

# Introduction to PHY207

Dr. Fly

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# Course Objectives

The objectives of this course are to:

1. Guide students to investigate physics problem by designing an experiment.
2. Introduce student different kinds of sensors used in experiment.
3. Train students work together in group to solve a problem.

# Course Learning Outcomes

At the end of this course, students will be able to:

1. Adapt an experimental setup to investigate physical problems or scenarios (P5,PLO3)
2. Work together in pairs or a group to plan, setup and implement the experimental investigation (A3, PLO4)
3. Work together in pairs or a group to plan, setup and implement the experimental investigation (A5,PLO5)
4. Construct charts and graphs using graphical software for the analysis of experimental findings(P4,PLO6)
5. Organize a functional team with diverse roles to tackle different aspects of the experimental investigation(A4,PLO8)

# Assessment

## Method of Assessment

## Total

- Instructor's Observation

40%

- Lab Report

30%

- Viva Voce(Progress/Final Presentation)

30%

- 

**TOTAL**

**100%**

# Roles of group members

Manager: Hosting meeting, Planning project and budget, Purchasing



Engineer: Designing and assembling equipment, setting up and conducting experiment

Programmer: Developing Apps for controlling equipment, collecting and analyzing data,



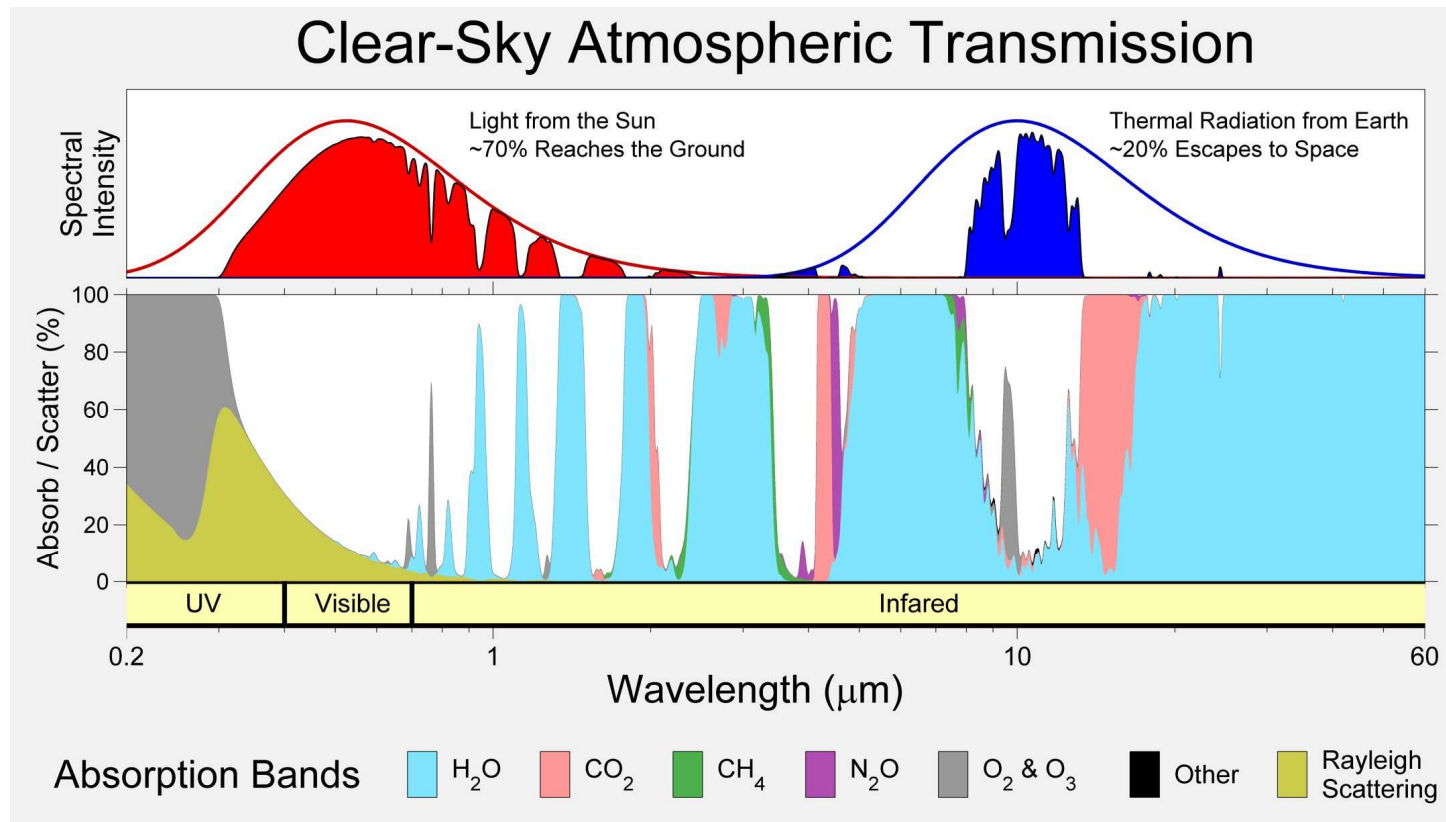
<b>Week 1</b>	<b>Online meeting</b> <b>Introduction to the course and Project</b> <ul style="list-style-type: none"> <li>- Spectroscopy and its applications.</li> <li>- Preparation of materials.</li> <li>- Suggestions:               <ol style="list-style-type: none"> <li>1. Determine the grating constant for different brands of cd or dvd.</li> <li>2. Beer-Lambert Law</li> <li>3. Measurement of Planck's constant</li> <li>4. Absorption spectrum</li> <li>5. Solar Spectrum</li> </ol> </li> </ul>
<b>Week 2</b>	<b>Online meeting</b> <b>Microsoft Teams Group meeting 1 (~ 30 minutes)</b> <ul style="list-style-type: none"> <li>- Hosted by each group in English</li> <li>- To brainstorm the project direction</li> <li>- To distribute the work</li> </ul>
<b>Week 3</b>	<b>Online meeting</b> <b>Introduction to</b> <ul style="list-style-type: none"> <li>- Error analysis</li> <li>- Scientific report writing</li> </ul> <p>* Each group should have got the essential parts for the project.          * Each group can add on any tools or devices subjected to the budget.</p>
<b>Week 4</b>	<b>Online meeting</b> <b>Microsoft Teams Group meeting 2 (~ 30 minutes)</b> <ul style="list-style-type: none"> <li>- Hosted by each group in English</li> <li>- Finalized the project direction with a preliminary design</li> </ul>
<b>Week 5</b>	<b>Progress Report -hosted by the lecturer</b> <ul style="list-style-type: none"> <li>- Students should have all the essential parts and a preliminary design.</li> <li>- Online meeting</li> </ul>
<b>Week 6</b>	<b>Microsoft Teams Group meeting 3 (~ 30 minutes)</b> <ul style="list-style-type: none"> <li>- Hosted by each group</li> </ul>

# Spectroscopy

(Electromagnetism + Modern Physics)

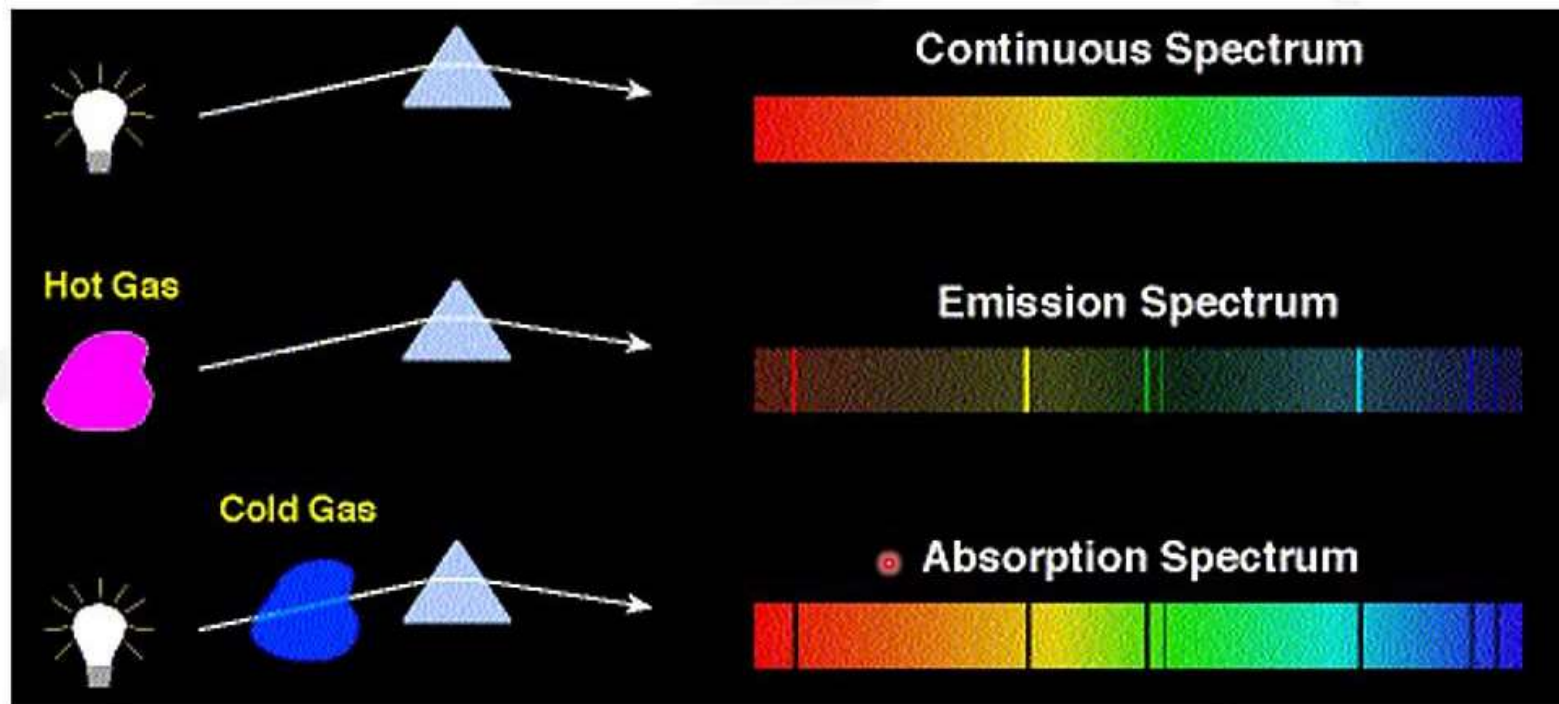
Dr. Fly

# Importance of Spectroscopy

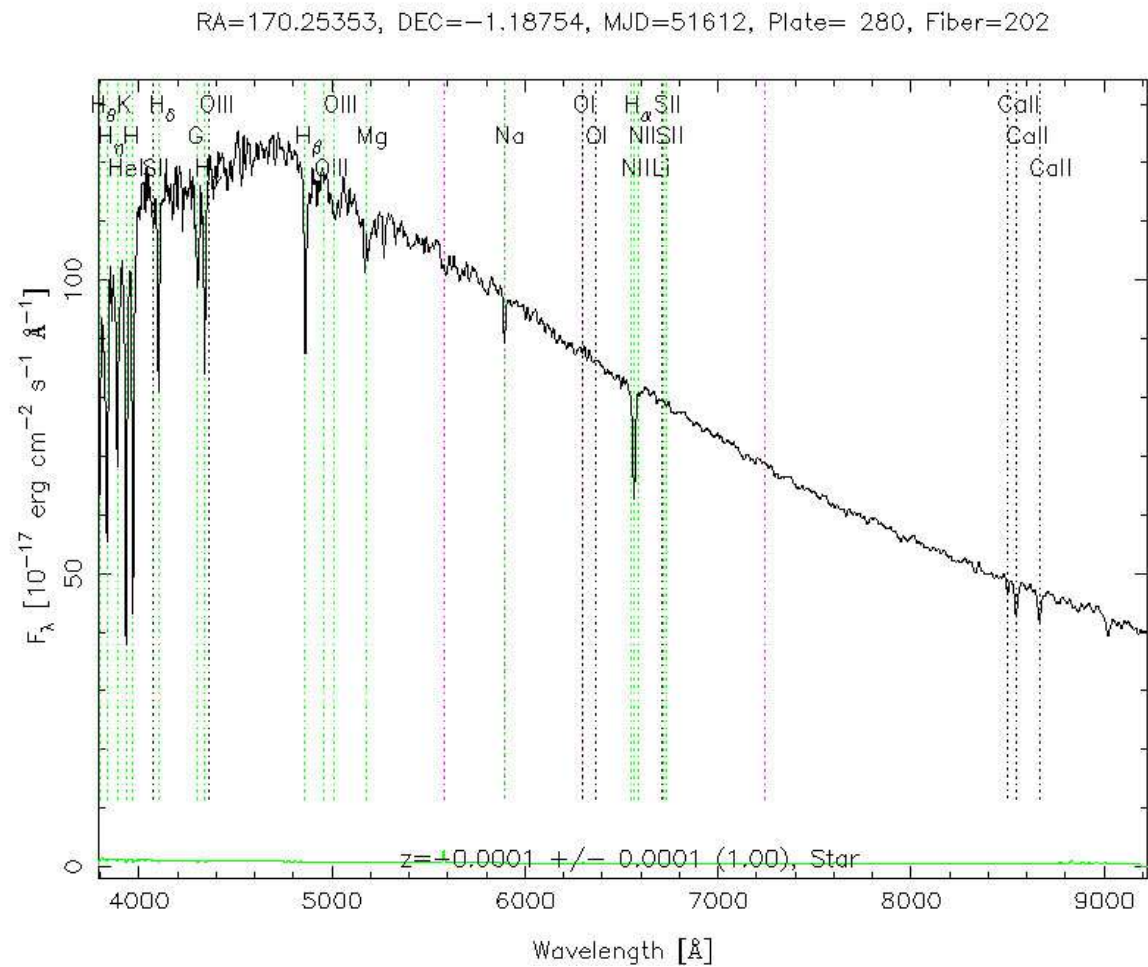




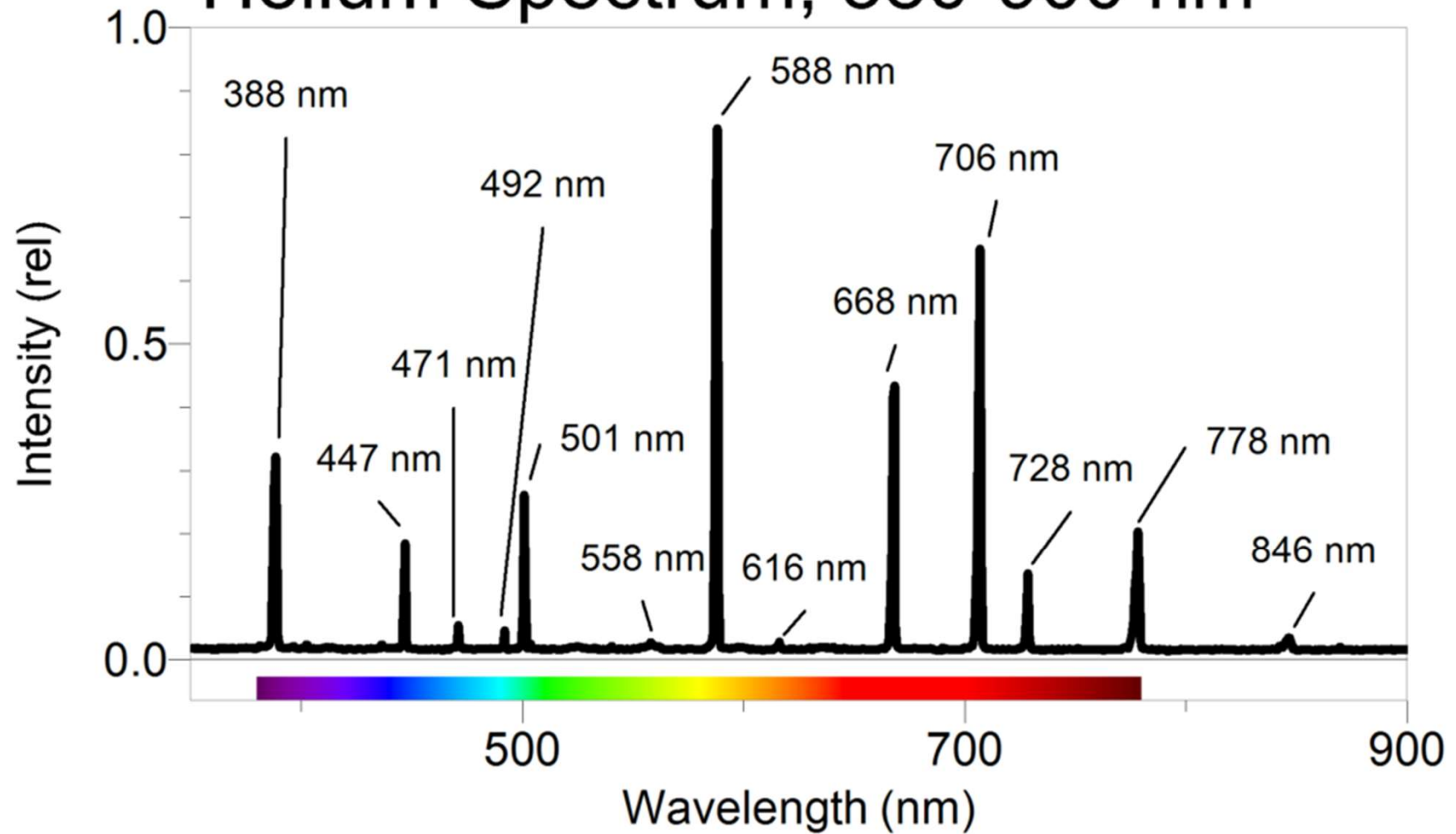
# Continuous, emission, and absorption spectra



# Spectrum of a Star



# Helium Spectrum, 380-900 nm



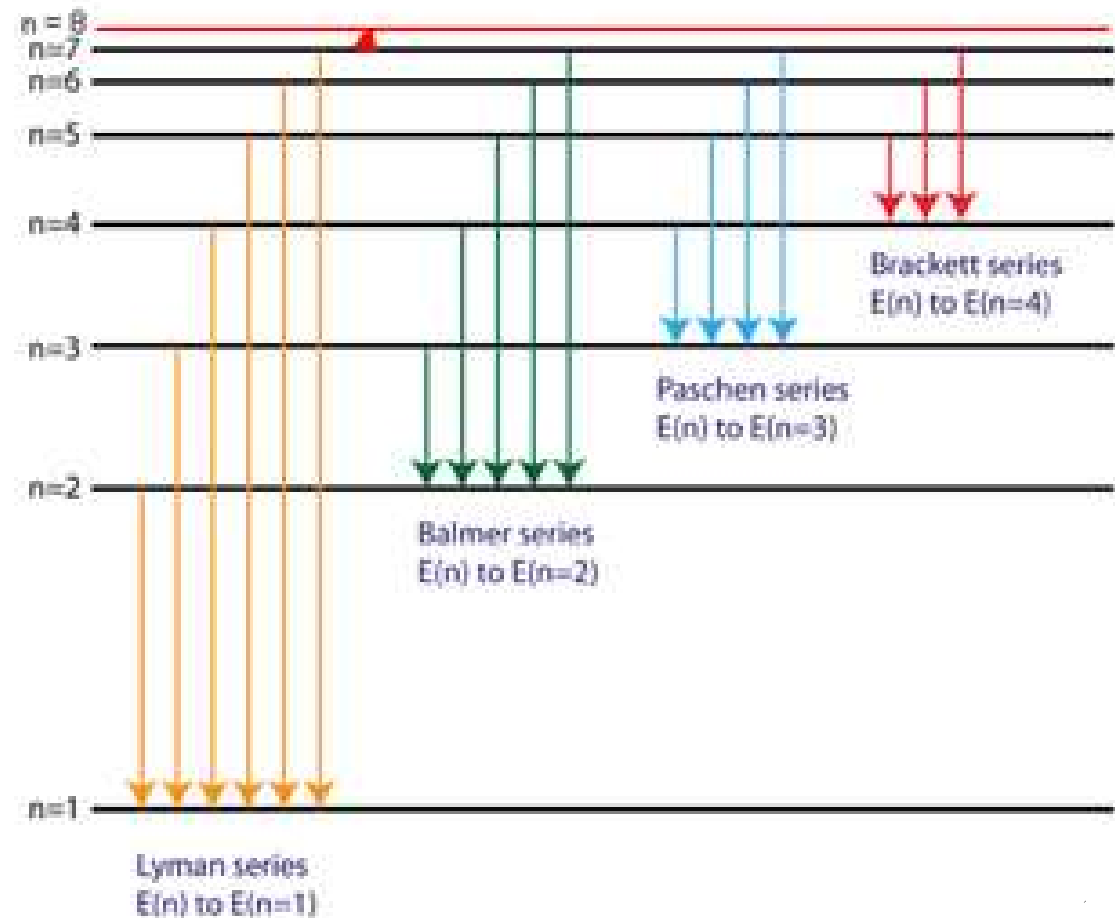
# Absorption and Emission

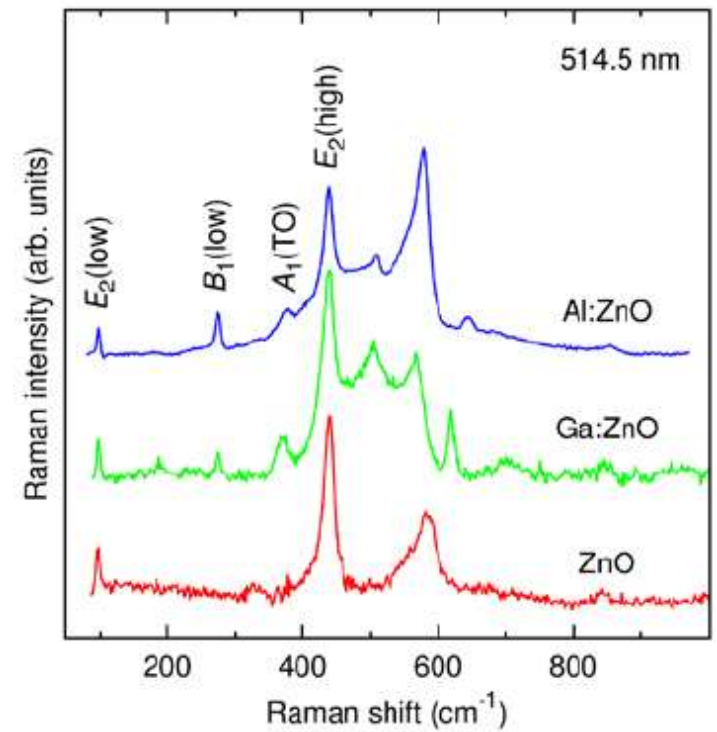
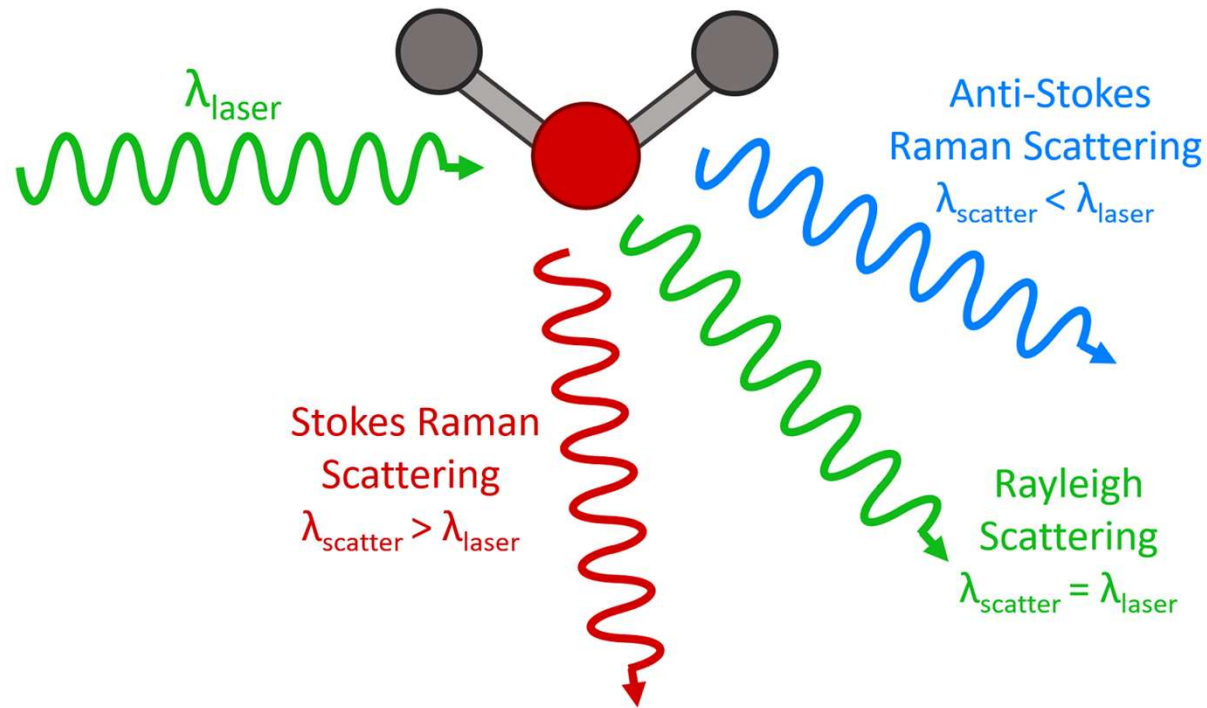


$$L = \underbrace{rmv}_{n\hbar}$$

$$E = \frac{1}{2}mv^2 - K\frac{e^2}{r} = -\frac{E_0}{n^2}$$

Electron transitions for the Hydrogen atom



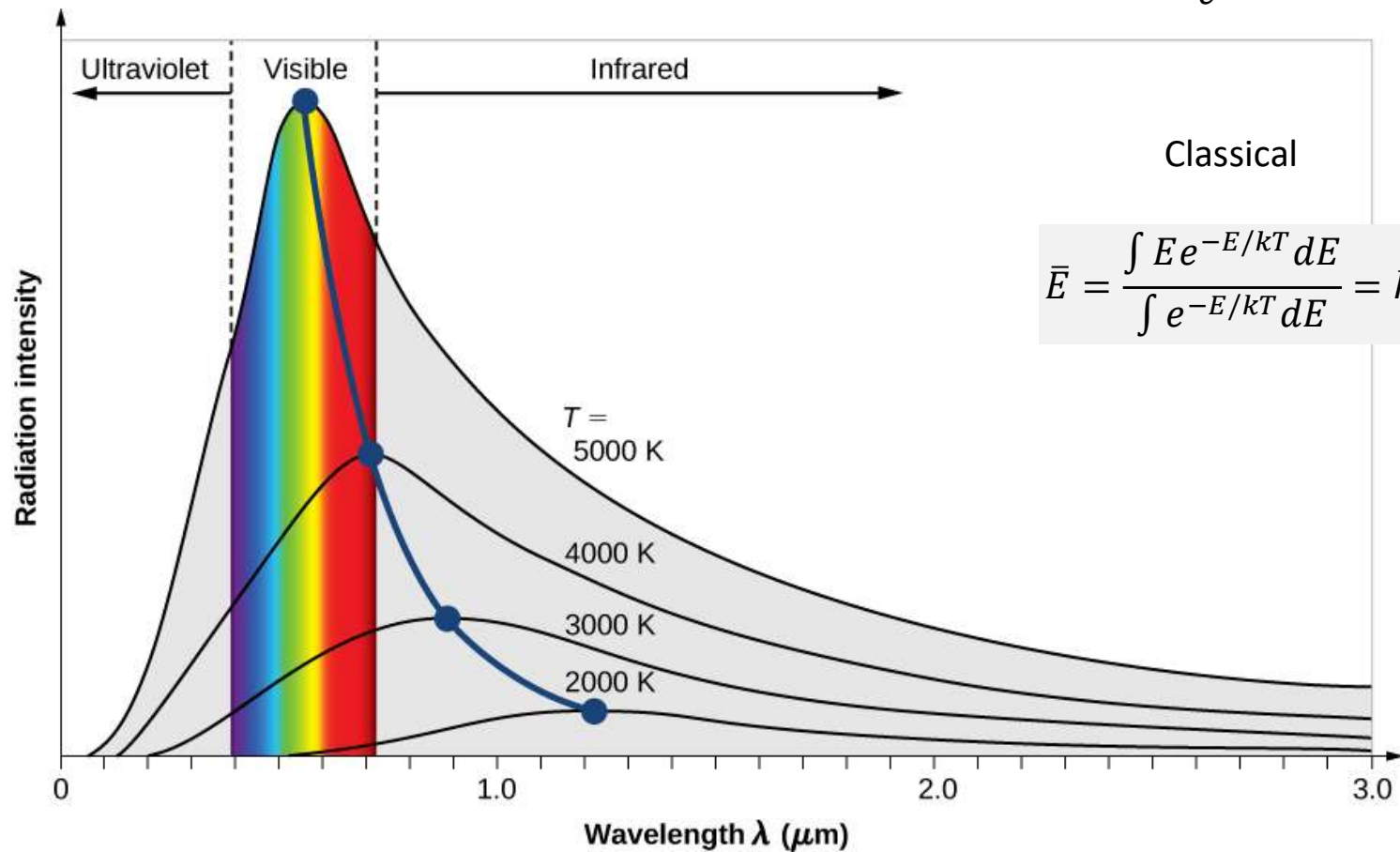


# Rayleigh and Raman Scattering

# Black body spectrum

Number of modes

$$N = \frac{8\pi f^2}{c^3}$$



Classical

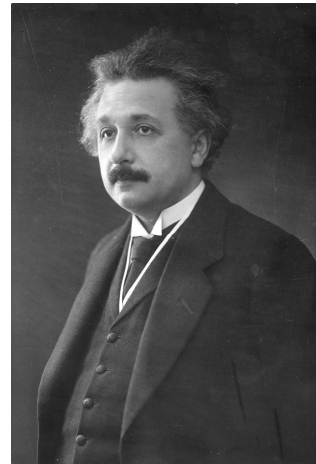
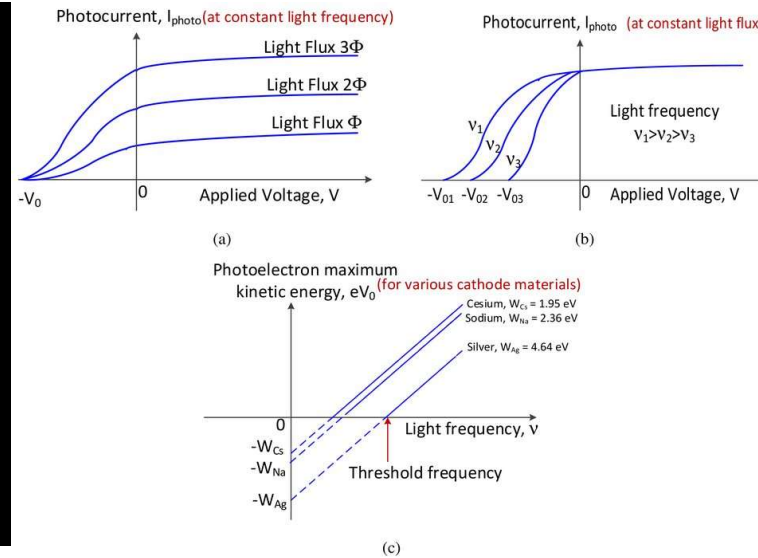
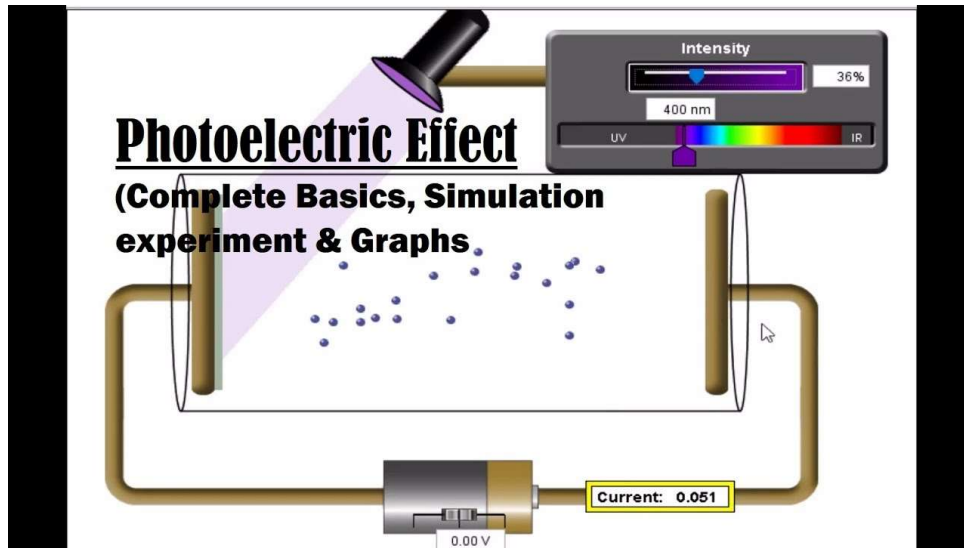
$$\bar{E} = \frac{\int E e^{-E/kT} dE}{\int e^{-E/kT} dE} = kT$$

Planck

$$\begin{aligned} \bar{E} &= \frac{\sum nhf e^{-nhf/kT}}{\sum e^{-nhf/kT}} \\ &= \frac{hf}{e^{hf/kT} - 1} \end{aligned}$$



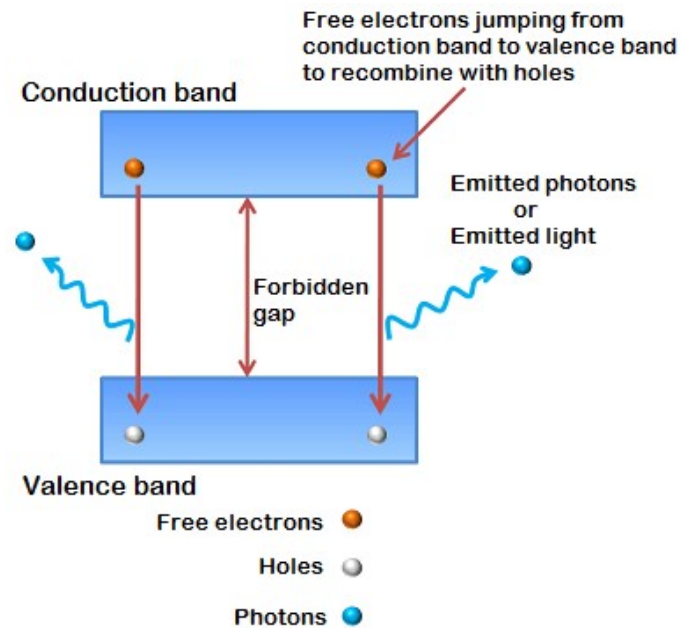
# Photoelectric experiment



$$eV = hf - W$$

$$h = \frac{\Delta V}{\Delta f}$$

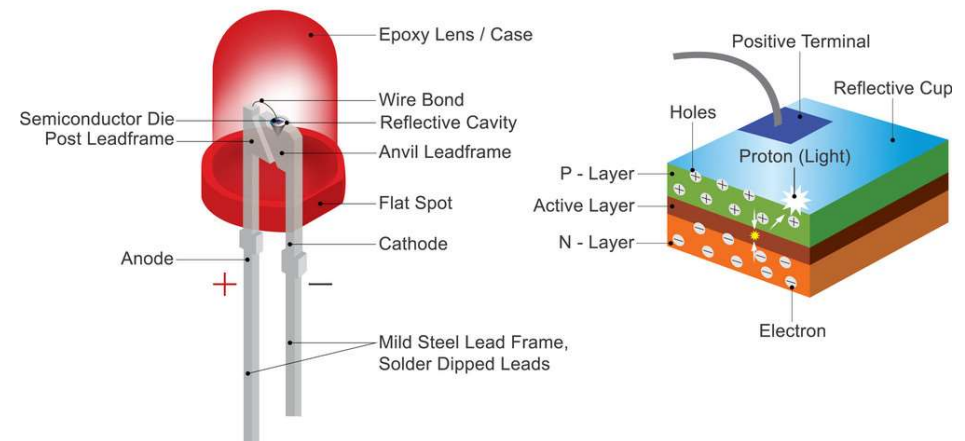
# LED



## Process of light emission in LED

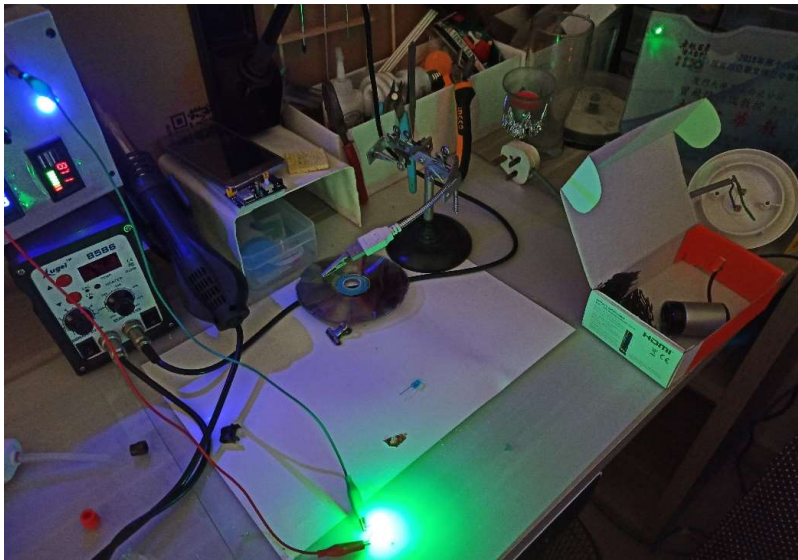
*Physics and Radio-Electronics*

## A Light-Emitting Diode (LED)





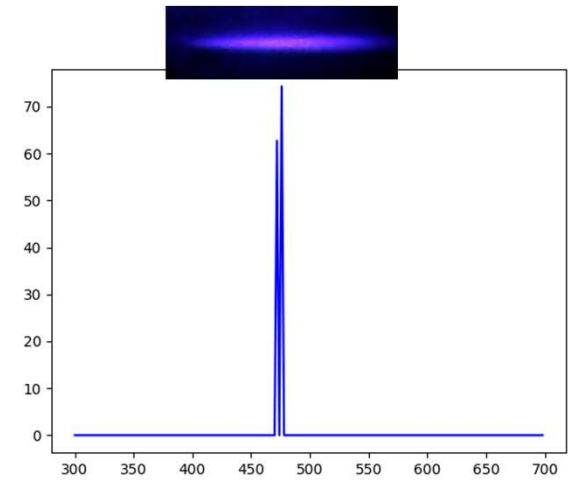
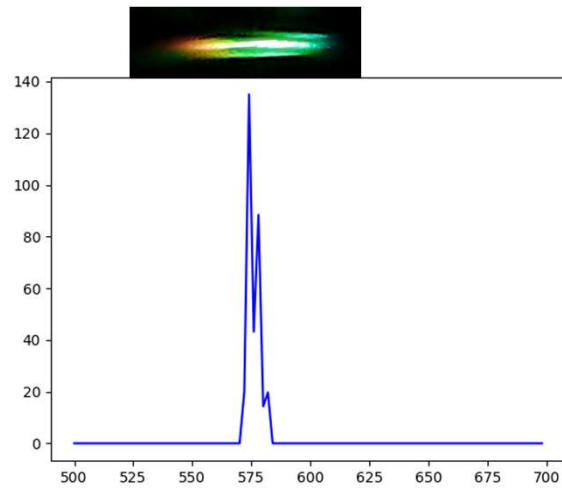
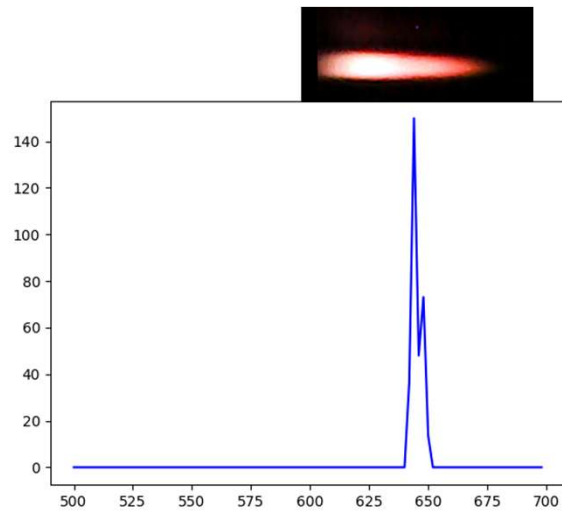
# DIY spectrometer



[\(54\) How to build a spectrometer from the College of Natural Sciences - CSU Online - YouTube](#)

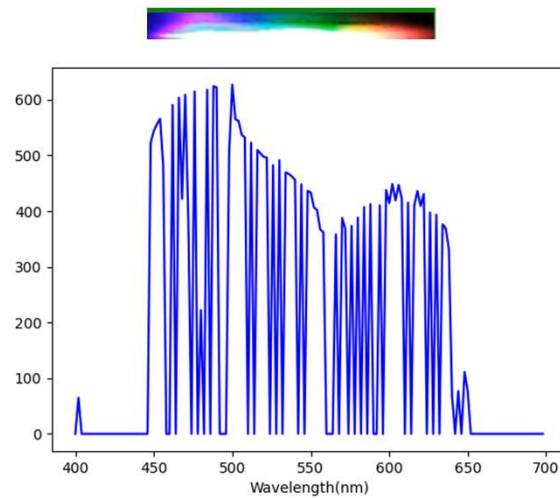
[\(54\) DIY Spectrograph - YouTube](#)

# Calibration with LED

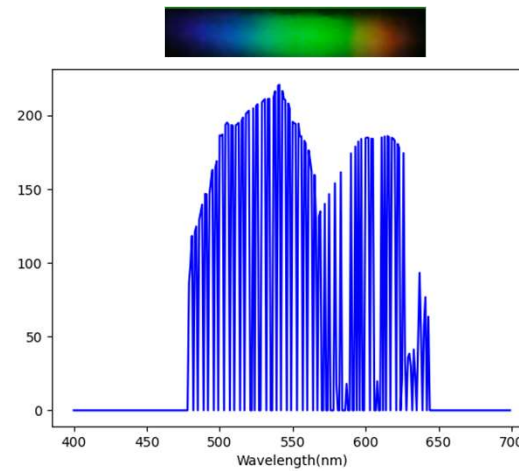


$$h \approx 7.0 \times 10^{-34} \text{ Js}$$

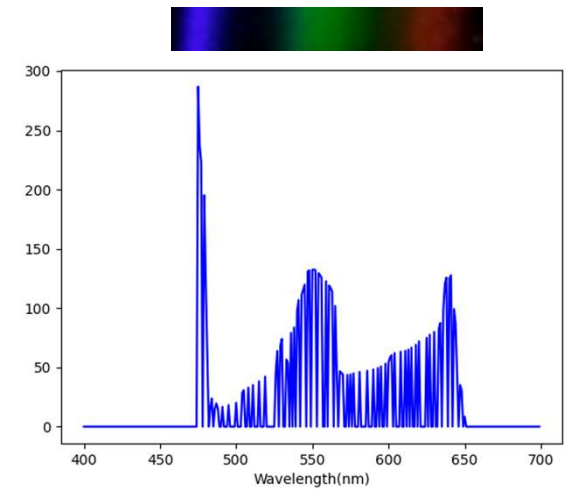
# White LED



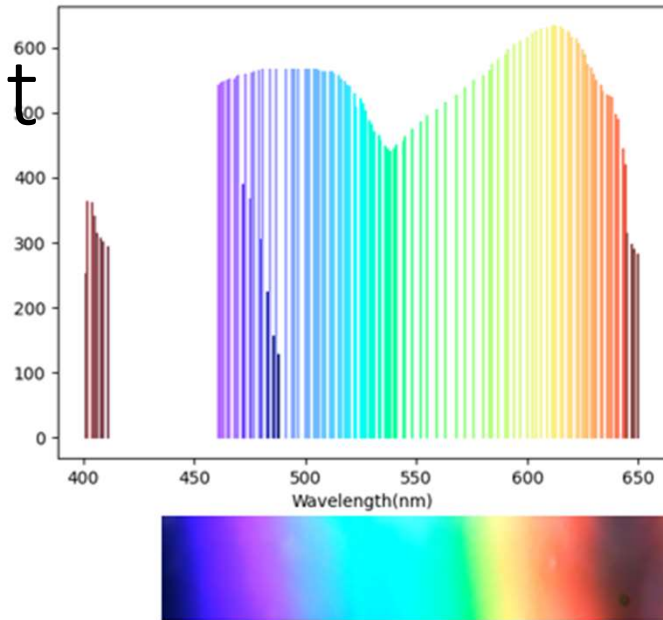
# Candle



# LED monitor



# Sunlight



# Solar Panel

