



A review of N-functionalized solid adsorbents for post-combustion CO₂ capture

Xiayi (Eric) Hu^{a,b,1,*}, Libin Liu^{a,1}, Xiao Luo^{c,1}, Gongkui Xiao^d, Elenica Shiko^e, Rui Zhang^a, Xianfeng Fan^{b,*}, Yefeng Zhou^{a,*}, Yang Liu^f, Zhaogang Zeng^g, Chao'en Li^h

^a College of Chemical Engineering, Xiangtan University, Xiangtan 411105, PR China

^b Institute for Materials and Processes, School of Engineering, University of Edinburgh, Mayfield Road, Edinburgh EH9 3JL, UK

^c College of Chemistry and Chemical Engineering, Hunan University, Changsha 410082, PR China

^d Department of Chemical Engineering, University of Western Australia, 35 Stirling Hwy, Crawley, WA 6009, Australia

^e Department of Chemical Engineering, University of Bath, Claverton Down, Bath BA2 7AY, UK

^f Hunan BG Well-point Environmental Science & Technology Co., Ltd, Changsha 41000, PR China

^g Xiangtan Sepiolite Technology Co., Ltd, Xiangtan 411100, PR China

^h CSIRO Energy, 71 Normanby Road, Clayton North, Victoria 3169, Australia

HIGHLIGHTS

- First holistic review of N-functionalized solid adsorbents for CO₂ capture.
- The criteria of developing high-efficiency and inexpensive N-functionalized solid adsorbents were summarized.
- The opportunities and challenges of N-functionalized solid adsorbents were discussed.

ARTICLE INFO

Keywords:

NFSAs
CO₂ capture
Functionalization
Amine
Solid porous-based materials
Adsorption performance

ABSTRACT

Over the past decade, amine-loaded solid adsorbents for capturing CO₂ from power plants have been widely studied. Various nitrogen (N) sources have been used for this purpose, and the current range of adsorbents, referred to here as N-functionalized solid adsorbent (NFSAs), are the subject of this review. The main synthesis methods of NFSAs are described and recent progress in the field discussed. Criteria for improving NFSAs performance are highlighted with reference to a variety of solid supports, providing guidance on the selection of highly efficient, inexpensive adsorbents. A thorough assessment of adsorption mechanisms and factors influencing the adsorption process is given. The review concludes by exploring future research and development opportunities, as well as pathways for commercializing NFSAs.

1. Introduction

About 40% of anthropogenic CO₂ emissions are produced by fossil fuel-based power plants. To meet the targets to maintain global temperatures within 2 °C above pre-industrial levels by the end of the century, emissions need to significantly decrease [1,2]. There are three approaches for capturing CO₂ from power plants: pre-combustion [3,4], post-combustion [5–7], and oxy-fuel combustion [8,9]. Post-combustion capture, which adsorbs CO₂ after fossil fuel combustion using wet/dry sorbents, is the most popular industrial method [10,11].

To date, CO₂ absorption by aqueous amine of 20–30 wt% is still the

state-of-the-art for post-combustion CO₂ capture processes, mainly due to the high CO₂ affinity of amines [12–14]. However, its application is restricted by several drawbacks, such as high equipment corrosion [7], intensive energy requirement for regeneration [12], tendency to degrade [15], and foaming issues [16]. Of the proposed alternative technologies for post-combustion CO₂ capture, adsorption-based technology is the most promising, due to its potentially lower cost while avoiding the drawbacks of aqueous amine. Adsorbents are the cornerstones for all adsorption processes, and numerous porous solid adsorbents have been developed for CO₂ adsorption in post-combustion gas conditions. However, these materials suffer either from low CO₂

* Corresponding authors at: College of Chemical Engineering, Xiangtan University, Xiangtan 411105, PR China (X. (Eric) Hu).

E-mail addresses: xiayihu@xtu.edu.cn (X.E. Hu), X.Fan@ed.ac.uk (X. Fan), zhouyef@xtu.edu.cn (Y. Zhou).

¹ These authors contribute equally to this paper.