

Element features

Table S1: 58 element features that are used to calculate compositional descriptors. We provide these feature information via our open-source software XenonPy [1]

Feature ID	Description	Unit	Reference
<code>atomic_number</code>	Atomic number		
<code>atomic_radius</code>	Atomic radius	pm	[2, 3]
<code>atomic_radius_rahm</code>	Atomic radius by Rahm et al.	pm	[4, 5, 3]
<code>atomic_volume</code>	Atomic volume	cm ³ /mol	[3]
<code>atomic_weight</code>	Atomic weight		[6, 7, 3]
<code>boiling_point</code>	Boiling temperature	K	[3]
<code>bulk_modulus</code>	Bulk modulus	GPa	[8]
<code>c6_gb</code>	C ₆ dispersion coefficient in a.u. (Gould & Bučko)	a.u.	[9, 10, 3]
<code>covalent_radius_cordero</code>	Covalent radius by Cordero et al.	pm	[11, 3]
<code>covalent_radius_pyykko</code>	Single bond covalent radius by Pyykko et al.	pm	[12, 3]
<code>covalent_radius_pyykko_double</code>	Double bond covalent radius by Pyykko et al.	pm	[13, 3]
<code>covalent_radius_pyykko_triple</code>	Triple bond covalent radius by Pyykko et al.	pm	[14, 3]
<code>covalent_radius_slater</code>	Covalent radius by Slater	pm	[2]
<code>density</code>	Density at 295K	g/cm ³	[3]
<code>dipole_polarizability</code>	Dipole polarizability	a.u.,	[15, 3]
<code>electron_negativity</code>	Pauling electronegativity		[16]
<code>electron_affinity</code>	Electron affinity	eV	[17, 18, 3]
<code>en_allen</code>	Allen's scale of electronegativity	eV	[19, 20, 3]
<code>en_ghosh</code>	Ghosh's scale of electronegativity		[21, 3]
<code>en_pauling</code>	Pauling's scale of electronegativity		[17, 3]
<code>first_ion_en</code>	First ionisation energy	eV	[17]
<code>fusion_enthalpy</code>	Enthalpy of fusion for elements at their melting temperatures	kJ/mol	[17]
<code>gs_bandgap</code>	DFT bandgap energy of T=0K ground state	eV	[22, 23]
<code>gs_energy</code>	DFT energy per atom (raw VASP value) of T=0K ground state	eV/atom	[22, 23]
<code>gs_est_bcc_latcnt</code>	Estimated BCC lattice parameter based on the DFT volume of the OQMD ground state for each element		[22, 23]
<code>gs_est_fcc_latcnt</code>	Estimated FCC lattice parameter based on the DFT volume of the OQMD ground state for each element	Å	[22, 23]
<code>gs_mag_moment</code>	DFT magnetic moment of T=0K ground state		[22, 23]
<code>gs_volume_per</code>	DFT volume per atom of T=0K ground state	Å ³ /atom	[22, 23]
<code>hhi_p</code>	HerfindahlHirschman Index (HHI) production values		[24]
<code>hhi_r</code>	HerfindahlHirschman Index (HHI) reserves values		[24]
<code>heat_capacity_mass</code>	Specific heat capacity at STP	J/mol-K	[17]
<code>heat_capacity_molar</code>	Molar heat capacity at STP	J/mol-K	[17]
<code>icsd_volume</code>	Volume per atom of ICSD phase at STP		[25, 26, 27]
<code>evaporation_heat</code>	Evaporation heat	kJ/mol	[3]

Table S1 continued

Feature ID	Description	Unit	Reference
heat_of_formation	Heat of formation	kJ/mol	[3]
lattice_constant	Lattice constant	Å	[3]
mendeleev_number	Mendeleev’s number		[28, 29, 3]
melting_point	Melting temperature	K	[3]
molar_volume	Molar volume	L/mol	[8]
num_unfilled	Number of unfilled valence orbitals		[16]
num_valance	Number of valence electrons		[16]
num_d_unfilled	Number of unfilled d valence orbitals		[16]
num_d_valance	Number of filled d valence orbitals		[16]
num_f_unfilled	Number of unfilled f valence orbitals		[16]
num_f_valance	Number of filled f valence orbitals		[16]
num_p_unfilled	Number of unfilled p valence orbitals		[16]
num_p_valance	Number of filled p valence orbitals		[16]
num_s_unfilled	Number of unfilled s valence orbitals		[16]
num_s_valance	Number of filled s valence orbitals		[16]
period	Period in periodic table		[3]
specific_heat	Specific heat at 20 °C	J/(g mol)	[3]
thermal_conductivity	Thermal conductivity at 25 °C	W/(m K)	[3]
vdw_radius	Van der Waals radius	pm	[17, 3]
vdw_radius_alvarez	Van der Waals radius according to Alvarez	pm	[30, 3]
vdw_radius_mm3	Van der Waals radius from the MM3 FF	pm	[31, 3]
vdw_radius_uff	Van der Waals radius from the UFF	pm	[32, 3]
sound_velocity	Velocity of sound	m/s	[8]
polarizability	Static average electric dipole polarizability	10 ⁻²⁴ cm ³	[17]

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