



A comprehensive review of metal corrosion in a supercritical CO₂ environment

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ARTICLE INFO

Keywords:

CCS
Supercritical CO₂
Corrosion
Effect factors

ABSTRACT

Carbon capture and storage (CCS) is the most effective way to reduce CO₂ emissions. In CCS, CO₂ transportation plays an important role for the transmission of CO₂ from the capturing facility to the location of permanent storage. In order to increase the CO₂ density and avoid two-phase flow, CO₂ needs to be transformed into the supercritical state. The supercritical CO₂ environment poses a threat to pipeline safety because it is a high pressure environment with gaseous impurities. In this paper, the influencing factors of pipeline corrosion under supercritical CO₂ environment and the current prevention and control methods for SC CO₂ corrosion are reviewed. To begin, the supercritical CO₂ corrosion environment and corrosion evaluation methods are introduced. Then, the effects of some factors, such as the water content, gaseous impurities, acids, alkalis, salts, temperature, pressure, flow rate, exposure angle, and pipeline steel, on corrosion are comprehensively reviewed. New research areas, such as the use of coatings and inhibitors, are also reviewed. Finally, future development directions for studying corrosion in a supercritical CO₂ environment are proposed.

1. Introduction

Recently, with the rapid development of industry, the combustion of chemical fuels, such as oil and coal, has produced large amounts of greenhouse gases, which have negative impacts on the global climate. With the increasing emphasis on global climate issues, how to reduce emissions of greenhouse gases, such as CO₂, is receiving increasing attention all over the world (Choi and Nešić, 2011; Sim et al., 2013b). “Safe and economical CO₂ capture and storage technology (CCS) is an important technology for solving global climate change” was proposed by the Global Climate UN – COP24 (Czernichowski-Lauriol et al., 2018). In response to this proposal, CCS is certain to be an important development direction (Bhave et al., 2017; Reiner, 2016).

The process of CCS involves the following three stages: capture of the CO₂ from the power plant or industrial process, transmission of the CO₂, and storage in a geological reservoir (Gibbins and Chalmers, 2008; Lilliestam et al., 2012; Pearson et al., 2013). In terms of the CO₂ transportation, the use of a pipeline is the most economical option. To increase the CO₂ density and avoid two-phase flow, the CO₂ transported through pipelines is typically compressed into a supercritical state (larger than 7.38 MPa, 31.1 °C) (Boot-Handford et al., 2014; Gale and Davison, 2003; Kruse and Tekiel, 1996; Liu et al., 2016; Sim et al.,

2013b; Vandeginste and Piessens, 2008). However, high pressure increases the solubility of CO₂ in water, resulting in the production of high concentrations of H₂CO₃ solution, which increases the corrosion rate. Moreover, other impurities such as H₂SO₄, HNO₃, HCl, SO_x and NO_x may also exist in the CO₂ gas. The extreme condition and various possible impurities pose serious challenges to pipeline transportation. The presence of these impurities reduces the solubility of water and lowers the pH of the aqueous phase, thereby altering the structure of the corrosion product film and further exacerbating the corrosion of the material. (Seevam et al., 2008). Substantial experimental and theoretical data are needed to aid in the corrosion protection of the supercritical CO₂ transport pipeline, especially regarding the effect of the interactions between NO₂ and other impurities on the evolution of localized corrosion.

The study of corrosion behaviour in a supercritical CO₂ environment has become a hot spot. (Kang et al., 2013; Kanniche et al., 2010; Merkel et al., 2010; Thiruvengkatachari et al., 2009). Based on the previous studies, in this paper, the research progress on corrosion in a supercritical CO₂ environment is comprehensively reviewed. The effects of different factors on the corrosion rate, corrosion products and corrosion progress in a supercritical CO₂ environment and the antiseptic methods in a supercritical CO₂ environment are categorically summarized. The

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<https://doi.org/10.1016/j.ijggc.2019.102814>

Received 18 March 2019; Received in revised form 11 July 2019; Accepted 17 August 2019

Available online 22 August 2019

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