

## Sub: Science Practicals (Std. IX)

### Experiment No. 1

**Aim:** To find the pH of the given solutions using pH paper.

**Procedure:** Take small quantity of the given sample in a test tube. Dip the pH paper in the solution. Observe the colour developed and note the pH in the observation table. Repeat the procedure for other samples.

#### Observation Table

Sr. No	Sample	Colour Produced	Approximate pH
1	Dilute HCl	Red	Less than 7
2	Dilute solution of NaOH	Blue	More than 7
3	Water	No change	7
4	Dilute solution of $\text{NaHCO}_3$	Greenish blue	More than 7

#### Interference:

(1) Acid samples (with pH less than 7) → **Dilute HCl**

(2) Basic samples (with pH more than 7) → **Dilute solution of NaOH, Dilute solution of  $\text{NaHCO}_3$**

(3) Neutral samples (with pH = 7) → **Water**

## Experiment No. 2

**Aim:** To verify Ohm's law.

(To study the dependence of current (I) on potential difference (V) across a resistor and determine its resistance.)

**Apparatus:** A resistor, a rheostat, dry cell or battery eliminator, a plug key, a voltmeter, ammeter and connecting wires.

### Observation Table

Sr. No	P.D. in volt (V)	P.D. in mV	Current in mA	R (Resistance)= V/ I
1	1	1000	100	10 ohms
2	2	2000	200	10 ohms
3	3	3000	300	10 ohms
4	4	4000	400	10 ohms

### Inference / Conclusion

1. The potential difference is directly proportional to the current through the conductor.
2. As the potential difference increases, the current increases proportionally, confirming Ohm's law.

### Circuit Diagram

