

Writing an Ultra-Simple Molecular Dynamics Program in Python

Source: <https://qiita.com/ulabiisutokyo/>

Assignment ~ Create or modify a Molecular Dynamics (MD) code

- 1.1. Confirm that the following programs (`crudemd1.py` to `crudemd3.py`) run correctly.
- 1.2. Try running the MD simulation after changing the initial conditions (initial positions, initial velocities, number of atoms, etc.).
- 1.3. In the original program (`crudemd3.py`), atoms that exit the simulation cell do not return. Modify the program so that atoms **bounce back** from the cell boundaries and return.

※ The length of one side of the simulation cell is $l/scl = 500/10 = 50 \text{ \AA}$.

0. Preparation ~ Create a File for Initial Atomic Configuration

- 0.1. Write the initial positions and velocities of three atoms in a file named `initial.d`.

```
#### initial.d
15 10 0 0
20 10 0 0
20 15 0 0
#### End of initial.d
```

- 0.2. Each line in `initial.d` contains four values: the x and y coordinates of an atom, and the x and y components of its velocity. These should be separated by a single half-width space.

- Notes:
 - Use only one space between values.
 - Do not insert a space at the beginning of each line.
 - Failing to follow these rules may cause errors during execution.
 - Also, `initial.d` must be placed in the same folder as the Python program being run.

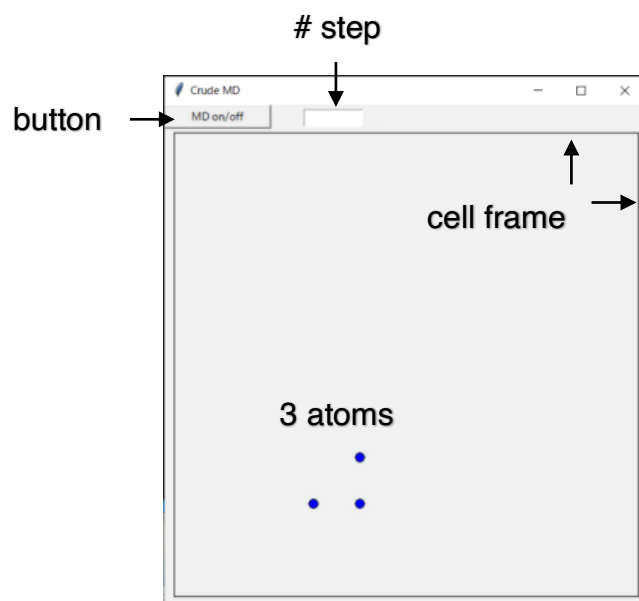


Figure 1 – Create the window as shown in the figure

1. Placing Atoms on the Canvas (`crudemd1.py`)

- 1.1. This step opens a canvas for drawing and reads the atomic positions and velocities from `initial.d`, placing the atoms at their initial positions.
- 1.2. An example program, `crudemd1.py`, is shown on the next page.
- 1.3. A dummy button is also included for later use (currently, pressing the button does nothing).
- 1.4. Since `initial.d` contains 3 lines, the number of atoms will be 3. You can increase the number of atoms by adding more lines.

```
#### crudemd1.py (anything after the # symbol is treated as a comment)
# ← In Python, # is considered as commented

import tkinter as tk

def dummy(event):          # Dummy button
    print('button is pressed')

def drawatom(x,y):         # Draw atom
    scl=10 # Scaling (magnification) factor
    rad=5  # Radius of sphere (atom)
    x1=x0+x*scl
    y1=y0+l-y*scl
    canvas.create_oval(x1-rad,y1-rad,x1+rad,y1+rad, fill='blue')

# creation of main window and canvas
win = tk.Tk()
win.title("Crude MD")
win.geometry("520x540")
x0=10 # Origin
y0=10 # Origin
l=500 # Size of canvas
canvas = tk.Canvas(win,width=l+x0*2,height=l+y0*2)
canvas.place(x=0, y=20)
canvas.create_rectangle(x0,y0,l+x0,l+y0)

# Declaration of arrays
rx = [] # Position of atom
ry = []
vx = [] # Velocity of atom
vy = []
fx = [] # Force on atom
fy = []

# button sample (dummy)
button_dummy = tk.Button(win,text="dummy",width=15)
button_dummy.bind("<Button-1>", dummy)
button_dummy.place(x=0,y=0)
```

```
# read initial position and velocity
f = open('initial.d', 'r')
i = 0
for line in f:
    xy = line.split(' ')
    rx = rx + [float(xy[0])]
    ry = ry + [float(xy[1])]
    vx = vx + [float(xy[2])]
    vy = vy + [float(xy[3])]
    print(rx)
    drawatom(rx[i],ry[i])
    i = i + 1

# Main Loop
win.mainloop()

#### End of crudemdl.py
```

2. Responding to Button Events and Moving Atoms (crudemd2.py)

2.1. This program extends the functionality of `crudemd1.py` with the following features:

- Detects button press events.
- When the button is pressed, all atoms begin moving diagonally upward in a continuous motion.

2.2. Note:

- This is **not yet a Molecular Dynamics (MD) simulation**, as the atoms are not moving due to interatomic forces or physical laws—only by a fixed directional displacement.

```
#### crudemd2.py

import tkinter as tk

def dummy(event):          # Button to switch on/off
    global imd              # to substitute the value into the variable
    if (imd == 0 ):
        imd = 1 # Run
    else:
        imd = 0 # Stop

def drawatom(x,y):         # Draw atom
    x1=x0+x*scl
    y1=y0+l-y*scl
    return canvas.create_oval(x1-rad,y1-rad,x1+rad,y1+rad, fill='blue')

def moveatom(obj,x,y):    # Move atom
    x1=x0+x*scl
    y1=y0+l-y*scl
    canvas.coords(obj,x1-rad,y1-rad,x1+rad,y1+rad)

# creation of main window and canvas
win = tk.Tk()
win.title("Crude MD")
win.geometry("520x540")
x0 =10 # Origin
y0 =10 # Origin
l =500 # Size of canvas
scl=10 # Scaling (magnification) factor
rad=5  # Radius of sphere

canvas = tk.Canvas(win,width=l+x0*2,height=l+y0*2)
canvas.place(x=0, y=20)
canvas.create_rectangle(x0,y0,l+x0,l+y0)

imd = 0 # MD on/off
```

```

# declaration of arrays
rx = [] # Position
ry = []
vx = [] # Velocity
vy = []
fx = [] # Force
fy = []
obj = []

# button sample (dummy)
button_dummy = tk.Button(win, text="dummy", width=15)
button_dummy.bind("<Button-1>", dummy)
button_dummy.place(x=0, y=0)

# Entry box (MD step)
entry_step = tk.Entry(width=10)
entry_step.place(x=150, y=5)

# read initial position and velocity
f = open('initial.d', 'r')
n = 0
for line in f:
    xy = line.split(' ')
    rx = rx + [float(xy[0])]
    ry = ry + [float(xy[1])]
    vx = vx + [float(xy[2])]
    vy = vy + [float(xy[3])]
    print(rx)
    obj = obj + [drawatom(rx[n], ry[n])] # obj[] holds pointer to object
    n = n + 1 # number of total atoms
print("number of atoms = ", n)

# Main Loop
step = 0
stepend = 1000

while step < stepend:
    if imd == 1:
        for i in range(n):
            rx[i] = rx[i] + 0.01
            ry[i] = ry[i] + 0.01
            moveatom(obj[i], rx[i], ry[i])
        step = step + 1
        entry_step.delete(0, tk.END)
        entry_step.insert(tk.END, step)
    win.update()

win.mainloop()

#### End of crudemd2.py

```

3. Toggle Molecular Dynamics with Button Press (`crudemd3.py`)

3.1. This program builds upon `crudemd2.py` and introduces actual Molecular Dynamics functionality:

- **Pressing the button toggles the MD simulation on and off.**
- Atom positions are updated based on **interatomic forces**, calculated using the **Morse potential**.
- Atom movement is computed using the **Verlet integration method**, a standard approach in MD simulations.

3.2. Note:

- This marks the transition to an actual MD simulation, where atoms interact physically rather than move arbitrarily.
- Please refer to Notes 1 and 2 at the end for details on the Morse potential and the Verlet method.

```
#### crudemd3.py

import tkinter as tk    # to create a window and a canvas
import math

def dummy(event):       # Switch MD on/off
    global imd           # to substitute the value into the variable
    if (imd == 0 ):
        imd = 1
    else:
        imd = 0

def drawatom(x,y):      # Draw atoms on canvas
    x1=x0+x*scl
    y1=y0+l-y*scl
    return canvas.create_oval(x1-rad,y1-rad,x1+rad,y1+rad, fill='blue')

def moveatom(obj,x,y):  # Move atoms on canvas
    x1=x0+x*scl
    y1=y0+l-y*scl
    canvas.coords(obj,x1-rad,y1-rad,x1+rad,y1+rad)

def v(rang):            # Morse potential
    ep = 0.2703 # epsilon [eV]
    al = 1.1646 # alpha [1/ang]
    ro = 3.253  # rho [ang]
    ev = 1.6021892e-19 # eV -> J (energy unit conversion)
    return ep*(math.exp(-2.0*al*(rang-ro))-2.0*math.exp(-al*(rang-ro))) *ev

def vp(rang):           # Derivative of Morse potential (= Force)
```

```

    ep = 0.2703
    al = 1.1646
    ro = 3.253
    ev = 1.6021892e-19
    return -2.0*al*ep*(math.exp(-2.0*al*(rang-ro))-math.exp(-al*(rang-ro)))
*ev*1.0e10

# creation of main window and canvas
win = tk.Tk()
win.title("Crude MD")
win.geometry("520x540")
x0=10    # Origin
y0=10    # Origin
l=500    # Size of canvas
scl=10   # Scaling (magnification) factor
rad=5    # Radius of sphere

canvas = tk.Canvas(win,width=l+x0*2,height=l+y0*2)
canvas.place(x=0, y=20)
canvas.create_rectangle(x0,y0,l+x0,l+y0)

imd = 0 # MD on/off

# declaration of arrays
rx = [] # Position [ang]
ry = []
vx = [] # Velocity [ang/s]
vy = []
fx = [] # Force [N]
fy = []
epot = [] # Potential energy [J]
obj = [] # Object (for visualization)
dt = 1.0e-16 # Time step [s]
wm = 1.67e-37 # Mass [1e-10 kg]

# button sample (dummy)
button_dummy = tk.Button(win,text="MD on/off",width=15)
button_dummy.bind("<Button-1>", dummy)
button_dummy.place(x=0,y=0)

# Entry box (MD step)
entry_step = tk.Entry(width=10)
entry_step.place(x=150,y=5)

# read initial position and velocity
f = open('initial.d', 'r')
n = 0
for line in f:
    xy = line.split(' ')
    rx = rx + [float(xy[0])]
    ry = ry + [float(xy[1])]
    vx = vx + [float(xy[2])]
    vy = vy + [float(xy[3])]
    fx = fx + [0]
    fy = fy + [0]
    epot = epot + [0]
    obj = obj + [drawatom(rx[n],ry[n])] # obj[] holds pointer to object

```



```

    n = n + 1 # number of total atoms
print("number of atoms = ",n)

# Main Loop of MD
step = 0
stepend = 100000

while step < stepend:
    if imd == 1:
        # Verlet(1)
        for i in range(n):
            rx[i] = rx[i] + dt * vx[i] + (dt*dt/2.0) * fx[i] / wm
            ry[i] = ry[i] + dt * vy[i] + (dt*dt/2.0) * fy[i] / wm
            vx[i] = vx[i] + dt/2.0 * fx[i] / wm
            vy[i] = vy[i] + dt/2.0 * fy[i] / wm
        # Force and energy
        for i in range(n):
            fx[i] = 0
            fy[i] = 0
            epot[i] = 0
        for i in range(n):
            for j in range(n):
                if (i != j):
                    rr = math.sqrt((rx[i]-rx[j])**2 + (ry[i]-ry[j])**2)
                    drx = rx[i] - rx[j]
                    dry = ry[i] - ry[j]
                    fx[i] = fx[i]-vp(rr)/rr*drx
                    fy[i] = fy[i]-vp(rr)/rr*dry
                    epot[i] = epot[i]+v(rr)/2.0
        # Verlet(2)
        for i in range(n):
            vx[i] = vx[i] + dt/2.0 * fx[i] / wm
            vy[i] = vy[i] + dt/2.0 * fy[i] / wm
            moveatom(obj[i],rx[i],ry[i])
        step = step + 1
        entry_step.delete(0,tk.END)
        entry_step.insert(tk.END,step)
    win.update()

win.mainloop()

#### End of crudemd3.py

```

Note 1: Morse Potential

The Morse potential is a type of interatomic potential function that defines the interaction between atoms. Its functional form is as follows: the potential energy $U(r_{ij})$ is determined based on the distance r_{ij} between a pair of atoms i and j .

$$U(r_{ij}) = \varepsilon \left[\exp(-2\alpha(r_{ij} - \rho)) - 2\exp(-\alpha(r_{ij} - \rho)) \right]. \quad (1-1)$$

When plotted, the Morse potential appears as shown in Figure 2. The parameters ε , ρ , and α are potential parameters corresponding to the **bond energy**, **equilibrium (most stable) distance**, and **spring constant** (curvature at the equilibrium point), respectively. By differentiating the potential U with respect to r_{ij} , the **magnitude of the force** f_{ij} exerted on atom i by atom j can be calculated. Since the force f_{ij} acts in the direction of \mathbf{r}_{ij} , its **direction is aligned** with the vector connecting the two atoms.

$$f_{ij} = -\frac{dU(r_{ij})}{dr_{ij}} = 2\alpha\varepsilon \left[\exp(-2\alpha(r_{ij} - \rho)) - \exp(-\alpha(r_{ij} - \rho)) \right], \quad (1-2)$$

$$\mathbf{f}_{ij} = f_{ij} \frac{\mathbf{r}_{ij}}{r_{ij}}. \quad (1-3)$$

The net force (resultant force) \mathbf{f}_i acting on atom i is the sum of \mathbf{f}_j for all atoms j .

$$\mathbf{f}_i = \sum_{j \neq i} \mathbf{f}_{ij}. \quad (1-4)$$

If the force \mathbf{f}_i acting on each atom is known, the atomic acceleration \mathbf{a}_i can be obtained using Newton's equation of motion: $\mathbf{f}_i = m_i \mathbf{a}_i$. By numerically integrating this equation, we can describe the motion of the atom—that is, how its **position** \mathbf{r}_{ij} and **velocity** \mathbf{v}_{ij} change over time. It is noteworthy that the **initial conditions** (initial position and velocity) must be explicitly provided beforehand. Since we are treating a **uniform system**, the mass m_i is assumed to be the **same for all atoms** i .

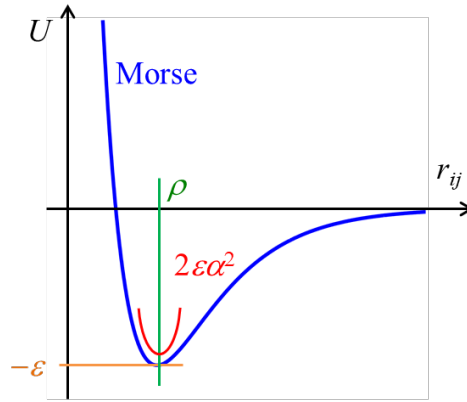


Figure 2 – Morse potential function form

Note 2: Verlet method

The Verlet method is a numerical solution method for differential equations. In molecular dynamics, it is used to calculate the time evolution of each atom's position and velocity. The purpose is to calculate $\mathbf{r}_i(t+\Delta t)$ and $\mathbf{v}_i(t+\Delta t)$ after a small amount of time Δt has passed from the position $\mathbf{r}_i(t)$ and velocity $\mathbf{v}_i(t)$ at time t .

$$\mathbf{r}_i(t + \Delta t) = \mathbf{r}_i(t) + \mathbf{v}_i(t)\Delta t + \frac{1}{2m_i}\mathbf{f}_i(t)(\Delta t)^2, \quad (2-1)$$

$$\mathbf{v}_i(t + \Delta t) = \mathbf{v}_i(t) + \frac{1}{2m_i}[\mathbf{f}_i(t) + \mathbf{f}_i(t + \Delta t)]\Delta t. \quad (2-2)$$

In the main loop of the program `crudemd3.py`, these two equations are used to update \mathbf{r}_i and \mathbf{v}_i at every step. Note that in the Eq. (2-1) for \mathbf{r}_i , the right-hand side only contains information about time t ($\mathbf{r}_i(t+\Delta t)$ can be determined using only the information about time t), whereas the Eq. (2-2) for $\mathbf{v}_i(t)$ uses both the force $\mathbf{f}_i(t)$ before time advances and the $\mathbf{f}_i(t+\Delta t)$ after time has advanced by Δt . Therefore, the program processes in the following order:

1. Update $\mathbf{r}_i(t)$ [Eq. (2-1), Verlet(1) in “`crudemd3.py`”]
2. Reflect the $\mathbf{f}_i(t)$ part of $\mathbf{v}_i(t)$ [Eq. (2-2), Verlet(1)]
3. Update $\mathbf{f}_i(t)$ ($t \rightarrow t+\Delta t$)
4. Reflect the $\mathbf{f}_i(t+\Delta t)$ part of $\mathbf{v}_i(t)$ [Eq. (2-2), Verlet(2)]

The Verlet method processing is divided into (1) and (2) in the program because the force \mathbf{f} needs to be updated in the middle of updating the velocity \mathbf{v} .

For reference, let us derive the formula for $\mathbf{r}_i(t+\Delta t)$. First, expand $\mathbf{r}_i(t+\Delta t)$ in terms of Δt .

$$\mathbf{r}_i(t + \Delta t) = \mathbf{r}_i(t) + \frac{d\mathbf{r}_i(t)}{dt}\Delta t + \frac{1}{2}\frac{d^2\mathbf{r}_i(t)}{dt^2}(\Delta t)^2 + \dots \quad (2-3)$$

By discarding the third-order and higher terms in this equation and using $\mathbf{v} = d\mathbf{r}/dt$, $\mathbf{a} = d^2\mathbf{r}/dt^2$ and $\mathbf{f} = m\mathbf{a}$, we obtain Eq. (2-1). Note that the derivation of Eq. (2-2) for \mathbf{v} is somewhat tricky, so we will omit it here (though if you are interested, feel free to give it a try).