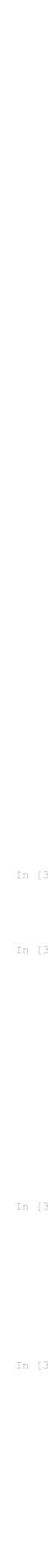
**Problem 3** In [ ]: # problem 3.1 The reactor in the problem is examined in an IMRCF.  $A(g) + B(g) \rightleftharpoons C(g) + D(g)$ D selectively permeates through the membrane at the rate  $R_D = k_d \cdot CD$ The following code is adapted from the provided IMRCF codes on canvas. For Total Concentration (y-axis) vs Catalyst Mass (x-axis)  $C_j = rac{F_j}{F} * C_0$ • The total molar flow rate is adapted from the provided code # Import packages import numpy as np import matplotlib.pyplot as plt from scipy.integrate import solve\_ivp from scipy.stats import linregress import pandas as pd In [4]: # Provided by Professor Eagan def diff(t,U): # Specify identity of each U for clarity FA = U[0] # volumetric feed of CO FB = U[1] # volumetric feed of H2O FC = U[2] # volumetric feed of CO2FD = U[3] # volumetric feed of H2# Perform calculations F = FA + FB + FC + FDC = C0# Concentration CA = FA/F\*C # COCB = FB/F\*C # H20CC = FC/F\*C # CO2CD = FD/F\*C # H2QC = CD\*CC/(CA\*CB)r = kf\*(CA\*CB)\*(1-QC/KC) # technically written as -rA # Define derivatives dFA = -r # flip sign forwardsdFC = r # retain negative for backwards dFD = r - kH2\*CD # accounting for permeations #Populate derivatives array dU=np.zeros(len(U)) dU[0] = dFAdU[1] = dFBdU[2] = dFCdU[3] = dFD# Reorganize list to fit required dimensions dU = np.array(dU).tolist() return dU # Specifications P0 = 101.325 # kPaR = 8.314 # kPa L / mol Kkf = 1.37 #L^2/mol kg min KC = 1.44 #mol/L # EQUILIBRIUM constant # Initial Concentration  $C0 = 0.4 #mol/dm^3$  $FA0 = 10 * C0/2 #dm^3/min$ # DOUBLE THE FLOW RATE HERE # FA0 = 2\*FA0FB0 = FA0# Set initial conditions U0 = [FA0, FB0, 0, 0]# Determine integration limits tlim = (0, 100)tlist = np.linspace(tlim[0], tlim[1], num=100) # Integrate solution = solve\_ivp(diff,tlim,U0,t\_eval=tlist) # Obtain ouputs and times U = solution.y.T W = solution.tFA = U[:, 0]FB = U[:,1]FC = U[:,2]FD = U[:,3]FT = FA + FB + FC + FDCA = FA/FT\*C0CB = FB/FT\*C0CC = FC/FT\*C0CD = FD/FT\*C0CT = FT/FT\*C0# Plot data fig = plt.figure() ax1 = plt.subplot(121)# Since the reactants are equimolar entering at the same concentration # it is reasonable to identify an overlap in the reactant A and B lines ax1.plot(W,FA,'b') # ax1.plot(W,FB,'g') ax1.plot(W,FC,"r") ax1.plot(W,FD,'c') ax1.plot(W,FT, "m") ax1.legend(['A and B','C','D','Total']) ax1.set xlabel('Catalyst mass (kg)') ax1.set ylabel('Fj (mol/min)') ax1.set title('Molar flow rates at Keq') ax2 = plt.subplot(122)ax2.plot(W,CA,'b') # ax2.plot(W,CB,'g') ax2.plot(W,CC,"r") ax2.plot(W,CD,'c') ax2.plot(W,CT,'m') ax2.legend(['A and B','C','D','Total']) ax2.set xlabel('Catalyst mass (kg)') ax2.set ylabel('Cj (mol/dm^3)') ax2.set\_title('Concentrations at Keq') fig.tight layout() plt.show() Molar flow rates at Keq Concentrations at Keq 4.0 0.40 3.5 0.35 0.30 3.0 € 0.25 (mol/min) 2.0 1.5 A and B A and B (mol/dm C C 0.20 D D Total Total 0.15 1.0 0.10 0.5 0.05 0.0 0.00 100 100 50 50 25 75 0 75 Catalyst mass (kg) Catalyst mass (kg) Problem 4 from scipy.stats import linregress import math # Problem 4.1 t = np.array([0,5,9,15,22,30,40,60])CA = np.array([2,1.6,1.35,1.1,0.87,0.70,0.53,0.35])deltat = np.diff(t)deltaC = np.diff(CA) negCA\_t = np.multiply(deltaC, -1)/deltat  $nat_lgCA = np.log(CA)$ x axis = np.delete(nat lgCA,0) # omit first index y\_axis = np.log(negCA\_t) slope, intercept, r\_value, p\_value, std\_err = linregress(x\_axis, nat\_negCA\_t) print("n is {:.5f} and therefore the reaction order is".format(slope), math.ceil(slope)) n is 1.42093 and therefore the reaction order is 2# Problem 4.2  $CA0 = CA[0] \# mol/dm^3$ x axis = t $y_axis = 1/CA - 1/CA0$ # Second Order  $nA_pos = round(slope + 0.5)$ fig = plt.figure() ax1 = plt.subplot(111)ax1.plot(x\_axis, y\_axis,"p") slope1, intercept1, r1\_value, p1\_value, std\_err1 = linregress(x\_axis, y\_axis) ax1.plot(x\_axis, slope1\*x\_axis,'--') ax1.legend(["actual data","linear approximation"]) ax1.set xlabel('t (min)') ax1.set ylabel('CA0 - Ca (M)') ax1.set\_title('2nd order at nA + 0.5') plt.show()  $print("y1 = {:.5f}x + {:.5f}".format(slope1, intercept1))$ print("R^2 ={:.3f}".format(r1\_value)) # First Order nA neg = round(slope) ax2 = plt.subplot(111) $y_axis = np.log(CAO/CA)$ ax2.plot(x\_axis, y\_axis, "p") slope2, intercept2, r2\_value, p2\_value, std\_err2 = linregress(x\_axis, y\_axis) ax2.plot(x\_axis, x\_axis\*slope2, '--') ax2.legend(["actual data","linear approximation"]) ax2.set\_xlabel('t (min)') ax2.set ylabel('ln(CA/CA0) (M)') ax2.set\_title('1st order at nA - 0.5') plt.show()  $print("y2 = {:.5f}x + {:.5f}".format(slope2, intercept2))$  $print("R^2 = {:.3f}".format(r2 value))$ 2nd order at nA + 0.5 actual data linear approximation 2.0 CA0 - CA (M) 0.5 0.0 40 50 10 20 30 60 t (min) y1 = 0.03902x + -0.12075 $R^2 = 0.993$ 1st order at nA - 0.5 1.75 actual data linear approximation 1.50 1.25 In(CA/CA0) (M) 1.00 0.75 0.50 0.25 0.00 30 40 60 10 20 50 t (min) y2 = 0.02894x + 0.11625 $R^2 = 0.991$ # Problem 4.3 from scipy.optimize import curve fit from scipy.stats.distributions import t as tdist In [324... # Model def model(x,\*parameters): # Indepdendent variables CA = xndtps = len(CA)# Parameters to be fit k = parameters[0] nA = parameters[1] # Calculate output value for each data point based on a model time\_model = np.zeros(ndpts) for i in range(ndpts): time model[i] = (-1/k)\*(((CA[i]\*\*(1-nA))/(1-nA)) - (2\*\*(1-nA))/(1-nA)) # Manually write the rate express return time model # Data () # see line 186 in Problem 4.1 ndpts = len(t)# Parameter Guess k g = 0.0390na g = slope  $guess = ([k_g, na_g])$ npars = len(guess) In [326... # Regressions xvars = CA popt,pcov = curve fit(model,xvars,t,guess,method='lm') # Confidence intervals alpha = 0.05 # 95% confidence interval dof = max(0,ndpts-npars) # number of degrees of freedom tval = tdist.ppf(1.0-alpha/2.0,dof) # student t value for the dof and confidence level ci = np.zeros([npars,2]) for i,p,var in zip(range(ndpts),popt,np.diag(pcov)): sigma = var\*\*0.5



ci[i,:] = [p-sigma\*tval,p+sigma\*tval]

sqr\_resid = (time\_predict[i]-t[i])\*\*2

pd.options.display.float\_format = "{:,.3f}".format

0.000

0.038

ax1.plot(t,time\_predict,'ob',mfc='none')

20

CAO = np.arange(1, 4.5, 0.5, dtype = "float")

print("nA is {:.4f}".format(slope4))

print("k is {:.4f}".format(np.exp(intercept4)))

 $neg_rA = np.multiply(rc, 0.5)$ nat\_rA = np.log(neg\_rA)  $nat_CA0 = np.log(CA0)$ 

30 Experimental Rate

rc = np.array([0.063, 0.114, 0.182, 0.264, 0.353, 0.419, 0.528])

slope4, intercept4, r4\_value, p4\_value, std\_err4 = linregress(x\_axis, y\_axis)

ax1.plot(x\_parity,y\_parity,'--b') ax1.set\_xlabel('Experimental Rate') ax1.set ylabel('Predicted Rate')

Value 95% CI Half Width 95% CI Half Width Rel %

soln['95% CI Half Width'] = ci\_width/2

print('SSR = ' + "{:.3e}".format(ssr))

time predict = np.zeros(ndpts)

ssr = ssr + sqr\_resid

for i in range(ndpts):

print('\n', soln, '\n')

SSR = 4.978e-01

0.033

# Plot data

fig = plt.figure() ax1 = plt.subplot(111)

fig.tight\_layout()

plt.show()

60

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# Problem 4.4

 $x_axis = nat_CA0$ y\_axis = nat rA

nA is 1.5433 k is 0.0313

Predicted Rate 30

In [328...

## Parity plots

nA 1.515

k

soln = pd.DataFrame(popt,index=param names,columns=['Value'])

# Generate matrix containing simulated rates and calculate the sum of squared residuals (ssr)

0.956

2.487

time predict[i]=(-1/k)\*(((CA[i]\*\*(1-nA))/(1-nA))-(2\*\*(1-nA))/(1-nA))

x\_parity = [min([min(t), min(time\_predict)]), max([max(t), max(time\_predict)])] y\_parity = [min([min(t), min(time\_predict)]), max([max(t), max(time\_predict)])]

soln['95% CI Half Width Rel %'] = ci\_width/2/popt\*100

ci width = ci[:,1]-ci[:,0]

# Create output dataframe param\_names = ['k','nA']

k, nA = popt

ssr = 0