (dashed lines) for $R_d = 1.5$. (A) c_l along a radial slice normal to the free surface in steady state (core center at $z/r_0 = -0.3$ corresponds to horizontal dotted line). (B) c_l at the core center ($z/r_0 = -0.3$) versus number of rotations (dashed lines correspond to $c_{l,\text{eq}}$ for each R_p).

4, 5. However, for cases with \bar{c}_l further away from $c_{l,\text{eq}}$ as determined by the value of R_p , it takes a longer time to reach steady-state.

The appearance of the equilibrium concentration in the core independent of \bar{c}_l seems to be a characteristic of steady-state segregation in tumbler flows. Apparently, as the particle mixture adjusts to its steady-state distribution, the core concentration relaxes to $c_{l,\text{eq}}$ while the “excess” particles (small particles for $c_l < c_{l,\text{eq}}$ and large particles for $c_l > c_{l,\text{eq}}$) are displaced to the tumbler periphery where their concentration is enhanced. An explanation is as follows. Starting from a fully mixed condition and during the initial transient, upward segregating particles rise in the flowing layer until the concentration in the lower portion of the flowing layer achieves the equilibrium concentration, $c_{l,\text{eq}}$. Once $c_{l,\text{eq}}$ is established locally at the bottom of the flowing layer, no further segregation occurs in this region. Since this layer with

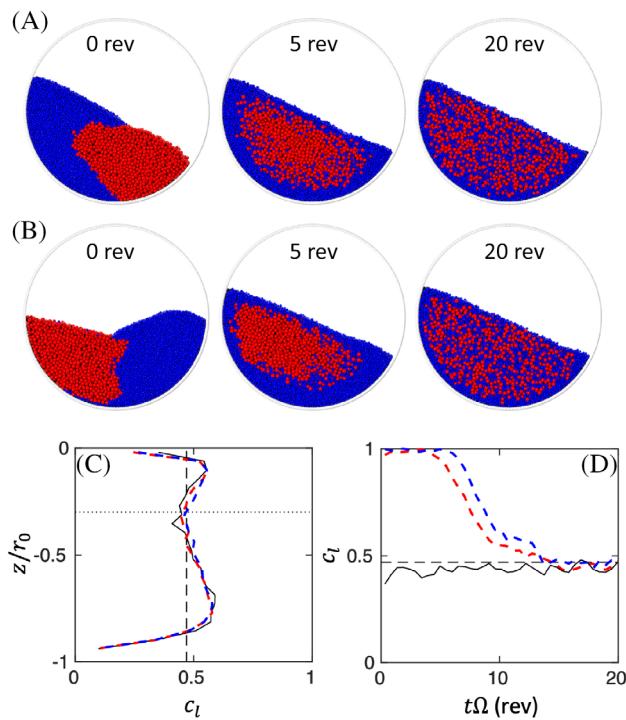


FIGURE 15 Initially fully segregated large heavy ($d_l = 3$ mm, $\rho_l = 4$ g/cm³, red) and small light ($d_s = 2$ mm, $\rho_s = 1$ g/cm³, blue) particles with $\bar{c}_l = c_{l,eq} = 0.47$ become well mixed at steady state. Left to right: initial segregated particle distribution, transient state after five rotations, and steady state at 20 rotations for (A) large heavy particles loaded into the tumbler before small light particles and (B) vice versa. (C) Average steady state ($t\Omega > 15$) radial large-particle concentration profile normal to free surface for well-mixed (black curve) and fully segregated [red and blue curves represent cases in (A) and (B), respectively] initial conditions. Vertical dashed line corresponds to $c_{l,eq} = 0.47$. (D) c_l at the core center ($z/r_0 = -0.3$) versus number of rotations. Horizontal dashed line corresponds to $c_{l,eq} = 0.47$.

concentration $c_{l,eq}$ is at the bottom of the flowing layer, it is deposited on the fixed bed at the core, while the particles that have segregated upward in the flowing layer are carried to the periphery of the tumbler to be deposited there. This process is reinforced over several rotations until steady-state is reached, which corresponds to a concentration near $c_{l,eq}$ in the core. We return to this phenomenon in the next section as a potential method to experimentally determine the equilibrium concentration for arbitrary binary particle mixtures.

One further consideration is that for the simulations described to this point, particles are initially well-mixed. However, in many practical industrial situations the two particle species are often initially segregated. Figure 15 shows the mixing of the same particle mixture in Figure 9B but starting from an initially fully segregated state shown in Figure 15A, accomplished by filling the left side of the tumbler first with large red particles followed by small blue particles as the tumbler starts to rotate. After five rotations (30 s) the system forms a large-particle enriched core with a small-particle enriched periphery. After 20 rotations (120 s), despite the initial complete segregation, particles

are well-mixed at steady state, just as occurs for the initially mixed case. Figure 15B shows the analogous situation except with the left side of the tumbler filled first with small blue particles followed by large red particles as the tumbler starts to rotate. Unlike the case in Figure 15A, the large red particles initially behave as a solid body due to slippage at the tumbler wall, while small particles slip under the large red particles at the bottom. Nevertheless, after about five rotations (30 s), the system forms a large-particle enriched core similar to that in Figure 15A, and after 20 rotations the particles are well-mixed. The steady state concentration profiles in Figure 15C are identical for the completely segregated and well-mixed initial conditions. However, the initially segregated cases take longer to reach equilibrium (15 rotations, $t\Omega \approx 15$, or 90 s), as shown in Figure 15D. For the well-mixed case with $\bar{c}_l = c_{l,eq} = 0.47$, the mixture almost immediately reaches $c_{l,eq}$, although for $\bar{c}_l \neq c_{l,eq}$ it takes about 5 rotations ($t\Omega \approx 5$, or 30 s), as shown in Figure 11B. The results in Figure 15 suggest that initial segregation conditions of the mixture or variations in the initial local concentration, with sufficient time, have little influence on the steady state segregation pattern, which always exhibits a core concentration close to $c_{l,eq}$.

4 | METHOD TO DETERMINE EQUILIBRIUM CONCENTRATION

The appearance of $c_{l,eq}$ in the core of the rotating tumbler bed suggests a simple methodology to experimentally determine $c_{l,eq}$ for an arbitrary binary particle mixture. The procedure is as follows. Fill a rotating tumbler to half full with an arbitrary concentration binary mixture and tumble until steady state segregation is reached. Remove a sample volume of particles from the core. The measured c_l for the sample should be close to $c_{l,eq}$. Then prepare a new mixture at the measured core concentration. It may be necessary to repeat these steps one or two times until no further variation in c_l at the core is observed.

The advantage of this approach is that the sizes and densities of the two particle species need not be known in advance, and, more significantly, no analog to Figure 3 is necessary (which requires many experiments or simulations). This makes the approach applicable for particles that may vary in properties other than just size or density, such as shape.¹⁶ Using this approach, a few simple experiments can determine the equilibrium concentration that minimizes segregation.

Figure 16 shows DEM simulations demonstrating this approach for determining the equilibrium concentration for example mixtures with $R_d = 2$ and $R_p = 3$, starting with either of two different initial global concentrations: $\bar{c}_l = 0.1$ (Figure 16A) or 0.5 (Figure 16B). Figures 16E,F show that c_l at the core center saturates at about 0.3 (dashed line) for both cases, indicating that $c_{l,eq} \approx 0.3$, which is close to $c_{l,eq} = 0.26$ for $R_d = 2$ and $R_p = 3$ determined from Figure 3. Noting that small deviations in the concentration from the actual value of $c_{l,eq}$ are inconsequential (see Figure 12), mixtures prepared at the measured $\bar{c}_l = 0.3$ are tested in both the rotating tumbler and the bounded heap. Figures 16C,D,G,H show that particles are relatively well-mixed

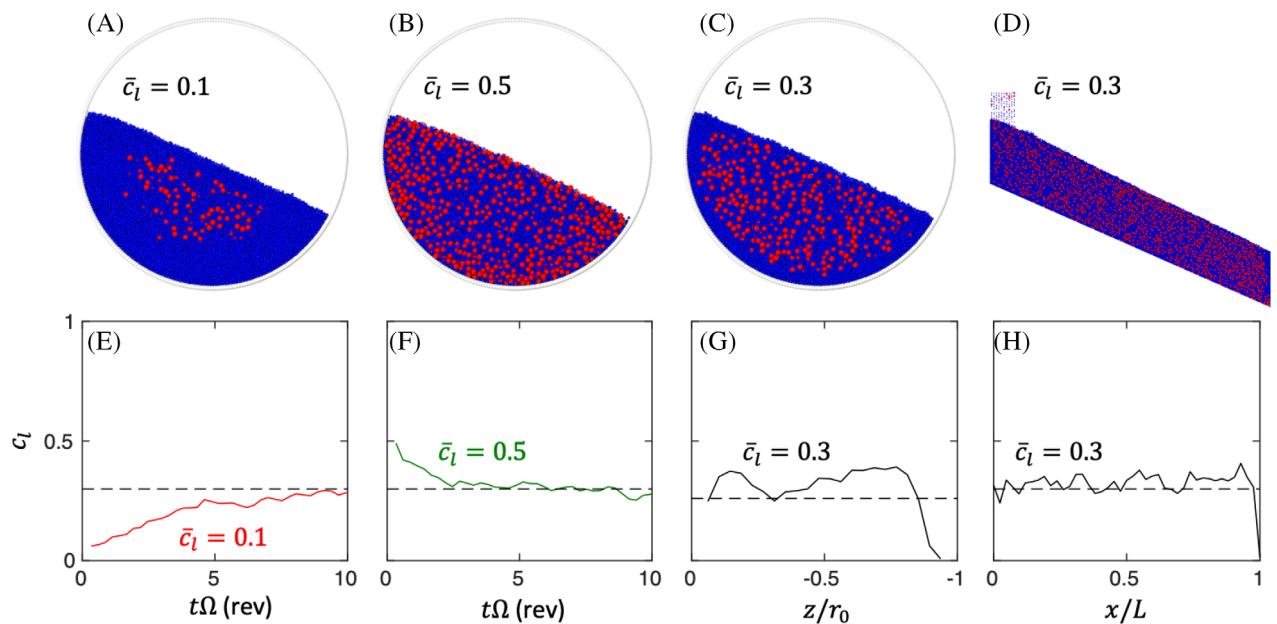


FIGURE 16 Simulations demonstrating practical determination of $c_{l,eq}$ in a rotating tumbler for different initial mixture concentrations of (A,C) $\bar{c}_l = 0.1$ and (B,F) $\bar{c}_l = 0.5$ by measuring the core concentration value in steady state. The measured value of $c_{l,eq} = 0.3$ [dashed line in (E-H)] is tested in (C,G) a tumbler flow and (D,H) a bounded heap flow, both of which show minimal segregation. Note that the concentration profile in (G) is rotated 90° from its orientation in previous figures so it can be easily compared with (H).

over the two domains. Thus, for these parameters, only one iteration of the rotating tumbler experiment is necessary to narrow in on the approximate value of $c_{l,eq}$, whether starting from too low of a concentration (Figures 16A,E) or too high of a concentration (Figures 16B,F). The second iteration, Figures 16C,G, confirms the value for $c_{l,eq}$. These results indicate that this simple practical approach for finding the equilibrium concentration for arbitrary bidisperse particle mixtures holds substantial promise.

5 | EXPERIMENT

Results to this point are based on DEM simulation, which for dense flows is usually quite accurate and relatively insensitive to particle and boundary interaction parameters. To validate the simulations, experiments are performed in a rotating tumbler ($\Omega = 4.08$ rev/min) with large steel and small clear glass spherical particles. The experimental set-up is similar to that in the simulation in Figure 9, except that the 15 cm diameter tumbler has two flat frictional endwalls made of clear acrylic and separated by 1.2 cm ($3.82d_l$) rather than the periodic endwall condition used in the simulations. One endwall is backlit to aid visualization of the particle distribution through the bulk. With the particle properties given in Table 1, the size ratio of large steel to small glass particles is $R_d = 2.04$ and the density ratio $R_\rho = 3.14$. The equilibrium concentration according to Figure 3 is $c_{steel,eq} \approx 0.27$.

Figure 17 shows images of the segregation pattern from experiment for different global concentrations of large steel particles \bar{c}_{steel} after 10 min of rotation, when the segregation pattern has reached a steady state, as well as DEM results for the same conditions as the

TABLE 1 Particle properties in experiments: diameter from caliper measurements of 300 particles; density from the mass of three sets of 100 particles and average particle diameter

Material	Color	Diameter (mm)	Density (g/cm ³)
Steel	Dark	3.14 ± 0.01	7.85 ± 0.02
Glass	Light	1.54 ± 0.08	2.50 ± 0.05

experiment. In this case, the DEM simulations include frictional endwalls with $\mu = 0.5$ separated by 1.2 cm, rather than periodic boundary conditions as in the other tumbler simulations in this article.

Although the visualization of the experimental results are less clear than the DEM simulation results, it is evident that the fundamental character of the particle distribution changes from mostly small glass particles surrounding a relatively mixed core at $\bar{c}_{steel} = 0.13$ to a generally mixed condition throughout the tumbler at $\bar{c}_{steel} = 0.27$ to mostly large steel particles surrounding a somewhat mixed core at $\bar{c}_{steel} = 0.40$ and $\bar{c}_{steel} = 0.50$. Focusing first on $\bar{c}_{steel} = 0.27$, which is at the predicted equilibrium concentration from Figure 3, a band of small particles appears at the tumbler periphery for both experiments and simulations, confirming that this phenomenon occurs not only in the simulations but also in the experiments. That said, the degree of mixing is substantially better for $\bar{c}_{steel} = 0.27$ than for the lower concentration ($\bar{c}_{steel} = 0.13$), where the band of small particles at the periphery is wide and quite pure, or the higher concentrations ($\bar{c}_{steel} = 0.40, 0.50$), where the mixture is dominated by large particles in a wide band near the periphery, although a narrow band of small particles immediately adjacent to the tumbler wall persists in the

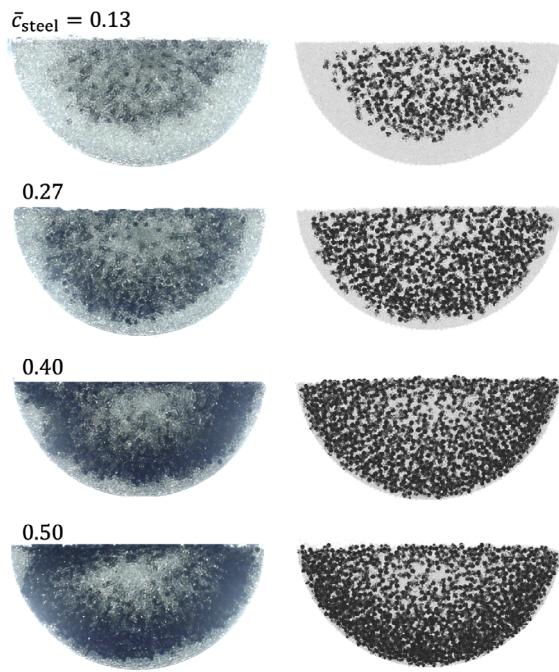


FIGURE 17 Segregation of large steel (dark) and small glass (bright) particles in a 15 cm rotating tumbler from experiments (left) and DEM simulations (right) with particle properties specified in Table 1. $R_d = 2.04$ and $R_p = 3.14$ for which $c_{\text{steel},\text{eq}} \approx 0.27$ according to Figure 3. Images from experiment and visualization of the simulations are obtained by backlighting one of the tumbler endwalls.

experiments. Thus, while more work is needed to resolve these discrepancies, the general result of a more mixed state appearing for the equilibrium concentration than other concentrations appears to be the case for these tumbler experiments.

6 | DISCUSSION

From both the DEM simulation results for rotating tumblers with $\bar{c}_l \approx c_{l,\text{eq}}$ (Figures 9–16) and the images from experiment ($\bar{c}_{\text{steel}} = 0.27$) in Figure 17B, it is evident that the large particle concentration decreases sharply near the tumbler wall such that the overall mixing is imperfect even at the equilibrium concentration (i.e., $\bar{c}_l = c_{l,\text{eq}}$). As mentioned earlier, the reduced large particle concentration near the tumbler wall has been noted previously^{66,71} but not explained. A possible explanation is that the equilibrium concentration in Figure 3 derived from depth-averaging heap flow segregation flux data as shown in Figure 2 may not be universal and could additionally vary locally with depth in the flowing layer. Specifically, our recent study on the forces on a single intruder particle in shear flow demonstrates that the equilibrium condition for single intruders depends on the local shear rate gradient and pressure.⁶⁴ Given that in free surface flows the streamwise particle velocity exponentially decreases with depth⁵⁹ and the local lithostatic pressure linearly increases with depth for a uniform density profile, it is reasonable to infer that the equilibrium concentration associated with non-mixing conditions may vary

locally with depth in the flowing layer. As a result, local segregation may occur despite the global concentration satisfying the equilibrium condition. Such local segregation appears to be negligible in heap flows, as the segregation is not fully developed before the particles deposit onto the fixed bed. However, in rotating tumblers, with particles repeatedly segregating in the flowing layer until reaching equilibrium, the variations in the local shear rate gradient and lithostatic pressure through the depth of the flowing layer may have sufficient time to result in greater segregation than in heap flows. This is evident in the concentration profile for $\bar{c}_l = c_{l,\text{eq}}$ in Figure 16G for a rotating tumbler deviating more from $c_{l,\text{eq}}$ than that for a bounded heap flow in Figure 16H.

A second reason for reduced large-particle concentration near the rotating tumbler wall may be the complex flow kinematics at the end of the downstream portion of the flowing layer where particles interact with the cylindrical tumbler wall before entering solid body rotation. The thickness of the small particle band near the tumbler wall decreases somewhat for simulations using smoother tumbler walls, suggesting that particle–wall interactions are important. The effect of the tumbler wall roughness is consistent with previous results indicating how the wall roughness can affect secondary flows in cylindrical tumblers⁷² and both secondary flows and segregation in spherical tumblers^{73,74} even far from the tumbler walls.

Despite the small particle band at the tumbler wall, evident visually in Figure 9B and in the concentration profile in Figure 11, it is clear that segregation is largely suppressed for $\bar{c}_l = c_{l,\text{eq}} = 0.47$ in this case. In contrast, for $\bar{c}_l = 0.3$, c_l decreases nearly to zero at the surface of the flowing layer and to zero (pure small particles) in a band at the tumbler periphery. For $\bar{c}_l = 0.7$, c_l increases at the surface of the flowing layer and at the periphery. Thus, despite the deviation of c_l from a uniform value for $\bar{c}_l = c_{l,\text{eq}}$ at the top of the flowing layer and near the cylindrical tumbler wall in this specific case as well as those in (Figures 12–16), most particles in the tumbler remain relatively well-mixed at the equilibrium concentration, whereas at other concentrations the particles tend to be substantially more segregated.

7 | CONCLUSION

For particle mixtures varying simultaneously in size and density, the two corresponding segregation mechanisms (percolation and buoyancy, respectively) interact with each other resulting in segregation behavior significantly different from size or density segregation alone. In particular, mixtures of large heavy and small light particles can exhibit an equilibrium concentration, $c_{l,\text{eq}}$, at which the two segregation mechanisms are locally balanced and the net segregation flux is zero. This leads to a methodology in which a particle system can be designed to reduce or even prevent segregation by specifying the optimal combination of particle size ratio R_d , density ratio R_p , and overall mixture concentration \bar{c}_l .

The near overlap of the equilibrium concentration curves in Figure 7 with the non-segregating region determined from heap flow simulations demonstrates not only the accuracy of the equilibrium

conditions predicted in Figure 3 but also the potential for designing minimally segregating granular mixtures by feeding particles initially mixed at the equilibrium concentration, that is, $\bar{c}_l = c_{l,eq}$. This potential to intentionally design minimally segregating mixtures also extends to bidisperse particle mixtures in rotating tumblers with global concentration $\bar{c}_l = c_{l,eq}$, even when particles are initially segregated. In addition, particles in the tumbler core tend toward the equilibrium concentration even when $\bar{c}_l \neq c_{l,eq}$. This leads to a methodology to experimentally find $c_{l,eq}$ using a rotating tumbler by measuring the concentration in the core.

One further consideration is the impact of species size-polydispersity. For the simulations used to obtain the equilibrium concentrations in Figure 3, each particle species has a $\pm 10\%$ variation in diameter, although simulations with no intraspecies variation in particle diameter have nearly the same segregation flux dependence and resulting equilibrium concentrations.⁵⁷ This is consistent with a previous study of heap flow segregation,⁷⁵ which shows that segregation of size-bidisperse particle mixtures with broad species-specific size distributions can be accurately modeled as mixtures of two size-monodisperse species, even if the size distributions of the two species overlap. However, this topic deserves more research in the context of SD-systems due to the concentration dependence of the segregation direction.

Given the similarities in flow kinematics among heap, rotating tumbler, and other geometries where a thin surface layer of particles flows relative to the bulk, we expect that the equilibrium condition in Figure 3 is generally applicable to surface flows in other geometries.^{60,62,76} However, the equilibrium condition in Figure 3 may not be universal and may change somewhat with flow kinematics, since the solid volume fraction, overburden pressure, shear rate, and shear rate gradient affect particle segregation. For example, the equilibrium condition for equal-volume particle mixtures under vibration is $R_d \approx R_p$,⁷⁷ which clearly differs from the $c_{l,eq} = 0.5$ curve in Figure 3. Likewise, we expect that the equilibrium concentration would be different in wall-driven shear flow between two planes due to the overburden pressure⁶⁵ or in chute flow where substantial slip can occur at the base of the flow or the velocity exhibits a Bagnold-type profile. Nevertheless, our results demonstrate the potential for specifying particular combinations of particle size ratio, density ratio, and concentration to promote mixing and minimize segregation in granular surface flows. Future studies connecting equilibrium conditions to flow kinematics and determining non-segregating parameter combinations analogous to Figure 3 for other flows such as chute flows and boundary-driven (wall or intruder) flows are warranted.

AUTHOR CONTRIBUTIONS

Yifei Duan: Conceptualization (equal); data curation (lead); formal analysis (lead); investigation (lead); methodology (equal); software (lead); validation (lead); visualization (lead); writing – original draft (lead); writing – review and editing (equal). **Jack Peckham:** Investigation (supporting); visualization (supporting). **Julio Ottino:** Funding acquisition (equal); investigation (equal); resources (equal); supervision (equal); writing – review and editing (equal). **Paul Umphanowar:** Conceptualization (equal); formal analysis (equal); investigation (equal);

methodology (equal); visualization (lead); writing – review and editing (equal). **Richard Lueptow:** Funding acquisition (equal); conceptualization (equal); investigation (equal); methodology (equal); writing – review and editing (equal).

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DATA AVAILABILITY STATEMENT

Data available upon request from corresponding author.

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ENDNOTE

* For application, the curve can be approximated as a linear function $R_d = 0.163(R_p - 1) + 1$.

REFERENCES

1. Savage SB, Lun CKK. Particle size segregation in inclined chute flow of dry cohesionless granular solids. *J Fluid Mech.* 1988;189:311-335.
2. Makse HA, Havlin S, King PR, Stanley HE. Spontaneous stratification in granular mixtures. *Nature.* 1997;386(6623):379-382.
3. May LBH, Golick LA, Phillips KC, Shearer M, Daniels KE. Shear-driven size segregation of granular materials: modeling and experiment. *Phys Rev E.* 2010;81(5):051301.
4. Golick LA, Daniels KE. Mixing and segregation rates in sheared granular materials. *Phys Rev E.* 2009;80(4):042301.
5. Bhattacharya T, McCarthy JJ. Chute flow as a means of segregation characterization. *Powder Technol.* 2014;256:126-139.
6. Kholo N, Wassgren C. Correlations for shear-induced percolation segregation in granular shear flows. *Powder Technol.* 2016;288:441-452.
7. Fan Y, Jacob KV, Freireich B, Lueptow RM. Segregation of granular materials in bounded heap flow: a review. *Powder Technol.* 2017;312:67-88.
8. Khakhar DV, McCarthy JJ, Ottino JM. Radial segregation of granular mixtures in rotating cylinders. *Phys Fluids.* 1997;9(12):3600-3614.
9. Tripathi A, Khakhar DV. Density difference-driven segregation in a dense granular flow. *J Fluid Mech.* 2013;717:643-669.
10. Liao CC, Hsiao SS, Nien HC. Density-driven spontaneous streak segregation patterns in a thin rotating drum. *Phys Rev E.* 2014;89(6):062204.
11. Liu S, McCarthy JJ. Transport analogy for segregation and granular rheology. *Phys Rev E.* 2017;96(2):020901.
12. Duan Y, Umphanowar PB, Ottino JM, Lueptow RM. Segregation models for density-bidisperse granular flows. *Phys Rev Fluids.* 2020;5:044301.
13. Alizadeh M, Hassanpour A, Pasha M, Ghadiri M, Bayly A. The effect of particle shape on predicted segregation in binary powder mixtures. *Powder Technol.* 2017;319:313-322.
14. Pereira GG, Cleary PW. Segregation due to particle shape of a granular mixture in a slowly rotating tumbler. *Granul Matter.* 2017;19(2):1-12.
15. Lu G, Müller CR. Particle-shape induced radial segregation in rotating cylinders. *Granul Matter.* 2020;22(2):1-14.

16. Jones RP, Ottino JM, Umbanhowar PB, Lueptow RM. Predicting segregation of nonspherical particles. *Phys Rev Fluids*. 2021;6(5):054301.
17. Kondic L, Hartley RR, Tennakoon SGK, Painter B, Behringer RP. Segregation by friction. *Europhys Lett*. 2003;61(6):742-748.
18. Gillemot KA, Somfai E, Börzsönyi T. Shear-driven segregation of dry granular materials with different friction coefficients. *Soft Matter*. 2017;13(2):415-420.
19. Standish N. Studies of size segregation in filling and emptying a hopper. *Powder Technol*. 1985;45(1):43-56.
20. Muzzio FJ, Shinbrot T, Glasser BJ. Powder technology in the pharmaceutical industry: the need to catch up fast. *Powder Technol*. 2002; 124(1-2):1-7.
21. Ottino JM, Lueptow RM. On mixing and demixing. *Science*. 2008; 319(5865):912-913.
22. Gray JMNT, Kokelaar BP. Large particle segregation, transport and accumulation in granular free-surface flows. *J Fluid Mech*. 2010;652: 105-137.
23. Gray JMNT, Ancey C. Multi-component particle-size segregation in shallow granular avalanches. *J Fluid Mech*. 2011;678:535-588.
24. Gray JMNT. Particle segregation in dense granular flows. *Annu Rev Fluid Mech*. 2018;50:407-433.
25. Umbanhowar PB, Lueptow RM, Ottino JM. Modeling segregation in granular flows. *Annu Rev Chem Biomol Eng*. 2019;10:129-153.
26. Alonso M, Satoh M, Miyanami K. Optimum combination of size ratio, density ratio and concentration to minimize free surface segregation. *Powder Technol*. 1991;68(2):145-152.
27. Jain N, Ottino JM, Lueptow RM. Regimes of segregation and mixing in combined size and density granular systems: an experimental study. *Granul Matter*. 2005;7(2-3):69-81.
28. Williams JC. The mixing of dry powders. *Powder Technol*. 1968;2(1): 13-20.
29. Drahun JA, Bridgwater J. The mechanisms of free surface segregation. *Powder Technol*. 1983;36(1):39-53.
30. Ottino JM, Khakhar DV. Mixing and segregation of granular materials. *Annu Rev Fluid Mech*. 2000;32(1):55-91.
31. Ristow GH. Particle mass segregation in a two-dimensional rotating drum. *Europhys Lett*. 1994;28(2):97-101.
32. Khakhar DV, McCarthy JJ, Ottino JM. Mixing and segregation of granular materials in chute flows. *Chaos*. 1999;9(3):594-610.
33. Pereira GG, Sinnott MD, Cleary PW, Liffman K, Metcalfe G, Šutalo ID. Insights from simulations into mechanisms for density segregation of granular mixtures in rotating cylinders. *Granul Matter*. 2011;13(1):53-74.
34. Metcalfe G, Shattuck M. Pattern formation during mixing and segregation of flowing granular materials. *Phys A: Stat Mech Appl*. 1996; 233(3-4):709-717.
35. Jenkins JT, Yoon DK. Segregation in binary mixtures under gravity. *Phys Rev Lett*. 2002;88(19):194301.
36. Félix G, Thomas N. Evidence of two effects in the size segregation process in dry granular media. *Phys Rev E*. 2004;70(5):051307.
37. Tunuguntla DR, Bokhove O, Thornton AR. A mixture theory for size and density segregation in shallow granular free-surface flows. *J Fluid Mech*. 2014;749:99-112.
38. Larcher M, Jenkins JT. The evolution of segregation in dense inclined flows of binary mixtures of spheres. *J Fluid Mech*. 2015;782:405-429.
39. Jing L, Ottino JM, Lueptow RM, Umbanhowar PB. Rising and sinking intruders in dense granular flows. *Phys Rev Res*. 2020;2(2):022069.
40. Gray JMNT, Thornton AR. A theory for particle size segregation in shallow granular free-surface flows. *Proc R Soc A*. 2005;461(2057): 1447-1473.
41. Tunuguntla DR, Weinhart T, Thornton AR. Comparing and contrasting size-based particle segregation models. *Comput Part Mech*. 2017;4(4): 387-405.
42. Fan Y, Schlick CP, Umbanhowar PB, Ottino JM, Lueptow RM. Modelling size segregation of granular materials: the roles of segregation, advection and diffusion. *J Fluid Mech*. 2014;741:252-279.
43. Schlick CP, Fan Y, Umbanhowar PB, Ottino JM, Lueptow RM. Granular segregation in circular tumblers: theoretical model and scaling laws. *J Fluid Mech*. 2015;765:632-652.
44. Jones RP, Isner AB, Xiao H, Ottino JM, Umbanhowar PB, Lueptow RM. Asymmetric concentration dependence of segregation fluxes in granular flows. *Phys Rev Fluids*. 2018;3(9):094304.
45. Gray JMNT, Ancey C. Particle-size and-density segregation in granular free-surface flows. *J Fluid Mech*. 2015;779:622-668.
46. Franklin SV, Shattuck MD. *Handbook of Granular Materials*. CRC Press; 2016.
47. Jenkins JT, Savage SB. A theory for the rapid flow of identical, smooth, nearly elastic, spherical particles. *J Fluid Mech*. 1983;130: 187-202.
48. GDR-MiDi. On dense granular flows. *Eur Phys J E*. 2004;14:341-365.
49. Pouliquen O, Cassar C, Jop P, Forterre Y, Nicolas M. Flow of dense granular material: towards simple constitutive laws. *J Stat Mech Theory Exp*. 2006;2006(7):P07020.
50. Jop P, Forterre Y, Pouliquen O. A constitutive law for dense granular flows. *Nature*. 2006;441(7094):727-730.
51. Kamrin K, Koval G. Nonlocal constitutive relation for steady granular flow. *Phys Rev Lett*. 2012;108(17):178301.
52. Henann DL, Kamrin K. A predictive, size-dependent continuum model for dense granular flows. *Proc Natl Acad Sci*. 2013;110(17):6730-6735.
53. Meier SW, Lueptow RM, Ottino JM. A dynamical systems approach to mixing and segregation of granular materials in tumblers. *Adv Phys*. 2007;56(5):757-827.
54. Ottino JM, Khakhar DV. Scaling of granular flow processes: from surface flows to design rules. *AIChE J*. 2002;48(10):2157-2166.
55. Dolgunin VN, Ukolov AA. Segregation modeling of particle rapid gravity flow. *Powder Technol*. 1995;83(2):95-103.
56. Dolgunin VN, Kudy AN, Ukolov AA. Development of the model of segregation of particles undergoing granular flow down an inclined chute. *Powder Technol*. 1998;96(3):211-218.
57. Duan Y, Umbanhowar PB, Ottino JM, Lueptow RM. Modelling segregation of bidisperse granular mixtures varying simultaneously in size and density for free surface flows. *J Fluid Mech*. 2021;918:A20.
58. Duan Y, Umbanhowar PB, Lueptow RM. Designing non-segregating granular mixtures. *EPJ Web Conf*. 2021;249:3011.
59. Fan Y, Schlick CP, Umbanhowar PB, Ottino JM, Lueptow RM. Kinematics of monodisperse and bidisperse granular flows in quasi-two-dimensional bounded heaps. *Proc R Soc A*. 2013;469(2157):20130235.
60. Isner AB, Umbanhowar PB, Ottino JM, Lueptow RM. Axisymmetric granular flow on a bounded conical heap: kinematics and size segregation. *Chem Eng Sci*. 2020;217:115505.
61. Xiao H, Umbanhowar PB, Ottino JM, Lueptow RM. Modelling density segregation in flowing bidisperse granular materials. *Proc R Soc A*. 2016;472(2191):20150856.
62. Isner AB, Umbanhowar PB, Ottino JM, Lueptow RM. Granular flow in a wedge-shaped heap: velocity field, kinematic scalings, and segregation. *AIChE J*. 2020;66(5):e16912.
63. Prasad I, Santangelo C, Grason G. Subjamming transition in binary sphere mixtures. *Phys Rev E*. 2017;96(5):052905.
64. Jing L, Ottino JM, Lueptow RM, Umbanhowar PB. A unified description of gravity-and kinematics-induced segregation forces in dense granular flows. *J Fluid Mech*. 2021;925:A29.
65. Fry AM, Umbanhowar PB, Ottino JM, Lueptow RM. Effect of pressure on segregation in granular shear flows. *Phys Rev E*. 2018;97:062906.
66. Pereira GG, Tran N, Cleary PW. Segregation of combined size and density varying binary granular mixtures in a slowly rotating tumbler. *Granul Matter*. 2014;16(5):711-732.
67. Nityanand N, Manley B, Henein H. An analysis of radial segregation for different sized spherical solids in rotary cylinders. *Metall Mater Trans B*. 1986;17(2):247-257.
68. Cantelaube F, Bideau D. Radial segregation in a 2d drum: an experimental analysis. *Europhys Lett*. 1995;30(3):133-138.

69. Mellmann J. The transverse motion of solids in rotating cylinders—forms of motion and transition behavior. *Powder Technol.* 2001; 118(3):251-270.
70. Zaman Z, D'Ortona U, Umbanhowar PB, Ottino JM, Lueptow RM. Slow axial drift in three-dimensional granular tumbler flow. *Phys Rev E.* 2013;88(1):012208.
71. Thomas N, D'Ortona U. Evidence of reverse and intermediate size segregation in dry granular flows down a rough incline. *Phys Rev E.* 2018;97(2):022903.
72. D'Ortona U, Thomas N, Lueptow RM. Mechanisms for recirculation cells in granular flows in rotating cylindrical rough tumblers. *Phys Rev E.* 2022;105(1):014901.
73. D'Ortona U, Thomas N, Zaman Z, Lueptow RM. Influence of rough and smooth walls on macroscale flows in tumblers. *Phys Rev E.* 2015; 92(6):062202.
74. D'Ortona U, Thomas N, Lueptow RM. Influence of rough and smooth walls on macroscale granular segregation patterns. *Phys Rev E.* 2016; 93(2):022906.
75. Gao S, Ottino JM, Umbanhowar PB, Lueptow RM. Modeling granular segregation for overlapping species distributions. *Chem Eng Sci.* 2021; 231:116259.
76. Deng Z, Fan Y, Theuerkauf J, Jacob KV, Umbanhowar PB, Lueptow RM. Modeling segregation of polydisperse granular materials in hopper discharge. *Powder Technol.* 2020;374:389-398.
77. Hong DC, Quinn PV, Luding S. Reverse Brazil nut problem: competition between percolation and condensation. *Phys Rev Lett.* 2001; 86(15):3423-3426.

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