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Review

Energizing the CO₂ utilization by chemo-enzymatic approaches and potentiality of carbonic anhydrases: A review



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ABSTRACT

During the last few decades, enormous emissions of greenhouse gases (GHGs) into the atmosphere by human activities, lead to global warming. Thus, it becomes essential to prevent the excessive emission or to develop new technologies to avoid successive accumulation of CO₂. Biological systems in nature have the capability to fix the atmospheric CO₂ but in the urban and industrially developed areas where a rate of CO₂ emission is very high, the biological system cannot capture and utilize the whole CO₂. Various chemicals and synthetic materials with CO₂ absorbing property are not eco-friendly or these are very expensive. Carbonic anhydrase (CA) is the fastest known enzymes containing zinc in its active site, convert CO2 to bicarbonate ions. It is one of a potent biological catalyst for CO2 conversion. Thus, in order to reduce the level of CO₂ the biocatalytic properties of microbial CA can be exploited. Literature survey showed that, more than fifty different microbial CAs have been explored for CO2 sequestration. The major advantages of CA to sequester CO₂ are economic viability and carbonation of CO₂ at a low concentration. Despite the higher rate of catalysis, the stability of CA is a major challenge for its industrial application. These difficulties have been partly solved by immobilizing the CA onto the bio-inspired surface, biochar, alginate, polyurethane foam and variety of nano-textured materials. A combination of enzyme and material which jointly capture and convert the CO2 into either carbon-rich compound of economic value or reduced carbon derivatives will plausibly energize the CO₂ utilization. In this review, we discussed the recent advances in chemical and materials used for CO2 capture, their advantages and limitations, utilization of microbial CA for CO₂ conversion, and its various applications.

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Contents

1.	Introduction	, 2
2.	Material collection and research analysis	. :
3.	Potential of microbial CA in CO ₂ conversion	. :
	Catalytic mechanism and properties of CA	
	Chemicals and materials facilitating the CO ₂ capture	
	5.1. Chemicals for CO ₂ capture	
	5.2. Materials for CO_2 capture	
	5.3. Hybrid systems for CO ₂ capture	
	5.4. Natural/biological system for CO_2 capture	
6.	Modification methods of CA for improved CO ₂ capture and conversion	
	6.1. Directed evolution and protein engineering of CA	
	6.2. CA as an immobilized biocatalyst for CO ₂ conversion	
7.	Analysis of different CO ₂ capture technologies used in various countries	

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