



Review of modelling and simulation strategies for evaluating corrosive behavior of aqueous amine systems for CO₂ capture

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ABSTRACT

Corrosion is one of the critical problems for process operations and plant life. Amine-based post-combustion CO₂ capture systems are known to be corrosive. This is because of the absorption of CO₂ by aqueous amine solution, which makes the system acidic and generates oxidizing agents through ionization leading to corrosion. Corrosion increases with CO₂ loading, process temperature, presence of oxygen (O₂), occurrence of sulfur compounds, low pH, velocity of solution and formation of some degradation products. Conducting experimental investigations to determine the rate of corrosion at various parts of process plant under a wide range of operating parameters is difficult and costly. Mathematical models and simulations play an important role in estimating the rate of corrosion under wide range of operating parameters for such systems. This study highlights the process of corrosion for amine-based CO₂ capture systems and development of corrosion models with critical analyses on their applications. Study also highlights the outlook of models and simulations.

1. Introduction

Industrialization, specially the coal dependency for energy is not going anywhere soon. Matter of the fact is that at present there is a list of 10,000 retired, operating and planned coal units, producing a total of almost 3,000 GW in 95 countries (Evans and Rosamund, 2019). It is reported that China and India are the leading countries with installation of new coal-fired power plants in last decade. Installation of new coal-fired power plants are reported to further thicken the atmospheric carbon dioxide (CO₂) layer beyond the repair. Records of Mauna Loa Observatory indicate that very unfortunately we already have crossed the mark of 415 ppm of CO₂ in atmosphere in May 2019 (Observatory, 2019). CO₂ is one the major greenhouse gases and it tends to accumulate in atmosphere leading to global warming and climate change. This means that the need for a retrofit technology for CO₂ capture cannot be ignored.

History of use of amines for sour gas treatment goes back to 1930s (Roger, 1930, 1931). The technology uses amines to capture CO₂ and

sulfur dioxide (SO₂). The technology has already matured on sour natural gases treatment before its application for CO₂ capture from flue gases (Milton, 1963). Retrofitting option of post-combustion CO₂ capture technology makes it feasible for CO₂ capture from flue gases (Rochelle, 2009). However, process operating conditions differ in flue gas treatment than natural gas (Gao et al., 2012; Soosaiprakasam, 2007; Wattanaphan et al., 2013). The amine-based CO₂ absorption is a chemical process. Mass transfer takes place between the gas and the liquid phase. The tall absorber and desorber vessels are used for the absorption and desorption. The major equipment involved in amine-based absorption process are absorber also called scrubber, regenerator also called stripper and desorber, rich-lean solution cross heat exchanger, reboiler, overhead condenser and cooler. A typical amine-based absorption process is exhibited in Fig. 1.

The emitted flue gases from coal-fired power plants are cooled before their entry to the major process equipment to enhance absorption efficiency and lower the risk of escape and degradation of solvent. In the process, aqueous amine reacts with CO₂ to produce water soluble ions,

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