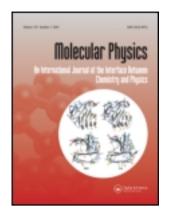
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INVITED TOPICAL REVIEW

Variational quantum approaches for computing vibrational energies of polyatomic molecules

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In this article, we review state-of-the-art methods for computing vibrational energies of polyatomic molecules using quantum mechanical, variationally-based approaches. We illustrate the power of those methods by presenting applications to molecules with more than four atoms. This demonstrates the great progress that has been made in this field in the last decade in dealing with the exponential scaling with the number of vibrational degrees of freedom. In this review we present three methods that effectively obviate this bottleneck. The first important idea is the *n*-mode representation of the Hamiltonian and notably the potential. The potential (and other functions) is represented as a sum of terms that depend on a subset of the coordinates. This makes it possible to compute matrix elements, form a Hamiltonian matrix, and compute its eigenvalues and eigenfunctions. Another approach takes advantage of this multimode representation and represents the terms as a sum of products. It then exploits the powerful multiconfiguration Hartree time-dependent method to solve the time-dependent Schrödinger equation and extract the eigenvalue spectrum. The third approach we present uses contracted basis functions in conjunction with a Lanczos eigensolver. Matrix vector products are done without transforming to a direct-product grid. The usefulness of these methods is demonstrated for several example molecules, e.g. methane, methanol and the Zundel cation.

Keywords: molecular vibrations; quantum methods; methane; methanol; Zundel

1. Introduction

The theory and computational modelling of molecular vibrations continues to be a central topic of interest in chemistry and in many related fields. In the broadest terms there are two aspects to this field. They are the molecular Hamiltonian, the sum of the kinetic energy operator (KEO), and the potential energy surface (PES), and the extraction of the vibrational information, i.e. eigenvalues, eigenfunctions, spectra, etc. from the Hamiltonian. This review will focus exclusively on so-called *ab initio* approaches to both aspects. That is, we will consider PESs that have been obtained from or heavily based on ab initio electronic energies and we will only consider quantum mechanical approaches to the vibrational dynamics that are based on variational theory as opposed to perturbation theory. Examples of applications will be given to molecules with more than four atoms to indicate the current state of the art.

In order to put this review in context it is useful to begin with a short and highly selective history of this quantum approaches to vibrational dynamics beginning in roughly the mid 1970s. It was at this time that variational approaches to obtain vibrational energies

were undertaken using the rigorous Watson Hamiltonian in mass-scaled rectilinear normal modes [1]. Whitehead and Handy's [2] seminal work on triatomic molecules using this approach is certainly noteworthy. They used a direct-product representation of the wavefunction and employed a quartic force field for the potential. The approach requires matrix elements of the Hamiltonian in this basis and multi-dimensional quadratures over the potential. (Quartic force fields originated much earlier in the field in the context of standard Rayleigh–Schrödinger second-order perturbation theory and were very much in focus during the 1970s and 1980s. Their calculation from *ab initio* electronic energies was and continues to be an important aspect of the field [3,4].)

Given any Hamiltonian describing the nuclear motion of f modes the direct-product representation of the f-mode wavefunction is given by $\{\phi^{(1)}(1)\} \times \{\phi^{(2)}(2)\} \times \cdots \times \{\phi^{(f)}(f)\}$. Each set of basis functions should be complete and orthonormal in the space of the mode. Typically they are eigenfunctions of a simple 1D Hamiltonian, e.g. a harmonic or Morse oscillator Hamiltonian, free-rotor Hamiltonian, etc.

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Without any further restrictions the size of the basis, and hence the order of the Hamiltonian matrix, is $O(\text{nbasis}^f)$ where nbasis is the size of each single-mode basis. A typical value of nbasis is 10 and therefore, if direct linear algebra methods are used, the basis size, $O(10^f)$, limits this approach to f=3, i.e. triatomic molecules. Quadratures of the potential in this basis also show exponential scaling of $O(10^f)$. (Actually more quadrature points per degree of freedom are needed than the maximum quanta in the basis to obtain accurate quadratures.) This approach is now referred to as the *Finite Basis Representation* (FBR).

A direct-product, grid-based representation of the wavefunction, known as the *Discrete Variable Representation* (DVR), eliminates the need to perform multi-dimensional quadratures over the potential (since in this representation the potential matrix is diagonal). However, to achieve a comparable accuracy to an FBR more grid points than basis functions are generally required. (Note that FBR also has a different and very specific meaning in the context of the DVR, where the two representations are related to each other by a unitary transformation. Our usage of FBR above does not make reference to a DVR and we use FBR here with this more general meaning.)

Even with some method to contract the size of the direct-product basis/grids, e.g. using a maximum sum of quanta or eliminating grid points where the potential exceeds some maximum value, the scaling of this representation of the wavefunction remains essentially exponential in the number of degrees. For non-routine cases, e.g. isomerisation, very high energy states, nbasis must be larger than 10, at least for some degrees of freedom, and restricting single-mode excitations may not be possible. In addition for these more challenging situations the choice of coordinates and thus the form of the Hamiltonian becomes a key consideration. These elements of the variational approach will recur throughout this review.

With respect to different choices of coordinates we note the pioneering variational calculations of Carter and Handy and co-workers using valence coordinates for triatomic molecules [5] and special classes of tetraatomic molecules [6,7]. A major aspect of this work was the derivation of the KEOs, which are quite complex. This effort resulted in two FBR-based codes RVIB3 (triatomics) and RVIB4 (tetraatomics) that are available through the CCP6 web site [8].

There are several useful alternatives to direct-product representations of the wavefunction. One is based on a Hartree representation of the wavefunction, followed by CI methods analogous to those used in electronic structure theory. The development of Vibrational Self-Consistent field (VSCF) and Virtual

space CI (VCI) approaches [9–12] (see also Section 7.4) using global or semi-global potential energy surfaces (not force fields) were reported, both in normal and curvilinear (Jacobi) coordinates for triatomic molecules. One of the earliest examples of this approach was to HCO where a global PES was used to obtain vibrational energies using the FBR initially in normal coordinates [13] and later in Jacobi coordinates [14], and where excellent agreement, as expected, from the two sets of calculations was seen. Also in the 1980s VSCF/VCI methods were used in a code, POLYMODE [15], which was based on the Watson Hamiltonian but which explicitly made use of quartic force fields in normal modes. This approach is quite attractive as matrix elements of the potential are analytical. This approach with many refinements has been further developed quite recently into a very useful computational tool [16].

Another very powerful alternative to the direct-product representation is a contraction method based on truncation-recoupling. In this approach eigenfunctions of subspace Hamiltonians are used to form multidimensional bases for final diagonalisation. Carter and Handy were perhaps the first to develop these ideas for triatomic molecules in the FBR. Bowman and Gazdy [17] developed a more sophisticated version of this approach in the FBR (which is in the spirit of Light and co-workers' version of truncation-recoupling in the DVR, described briefly below) where effective Hamiltonians for two degrees of freedom are obtained by averaging the full Hamiltonian over each basis function for the third degree of freedom.

For three degrees of freedom the simple version of this strategy is as follows. Let the full Hamiltonian be decomposed as

$$H = H_{2D} + H_{1D} + V_{c}(1, 2, 3),$$

$$H_{2D} = T_{1} + T_{2} + V(1, 2, 0),$$

$$H_{1D} = T_{3} + V(0, 0, 3),$$

$$V_{c}(1, 2, 3) = V(1, 2, 3) - V(1, 2, 0) - V(0, 0, 3), \quad (1)$$

where T_i are the kinetic energy operators (assumed separable) and V is the full potential. Let the 2D eigenfunctions of $H_{\rm 2D}$, denoted $\psi_m^{\rm 2D}(1,2)$, be obtained from a standard direct-product basis of say dimension $n_1 \times n_2$ and let $\phi_n(3)$ be the eigenfunctions of $H_{\rm 1D}$. Then the truncated 3D basis is given by $\{\psi^{({\rm 2D})}(1,2)\} \times \{\phi^{(3)}(3)\}$, where the number of 2D basis functions is much less than $n_1 \times n_2$ and thus the dimension of this basis is much less than that of the underlying direct-product basis. A further advantage is that the 2D eigenfunctions contain two-mode correlation and are therefore a very good basis especially for

cases where the third mode may be the least coupled of the three modes.

For a number of very challenging problems, perhaps best exemplified by the HCN/HNC isomerisation, neither normal coordinates nor valence coordinates are suitable to describe this large-amplitude motion. Even with better suited (atom-diatom Jacobi) coordinates the standard variational approaches in use in the 1980s were not able to obtain well-converged energies for this system. Light and co-workers [18-20] introduced two seminal ideas that conquered this problem and greatly advanced the field. The first was so the so-called Discrete Variable Representation (DVR) of the Hamiltonian, based on a compact underlying quadrature scheme, and the second was the use of a sequential diagonalisation in a subspace followed by a truncation-recoupling for the final diagonalisation step. (In passing we note that an FBR using a 'moveable basis' [21] was applied successfully to HCN/HNC isomerisation using an accurate PES [22].) A code based on the DVR approach for triatomic molecules, DVR3D, written by Tennyson and co-workers is available at the CCP6 web site [8].

This general truncation-recoupling strategy has largely been responsible for the success in the 1990s and to the present in doing variational calculations of the vibrational energies of tetraatomic molecules. Perhaps one of the most notable successes of this strategy was to the challenging acetylene/vinylidene isomerisation. In two approaches Jacobi coordinates in diatom-diatom C₂-H₂ coordinates were used in order to make maximum use of the permutation-inversion symmetry [23,24]. Also the KEO is relatively simple in these coordinates. In the work of Zou et al. [23] a 4D basis in the three angular variables plus R (the distance between the C₂ and H₂ centres of mass) was combined with a 2D basis in the remaining diatom distances $r_{\rm CC}$ and $r_{\rm HH}$. This challenging system was also tackled with this and two other sets of curvilinear coordinates [25] in order to illustrate the differences in efficiency of the various coordinates for the different classes of eigenstates, i.e. acetylene-like, vinylidene-like and totally delocalized states. A code from this group [25] for tetraatomic molecules, WAVR4, is available at the CCP6 web site [8].

There are important differences between the contraction scheme of Carter and Handy and that of Light and co-workers (see Section 5). In the scheme of Carter and Handy the contracted basis is obtained by computing eigensolutions of a small number of reduced dimensional Hamiltonians equal to the number of groups of coordinates. In the scheme of Light and co-workers the number of reduced Hamiltonians for which one must obtain eigensolutions is much larger

and depends on the number of primitive basis functions. The scheme of Light *et al.* is not practical for moleules with more than four atoms. Computing and storing eigenvectors of all the reduced Hamiltonians becomes unpractical (see Section 5).

Contracted basis functions are one way of dealing with the $nbasis^f$ basis size problem emphasized above. Iterative eigensolvers are another. Although for a fouratom molecule it is impossible to use a nbasis^t direct product basis with a standard diagonalisation method, it is possible to use it with an iterative eigensolver. See Section 4. Iterative eigensolvers enable one to compute a spectrum without storing or even computing Hamiltonian matrix elements. Iterative does not mean that energy levels are computed one by one, rather it means that they are computed from a series of matrix-vector products. Because matrix-vector products can be evaluated without first computing Hamiltonian matrix elements there is no need to do multi-dimensional integrals. Not surprisingly, it is advantageous to use both contracted basis functions and iterative eigensolvers but to do so efficiently one must employ a good scheme for doing the matrixvector products.

Finally, and although it is not the focus of this review, we note again that second-order perturbation theory (PT), which has a longer history than variational theory, has been widely applied and developed with the use of quartic force fields [26]. Extensions of PT beyond second order have also been developed and have been widely applied, for example by Sibert and co-workers [27,28]. Also, PT based on the VSCF Hamiltonian has been developed by Gerber and co-workers [29,30] and further developed and applied by Christiansen [31–33], Benoit and co-workers [34–36] and Barone [37].

Our discussion so far has been focused on the time-independent picture. We now will briefly discuss the time-dependent approach as well. People became aware of the power of the time-dependent approach largely through the seminal papers of Heller [38,39], (although the work of McCullough and Wyatt [40,41] should also be mentioned). Heller's approach is very approximate but it made the time-dependent picture popular and shortly after the appearance of his papers new techniques, like integrators dedicated to the time-dependent Schrödinger equation [42], were developed. The excellent review of Kosloff [43] reflects the state of the art of accurate time-dependent methodologies 20 years ago.

The domain of time-dependent methods were mainly scattering or half-scattering (e.g. photodissociation) problems. Time-dependent methods also proved to be very useful for simulating low-resolution spectra, but the generation of high resolution spectra – the topic of the present review – required long propagation times making the time-dependent approach less efficient. The situation improved somewhat with the advent of filter diagonalisation [44–49]. However, the calculation of eigenenergies seemingly remained the natural domain of methods based on the time-independent Schrödinger equation.

One of the advantages of the time-dependent approach becomes apparent when representing multidimensional wavefunctions. A time-dependent wavepacket, at each instant of time, is in general less structured and hence more easily approximated by parametrized functions or basis sets than a timeindependent eigenstate. For example, it is well known that time-dependent Hartree provides in general better approximate eigenenergies than time-independent Hartree (also known as VSCF). The multiconfiguration time-dependent Hartree (MCTDH) method, discussed in Section 7, is a powerful way to solve the time-dependent Schrödinger equation efficiently. For larger systems its high efficiency can compensate for the weakness of the time-dependent approach in computing vibrational energies. For the Zundel cation, $H_5O_2^+$, this is shown in Section 8.2.

A review of recent progress up to the present is the subject of the following sections. We begin with a review of representations of the potential and the kinetic energy operator in Sections 2 and 3, with a focus on molecules containing more than four atoms. We briefly describe eigensolvers in Section 4, followed by three sections describing approaches and codes based on variational solutions of the time-independent and timedependent Schrödinger equation. Section 5 is focused on contracted iterative techniques for solving the time-independent Schrödinger equation based on polyspherical coordinates. Section 6 describes the approach based on the Watson Hamiltonian, the *n*-mode representation of the potential and the VSCF/ VCI theory, already mentioned; however, it uses a new representation of the potential reviewed in the following section. The third approach, based on the time-dependent Schrödinger equation, is the Multi-Configuration-Time-Dependent Hartree method which is described in Section 7. Section 8 presents examples of applications of these methods to challenging molecules of more than four atoms. We conclude with a summary and short comments on possible future directions for the field.

2. Representations of the potential

If the number of vibrational coordinates is 6 (or less) it is practical to do full-dimensional quadratures of the potential, as is often done in the FBR, to compute the

vibrational energies. It is also practical for this number of coordinates to store the potential on a 6D direct product grid and use a product (FBR or DVR) approach that avoids full-dimensional quadratures (see for example [50–56]). For larger molecules the dimensionality of the quadratures and the size of the product grid make these approaches currently impractical. Solutions to these two problems have been proposed. One, which addresses the quadrature problem, is to represent the potential in a fashion so that only quadratures of lower dimension than the full dimension are required. The other, which addresses the grid size problem, is to compute a reduced matrix, called **F**, as explained in detail in Section 5.

This section reviews two methods of representing the potential. The first (see Section 2.1) is the so-called *n*-mode representation. This general representation for use in vibrational calculations was proposed in 1997 [57] and in independent work in a more general context in 1998 [58]; the restricted two-mode version appeared in 1996 [59] and was used in the context of second-order perturbation theory. The second (see Sections 2.2–2.6) is a product representation of the potential. Such product forms, which may be generated by the *potfit* algorithm, are extensively used in MCTDH.

2.1. n-mode representation

In the *n*-mode representation (*n*MR) the potential $V(\mathbf{Q})$ is given by

$$V(\mathbf{Q}) = V^{(0)} + \sum_{i} V_{i}^{(1)}(Q_{i}) + \sum_{ij} V_{ij}^{(2)}(Q_{i}, Q_{j}) + \sum_{ijk} V_{ijk}^{(3)}(Q_{i}, Q_{j}, Q_{k}) + \cdots,$$
(2)

where the one-mode representation of the potential contains only $V_i^{(1)}(Q_i)$ terms, which are the cuts through the hyperspace of normal coordinates with just one coordinate varying at a time, the two-mode representation contains those terms plus the $V_{ii}^{(2)}(Q_i,Q_i)$ terms, etc. Each of the terms in each summation is an *intrinsic p*-mode potential, because they each vanish if one of the variables is zero. However, if full two- or three-mode grids are used the above representation can still be used provided the overcounting of lower-dimensional grids embedded in higher dimensional ones is accounted for [57]. In order for this representation to be useful it must be truncated at a value of n less than the total number of modes. In the original implementation of this representation in the code MULTIMODE [57], which uses VSCF/VCI methods to obtain eigenvalues and eigenfunctions of the exact Watson Hamiltonian, the maximum value of n was 4; the current maximum value in the code is 6.

The *n*-mode representation has a number of useful features. First, matrix elements of the potential can be done relatively easily for up to a 6MR of the full potential. Second, each *p*-mode potential can be of arbitrary polynomial order thus eliminating the issues of negative divergences of say a quartic force field in the limit of large values of normal coordinates. Another feature of this representation of the potential is that convergence of vibrational energies obtained with increasing *n*MR can be systematically monitored. This important aspect of the *n*MR of the potential will be demonstrated in Section 8.1 for CH₄.

Several approaches have been taken to obtain these p-mode potentials. If one has a global or semi-global PES they can be obtained directly from it. However, they can also be obtained directly from ab initio calculations. For two-mode potentials this is feasible, but for larger p-mode potentials this becomes problematic as the number of grid points grows rapidly. For example even for tetraatomic molecules there are 20 three-mode grids and if one needed say eight electronic energies per mode this would amount to 10,000 energies. For the 15 four-mode grids 61,000 electronic energies would be needed which is bordering on impractical. There are ways to mitigate this bottleneck. One is to use sparse data on the various grids and then to interpolate the energy on each p-mode grid. For example if one uses five energies per mode the above numbers become 2500 and 9375 for three- and fourmode grids. This strategy was proposed by Carter et al. [60] who suggested least-squares polynomial fitting or multi-dimensional spline intepolation as well as a less well-known approach which makes explicit use of gradient data, namely reduced Hermite interpolation. Even using interpolation the number of electronic energies needed for three-mode grids can be quite large for molecules larger than tetraatomics. One effective so-called 'multi-resolution' approach to deal with this has been proposed. In this approach lower levels of ab initio theory are used for say the three-mode grids compared to the two-mode ones [61,62]. Another strategy is to consider only a subset of p-mode potentials, as was done in recent applications to the 12-atom molecule n-methyl acetamide [63,64].

Defining the mode terms to minimize the error in all of configuration space is another technique to obtain the *p*-mode potentials. Using the mode terms defined after Equation (2) is equivalent to minimizing the error along 1D slices, 2D surfaces etc. Instead, it is possible to use a random sampling high dimensional model representation [65]. In this case the different terms are not associated with different grids but the

error is minimized for a set of points drawn from some suitable probability distribution and all mode terms are fit from a single set of ab initio points. For example, one can choose the probability distribution to favour low-lying regions of the potential and use neural networks (NNs) to fit the mode terms. Details are given in [66]. Beginning with the one-mode term, the mode terms are determined sequentially. A fit for the s-mode term is built by fitting a NN to the difference of the potential and the sum of the previously determined mode terms. Note that unlike the original approach of [67] the mode terms can be determined without doing the multi-dimensional integrals by Monte Carlo methods. For HOOH, using this approach we are able to determine all of the four-mode terms from about 10,000 points. Levels computed on the surface are good to about $1 \,\mathrm{cm}^{-1}$.

We conclude this subsection by noting that the *n*-mode representation can also be applied to any multi-dimensional function. In particular it has been applied to the components of the dipole moment, which is needed in the calculation of the IR spectrum. This has been implemented in MULTIMODE, where matrix elements of a property can be evaluated, using an *n*-mode representation of that property [68].

2.2. Separable representation of potential: the product form

As noted already, the set-up of the Hamiltonian matrix can become a major part of the work to be performed when investigating molecular vibrations. Multi-dimensional integrals have to be done and when there are f=6 or more degrees of freedom (DOF) then these integrals are difficult to perform. The computation of the matrix elements of \hat{H} is considerably simplified if the Hamiltonian is of product form, i.e. if it can be written as

$$\hat{H} = \sum_{r=1}^{s} c_r \prod_{\kappa=1}^{f} \hat{h}_r^{(\kappa)}, \tag{3}$$

where $\hat{h}_r^{(\kappa)}$ operates on the κ th DOF only and where c_r is a number. Then multi-dimensional integrals can be written as a sum of products of one-dimensional integrals,

$$\left\langle \Phi_{J} \middle| \hat{H} \middle| \Phi_{L} \right\rangle = \sum_{r=1}^{s} c_{r} \left\langle \chi_{j_{1}}^{(1)} \middle| \hat{h}_{r}^{(1)} \middle| \chi_{l_{1}}^{(1)} \right\rangle \dots \left\langle \chi_{j_{f}}^{(f)} \middle| \hat{h}_{r}^{(f)} \middle| \chi_{l_{f}}^{(f)} \right\rangle. \tag{4}$$

Here $\chi_j^{(\kappa)}$ denotes the *j*th basis function of the basis for the κ th DOF. Often these basis functions are chosen as DVR functions, which simplifies the evaluation of matrix elements of local operators.

The multi-dimensional basis functions (configurations) Φ_J are products of the one-dimensional ones, i.e. we assume a direct product basis. The composite index $J = (j_1, \dots, j_f)$ collects all indices of one configuration.

As one-dimensional integrals are done very fast, the computation of the Hamilton matrix elements according to Equation (4) is much faster than doing multi-dimensional integrals. The storage requirements are also much smaller. However, one may ask how realistic it is to assume a Hamiltonian of product form. Well, kinetic energy operators (KEO) are mostly of product form. In particular when using polyspherical coordinates, the KEO is of product form. The potential energy surface (PES), however, is often not of product form, although a Taylor expansion of the potential as well as many model potentials are of product form. Hence a polynomial fit of the potential – when done in the same coordinates as the dynamical calculation - is of desired product form. In the following we will discuss a method, called potfit [69–71], which allows one to transform a general PES to product form, while controlling the error which is introduced by this re-fitting procedure.

Before we discuss *potfit* we remark that several methods for calculating vibrational energies, e.g. RVIB4 [7,8], rely on product form. For MCTDH, where the integrals have to be performed at every timestep, the product form is vital (except when using the CDVR method of Manthe [72]). Hence *potfit* was developed for MCTDH, but it can certainly be useful for other quantum dynamical methods as well.

2.3. The POTFIT algorithm

A direct way to the product form is an expansion in a product basis. Hence we approximate some given potential V by

$$V^{\text{app}}(q^{(1)}, \dots, q^{(f)})$$

$$= \sum_{j_1=1}^{m_1} \dots \sum_{i_r=1}^{m_p} C_{j_1 \dots j_r} v_{j_1}^{(1)}(q^{(1)}) \dots v_{j_r}^{(f)}(q^{(f)}), \qquad (5)$$

where $q^{(\kappa)}$ is the coordinate of the κ th DOF. The basis functions $v_{j_{\kappa}}^{(\kappa)}(q^{(\kappa)})$ are called *single-particle potentials* (SPP). The expansion orders, m_{κ} , must be chosen large enough to achieve an accurate expansion. On the other hand they should be as small as possible, because the numerical effort of the integral calculation scales linearly with the number of potential terms, i.e. with the product of the expansion orders. Hence both the expansion coefficients and the SPPs should be optimized to provide the best approximate potential for a given set of expansion orders.

Before we turn to analyse this optimisation problem, we simplify the problem somewhat. When DVRs are used to represent the wavefunctions, one needs to know the potential only at grid points. This allows us to work in finite dimensional discrete vector spaces. The full potential is now represented by a tensor

$$V_{i_1...i_f} = V(q_{i_1}^{(1)}, \dots, q_{i_f}^{(f)}),$$
 (6)

where $q_i^{(\kappa)}$ denotes the coordinate of the *i*th grid point of the κ th grid. The approximate tensor is written as

$$V_{i_1...i_f}^{\text{app}} = \sum_{i_1=1}^{m_1} \dots \sum_{i_r=1}^{m_f} C_{j_1...j_r} v_{i_1j_1}^{(1)} \dots v_{i_rj_r}^{(f)}, \tag{7}$$

with $v_{ij}^{(\kappa)} = v_j^{(\kappa)}(q_i^{(\kappa)})$. The SPPs are assumed to be orthonormal on the grid, $\sum_i v_{ij}^{(\kappa)} v_{i\ell}^{(\kappa)} = \delta_{j\ell}$. Throughout this section we will use the letters i and k to label grid points and j and ℓ to label SPPs.

The task is now to determine optimal coefficients and SPPs. To this end we minimize

$$\Delta^{2} = \sum_{i_{1}=1}^{N_{1}} \dots \sum_{i_{f}=1}^{N_{f}} \left(V_{i_{1}\dots i_{f}} - V_{i_{1}\dots i_{f}}^{\text{app}} \right)^{2} = \sum_{I} (V_{I} - V_{I}^{\text{app}})^{2},$$
(8)

where I denotes a composite index which runs over all grid points. Minimizing Δ^2 by varying only the coefficients yields

$$C_{j_1...j_f} = \sum_{i_1=1}^{N_1} \dots \sum_{i_r=1}^{N_f} V_{i_1...i_r} v_{i_1j_1}^{(1)} \dots v_{i_rj_r}^{(f)}, \tag{9}$$

i.e. the coefficients are given by overlap.

More difficult is to find optimal SPPs. Within potfit one defines the SPPs as eigenvectors of the potential density matrices

$$\varrho_{kk'}^{(\kappa)} = \sum_{I}^{\kappa} V_{i_1 \dots i_{\kappa-1} k i_{\kappa+1} \dots i_f} V_{i_1 \dots i_{\kappa-1} k' i_{\kappa+1} \dots i_f}, \qquad (10)$$

where the upper index κ at the summation symbol indicates that the summation is not over the κ th DOF. The eigenvectors are ordered according to their eigenvalues, i.e. one neglects SPPs with small eigenvalue and hence small contribution to the potential. This approach is known to be optimal for the two-dimensional case [73]. For the general f-dimensional case (f > 2), potfit is not optimal but provides SPPs which are close enough to the optimal ones to be useful. Fully optimal SPPs can be found – in principle – by minimizing Δ^2 , Equation (8), with the aid of a general minimisation procedure, e.g. simulated annealing. But such a procedure is much too costly.

The error of the approximate potential is checked by comparing it with the exact potential on the full product grid. But a simple estimate of the error based on the eigenvalues, λ_j , of the potential density matrices can also be given.

$$\Delta^2 \le \sum_{\kappa=1}^f \sum_{i=m_{\nu}+1}^{N_{\kappa}} \lambda_j^{(\kappa)}. \tag{11}$$

This estimate is very useful for determining appropriate expansion orders. It also shows that the L^2 error vanishes uniformly with increasing expansion orders. If all $m_{\kappa} = N_{\kappa}$ one recovers the *exact* potential on the grid points.

2.4. Contraction over one degree of freedom

The number of expansion terms, $s = \prod_{\kappa=1}^f m_{\kappa}$ appearing in Equation (7), should be as small as possible, because this number determines the effort of the integral calculation using this product-form potential. At virtually no cost one can reduce the number of expansion terms by one expansion order m_{κ} . To this end we define *contracted expansion coefficients*

$$D_{j_1...j_{\kappa-1}i_{\kappa}j_{\kappa+1}...j_f}^{(\kappa)} = \sum_{j_{\kappa}=1}^{m_{\kappa}} C_{j_1...j_f} v_{i_{\kappa}j_{\kappa}}^{(\kappa)}.$$
 (12)

As the expansion order m_{κ} will no longer appear in the working equation for $V^{\rm app}$, one may set $m_{\kappa} = N_{\kappa}$ and use the full set of SPPs for this particular DOF. In this case one performs a unitary transformation on the κ th index of the potential to obtain the coefficient C, and then performs the inverse unitary transformation on the κ th index of the coefficient. Hence effectively there is no transformation on the κ th index and $D^{(\kappa)}$ is conveniently computed as

$$D_{j_{1}...j_{\kappa-1}i_{\kappa}j_{\kappa+1}...j_{f}}^{(\kappa)} = \sum_{I}^{\kappa} V_{i_{1}...i_{f}} v_{i_{1}j_{1}}^{(1)} \cdots v_{i_{\kappa-1}j_{\kappa-1}}^{(\kappa-1)} v_{i_{\kappa+1}j_{\kappa+1}}^{(\kappa+1)} \cdots v_{i_{f}j_{f}}^{(f)}.$$
(13)

Note that the κ th potential density matrix and the C-tensor are no longer needed. Using contraction, the approximate potential is written as

$$V_{i_{1}...i_{f}}^{\text{app}} = \sum_{j_{1}=1}^{m_{1}} \dots \sum_{j_{\kappa-1}=1}^{m_{\kappa-1}} \sum_{j_{\kappa+1}=1}^{m_{\kappa+1}} \dots \sum_{j_{f}=1}^{m_{f}} D_{j_{1}...j_{\kappa-1}i_{\kappa}j_{\kappa+1}...j_{f}}^{(\kappa)} \times v_{i_{1}j_{1}}^{(1)} \dots v_{i_{\kappa-1}j_{\kappa-1}}^{(\kappa-1)} v_{i_{\kappa+1}j_{\kappa+1}}^{(\kappa+1)} \dots v_{i_{f}f}^{(f)}.$$

$$(14)$$

This contraction over the κ th DOF is a very helpful trick as it substantially reduces the numerical effort of the following integral calculation without affecting

the accuracy of the product expansion. One should contract over that DOF which otherwise would require the largest expansion order.

2.5. Including weights

inclusion of weights, i.e. minimizing $\Delta^2 = \sum_I (V_I - V_I^{\text{app}})^2 w_I^2$, is often inevitable for obtaining an accurate product representation of the physically relevant part of the potential without going to high expansion orders. The inclusion of separable weights, $w_I = w_{i_1}^{(1)} \cdots w_{i_f}^{(f)}$, is very simple [69], but separable weights are not so helpful and hence not discussed here. The inclusion of non-separable weights unfortunately leads to complicated equations of little use. There is, however, a nice trick [70,71], which allows one to emulate non-separable weights by an iterative procedure. To this end we introduce a reference potential V^{ref} such that the weighted difference between the potential and its product representation is identical to the difference between the reference potential and the product representation

$$(V_I - V_I^{\text{app}}) w_I = V_I^{\text{ref}} - V_I^{\text{app}}. \tag{15}$$

Then one simply can 'potfit' the reference potential to obtain a product representation, V^{app} , which is (almost) optimal with respect to the weighted sum of squared differences. Obviously, V^{ref} is given by

$$V_I^{\text{ref}} = w_I V_I + (1 - w_I) V_I^{\text{app}}.$$
 (16)

The definition of the reference potential depends on $V^{\rm app}$ which in turn depends on the reference potential. Hence, the equations must be solved iteratively. One first potfits V and evaluates the reference potential. Then the reference potential is potfitted and with the new $V^{\rm app}$ a new reference potential is built. The process is iterated until some break-off criterion is satisfied.

When emulating non-separable weights we always have used a special form of the weights. The weights are set to one within the so-called *relevant region* and are zero otherwise. The relevant region is usually defined by an energy criterion, i.e. it is the region where the potential is lower than some suitably chosen energy threshold. Restrictions on coordinates can be set as well when defining the relevant region. With such a definition of the weights, i.e. zero or one, Equation (16) has a vivid interpretation. The reference potential is the original potential within the relevant region and the fitted potential otherwise. Moreover, with this choice of the weights we always observed a lowering of the weighted L^2 error with each iteration.

2.6. Computational effort and memory requirements

The potfit expansion was introduced to reduce the numerical labour when evaluating the integrals. Consider the computation of the matrix element $\langle \Phi_J | V | \Phi_L \rangle$. Doing this integral on the primitive grid requires N^f multiplications. (Here we assume, for the sake of simplicity, that all DOFs have the same number of grid points, $N_{\kappa} = N$.) Doing the integral with a potfit expansion requires sfN multiplications. The number of potential terms is, due to contraction, $s = m^{f-1}$, where, similar to above, $m_{\kappa} = m$ is assumed. The gain is hence

$$gain_{CPU} = \frac{1}{f} \left(\frac{N}{m} \right)^{f-1}.$$
 (17)

A potfit expansion does not only speed-up the calculation, it also compacts the representation of the potential leading to a much lower memory demand. The full potential consists of N^f data points, whereas a potfit expansion, Equation (14), takes $Nm^{f-1} + Nm(f-1)$ data points. For f > 2 the second term is negligible in comparison with the first one, and one arrives at a memory gain

$$gain_{mem} = \left(\frac{N}{m}\right)^{f-1}.$$
 (18)

As an example let us consider a 6D problem where each DOF is represented by 25 grid points. Assuming m=6 (m=5), i.e. 7776 (3125) potfit terms after contraction, one has a CPU-gain of 209 (521), which is a quite remarkable speed-up. The potential consists of $N^f = 2.4 \times 10^8$ points and requires 1.8 GB of storage. The potfit consumes only 1.5 MB (615 KB). A potfit representation is hence very compact. This is an important feature when turning to larger systems where the potential evaluated at the grid points ceases to fit into memory. Unfortunately, potfit cannot solve this problem. Although the potfit representation is very compact, to arrive at this representation one has to perform sums over all grid points, see Equations (10) and (13). Hence, using today's workstations, potfit is limited to problems with less than 10⁹ grid points, i.e. in general to systems with at most six or seven degrees of freedom. One way out of this dilemma is to switch to a (in general more approximate) n-mode representation (see Section 2.1). One may then potfit the *n*-mode

When the PES is built from scratch one may fit the *ab initio* points by a model function which is of product form. As already noted, polynomial fits are of product form. New ideas on high dimensional potential fitting in product form using neural network techniques have recently been discussed by Manzhos and Carrington [74].

Finally, we note that in MCTDH potfit is used with a slight modification. MCTDH (see Section 7.2) makes use of mode combination and it is thus well-founded to expand the potential in the same combined modes. The changes to the above equations are trivial. One merely has to replace the number of degrees of freedom, f, by the number of particles, p, and to interpret N_{κ} as the number of points of the combined grid of the κ th particle.

3. Representations of the kinetic energy operator

One must choose coordinates to use for the kinetic energy operator (KEO). Basis functions are almost always functions of the coordinates in terms of which the KEO is represented. As long as the transformation between the coordinates used for the KEO and the coordinates used for the potential is known, the potential can be a function of any (even redundant) coordinates.

In this section, we focus on two quite different choices for coordinates, one is rectilinear normal coordinates and the other is curvilinear polyspherical coordinates. We point out the advantages and disadvantages of each and explain why each set is very much in use today. We consider normal coordinates first.

As noted already, early variational calculations of vibrational levels of triatomic molecules were done in rectilinear normal coordinates [2] at least in part because the exact (Watson) Hamiltonian in these coordinates was well known [1]. The Watson KEO (atomic units are used throughout this article) is

$$\hat{K}_{W} = \frac{1}{2} \sum_{\alpha,\beta} (\hat{J}_{\alpha} - \hat{\pi}_{\alpha}) \mu_{\alpha\beta} (\hat{J}_{\beta} - \hat{\pi}_{\beta})$$

$$-\frac{1}{2} \sum_{k} \frac{\partial^{2}}{\partial Q_{k}^{2}} - \frac{1}{8} \sum_{\alpha} \mu_{\alpha\alpha}, \qquad (19)$$

where

$$\mu_{\alpha\beta} = (\mathbf{I}^{\prime^{-1}})_{\alpha\beta}; \quad \mathbf{I}^{\prime}_{\alpha\beta} = \mathbf{I}_{\alpha\beta} + \sum_{k,l,m} \zeta_{km}^{\alpha} \zeta_{lm}^{\beta} Q_k Q_l \quad (20)$$

and where $I_{\alpha\beta}$ is the inertia tensor and ζ_{km}^{α} are Coriolis parameters defined for example in [1]. The vibrational angular momentum terms, π_{α} , are

$$\pi_{\alpha} = -i \sum_{k,l} \zeta_{kl}^{\alpha} Q_k \frac{\partial}{\partial Q_l}.$$
 (21)

Note that we include the last 'Watson' term in the expression for the KEO even though it does not contain differential operators.

The generality of the Watson KEO and thus the full Hamiltonian, which is just the sum of the KEO plus the potential V, was and is a major advantage and was a motivating factor for the development of variational codes based on it. An early example of such a code is POLYMODE [15] which was written in the mid 1980s and which uses quartic force fields. A number of other codes now exist based on the exact Watson Hamiltonian and also making use of the n-mode representation of the potential described in the previous section [57,61,62,75–81].

The $\pi^t \mu \pi$ term causes non-separability of the exact Watson KEO, even for J=0. In MULTIMODE [57,75–77] matrix elements of this term are computed by representing the μ matrix elements with an n-mode expansion analogous to the one used for the potential, but with an order smaller than that used for V. The $\pi^t \mu \pi$ terms are of the order of the rotation constants, and thus are generally small compared to other terms in the KEO and in V. Thus, in numerous applications in the literature these terms are neglected. In practice one can accurately treat the $\pi^t \mu \pi$ term, as done in MULTIMODE. If the basis is so large that the matrix representing $\pi^t \mu \pi$ cannot be stored in memory then matrix-vector products must be computed term by term and the fact that there are many $\pi^t \mu \pi$ terms slows the calculation.

There are well-known limitations of normal coordinates and thus to the Watson Hamiltonian. First, the above KEO for non-linear molecules is singular at linear shapes where $\pi^t \mu \pi$ is singular. If a molecule approaches linear geometries this can significantly affect results of variational calculations. Carter and Bowman investigated this in detail for highly excited states of H₂O, which has an energy at the linear geometry of roughly 11,100 cm⁻¹ above the minimum. They examined the deviation of the energies of highly excited pure bend overtone states from exact results and found deviations ranging from 0.3 to 200 cm⁻¹ for the third overtone to the sixth overtone [82]. For larger non-linear molecules this is of less concern, of course. Second, there is not a one-to-one mapping between points in normal coordinate space and molecular shapes. A simple example: points in normal coordinate space represent molecules with negative bond lengths. Third, normal coordinates describe large amplitude motions poorly and therefore basis functions that are functions of normal coordinates do not compactly represent wavefunctions of molecules undergoing large amplitude motion. This is in part due to the fact that normal coordinates are defined with respect to a single reference configuration. The fact that normal coordinates are rectilinear (and not curvilinear) also plays a role. There are, however, cases for which one can compute, using normal coordinates, accurate tunnelling splittings in multi-dimensions, e.g. for H₃O⁺ [83] and NH₃ [84]. In these cases saddle point normal modes were used since they span the two symmetric minima and the saddle point separating them.

It should be pointed out that reaction path [85] (and reaction surface [86]) Hamiltonians enable one to overcome these disadvantages by using one or more curvilinear coordinates and normal coordinates defined with respect to displacements from a flexible reference. The associated KEOs are complex but there is a version of MULTIMODE based on this Hamiltonian and the *n*-mode representation of the potential and it has been used successfully in numerous applications [87–91].

For ro-vibrational calculations the Watson KEO has the important advantage that it uses an Eckart molecule-fixed axis system [92] that significantly reduce vibration—rotation coupling. Deriving a curvilinear KEO in an Eckart frame is possible but the operator one obtains is not simple [93–95].

For a description of 'floppy' molecules, or in general large amplitude motion, geometrically defined curvilinear coordinates are better than normal coordinates. General KEOs also exist in curvilinear coordinates [96,97] but they are not as simple as the Watson KEO without the $\pi^t \mu \pi$ term. What criteria should be considered when coordinates are chosen? (i) Coordinates (and their domains) must be chosen so that there is a one-to-one correspondence between a geometry of the molecule and a set of coordinate values. (ii) It is advantageous to choose coordinates to minimise coupling and therefore facilitate choosing good basis functions. (iii) Coordinates should also be chosen to allow one to take advantage of symmetry. (iv) It is often helpful to choose coordinates to simplify the kinetic energy operator. (v) If potential integrals are evaluated by Gauss quadrature [98] the number of points required to converge the integrals depends on the degree of the polynomial expansion of the potential (even if the potential is not explicitly expanded) and it is therefore good to choose coordinates so that the potential, when expanded in terms of polynomials of the coordinates, converges quickly.

A simple and general KEO can be derived by using the polar coordinates associated with any set of N-1 vectors that specifies the shape and orientation of the molecule. Chapuisat and Iung [99] were perhaps the first to adopt this sort of approach, which was also used recently in [50,100–103]. Polar coordinates have the key advantage of satisfying criteria (iv) and (i).

For some molecules they will also satisfy criteria (ii) and (iii). By using the $\mathbf{J} - \mathbf{L}$ rule [104] it is possible to refine the coordinate choice to satisfy better criteria (ii) and (iii). The KEO is much simpler if one uses 'orthogonal' vectors. The term 'orthogonal' is usually taken to mean that the mass-weighted vectors are related by an *orthogonal* transformation to the mass-weighted nuclear position vectors. Note that the KEO in the length and angle coordinates obtained from these vectors has mixed second derivative terms.

How does one obtain the KEO in terms of the polar coordinates associated with a set of orthogonal vectors? Most methods of deriving kinetic energy operators follow one of two routes: (1) obtain the classical kinetic energy expression and then quantize it (usually using the procedure of Podolsky [105]) or (2) use the chain rule to transform a known quantum mechanical KEO (usually the space-fixed Cartesian KEO) to derive a new KEO. Chapuisat and co-workers [99–101,106–108] have advocated the first route. Sutcliffe and Tennyson [109–113] and Handy [114] were early proponents of the second route. Mladenović [102,103] has also used the chain rule route.

Regardless of which route is taken one starts from the KEO for N nuclei in space-fixed Cartesian coordinates. If the quantum mechanical operator is transformed one begins with

$$\hat{T}N = -\frac{1}{2} \sum_{i=0}^{N-1} \frac{1}{M_i} \left(\frac{\partial^2}{\partial X_i^2} + \frac{\partial^2}{\partial Y_i^2} + \frac{\partial^2}{\partial Z_i^2} \right)$$

$$\equiv -\frac{1}{2} \left(\hat{T}_X + \hat{T}_Y + \hat{T}_Z \right), \tag{22}$$

where i is a nuclear label and M_i a nuclear mass. For \hat{T}_X one then does three transformations: (i) to mass-weighted coordinates $\{\bar{X}_i = M_i^{1/2}X_i\}$; (ii) to coordinates $\{\bar{P}_\alpha\}$ linearly related to $\{\bar{X}_i\}$ by an $N \times N$ orthonormal transformation; and (iii) to mass-unweighted coordinates $\{P_\alpha = \mu_\alpha^{-1/2}\bar{P}_\alpha\}$. The third step introduces arbitrary masses $\{\mu_\alpha\}$:

$$\hat{T}_X = -\frac{1}{2} \sum_{\alpha=0}^{N-1} \frac{1}{\mu_\alpha} \left(\frac{\partial^2}{\partial P_\alpha^2} \right). \tag{23}$$

Applying an identical transformation to \hat{T}_Y and \hat{T}_Z (by introducing coordinates $\{Q_\alpha\}$ and $\{R_\alpha\}$), we obtain N vectors $\{\mathbf{r}_\alpha\}$ (each with space-fixed Cartesian components $(P_\alpha, Q_\alpha, R_\alpha)$) that are linear combinations of the space-fixed Cartesian nuclear position vectors with coefficients that are elements of an $N \times N$ matrix

$$\mathbf{J} = \mu^{-1/2} \mathbf{U} \mathbf{M}^{1/2}, \tag{24}$$

where **M** and μ are diagonal matrices of masses, and **U** is orthonormal [50]. \mathbf{r}_{N-1} is chosen to be the position

of the nuclear centre of mass. Natural coordinates for describing the shape and orientation of the molecule are: the lengths of the remaining N-1 vectors r_0, r_1, \ldots r_{N-2} ; N-2 polar angles, $\theta_{\alpha}(\alpha=1\cdots N-2)$ between r_0 and r_{α} ; N-3 angles $\phi_{\beta}(\beta=2\cdots N-2)$ between the plane that contains r_0 and r_1 and the planes that contain r_0 and r_{β} ; and three Euler angles that specify the orientation of the molecule-fixed axis system with respect to the space-fixed axes. The molecule-fixed z axis is parallel to r_0 and the molecule-fixed y axis is along $\mathbf{r}_0 \times \mathbf{r}_1$. The KEO in these coordinates is given in many papers [100,102,115,116]. One convenient form [117] is

$$T = T_{\rm s} + T_{\rm br} + T_{\rm cor} \tag{25}$$

with

$$T_{\rm br} = T_{\rm br,diag} + T_{\rm br,off}$$
 (26)

and

$$T_{s} = -\sum_{k=0}^{N-2} \frac{1}{2\mu_{k}} \frac{\partial^{2}}{\partial r_{k}^{2}},$$

$$T_{br,diag} = [B_{0}(r_{0}) + B_{1}(r_{1})]$$

$$\times \left[-\frac{1}{\sin\theta_{1}} \frac{\partial}{\partial \theta_{1}} \sin\theta_{1} \frac{\partial}{\partial \theta_{1}} + \frac{1}{\sin^{2}\theta_{1}} (J_{z} - L_{z})^{2} \right]$$

$$+ \sum_{k=2}^{N-2} [B_{0}(r_{0}) + B_{k}(r_{k})] l_{k}^{2} + B_{0}(r_{0})$$

$$\times \left[J^{2} - 2(J_{z} - L_{z})^{2} - 2J_{z}(L_{z}) + 2 \sum_{k \neq k'=2}^{N-2} l_{kz} l_{k'z} \right],$$

$$T_{br,off} = B_{0}(r_{0})$$

$$\times \left[(L_{+})a_{1}^{-} + (L_{-})a_{1}^{+} + \sum_{k \neq k'=2}^{N-2} (l_{k+}l_{k'-} + l_{k-}l_{k'+}) \right],$$

$$T_{cor} = -B_{0}(r_{0}) [J_{-}(a_{1}^{+} + L_{+}) + J_{+}(a_{1}^{-} + L_{-})], \qquad (27)$$

where

$$B_i(r_i) = \frac{1}{2\mu_i r_i^2},$$
 (28)

$$L_z = \sum_{k=2}^{N-2} l_{kz},$$
 (29)

$$L_{-} = \sum_{k=2}^{N-2} l_{k-}, \tag{30}$$

$$L_{+} = \sum_{k=2}^{N-2} l_{k+}, \tag{31}$$

$$l_{i+} = l_{ix} \pm i l_{iy} \quad (i = 2, \dots, N-2)$$
 (32)

$$J_{+} = J_{x} \pm i J_{y}, \tag{33}$$

$$a_1^{\pm} = \pm \frac{\partial}{\partial \theta_1} - \cot \theta_1 (J_z - L_z)$$
 (34)

and $l_{kx}, l_{ky}, l_{kz}, l_k^2$ are the usual angular momentum operators [118]. Note that angular momentum vectors for the vectors not used to define the molecule-fixed frame contribute to L. a_1^+ and a_1^- act as raising and lowering operators [117]. Although a_1^+ and a_1^- do not commute with $l_{k\pm}$ and J_{\pm} , the KEO is hermitian. The volume element associated with this KEO is $\sin \theta_1$ $\sin \theta_2 \cdots$, $\sin \theta_{N-2} dr_0 dr_1 \cdots dr_{N-2} d\theta_1 \cdots d\theta_{N-2} d\phi_2 \cdots$ $d\phi_{N-2}$. Note that this KEO is valid for any choice of the orthogonal vectors. Changing the definition of the vectors changes the masses that appear in the final KEO. For four-atom molecules several choices are given explicitly in [50]. There are momentum crossterms between angles but none between angles and lengths. This general 'polyspherical' KEO is remarkably compact if it is written in terms of angular momentum operators. In terms of derivatives with respect to coordinates it is more lengthy.

A disadvantage of the above KEO is its lack of flexibility: one must place the molecule-fixed z axis along one of the \mathbf{r}_i vectors and one must choose as vibrational coordinates polyspherical coordinates obtained from the N-1 \mathbf{r}_i vectors. If r_j ($j=2,\ldots,N-2$) are orthogonal to r_0 and r_1 a general (and more flexible) molecule-fixed KEO can be derived, even if r_0 and r_1 are themselves not orthogonal, by attaching the molecule-fixed axis system to the plane spanned by r_0 and r_1 . This is done as follows [104]. (i) Write the KEO obtained by adding Equation (23) and its Y and Z counterparts, and removing the centre of mass term, as

$$K = -\frac{1}{2\mu_0} \frac{\partial^2}{\partial \mathbf{r}_0^2} - \frac{1}{2\mu_1} \frac{\partial^2}{\partial \mathbf{r}_1^2} - \frac{1}{\mu_{01}} \frac{\partial}{\partial \mathbf{r}_1} \frac{\partial}{\partial \mathbf{r}_0} - \frac{1}{2} \sum_{i=2}^{N-2} \frac{1}{\mu_i} \frac{\partial}{\partial \mathbf{r}_i} \frac{\partial}{\partial \mathbf{r}_i}.$$
 (35)

(ii) From the first three terms in Equation (35) derive a ro-vibrational KEO for a triatomic-like system in terms of any vibrational coordinates q_1 , q_2 , and q_3 and in *any* molecule-fixed frame attached to the two embedding vectors \mathbf{r}_1 and \mathbf{r}_0 . (iii) Substitute $\mathbf{J} - \mathbf{L}$ for \mathbf{J} in the triatomic-like KEO. This yields,

$$K = K_{\text{tri}}(q_1, q_2, q_3, \mathbf{J} - \mathbf{L}) - \frac{1}{2} \sum_{i=2}^{N-2} \frac{1}{\mu_i} \frac{\partial}{\partial \mathbf{z}_i} \frac{\partial}{\partial \mathbf{z}_i}, \quad (36)$$

where the \mathbf{z}_i are the non-embedding vectors in the molecule-fixed frame. (iv) Transform from the z_{ia} coordinates (Cartesian components of the z_i) to some non-Cartesian coordinates that describe the length and orientation of the $N-3\mathbf{z}_i$ vectors with respect to the molecule-fixed frame attached to \mathbf{r}_1 and \mathbf{r}_0 . Here I is the total angular momenta of the non-embedding vectors in the molecule-fixed frame. This recipe enables one to obtain a KEO for an N-atom molecule from the KEOs of smaller molecules. This procedure works only because the $\mathbf{r}_i(j=2,\ldots,N-2)$ vectors are orthogonal to the two embedding vectors [104]. This KEO is simple and general despite the fact that we do not require that the z axis be placed along one of the \mathbf{r}_i vectors. This is important because either to reduce ro-vibrational coupling or to facilitate exploiting symmetry it is sometimes better not to have the z axis along a \mathbf{r}_i vector. In [119] ideas related to those of [104] are used to derive a KEO for NH₃ using hyperspherical coordinates for the H₃ group. However in [119] the molecule-fixed frame is not attached to the triatomic-like group as it is in [104].

4. Eigensolvers

Once one has chosen a KEO (i.e. coordinates) and obtained a potential it remains to determine eigenvalues and eigenfunctions of the Hamiltonian operator. To do so one selects a basis and computes eigenvalues and eigenvectors of a finite matrix representation of the operator. In some cases it is possible to explicitly build the matrix. If the basis is too large it is necessary to extract eigenvalues using techniques that require only matrix elements of factors or terms of the full Hamiltonian. There are two classes of methods for solving the matrix eigenvalue problem: direct and iterative methods. Direct methods can be used if the basis is small enough that it is possible to store the Hamiltonian matrix in memory. For larger basis sets iterative methods are necessary.

4.1. Direct methods

The standard direct method uses a combination of Householder transformations and the QR algorithm [98]. To use it one must store, in memory, the matrix whose eigenvalues are desired, i.e. N^2 numbers, where N is the size of the matrix, and the cost of the calculation scales as N^3 . If N is larger than about 50,000 both the memory cost and the CPU cost are significant. Nevertheless, the Householder direct eigensolver can be used to compute energy eigenvalues. To do so one must choose good basis functions so as to minimize N.

4.2. Iterative methods

The unfavourable scaling of the Householder algorithm has prompted the investigation and development of iterative methods for calculating spectra [45,46,50,120–131]. An iterative method requires only the computation of matrix-vector products [132]. Hamiltonian matrix-vector products can be computed without storing the Hamiltonian matrix and without even calculating its matrix elements [50,133,134]. One needs only enough memory to store a few vectors with as many components as there are basis functions.

The most popular iterative methods employed to calculate spectra are variants of the Lanczos algorithm [132,135], the filter diagonalisation method [45,46,126,129], or the Davidson algorithm [136]. In the Lanczos method, eigenvalues of an $N \times N$ symmetric matrix H are calculated by diagonalising an associated $M \times M$ tridiagonal matrix T_M which is generated recursively, having dimension M after M iterations. To obtain T_M one must compute M matrixvector products. If M is large enough, among the eigenvalues of T_M there will be eigenvalues of the original matrix H [135]. There are several refined versions of the Lanczos algorithm designed to accelerate the convergence of selected eigenvalues [137,138]. These methods facilitate the calculation of low-lying energy levels and allow calculation of eigenvectors without re-doing matrix-vector products but they require considerably more memory than a straightforward Lanczos method.

The goal of the filter diagonalisation method is to calculate energy levels and wavefunctions in a selected (not necessarily low-lying) spectral window. One builds a matrix representation of the Hamiltonian using basis vectors $\mathbf{z}_n = f(\mathbf{H}, n)\mathbf{v}, n = 1, 2, \dots, L$, where \mathbf{v} is a starting vector and $f(\mathbf{H}, n)$ is chosen so that the filtered basis vectors have only very small overlaps with eigenvectors of H whose eigenvalues are outside the window in which one wishes to calculate eigenvalues [126]. v and H are representations in some original basis of a starting ket and the Hamiltonian operator. In the filtered basis one solves a generalized eigenvalue problem to obtain energy levels. Wall and Neuhauser pointed out that if Hamiltonian and overlap matrix elements are calculated directly, eigenvalues of the Hamiltonian may be computed without computing (and storing) the \mathbf{z}_n vectors [45]. Mandelshtam and Taylor have developed a reliable Chebyshev FD algorithm [129]. Using their low-storage approach to filter-diagonalisation one is able to calculate energy levels using as little memory as one requires using a Lanczos approach. The Cullum & Willoughby (C&W) Lanczos approach and the filter-diagonalisation method extract energy levels from the same Krylov space. Both may be viewed as methods for dealing with numerical error introduced as the space is generated. Formally, the space spanned by the Lanczos vectors is larger than (and includes) the space spanned by the filtered \mathbf{z}_n vectors. The C&W Lanczos method requires about as many (in general somewhat fewer) matrix-vector products than FD and it is simpler to implement [139,140].

The Davidson algorithm is a preconditioned variant of the Lanczos algorithm [136]. It works best if one has both (i) an easily invertable matrix, the preconditioner, whose eigenvalues are close to those of the full Hamiltonian matrix and (ii) a good starting vector. It is common to use a diagonal preconditioner. The Davidson algorithm favours the convergence of the lowest or (with a block version) a number of the lowest eigenvalues. It is used in the MULTIMODE program [75,76] and the improved relaxation method [141]. The Davidson type approaches work well if it is possible to choose a basis that facilitates finding a good preconditioner.

If one opts for an iterative method one needs a routine for computing matrix-vector products. The matrix-vector product is the heart and soul of any iterative procedure. To compute matrix-vector products one should take advantage of any structure of the basis. The MCTDH basis is a direct product and this structure is very favourable. The basis used in MULTIMODE is selected from a direct product basis but has no exploitable structure. In the contracted iterative methods outlined in the next section the basis is a product of reduced dimension functions and this structure is used to evaluate matrix vector products.

In the next sections, we review methods and codes in detail. We begin with Hamiltonians in polyspherical coordinates and emphasize the use of truncation-recoupling. Following that we describe variational methods based on the Watson Hamiltonian, the *n*-mode representation of the potential and the VSCF/VCI approach to obtain eigenvalues and eigenfunctions. Section 7 describes the MCTHD approach.

5. Contracted iterative methods

The simplest way to compute an eigenvalue spectrum is to use a time-independent direct-product basis and a direct eigensolver. For a triatomic molecule this is a perfectly adequate approach. To solve more difficult problems one must use a better eigensolver, use a better basis or do both. The first calculations of

spectra done with iterative eigensolvers were done with direct-product bases [50,142,143]. Using the Lanczos algorithm it is straightforward to do calculations with millions of direct-product basis functions. The calculation can be done without representing the potential using either a potfit or a multimode expansion, regardless of whether one uses a DVR or an FBR. The sums required to evaluate matrix-vector products are done sequentially so that one obtains an eigenvalue spectrum without computing multi-dimensional integrals. Even the advantages of iterative methods are not enough to make it possible to compute spectra of five-atom molecules with a direct-product basis. To do so one must improve the basis. There are three ways in which the basis set can be improved: (1) only the most important basis functions can be selected from a huge direct-product basis; (2) contracted basis functions that are not simple products can be used; (3) time-dependent basis functions can be used. The MULTIMODE package of Carter, Bowman, Handy, and co-workers uses strategy (1) and very recently strategy (2). MCTDH uses strategy (3) and also (2). In this section we sketch an approach that uses strategy (2), while strategy (3) is discussed in Section 7.

Contracted basis functions are chosen to reduce the number of basis functions required to obtain converged energy levels. It has been clearly demonstrated that if they are used with a direct eignensolver they are powerful tools for computing vibrational energy levels [6,144]. Contracted basis functions are usually obtained by diagonalizing reduced-dimension Hamiltonian matrices. There are two popular ways to define the effective reduced-dimension Hamiltonian matrices: (1) one diagonalizes blocks of the full Hamiltonian matrix in a (primitive) product basis representation; (2) one diagonalizes matrices that represent the Hamiltonian with one or more coordinates fixed. The first route leads to nondirect-product contracted basis functions [144-151]. The second route leads to basis functions that are direct-products of functions of different coordinates or groups of coordinates [6,152]. An example (see also Equation (1)) should make this clearer. For a triatomic molecule, a contraction scheme of type 1 is obtained by diagonalizing (stretch) blocks $\langle \alpha(\theta)|\hat{H}(\theta,r_1,r_2)|\alpha(\theta)\rangle$ to get eigenfunctions $\phi_n^{\alpha}(r_1, r_2)$ where $\alpha(\theta)$ is a DVR bend function, $H(\theta, r_1, r_2)$ is the full Hamiltonian operator, and θ , r_1 , r_2 are the bend and stretch coordinates. The contracted basis functions are then $\alpha(\theta)\phi_n^{\alpha}(r_1, r_2)$. A contraction scheme of type 2 is obtained by combining the eigenfunctions of $H(\theta_e, r_1, r_2)$, denoted $\psi(r_1, r_2)$ and eigenfunctions of $H(\theta, r_1^e, r_2^e)$, denoted $\chi(\theta)$. The contracted basis functions are $\psi(r_1, r_2)\chi(\theta)$. r_1^e , r_2^e , and θ_e are selected specific values of r_1 , r_2 , and θ . Often equilibrium values will be good choices. Regardless of the route one takes, only some of the eigenvectors of the reduced-dimension Hamiltonian matrices are retained: such methods are sometimes called diagonalisation-truncation-recoupling methods [17,52].

The type 1 approach has been used extensively by Bacic, Bowman, Light, Tennyson and their coworkers [17,52,144,145,147,148]. Luckhaus [151] and Mladenović [149,150] have recently used type 1 contractions in 6D calculations. Both these authors use contracted basis functions with a direct eigensolver (Householder transformations). The type 2 approach was pioneered by Carter and Handy [6,152] and has been used with a direct eigensolver to compute rovibrational energy levels of four-atom molecules, e.g. C_2H_2 [153], H_2O_2 [154], NH_3 (vibration only) [7].

Is it efficient to use contracted basis functions (of either type) with an iterative eigensolver? This seems like an obvious combination. Why not profit from both the smallness of the contracted basis and the reduced memory requirement of iterative eigensolvers? The combination is effective, but finding a good scheme for evaluating matrix-vector products in the contracted basis requires some thought. The ideas used for evaluating matrix-vector products in a directproduct basis exploit the special structure of the product basis to make it possible to do summations (over basis or quadrature indices) sequentially [50]. If one uses contracted basis functions at least some of this product structure is lost. The contracted basis + Lanczos combination is expected to be better than the product basis + Lanczos combination because: (1) the Lanczos vectors are smaller (reducing the memory cost) and (2) the number of required Lanczos iterations is smaller because it depends on the spectral range of the matrix which will be smaller if contracted functions are used. To realize these advantages one must have a good means of computing contracted basis matrix-vector products.

Type 1 contracted functions can be used with iterative methods [125,146,155,156]. A type 1 basis large enough to compute a desired portion of an eigenvalue spectrum will almost always be smaller than its type 2 counterpart. Nevertheless, iterative calculations with a type 2 basis will, in most cases, be more efficient. To understand why consider the type 1 contracted matrix-vector product for a triatomic. We label with j functions of q_1 and q_2 obtained by diagonalising a two-dimensional Hamiltonian for each DVR point $(q_3)_{\gamma}$. If the two-dimensional Hamiltonian is diagonalised in a direct-product q_1 q_2 basis and α and β are DVR labels for the q_1 and q_2 DVR basis functions then the matrix of eigenvectors is the

transformation matrix whose elements are $C^{\gamma}_{\alpha\beta,j}$. In a basis of functions labelled by j and γ the Hamiltonian (written in Radau, symmetrized Radau, or Jacobi coordinates) matrix elements are

$$\langle j'\gamma'|\hat{H}|j\gamma\rangle = E_j^{\gamma}\delta_{j'j}\delta_{\gamma\gamma'} + \sum_{\alpha\beta} C_{\alpha\beta,j'}^{\gamma'}\mu((q_1)_{\alpha},(q_2)_{\beta})l_{\gamma'\gamma}C_{\alpha\beta,j}^{\gamma}, \qquad (37)$$

where μ is an inverse moment of inertia function and $l_{\gamma'\gamma}$ is a DVR matrix element of an operator proportional to $(\partial^2/\partial\theta^2) + \cot\theta(\partial/\partial\theta)$. In this representation applying the Hamiltonian to a vector implicates only one non-trivial matrix-vector product. When sums are evaluated sequentially the cost of the matrix-vector product,

$$\sum_{\alpha\beta} C_{\alpha\beta,j}^{\gamma'} \mu((q_1)_{\alpha}, (q_2)_{\beta}) \sum_{\gamma} l_{\gamma'\gamma} \sum_{j} C_{\alpha\beta,j}^{\gamma} v_{\gamma j}, \qquad (38)$$

scales as $2(n_j n_\alpha n_\beta n_\gamma + n_\gamma^2 n_\alpha n_\beta + n_\alpha n_\beta n_\gamma)$ (n_i is the number of eigenfunctions of the two-dimensional Hamiltonians which are retained). If $n_{\alpha} = n_{\beta} = n_{\gamma} = n$ and n_i is n, the cost of each matrix-vector product scales as n^4 . In general, the cost of the type 1 contracted matrix-vector products will depend on the number of retained basis functions and is hard to estimate. However, it is clear that the *number* of matrix-vector products required to obtain a converged eigenvalue spectrum will be less than would be necessary if one used a product basis. This type 1 approach has several important disadvantages. (1) Many reduced-dimension problems must be solved. In the triatomic case there is only one bend angle, but for a five-atom molecule there are five angles and a type 1 contraction, similar to the one outlined above for triatomics, would require solving a stretch problem for millions of (multi-dimensional) DVR bend functions. (2) For a five-atom molecule it would be impossible to store, if it were possible to compute, the stretch eigenfunctions for all the bend DVR points. (3) The type 1 contraction presented above derives its efficiency from the use of a DVR for the bend and exploitation of an adiabatic bend/stretch separation. Multi-dimensional DVRs that allow one to cope with KEO singularities are not known. (Note that disadvantages (1) and (2) would also cripple a type 1 approach in which a bend problem is solved for each multdimensional stretch DVR point.) (4) To evaluate the matrix-vector product in Equation (38), the vector is transformed to a simple product representation in which it is easy to apply the KEO and then back transformed to the contracted representation. This inevitably entails storing a vector in the product representation. Thus, despite the fact that the vector in the contracted basis is not large, one must

store a large vector to evaluate matrix-vector products. Type 2 contracted functions do not have these disadvantages. We have called such functions 'simply contracted' basis functions [125,157]. A type 2 contraction technique will now be described.

In orthogonal polyspherical coordinates the Hamiltonian (for J=0) of any molecule can be written as [100.102,108]

$$H = T_{\text{ben}}(\theta, r) + T_{\text{str}}(r) + V(\theta, r)$$
 (39)

with

$$T_{\text{ben}}(\theta, r) = \sum_{i} B_{i}(r) T_{\text{b}}^{(i)}(\theta),$$

$$T_{\text{str}}(r) = \sum_{i} \frac{1}{2\mu_{i}} \frac{\partial^{2}}{\partial r_{i}^{2}}.$$
(40)

 θ represents all of the bend coordinates and r represents all of the stretch coordinates. The functions $B_i(r)$ and the operators $T_b^{(i)}(\theta)$ are known [100,102].

We first construct contracted bend functions from a Hamiltonian obtained by freezing all the stretch coordinates at equilibrium (or more generally reference) values and contracted stretch functions from a Hamiltonian obtained by freezing all the bend coordinates at equilibrium (or reference) values. Products of the bend contracted functions and stretch contracted functions are our final basis functions.

The reduced-dimension Hamiltonian for the bend contraction is

$$H^{(b)} = T_{ben}(\theta, r_e) + V(\theta, r_e). \tag{41}$$

Its wavefunctions are denoted by

$$X_{b}(\theta) = \sum_{l} C_{lb} f_{l}(\theta)$$
 (42)

and the energies by E_b . The f_l are primitive FBR or DVR bend basis functions (if there is more than one bend coordinate l is a composite index) and the number of retained bend wavefunctions is denoted by n_b .

Similarly, the reduced-dimension Hamiltonian for the stretch contraction is

$$H^{(s)} = T_{\rm str}(r) + V(\theta_{\rm e}, r).$$
 (43)

with the wavefunctions denoted by

$$Y_{s}(r) = \sum_{\alpha} D_{\alpha s} g_{\alpha}(r) \tag{44}$$

and the energies by E_s . The g_{α} are primitive DVR stretch basis functions (if there is more than one stretch coordinate α is a composite index representing a multi-dimensional DVR function) and the number

of retained stretch wavefunctions is denoted by $n_{\rm s}$. $\theta_{\rm e}$ and $r_{\rm e}$ represent equilibrium values of (respectively) all the bend coordinates and all the stretch coordinates. The final basis is a product of the retained stretch and bend eigenfunctions

$$|bs\rangle = |X_b\rangle |Y_s\rangle. \tag{45}$$

To facilitate calculation of its matrix elements in this basis we write the full Hamiltonian

$$H = H^{(b)} + H^{(s)} + \Delta T + \Delta V,$$
 (46)

where

$$\Delta V(r,\theta) = V(r,\theta) - V(r_e,\theta) - V(r,\theta_e) \tag{47}$$

and

$$\Delta T = \sum_{i} \Delta B_{i}(r) T_{b}^{(i)}(\theta)$$
 (48)

with

$$\Delta B_i(r) = B_i(r) - B_i(r_e). \tag{49}$$

Since $H^{(b)}$ and $H^{(s)}$ are diagonal in the product contracted basis and the diagonal matrix elements are known (they are the eigenvalues of $H^{(b)}$ and $H^{(s)}$) one only needs to calculate matrix elements of ΔT and ΔV in the product contracted basis. If it were possible to store the Hamiltonian matrix in the the product contracted basis $|bs\rangle$ a direct linear algebra method could be used to compute the energy levels. If this is not possible one *must* use an iterative method and evaluate matrix-vector products.

The iterative method we prefer is the Lanczos algorithm. We use it to solve the bend problem, the stretch problem and to compute the final solutions in the basis of products of bend and stretch solutions. The Lanczos algorithm is simple and easy to use. There are no problems with spurious eigenvalues [135]. From a single set of matrix-vector products we can extract a large number of eigensolutions. It is necessary to store only a few vectors and there is no need for a good initial vector.

Matrix-vector products for ΔT

Due to the product structure of Equation (48) it is best to store the bend and stretch matrices separately. The matrix-vector product for ΔT can be performed sequentially without summing over n_b bend functions and n_s stretch functions simultaneously. Note that it is not necessary to store any vectors in the primitive basis.

Specifically, the ΔT matrix-vector product can be calculated

$$u'_{b's'} = \sum_{i,s} \langle s' | \Delta B_i | s \rangle \sum_b \langle b' | T_b^{(i)} | b \rangle u_{bs}.$$
 (50)

The ΔB_i and $T_b^{(i)}$ matrices are both small and are stored. The ΔB_i matrix elements are computed from

$$\langle s'|\Delta B_i|s\rangle = \sum_{\alpha} D_{\alpha s'} \Delta B_i(r_{\alpha}) D_{\alpha s}.$$
 (51)

The $T_b^{(i)}$ matrix elements are computed from

$$\langle b'|T_b^{(i)}|b\rangle = \sum_{l,l'} C_{l'b'} \langle f_l|T_b^{(i)}|f_l\rangle C_{lb}. \tag{52}$$

Assuming that $n_b > n_s$ (which almost always holds) the memory cost of the ΔT matrix-vector product is n_b^2 and the CPU cost is $n_b^2 n_s$. The memory and CPU costs of the ΔV matrix-vector product are much higher [157].

5.1. Matrix-vector products for ΔV

In this subsection, we present a method for calculating the ΔV matrix-vector product [157]. If one uses an FBR primitive bend basis a matrix element of ΔV in the product contracted basis is

$$\langle b's'|\Delta V(\theta,r)|bs\rangle = \sum_{\substack{l'l\\\alpha\beta}} C_{l'b'}C_{lb}D_{\alpha s'}D_{\alpha s}\langle l'\alpha|\Delta V(\theta,r)|l\alpha\rangle$$

$$= \sum_{\substack{l'l\\\alpha\beta}} C_{l'b'}C_{lb}D_{\alpha s'}D_{\alpha s}T_{l'\beta}T_{l\beta}\Delta V_{\beta\alpha},$$
(53)

where

$$T_{l\beta} = w_{\beta}^{1/2} f_l(\theta_{\beta}),$$

$$\Delta V_{\beta\alpha} = \Delta V(\theta_{\beta}, r_{\alpha}).$$
 (54)

 $(\theta_{\beta}, w_{\beta})$ are quadrature points and weights for the angles. Note that if there is more than one bend coordinate l and β are composite indices and if there is more than one stretch coordinate α is a composite index. To do matrix-vector products of ΔV efficiently there is no need to represent it either as a sum of products or in multimode form. It is sometimes advantageous to rewrite Equation (53) as

$$\langle b's'|\Delta V(\theta,r)|bs\rangle = \sum_{\alpha\beta} \tilde{C}_{\beta'b'} \tilde{C}_{\beta b} D_{\alpha s'} D_{\alpha s} \Delta V_{\beta \alpha}, \quad (55)$$

where

$$\tilde{C}_{\beta b} = \sum_{l} C_{lb} T_{l\beta}. \tag{56}$$

A potential matrix element in the product contracted basis is rewritten

$$\langle b's'|\Delta V(\theta,r)|bs\rangle = \sum_{\alpha\beta} \tilde{C}_{\beta'b'} \tilde{C}_{\beta b} D_{\alpha s'} D_{\alpha s} \Delta V_{\beta \alpha}$$
$$= \sum_{\alpha} F_{b'b,\alpha} D_{\alpha s'} D_{\alpha s}, \tag{57}$$

where we have introduced an F matrix defined by,

$$F_{b',b,\alpha} = \langle b' | \Delta V(\theta, r_{\alpha} | b) = \sum_{\beta} \tilde{C}_{\beta b'} \tilde{C}_{\beta b} \Delta V_{\beta \alpha}.$$
 (58)

To use this method one calculates and stores all the $F_{b'b,\alpha}$. One then does the ΔV matrix-vector product,

$$u'_{b's'} = \sum_{bs} \langle b's' | \Delta V | bs \rangle u_{bs}. \tag{59}$$

This is most efficiently done in three steps:

$$u_{b\alpha}^{(1)} = \sum_{s} D_{\alpha s} u_{bs},$$

$$u_{b'\alpha}^{(2)} = \sum_{b} F_{b'b\alpha} u_{b\alpha}^{(1)},$$

$$u_{b's'}^{(2)} = \sum_{\alpha} D_{\alpha s'} u_{b'\alpha}^{(2)}.$$
(60)

Full-dimensional integrals are never computed. The CPU cost of the matrix-vector product in Equation (60) is $n_{\alpha}(n_b^2 + 2n_b n_s)$. Assuming $n_b > n_s$ the CPU cost scales as $n_{\alpha}n_b^2$. The calculation of **F** can be easily parallelized. For example, one can allocate the stretch DVR (α) points to different computers or processors and allow each to calculate a block of **F** for the allocated α points.

In previous approaches for using contracted basis functions in conjunction with iterative eigensolvers one evaluated a matrix-vector product by transforming from the contracted basis to a direct product basis, applying an operator, and then transforming back. This has the disadvantage that it requires storing a vector with as many components as there are product basis functions. To solve the Schrödinger equation using such an approach one would need to store a vector with N^{j} components, where N is a representative number of basis functions (or DVR points) for a single coordinate. If $N \sim 20$ and f = 9 this requires more than 4000 GB! Using the approach outlined above one needs much less memory. It is necessary only to store **F** or $n_b^2 n_\alpha/2$ numbers. For the methane calculation reported in Section 8.1 this requires less than 4GB. The cost of the most expensive matrix-vector product scales as $n_{\alpha}n_{\rm h}^2$, which is obviously smaller if n_b is small, i.e. if the bend contraction works wells. Typically the matrix-vector products are much less costly than direct-product

matrix-vector products would be if it were possible to store the direct-product vector.

These ideas were first applied to compute energy levels of H₂O₂ for which it is possible to do Lanczos calculations with a product basis [157] and it was demonstrated that for this problem the method is efficient. Later the same approach was used for methane for which a direct-product calculation is impossible [158]. For this molecule the bend calculation is the hardest part [54]. Later some J > 0 levels were also computed [117]. Similar ideas based on contraction and truncation were used to compute highlying vibrational levels of the vinylidene/acetylene system [23–25]: Tremblay and Carrington [24] used methods most closely related to those described in this section. Yu has studied several molecules using similar ideas [159,160]. He contracts only the bend part and not the stretch part, but the greatest benefit is derived from the bend contraction. Other scientists [161,162] have modified and applied the approach of [157].

6. VSCF/VCI with the Watson Hamiltonian

The vibrational SCF approach to obtain eigenfunctions and eigenvalues of the Watson Hamiltonian begins with a Hartree product for the total wavefunction

$$\Psi_{n_1...n_N}^{VSCF}(Q_1, \ldots Q_N) = \prod_{i=1}^N \phi_{n_i}^{(i)}(Q_i),$$
 (61)

where the modal wavefunctions $\phi_{n_i}^{(l)}(Q_i)$, subject to the usual normalisation constraint, are solutions of the coupled equations

$$\left[T_{l} + \left\langle \prod_{i \neq l}^{N} \phi_{n_{i}}^{(l)}(Q_{i}) \middle| T_{c} + V \middle| \prod_{i \neq l}^{N} \phi_{n_{i}}^{(l)}(Q_{i}) \right\rangle - \epsilon_{n_{l}}^{(l)} \right] \phi_{n_{l}}^{(l)}(Q_{l}) = 0,$$
(62)

where

$$T_{l} \equiv -\frac{1}{2} \sum_{k} \frac{\partial^{2}}{\partial Q_{k}^{2}}, \quad T_{c} \equiv \frac{1}{2} \sum_{\alpha,\beta} \hat{\pi}_{\alpha} \mu_{\alpha\beta} \hat{\pi}_{\beta} - \frac{1}{8} \sum_{\alpha} \mu_{\alpha\alpha} \quad (63)$$

and the integration is over the coordinates of N-1 modes. These equations are solved in the usual iterative fashion.

After achieving convergence each VSCF effective Hamiltonian has a set of virtual eigenfunctions which can be used in a standard 'CI' procedure to obtain eigenvalues and eigenfunctions of the Hamiltonian. This approach is denoted VSCF/VCI. (The non-orthogonal basis of VSCF states can also be used in a CI [10,11]; however, since this is not routinely done now we do not describe this any further.)

It is immediately clear that matrix elements of the Hamiltonian, both the potential and $T_{\rm c}$ for molecules larger than tetraatomics are not feasible for general potentials. The n-mode representation of the potential, and for $T_{\rm c}$ are used to extend the approach to larger molecules. Current typical computational power limits n to a maximum of 6. For $T_{\rm c}$ a relatively small value of n is sufficient, e.g. n=2 or 3 for most applications since this term is generally small compared to V and also more nearly separable than V.

The implementation of the *n*-mode representation of the potential together with the rigorous Watson Hamiltonian was first done in the code MULTMODE [57,76,77]. There now exist several other codes that use the *n*-mode representation of the potential to obtain eigenfunctions and eigenvalues of the rigorous Watson Hamiltonian using VSCF/VCI methods [62,78].

The choice of n for the nMR of V also has consequences for selecting the CI excitation space. This can be seen by considering a potential matrix element for a specific p-mode potential, denoted compactly as $V^{(p)}(\{Q_p)\}$ for a tetraatomic where N=6 and where p is less than N of course. The matrix element is of the reduced form

$$\langle n'_{1}n'_{2}n'_{3}n'_{4}n'_{5}n'_{6} | V^{(p)}(\{\mathbf{Q}_{p}\}) | n_{1}n_{2}n_{3}n_{4}n_{5}n_{6} \rangle$$

$$= \delta_{n'_{i},n_{i}} \dots \delta_{n'_{j},n_{j}} V^{(p)}_{\mathbf{n}'_{p},\mathbf{n}_{p}},$$
(64)

where there are 6-p Kronecker delta functions. Clearly then for an nMR of V the matrix V is built from a series of p-mode terms ranging from a very sparse two-mode component to the increasingly less sparse components as p increases to the value of n.

A detailed scheme based on the structure of the nMR of V has been adopted in MULTIMODE [75]. Other schemes have also been suggested, in particular we note the approach of Rauhut [81] where the reference VSCF state from which a virtual space is selected is chosen to be 'close' to energies of the states of interest. In the approach taken by Carter, Bowman and Handy and co-workers in MULTIMODE, the VSCF reference state is typically the ground vibrational state, although this is not a limitation of the code.

A truncation-recoupling contraction scheme considering the nMR of V has very recently been implemented in MULTIMODE by Carter, Handy and Bowman [163,164]. In this approach the potential is given by three different nMRs as follows:

$$V(\mathbf{Q}_{s}, \mathbf{Q}_{b}) = V^{n_{s}MR}(\mathbf{Q}_{s}) + V^{n_{b}MR}(\mathbf{Q}_{b}) + V^{n_{sb}MR}(\mathbf{Q}_{s}, \mathbf{Q}_{b})$$
(65)

where Q_s denotes system modes, and Q_b denotes bath modes. We note that partitioning modes into groups was suggested by Scribano and Benoit for the 2MR of the potential [36].

Assuming the same decomposition holds for all other terms in the exact Hamiltonian the full (assumed real) Hamiltonian matrix is given by

$$\mathbf{H} = \mathbf{H}_{s}^{n_{s}MR} + \mathbf{H}_{b}^{n_{b}MR} + \mathbf{H}_{sb}^{n_{sb}MR},$$
 (66)

where the notation should be clear ($\mathbf{H}_s^{n_s MR}$ and $\mathbf{H}^{n_b MR_b}$ are direct-product matrices of the same order) and where we note that the dimension of \mathbf{H} could be quite large. Proceeding as usual with a contraction scheme (see the Introduction and also Section 5) the matrices $\mathbf{H}_s^{n_s MR}$ and $\mathbf{H}_b^{n_b MR}$ are separately diagonalized and a truncated basis of the multi-dimensional eigenfunctions is used in the final step to obtain eigenfunctions of the full Hamiltonian. matrix. In this multi-dimensional basis the H-matrix is given by

$$\langle s'b'|H|sb\rangle = \varepsilon_{s}\delta_{b'b} + \varepsilon_{b}\delta_{s's} + \langle s'b'|H_{sb}|sb\rangle, \tag{67}$$

where ε_s and ε_b are the eigenvalues of $\mathbf{H}_s^{n_s MR}$ and $\mathbf{H}_b^{n_b MR}$, respectively.

As usual for this scheme to be effective the dimension of the H-matrix in the multi-dimensional basis must be much smaller than the dimension of the underlying basis used to obtain the system and bath eigenfunctions. A further, hoped-for efficiency comes from using different levels of mode coupling in the potentials describing the 'system' and the 'bath'. It is also possible to treat the off-diagonal $\langle s'b'|H_{sb}|sb\rangle$ terms if they are small with PT. In addition, off-diagonal terms can also be ignored entirely. The approach ignoring the off-diagonal terms but where the number of modes in the system is changed has already been investigated recently for the water dimer and trimer [163,164].

Before we conclude this subsection we note that MULTIMODE can do 'exact' calculations for J > 0, the quotes are added because the nMR is an approximation, albeit a highly controlled one. This capability was first demonstrated for HO_2 and H_2O [82], for which a 3MR is exact. A small selection of results of those test calculations for are shown in Table 1 where excellent agreement with benchmark results for HO_2 [165] is seen.

7. MCTDH

7.1. Introduction

Multiconfiguration Time-Dependent Hartree (MCTDH) is an algorithm for propagating

Table 1. J=0 energies and rotation constants from J=1 energies (cm⁻¹) for the first fifteen vibrational states of HO₂. Benchmark results are from an independent calculation done in Jacobi coordinates.

| | | MULTIM | IODE | | | Benchmark | | |
|-----|---------|--------|------|------|---------|-----------|------|------|
| No. | J=0 | A | В | C | J=0 | A | В | С |
| 1 | 1065.49 | 20.46 | 1.12 | 1.06 | 1065.49 | 20.46 | 1.12 | 1.06 |
| 2 | 1296.38 | 20.20 | 1.09 | 1.03 | 1296.38 | 20.20 | 1.10 | 1.04 |
| 3 | 2091.12 | 21.25 | 1.12 | 1.06 | 2091.12 | 21.24 | 1.13 | 1.06 |
| 4 | 2359.53 | 19.94 | 1.07 | 1.01 | 2359.54 | 19.94 | 1.07 | 1.01 |
| 5 | 2516.68 | 20.96 | 1.09 | 1.04 | 2516.68 | 20.95 | 1.11 | 1.04 |
| 6 | 3081.02 | 22.10 | 1.12 | 1.06 | 3081.02 | 22.08 | 1.14 | 1.06 |
| 7 | 3333.68 | 19.71 | 1.05 | 1.00 | 3333.69 | 19.70 | 1.05 | 0.99 |
| 8 | 3387.01 | 19.67 | 1.13 | 1.06 | 3387.01 | 19.66 | 1.13 | 1.06 |
| 9 | 3589.80 | 20.63 | 1.07 | 1.02 | 3589.80 | 20.61 | 1.09 | 1.02 |
| 10 | 3725.78 | 21.75 | 1.09 | 1.03 | 3725.77 | 21.73 | 1.12 | 1.04 |
| 11 | 4041.28 | 23.03 | 1.1 | 1.06 | 4041.29 | 23.00 | 1.14 | 1.06 |
| 12 | 4368.56 | 19.49 | 1.02 | 0.97 | 4368.57 | 19.49 | 1.03 | 0.97 |
| 13 | 4421.44 | 20.27 | 1.06 | 1.00 | 4421.46 | 20.25 | 1.07 | 1.00 |
| 14 | 4615.59 | 19.54 | 1.09 | 1.03 | 4615.59 | 19.53 | 1.10 | 1.03 |
| 15 | 4688.32 | 21.36 | 1.08 | 1.02 | 4688.34 | 21.34 | 1.10 | 1.02 |

wavefunctions of several dimensions. The MCTDH algorithm was introduced in 1990 by Meyer *et al.* [166]. A first comprehensive description of the method – together with the first non-trivial application (photodissociation of NOCl) – appeared two years later [167]. The basic theory of MCTDH has been discussed in great detail in two review articles [71,168]. Hence in the following only a brief overview of MCTDH theory is given, highlighting the features that give the method its power and flexibility.

In contrast to the other methods discussed in this review article, MCTDH is a time-dependent method. As such it is well suited to study problems where the essential physics is finished after a rather short propagation time, e.g. photodissociation or scattering processes. Moreover, the time-dependent approach has several advantages. The time-dependent Schrödinger equation poses an initial value rather than an eigenvalue problem. This makes it easier to concentrate on computing only desired quantities, a feature which is the more desirable the larger the system. However, turning to spectroscopy of bound systems, where the vibrations 'live' forever and the autocorrelation functions never vanish, it is at the first glance not obvious that a time-dependent method can compete with time-independent ones. Here, filter-diagonalisation (FD) [49,126,169] (see also Section 7.3) helps, as it allows one to compute accurate eigenvalues while requiring reasonably short propagation times.

Finally, a recent development of MCTDH, *improved relaxation* [141], allows one to solve the time-independent Schrödinger equation, i.e. to compute eigenvalues and eigenstates, while using all the

MCTDH machinery. However, improved relaxation fails to converge if the density of states is too high. For a larger system, more than four atoms, say, improved relaxation can determine only low lying states. It is an excellent method for obtaining ground states.

7.2. MCTDH theory

7.2.1. Wavefunction ansatz and equations of motion

The basis of the MCTDH method is the use of the following wavefunction ansatz to solve the time-dependent Schrödinger equation for a physical system with f degrees of freedom (DOFs) described by coordinates q_1, \ldots, q_f :

$$\Psi(q_1, \dots, q_f, t) = \Psi(Q_1, \dots, Q_p, t)
= \sum_{j_1=1}^{n_1} \dots \sum_{j_p=1}^{n_p} A_{j_1 \dots j_p}(t) \varphi_{j_1}^{(1)}(Q_1, t) \dots \varphi_{j_p}^{(p)}(Q_p, t)
= \sum_{j_1=1}^{n_1} A_j \Phi_j.$$
(68)

Equation (68) is a direct product expansion of p sets of orthonormal time-dependent basis functions $\{\varphi^{(\kappa)}\}$, known as *single-particle functions* (SPFs). The coordinate for each set of n_{κ} functions is a composite or 'logical' coordinate of one or more 'physical' coordinates

$$Q_{\kappa} = (q_a, q_b, \ldots). \tag{69}$$

Thus, the basis functions are d-dimensional, where d is the number of system coordinates that have been

combined together and treated as one 'particle'. (Typically d=1–4.) The third line of Equation (68) defines the composite index $J=(j_1,\ldots,j_p)$ and the *Hartree product* Φ_J . The ansatz looks similar to the standard wavepacket expansion [43,170,171], except that the SPFs provide a time-dependent basis set.

The SPFs are numerical functions and are represented in a standard way through a primitive basis

$$\varphi_{j_{\kappa}}^{(\kappa)}(Q_{\kappa},t) = \sum_{\ell_{1}=1}^{N_{1,\kappa}} \cdots \sum_{\ell_{d}=1}^{N_{d,\kappa}} c_{j_{\kappa}\ell_{1}\cdots\ell_{d}}^{(\kappa)}(t) \chi_{\ell_{1}}^{(1,\kappa)}(q_{1,\kappa}) \cdots \chi_{\ell_{d}}^{(d,\kappa)}(q_{d,\kappa}).$$
(70)

As primitive basis $\{\chi\}$ we usually employ a DVR.

The ansatz Equation (68) is not unique as both the coefficients and the SPFs are time-dependent. A modification of the wavefunction accomplished by changing the coefficients may alternatively be achieved by changing the SPFs. Hence one needs to introduce constraints which do not narrow the variational space but uniquely distribute time changes of the wavefunction to coefficients and SPFs. The simplest and most often used constraint reads

$$\left\langle \varphi_{j}^{(\kappa)} \middle| \dot{\varphi}_{\ell}^{(\kappa)} \right\rangle = 0.$$
 (71)

This constraint has a vivid interpretation. Any variation that can be done by changing the A-coefficients will be done by the A-coefficients. Only when the wavefunction moves into a direction which is not covered by the current SPF basis, the SPFs will change.

Using the ansatz Equation (68) and the constraint Equation (71), one obtains as variational solution to the time-dependent Schrödinger equation a coupled set of equations, one for the expansion coefficients:

$$i\mathbf{A} = \mathcal{K}\mathbf{A},$$
 (72)

and one for each set of SPFs

$$i\dot{\boldsymbol{\varphi}}^{(\kappa)} = (1 - P^{(\kappa)})(\boldsymbol{\rho}^{(\kappa)})^{-1} \mathcal{H}^{(\kappa)} \boldsymbol{\varphi}^{(\kappa)}. \tag{73}$$

A matrix notation has been used with the A-coefficients and SPFs written as vectors, i.e. $\varphi^{(\kappa)} = (\varphi_1^{(\kappa)}, \dots, \varphi_{n_{\kappa}}^{(\kappa)})^{\mathrm{T}}$.

The matrix K is the Hamiltonian operator represented in the basis of Hartree products

$$\mathcal{K}_{JL} = \langle \Phi_J \mid H \mid \Phi_L \rangle. \tag{74}$$

Thus Equation (72) has a similar form as the equations of motion for standard wavepacket propagation. The difference is that the Hamiltonian matrix is time-dependent due to the time-dependence of the SPFs.

The equations of motion for the SPFs contain three new entities. The first is the projector onto the space spanned by the SPFs

$$P^{(\kappa)} = \sum_{j} \left| \varphi_{j}^{(\kappa)} \rangle \langle \varphi_{j}^{(\kappa)} \right|. \tag{75}$$

The operator $(1 - P^{(\kappa)})$ ensures that the time-derivative of the SPFs is orthogonal to the space spanned by the functions.

For the other two new entities it is useful to introduce the *single-hole function*, $\Psi_a^{(\kappa)}$, which is the wavefunction associated with the *j*th SPF of the κ th particle. As the total wavefunction lies in the space spanned by the SPFs one can make use of the completeness relation and write

$$\Psi = \sum_{a} |\varphi_{a}^{(\kappa)}\rangle\langle\varphi_{a}^{(\kappa)}|\Psi\rangle = \sum_{a} \varphi_{a}^{(\kappa)} \Psi_{a}^{(\kappa)}.$$
 (76)

Using this notation, the *mean-field operator matrix*, $\mathcal{H}^{(\kappa)}$ can be written as

$$\mathcal{H}_{ab}^{(\kappa)} = \langle \Psi_a^{(\kappa)} \mid H \mid \Psi_b^{(\kappa)} \rangle. \tag{77}$$

The integration in the brackets is over all particles except κ . This operator on the κ th particle correlates the motion between the different sets of SPFs.

Finally, the *density matrix* $\rho^{(\kappa)}$ is

$$\rho_{ab}^{(\kappa)} = \langle \Psi_a^{(\kappa)} \mid \Psi_b^{(\kappa)} \rangle. \tag{78}$$

The density matrices, which enter the equations of motion for the SPFs, Equation (73), can be used to provide a useful measure of the quality of the calculation. In an analogous way to the use of density matrices in electronic structure theory, the eigenfunctions of this matrix are termed *natural orbitals* and the eigenvalues provide populations for these functions. The lower the population, the less important the function. The space spanned by the natural orbitals is equivalent to that of the original SPFs, and if the population of the highest natural orbital is such that the function is effectively not required for an accurate description of the evolving wavepacket, the MCTDH wavefunction is converged.

7.2.2. Efficiency and memory requirements

Standard wavepacket dynamics uses a wavefunction ansatz like that of Equation (68), except with a set of time-independent basis functions for each DOF rather than a set of time-dependent functions for each particle. While the number of basis functions may vary for each DOF, if N is representative of this number, then the wavefunction is represented by N^f expansion coefficients. This is the basis of the exponential

increase of computer resources with system size that plagues wavepacket dynamics. Assuming N=32, a four-dimensional system using double-precision complex arithmetic requires 16 MB of memory just to store one wavefunction, while a six-dimensional system requires 16 GB, and an eight-dimensional one 16 TB. Clearly this scaling severely limits the size of system treatable by these methods.

An MCTDH wavefunction, on the other hand, is very compact, since the number of SPFs per particle is usually much smaller than the number of grid points used to represent a SPF. The number of data-points needed to store one MCTDH wavefunction is

memory
$$\sim n^p + pnN^d$$
, (79)

where n and N are characteristic of the number of SPFs and grid points per DOF, respectively. The first term is the number of A-coefficients, and the second accounts for representing of the SPFs on the primitive grid. Of course, the number of SPFs, n, necessary for convergence increases with d, the degree of combination, but this increase is slow, often almost linear in d. To give an example we assume an eight-dimensional wavefunction with p=4, d=2, n=16, N=32. This wavefunction takes only 2 MB.

The reduction of memory requirement is paralleled by a reduction of numerical effort. The effort of the standard method scales like N^{f+1} while the MCTDH effort can be estimated by a sum of two terms:

effort
$$\sim c_1 p^2 n^{p+1} + c_2 p n N^{2d}$$
, (80)

where c_1 and c_2 are constants of proportionality. The first term is due to building the mean-field matrices and calculating the time-derivative of the A-coefficients. The second term is due to the operation of the Hamiltonian on the SPFs.

Thus if p is large, the effort for the algorithm is dominated by the building of the mean-field matrices. If p is small and d large, the second-term, the one for the propagation of the SPFs, dominates due to the high dimensionality of the functions. Again we see the trade between the effort required for the coefficients and the SPFs which can be altered by suitably combining DOFs together into particles, balancing the reduced effort due to low p with increased effort due to increasing N^d .

The evaluation of the Hamiltonian matrix elements \mathcal{K}_{JL} and the building of the mean fields requires f and (f-d) dimensional integrals to be done at every time step. If one would do these integrals by quadrature over the primitive grid, MCTDH would not be able to compete with other methods. A fast method for evaluating the multi-dimensional integrals is essential

for the success of MCTDH. One such fast algorithm is the CDVR method developed by Manthe [72]. However, we prefer to make use of a product representation of the Hamiltonian as already discussed in Section 2.2. The estimate of the effort, Equation (80), was derived while assuming a Hamiltonian in product form. The constants of proportionality, c_1 and c_2 , increase linearly with s, the number of terms of the product expansion Equation (3).

Finally we note that one usually employs the constant mean-field (CMF) integration scheme [71,172] when solving the MCTDH equations of motion. CMF takes advantage of the fact that the mean fields vary less rapidly than the SPFs or the A-coefficients and hence may be built using a coarser time-grid than the one used for propagating the A-coefficients and SPFs. The CMF integration scheme usually speeds up a calculation by factors between 3 and 12 as compared to solving the whole coupled set of equations with a standard all-purpose integrator.

7.3. Fourier spectra and filter diagonalisation

A photo-absorption spectrum, $\sigma(\omega)$, can be calculated from the Fourier transform of an autocorrelation function

$$\sigma(\omega) \propto \omega \operatorname{Re} \int_0^T \mathrm{d}t \, g(t) C(t) \, \exp[\mathrm{i}(E_0 + \omega)t],$$
 (81)

where E_0 denotes the ground-state energy and where the autocorrelation function is defined as

$$C(t) = \langle \Psi(0) | \Psi(t) \rangle.$$
 (82)

A derivation is given e.g. in [173]. See also [174,175] for a formula which provides absolute intensities. The initial state to be propagated, $\Psi(0)$, is the ground state operated by the transition operator, $\Psi(0) = \hat{\mu} \Psi_{GS}$. The damping function g(t) is introduced to reduce spurious effects (Gibbs phenomenon) introduced by the finite propagation time T. A convenient choice is

$$g(t) = \cos\left(\frac{\pi t}{2T}\right)\Theta\left(1 - \frac{|t|}{T}\right),\tag{83}$$

where Θ denotes the Heaviside step function. The damping function smoothly forces the integrand to be 0 at T. The effect of the filter g(t) is that the exact spectrum is convoluted with $\tilde{g}(\omega)$, the Fourier transform of g(t).

$$\tilde{g}(\omega) = \frac{4\pi T \cos(\omega T)}{(\pi - 2\omega T)(\pi + 2\omega T)}.$$
(84)

The full width at half maximum of this function is 27.3 cm⁻¹ ps. Hence one has to propagate for 27 ps to achieve a resolution of 1 cm⁻¹. Although one may save a factor of 2 by using the 'T/2 trick' [71,176,177], such a long propagation is in general not realistic. Fourier spectra mostly show a resolution between 10 and 100 cm⁻¹. There are several applications for which such a resolution is sufficient, e.g. when spectra are congested and only the envelope but not the individual lines are of interest [178–183].

To improve the situation, Filter Diagonalisation (FD) was proposed by Neuhauser [126]. This hybrid approach promises to combine the advantages of wavepacket propagation with obtaining accurate eigenvalues by diagonalisation. The idea is to build a suitable small basis to represent the Hamiltonian in a particular energy range. To do this a wavepacket is propagated and Fourier transformed on a set of, typically equidistant, energies E_1, \ldots, E_L . The Hamiltonian is then diagonalized in this set of filtered states. In practice one does not Fourier transform the wavepacket (which would be very elaborate), but works with the autocorrelation function only [49,126]. FD is hence a more clever way to analyse the autocorrelation function as compared to just Fourier transform it.

One problem is that FD can produce spurious eigenvalues. This can be a serious problem if there are errors in the propagation, something which cannot be avoided in MCTDH. A way of removing the spurious eigenvalues is to solve the eigenvalue problem using different filtering functions and different variational principles. Eigenvalues that appear in all the calculations are real, the others spurious [169].

This MCTDH/FD scheme works well for small systems like CO₂ [169] or HO₂ [184]. For large systems, however, one is plagued with a large number of spurious eigenvalues and it is difficult to identify and remove them. For large systems one hence adopts a different strategy. One no longer tries to generate the full spectrum by one calculation, but concentrates on the computation of the eigenenergy of a selected state. An initial state is built which is as close to the sought state as possible. The FD spectrum of such a state will show one strong line, the sought eigenstate, and several small lines, real or spurious, of no further interest. This approach, which may be called *filtering by the initial state*, was successfully applied in [175,185].

7.4. Improved relaxation

Improved relaxation [141,168] is essentially a multi-configuration self-consistent field (MCSCF) approach.

Initially this method was implemented as a modification of the well-known relaxation method [186], which is based on propagation in negative imaginary time. Using the CMF propagation scheme, the propagation of the coefficients is decoupled from the propagation of the SPFs. The modification consisted in determining the A-vector no longer by relaxation but by diagonalisation of the Hamiltonian matrix K. Hence the name: improved relaxation.

To derive the working equations of the improved relaxation algorithm in a rigorous way, we employ the variational principle of time-independent quantum mechanics

$$\delta \left\{ \langle \Psi | H | \Psi \rangle - E \left(\sum_{J} A_{J}^{*} A_{J} - 1 \right) - \sum_{\kappa=1}^{p} \sum_{j,\ell=1}^{n_{\kappa}} \epsilon_{j\ell}^{(\kappa)} \left(\langle \varphi_{j}^{(\kappa)} | \varphi_{\ell}^{(\kappa)} \rangle - \delta_{j\ell} \right) \right\} = 0, \quad (85)$$

where it is assumed that Ψ is of MCTDH form, Equation (68). The Lagrange multipliers E and $\epsilon_{jl}^{(\kappa)}$ are introduced to keep the A-vector normalized and the SPFs orthonormal.

Varying A_J^* yields

$$\sum_{L} \mathcal{K}_{JL} A_L = E A_J. \tag{86}$$

Hence the coefficient vector is obtained as an eigenvector of the Hamiltonian represented in the basis of the configurations Φ_J .

From Equation (76) it follows that a variation of the total wavefunction with respect to a SPF yields a single-hole function. Using this and varying the expression (85) with respect to $\langle \varphi_k^{(\kappa)} |$ gives [187,188]

$$\sum_{l=1}^{n_{\kappa}} \mathcal{H}_{j\ell}^{(\kappa)} \varphi_{\ell}^{(\kappa)} = \sum_{\ell=1}^{n_{\kappa}} \epsilon_{j\ell}^{(\kappa)} \varphi_{\ell}^{(\kappa)}. \tag{87}$$

Projecting Equation (87) onto $\varphi_k^{(\kappa)}$, one arrives at

$$\epsilon_{jk}^{(\kappa)} = \sum_{\ell=1}^{n_{\kappa}} \left\langle \varphi_k^{(\kappa)} \middle| \mathcal{H}_{j\ell}^{(\kappa)} \middle| \varphi_\ell^{(\kappa)} \right\rangle, \tag{88}$$

and from Equations (87) and (88) it then follows

$$(1 - P^{(\kappa)}) \sum_{\ell=1}^{n_{\kappa}} \mathcal{H}_{j\ell}^{(\kappa)} \varphi_{\ell}^{(\kappa)} = 0.$$
 (89)

A variationally optimal solution is found when Equations (86) and (89) are satisfied simultaneously.

Since Equation (89) holds for each *j*, it must hold for any linear combination of these equations as well.

To arrive at a form similar to the equation of motion (73) we insert the inverse of the density matrix

$$\frac{\partial \varphi_j^{(\kappa)}}{\partial \tau} := -(1 - P^{(\kappa)}) \sum_{k,\ell=1}^{n_{\kappa}} (\rho^{(\kappa)})_{jk}^{-1} \mathcal{H}_{k\ell}^{(\kappa)} \varphi_{\ell}^{(\kappa)} = 0, \quad (90)$$

where $\tau = -it$. This suggests a convenient way to solve the variational Equation (89). The SPFs are relaxed, i.e. propagated in negative imaginary time, until their time derivative is sufficiently small.

The improved relaxation method proceeds as follows. First the user has to define an initial state which should have some overlap with the eigenstate one wants to compute. The Hamiltonian is then diagonalized in the basis of the configurations Φ_J of the initial state. After the diagonalisation, the mean-fields $\mathcal{H}^{(\kappa)}$ are built and the SPFs are relaxed over a suitable time interval. The Hamilton matrix K is then rebuilt in the new configurations (Hartree products) and diagonalized. The whole process is iterated until convergence is achieved. If the ground state is computed, the selection of the eigenvector of the Hamiltonian matrix is simple: one takes the eigenvector with lowest energy. When excited states are computed, that eigenvector is taken which corresponds to the wavefunction with largest overlap with the initial state.

As the dimension of the space spanned by the configurations is rather large – typical values range from 2000 to 2000,000 – the Davidson algorithm [136] is employed for the diagonalisation of \mathcal{K} . The preconditioner for the Davidson step is, as usual, the diagonal of the matrix. When excited states are computed one may additionally improve the preconditioner by inverting, say, a 1000×1000 , block around the energy of interest. This accelerates the convergence of the Davidson iterations.

An MCTDH propagation always works, whatever the numbers of SPFs. If there are too few configurations, the propagation will be less accurate but usually still describes the overall features rather well. This is in contrast to *improved relaxation* which fails to converge when the configuration space is too small. There is no problem in computing the ground state, this will always work, but converging to excited states becomes more difficult the higher the excitation energy, or — more precisely — the higher the density of states.

After each diagonalisation of \mathcal{K} one obtains an approximation to the sought exact eigenstate of the Hamiltonian. This approximate eigenstate contains, of course, contributions from many exact eigenstates. Problematic are here contributions from eigenstates with energies below the one sought. If these contributions are not small, the algorithm 'notices' that the energy can be lowered by optimizing the SPFs for the

low-energy contributions to the approximate eigenstate. The energy hence decreases in the following improved relaxation iterations until – based on the overlap criterion – another eigenvector of \mathcal{K} is taken. In this case, the energy jumps up and is lowered again in the following iteration steps. The energy keeps on oscillating [141] and a convergence cannot be achieved. Presently the only known solution to this problem is to increase the numbers of SPFs. A larger configuration space, however, quickly makes the calculation unfeasible, at least for large systems. Note that if convergence is achieved, the computed energies are variational, i.e. they are upper bounds to the exact ones [189,190].

7.5. Example

Improved relaxation has been applied quite successfully to a number of problems, see e.g. [141,175,191]. For four-atom systems (6D) it is in general possible to compute all eigenstates of interest. For a system as large as $H_5O_2^+$ (15D) it was only possible to converge low-lying states [175]. In this case filter diagonalisation was used to compute the energies of high lying states.

The improved relaxation algorithm may be used in block form, i.e. one may start with a block of initial vectors which then converge collectively to a set of eigenstates. Formally the different wavefunctions are treated as electronic states of one 'super wavefunction'. The mean-fields are state-averaged mean-fields and the Davidson routine is replaced by a block-Davidson one.

As an example [192] we discuss the results of computing the lowest 40 eigenstates of HONO by block improved relaxation. The HONO molecule [185,191,193] is described in valence coordinates. The six DOFs are combined to four particles where the first two (uncombined) particles represent the out-of-plane torsion and the O-H stretch. The underlying primitive grid consists of 28×10^6 points. The results of two calculations, using different numbers of SPFs, are shown in Table 2.

The first run used the SPF set 10/5/30/20 (i.e. 30,000 configurations altogether) and took 3 h CPU time and 560 MB RAM on a 3.2 GHz Pentium 4 PC, the second run with 12/5/42/28 SPFs took 11 h CPU time and 1340 MB RAM. The second calculation was done mainly to show that already the first one is sufficiently converged. The energy is lowered by less than $0.3 \, \mathrm{cm}^{-1}$ for all states except one, the 35th excited state is lowered by $0.758 \, \mathrm{cm}^{-1}$.

Block-relaxation requires more SPFs to converge than single-vector relaxation, because in blockrelaxation the SPFs are optimized to represent all

Table 2. The lowest 40 vibrational eigenvalues of HONO. The results of two calculations and their differences are displayed (see text). All energies are in cm⁻¹. A ground state energy of 4367.634515 cm⁻¹ is subtracted from all entries of column 2 and 3.

| # | 10/5/30/20 | 12/5/42/28 | Diff. |
|-----|------------|------------|-------|
| 0 | 0.000 | 0.000 | 0.000 |
| 1 | 93.974 | 93.972 | 0.002 |
| 2 3 | 600.873 | 600.871 | 0.002 |
| 3 | 710.625 | 710.621 | 0.004 |
| 4 | 796.000 | 795.997 | 0.003 |
| 5 | 944.116 | 944.108 | 0.008 |
| 6 | 1055.385 | 1055.384 | 0.001 |
| 7 | 1188.079 | 1188.070 | 0.009 |
| 8 | 1267.609 | 1267.598 | 0.011 |
| 9 | 1306.604 | 1306.595 | 0.009 |
| 10 | 1312.761 | 1312.736 | 0.025 |
| 11 | 1385.263 | 1385.247 | 0.016 |
| 12 | 1405.545 | 1405.510 | 0.035 |
| 13 | 1547.458 | 1547.431 | 0.027 |
| 14 | 1574.851 | 1574.821 | 0.030 |
| 15 | 1640.887 | 1640.884 | 0.003 |
| 16 | 1690.034 | 1690.006 | 0.028 |
| 17 | 1726.050 | 1726.009 | 0.041 |
| 18 | 1761.638 | 1761.581 | 0.057 |
| 19 | 1779.466 | 1779.377 | 0.089 |
| 20 | 1829.017 | 1829.013 | 0.004 |
| 21 | 1858.242 | 1858.210 | 0.032 |
| 22 | 1897.807 | 1897.580 | 0.227 |
| 23 | 1902.886 | 1902.838 | 0.048 |
| 24 | 1961.701 | 1961.558 | 0.143 |
| 25 | 2002.404 | 2002.323 | 0.081 |
| 26 | 2025.384 | 2025.381 | 0.003 |
| 27 | 2049.045 | 2048.967 | 0.078 |
| 28 | 2120.019 | 2120.002 | 0.017 |
| 29 | 2136.567 | 2136.276 | 0.291 |
| 30 | 2154.080 | 2153.897 | 0.183 |
| 31 | 2210.633 | 2210.622 | 0.011 |
| 32 | 2240.933 | 2240.825 | 0.108 |
| 33 | 2291.196 | 2291.096 | 0.100 |
| 34 | 2306.477 | 2306.460 | 0.017 |
| 35 | 2322.512 | 2321.754 | 0.758 |
| 36 | 2339.333 | 2339.225 | 0.108 |
| 37 | 2339.687 | 2339.416 | 0.271 |
| 38 | 2370.709 | 2370.415 | 0.294 |
| 39 | 2376.419 | 2376.401 | 0.017 |

considered states simultaneously, while in single-vector relaxation they are optimized to represent a particular state. However, we found that the required increase in the number of SPFs is rather small. In fact, block-relaxation is in general more efficient than performing single-vector relaxations for each of the block states. But the memory consumption of block-relaxation is considerably larger (560 MB compared to 20 MB for the present example). This may limit the applicability of block-improved-relaxation when turning to larger systems.

8. Applications

We now present applications of these various approaches to CH_4 , CH_3OH , and $H_5O_2^+$.

8.1. *CH*₄

First we present benchmark calculations of the vibrational energies of CH₄ using the contractediterative method outlined in Section 5. This method has been applied to several molecules. It was first tested on HOOH where its efficiency was established [157]. Yu has contracted the bend but not the stretch basis to compute energy levels for several molecules [159,160,194]. Lee and Light contracted both bend and stretch bases [161,162]. In this subsection we summarize calculations done on methane using the contracted basis-iterative method [54,158]. To compute vibrational energy levels of methane we use 4+1 Radau vectors [195] and their associated polyspherical coordinates. There are three polar angles, θ_i (i = 1, 2, 3) and two dihedral angles $\phi_i(i=2,3)$. The coordinates are defined explicitly in [158]. The calculation consists of three basic steps: (1) solve the stretch problem; (2) solve the bend problem; (3) combine solutions of the stretch and bend problems to determine energy levels of methane.

The stretch KEO is simple,

$$T_{\text{str}} = \sum_{i=0}^{3} -\frac{1}{2\mu_i} \frac{\partial^2}{\partial r_i^2}.$$
 (91)

Stretch eigenfunctions are computed with a primitive basis of products of 10 1DPODVR functions [196,197]. These in turn are built from a sinc DVR basis in the range [1.2 a_0 , 7.0 a_0] [198]. We discard all the PODVR points whose potential energy is above 30,000 cm⁻¹. This reduces the basis size from 10,000 to 5049. We retain all the stretch levels up to 20,000 cm⁻¹. We calculate stretch levels separately for the two irreps of G_6 . This is done using the symmetry adapted Lanczos (SAL) algorithm [199,200].

The bend KEO is easily extracted from Equation (27). We use a parity adapted basis of products of spherical harmonic basis functions with $l_{\rm max} = m_{\rm max} = 25$ and the SAL to compute eigenvalues and eigenvectors for different irreps of G_6 [201]. The spherical harmonic basis has the advantage that it enables one to easily compute matrix elements of the bend KEO using a very compact form (see Equation (27)). It also makes it possible to calculate wavefunctions with non-zero amplitude at singularities of the bend KEO (not a problem for methane). Odd parity bend wavefunctions are computed and

the SAL is used to obtain levels labelled by irreducible representations of the G_6 group. We retain bend eigenfunctions up to the polyad $n_2 + n_4 = 6$, i.e. all those that differ from the bend ground state by less than $9098.0 \,\mathrm{cm}^{-1}$. This basis is somewhat larger than the one used in [158]), but the lower levels in [158] are all well converged. G_2 symmetry is exploited when solutions of the bend and stretch subproblems are coupled to obtain the final energy levels. See [158] for more detail.

To use the eigenfunctions of the bend and stretch problems to compute energy levels and wavefunctions of methane we must compute and store them. Computing accurate eigenvalues with the Lanczos algorithm is extremely easy, but to obtain the corresponding eigenvectors one must be careful to limit the influence of numerical error. To do so one determines, separately for each eigenvector, the number of Lanczos vectors to be used to compute it. This is the iteration depth [54,135]. To store the eigenvectors we use a technique called compaction [125,155,158]. We need C (see Section 5) to compute the final levels. Rather than storing C we store the important rows of $\tilde{\mathbf{C}}$, where

$$\tilde{C}_{\beta b} \equiv \sum_{l} T_{l\beta} C_{lb}. \tag{92}$$

All elements of rows of $\tilde{C}_{\beta b}$ that correspond to quadrature points at which the potential is high will be small and can be discarded. For methane the ratio of the number of retained quadrature points to the number of direct product quadrature points is about 0.01. Storing the compacted version of \tilde{C} makes it much easier to use the contracted basis.

We use the T8 potential surface described in [202]. The size of the basis of products of stretch and bend functions is 113,360. These functions are divided into two symmetry groups. There are 57,548 A symmetry functions and 55812 B symmetry functions. The final basis set is about 6 orders of magnitude smaller than the original primitive basis! Energy levels are reported in Table 3. The column labelled Basis 1 is obtained with basis 1 of [158]. Levels above about 5100 cm⁻¹ are better converged than those given in [158] where not all the bend basis functions in the $n_2 + n_4 = 6$ polyad were included. The states for which the energy in [158] differs most from the energy reported here are those with the largest $n_2 + n_4$ values. Many high stretch overtone and combination states are also well converged but are not included in the table. There is good agreement between experimental and computed levels, indicating that the potential is realistic.

Using this contracted basis-iterative eigensolver approach it has also been possible to compute

vibrational energy levels of CH₅⁺. The details of this calculation will be published soon [203]. As was done for methane we use Radau vectors and their associated angles and a large spherical harmonics basis. A primitive bend basis with more than 200 million functions is required to converge the bend levels. The energy level pattern differs from those obtained previously.

These results for CH₄ were recently used as benchmarks to test the accuracy of vibrational energies obtained using MULTIMODE with nMR, n = 2, 3, 4, 5representations of the potential (and the vibrational angular momentum terms) [204]. The MULTIMODE calculations did not make full use of the T_d symmetry and instead were done with $C_{2\nu}$ symmetry. Rather than repeat the lengthy table of comparisons in [204] we graphically show the convergence and accuracy of MULTIMODE energies as n varies by plotting the root mean square (rms) deviation from the benchmark energies in Figure 1. As seen the rms difference decreases by roughly an order of magnitude from $32\,\mathrm{cm}^{-1}$ for the 2MR calculations to less than $0.5\,\mathrm{cm}^{-1}$ for the 5MR calculations which are thus in quite good agreement with the benchmark energies. For several states the error in the 2MR calculations is between 70 and 80 cm⁻¹. The sizes of the four VCI matrices, obtained with a direct-product basis but with excitation restrictions, for the 4MR and 5MR calculations were 54,522, 47,312, 47,312 and 42,459. These are moderately large and so the block Davidson method was used to get the eigenvalues. We also examined the degeneracy of E and F fundamentals and showed that correct degeneracy to within less than 1 cm⁻¹ was achieved with a 4MR and to within less than 0.1 cm⁻¹ with 5MR.

As noted above MULTIMODE can also perform VCI calculations for J > 0 and energies (cm⁻¹) for J = 1 agree well with the benchmark results, as shown in Table 4.

Next we briefly consider the most recent MULTIMODE-RPH calculations for CH₃OH [91]. After the validation of the MM-RPH approach for H₂O₂ [87] methanol was one of the first new applications of the method [88], using a fairly low-level of electronic structure obtained along the reaction path, i.e. the torsion coordinate. In the latest work a semi-global potential energy surface fit to roughly 20,000 CCSD(T)/aug-cc-pVTZ energies was used [91]. Convergence studies indicated that the energies are converged to within a cm⁻¹ or less with a 5MR for the potential and a 3MR for vibrational angular momentum terms. This is quite impressive given that this is a floppy 12 degree of freedom molecule.

Table 3. Vibrational energy levels of CH_4 in cm^{-1} .

| A1 5248.437 F2 5383.087 F1 5401.639 | 33 F1 33 F1 33 F1 33 F1 33 F1 | (00)(04) A1 | | |
|---|---|----------------------------|--|----------|
| 5383.08 | | | | |
| | | (00)(13) F2 (00)(13) F1 | 0.000 (00)(13) F2 0.004 (00)(13) F1 | 0.009 |
| 5435.438 | | (00)(13) E | | 0.003 |
| 5440.667 | | (00)(13) F2 (00)(13) F1 | 0.003 (00)(13) F2 0.009 (00)(13) F1 | |
| 5455.056 | | (00)(13) F2 | | 0.008 |
| 5473.525 | | (00)(13) F1 | | 3 -0.001 |
| 5493.222 | | (10)(02) A1 | <u> </u> |) -0.001 |
| 5533 970 | | (10)(02) F2 | 0.005 (10)(02) F2 0.004 (10)(02) E | _ |
| 5585.364 | | (01)(02) F2 | | 0.014 |
| 5603.628 | | (01)(02) A1 | | |
| 5612.274 | | (01)(02) F2 | | 0.012 |
| 5612.441 | | (01)(02) F1 | | 0.013 |
| 5613.648 | | (01)(02) E | | 0.044 |
| 5620.131 | | (00)(22) A1 | 0.045 (00)(22) A1 | _ |
| 5623.415 | | (01)(02) F1 | | 0.042 |
| 5625.321 | | (01)(02) F2 | | 0.045 |
| 5625.411 | | (00)(22) E | | 0.039 |
| 5651.123 | | (00)(22) F2 | | 0.038 |
| 5662.914 | | (00)(22) E | | 0.001 |
| 5664.262 | | (00)(22) F1 | | 0.000 |
| 5672.243 | | (00)(22) A2 | _ | 0.002 |
| 5676.626 | | (00)(22) F2 | | 0.002 |
| 5689.821 | | (00)(22) A1 | = | 0.003 |
| 5699.177 | | (00)(22) E | 0.033 $(00)(22)$ E | |
| 5743 523 | | (10)(11) F1 | | 0.030 |
| 5782.245 | | (20)(00) A1 | | 0.001 |
| 5817.233 | | (01)(11) F2 | | 0.001 |
| 5820.510 | | (01)(11) F1 | _ | 0.002 |
| 5828.639 | | (01)(11) E | | 0.013 |
| 5830.893 | | (01)(11) A1 | = | 0.013 |
| 5838.452 | | (01)(11) E | | 0.013 |
| 5838.477 | | (01)(11) A2 | _ | 1.430 |
| 5839.469 | | (01)(11) F2 | | 1.367 |
| 5842.714 | | (01)(11) F1 | | 1.414 |
| 5853.956 | | (11)(00) F2 | _ | 1.296 |
| 5872.466 | | (00)(31) F2 | 1.317 (00)(31) F2 | _ |

Table 3. Continued.

| (0)(10) F1 (862,687 (862,290 (0.457 (0.0)(12) F1 (118,994 (111,495 (1.0)(1.0) F1 (116,997 (117,418) (1.0)(1.0) F1 (116,997 (117,418) (1.0)(1.0) F1 (117,418) (1.0)(1.0) F2 (1.0) | $(n_1n_3)(n_2n_4)$ Sym | Basis I | Basis II | Basis I – II | $(n_1n_3)(n_2n_4)$ Sym | Basis I | Basis II | Basis I – II | $(n_1n_3)(n_2n_4)$ Sym | Basis I | Basis II | Basis I – II |
|--|------------------------|----------|----------|--------------|------------------------|----------|----------|--------------|------------------------|----------|----------|--------------|
| NE 6864,175 (682.56) 6864,175 (682.66) (1)(10) E1 7368,478 7364,175 74 6864,175 (682.94) 0.255 (10)(12) E2 (10)(12) E2 7136,117,288 1.25 (10)(10) E2 738,438 736,478 74 736,478< | (01)(03) F1 | 6862.637 | 6862.290 | 0.347 | (00)(32) E | 7133.304 | 7131.695 | 1.609 | | 7364.787 | 7364.377 | 0.410 |
| A.J. 686,172 686,172 686,173 686,173 686,173 686,173 686,173 686,173 686,173 687,110 687,134 687,110 | (01)(03) E | 6862.856 | 6862.400 | 0.456 | (01)(12) F1 | 7136.597 | 7136.117 | 0.480 | (11)(10) F1 | 7365.047 | 7364.825 | 0.222 |
| F. P. 6880, 594 (2007) | (01)(03) A1 | 6864.172 | 6863.915 | 0.257 | (01)(12) F2 | 7139.734 | 7137.288 | 2.446 | (11)(10) F2 | 7368.767 | 7368.408 | 0.359 |
| FP 686,884 686,854 686 | (00)(23) F2 | 6887.011 | 6882.949 | 4.062 | (00)(32) A1 | 7140.679 | 7139.156 | 1.523 | | 7391.612 | 7389.133 | 2.479 |
| Fig. 6907.357 690.2587 690.2587 600.00(41) Fig. 748.1527 748.514 718.514 | (01)(03) F2 | 6896.894 | 8896.508 | 0.386 | | 7147.895 | 7144.203 | 3.692 | (00)(41) F1 | 7404.788 | 7402.065 | 2.723 |
| θ (8) (10) (8) (8) (8) (8) (8) (8) (8) (8) (9) (10) (10) (10) (10) (10) (10) (10) (10 | (00)(23) F1 | 6907.072 | 6902.987 | 4.085 | (01)(12) F1 | 7149.514 | 7148.974 | 0.540 | (00)(41) F2 | 7418.752 | 7416.101 | 2.651 |
| FE 6910.08 609.07.60 6037 (01/01) F 7151.121 7160.926 (02)10.18 744.77 744.167 <th< td=""><td>(01)(03) E</td><td>6907.488</td><td>6907.256</td><td>0.232</td><td>(11)(01) A1</td><td>7150.344</td><td>7150.077</td><td>0.267</td><td>(00)(41) F1</td><td>7432.489</td><td>7429.666</td><td>2.823</td></th<> | (01)(03) E | 6907.488 | 6907.256 | 0.232 | (11)(01) A1 | 7150.344 | 7150.077 | 0.267 | (00)(41) F1 | 7432.489 | 7429.666 | 2.823 |
| F1 6915 098 6014 299 7.79 (11)(1) F1 7157,548 7157,548 7157,548 7157,948 7157,948 7157,748 7157,948 7157,948 7157,948 7150,000 729 (11)(1)(1) F1 718,128 0.237 (10)(30) A2 746,6396 745,718 746,6108 746,1019 746,6108 746,6108 746,1010 746,314 746,31 | (01)(03) F2 | 6910.086 | 6906.769 | 0.317 | (11)(01) F2 | 7151.121 | 7150.926 | 0.195 | (02)(10) E | 7440.532 | 7440.358 | 0.174 |
| A2 698 001 7 601 7 40 0.541 (11)(01) Ε 7188.182 71579 45 0.237 (10)(30) A 1 746.53 4 746.5108 F 1 6921.095 6917.713 3.32 (00)(32) F 1 7188.723 74.243 (10)(30) A 1 746.534 7477.110 F 2 6922.736 692.737 6.540 (00)(32) F 1 7181.632 4.243 (10)(30) B 1 7477.209 7477.110 F 2 6922.738 692.490 0.560 (00)(32) F 1 7183.202 4.259 (00)(10) F 1 753.394 753.394 A1 692.738 692.490 6.50 (00)(22) F 1 7193.20 7199.467 3.850 (02)(10) F 1 753.394 | (01)(03) F1 | 6915.098 | 6914.299 | 0.799 | (11)(01) F1 | 7157.548 | 7156.866 | 0.682 | (00)(41) F2 | 7446.767 | 7444.167 | 2.600 |
| 6921.085 6917.880 4.465 (10)(32) FI 7168.870 716.53.20 3.640 (10)(30) AI 746.6108 4.466.108 6922.285 6917.880 4.465 (10)(32) FI 718.882 4.253 (10)(10) FI 748.828 747.110 6922.270 6922.270 6.560 (00)(32) FI 718.128 717.380 4.253 (02)(10) FI 748.828 7487.111 6955.215 6949.996 5.229 (00)(32) FI 718.028 4.253 (02)(10) FI 750.103 753.991 6955.216 695.229 (00)(32) FI 720.524 7223.755 4.253 (02)(10) FI 750.599 750.599 695.216 695.197 60.00(32) FI 720.524 7224.524 720.101 750.599 750.101 750.599 695.217 60.01(20) FI 720.23.755 7224.524 720.101 774.44 701.01 750.594 750.594 750.594 750.594 750.594 750.504 750.504 750.504 750.504 750.504 750.504 750.504 | (01)(03) A2 | 6918.017 | 6917.476 | 0.541 | (11)(01) E | 7158.182 | 7157.945 | 0.237 | (10)(30) A2 | 7465.976 | 7465.778 | 0.198 |
| 6922.285 6912.285 6912.285 6912.285 692.285 6912.285 692.297 | (01)(03) F1 | 6921.095 | 6917.713 | 3.382 | (00)(32) F1 | 7168.870 | 7165.230 | 3.640 | (10)(30) A1 | 7466.304 | 7466.108 | 0.196 |
| 6922.779 692.210 0.569 (00)(32) F2 7181.629 717.360 4.269 (02)(10) F2 749.88.78 498.781 6922.479 692.210 0.569 (00)(32) F2 7180.482 735.954 7531.091 7501.091 7501.091 7501.092 7501.002 692.41 692.41 6949.996 5.229 (00)(32) F2 7205.49 738.50 (02)(10) A2 753.934 753.941 753.954 753.954 753.954 753.954 753.954 753.954 753.954 753.954 753.954 754.959 756.056 756.056 756.056 750.056< | (00)(23) F2 | 6922.285 | 6917.880 | 4.405 | (00)(32) E | 7181.075 | 7176.832 | 4.243 | (10)(30) E | 7477.209 | 7477.110 | 0.099 |
| 6925.417 692.4907 0.510 (00)(32) FI 7199.048 7185.795 4.253 (02)(10) FI 7501.091 7501.021 6925.417 692.641 6.540 (00)(32) FI 7199.467 4.229 (02)(10) FI 7501.091 7501.021 6955.225 6945.996 5.239 (00)(32) FI 7205.049 7200.832 4.217 (02)(10) AI 756.929 756.929 6957.415 6951.979 5.436 (10)(21) F2 7222.944 6811 (01)(30) F2 756.925 756.949 6962.104 6956.920 5.184 (00)(32) E 7226.625 4.20 (01)(30) F2 756.925 756.949 696.01 696.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.01 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 6.02 <td>(01)(03) A1</td> <td>6922.779</td> <td>6922.210</td> <td>0.569</td> <td>(00)(32) F2</td> <td>7181.629</td> <td>7177.360</td> <td>4.269</td> <td>(02)(10) F2</td> <td>7498.828</td> <td>7498.781</td> <td>0.047</td> | (01)(03) A1 | 6922.779 | 6922.210 | 0.569 | (00)(32) F2 | 7181.629 | 7177.360 | 4.269 | (02)(10) F2 | 7498.828 | 7498.781 | 0.047 |
| 6927.081 6926.541 0.540 (00)(32) F1 7193.320 7189.028 4.920 (01)(10) A2 7533.941 7533.941 6957.225 6949.996 5.229 (00)(32) F2 7205.344 3850 (02)(10) A1 7546.912 7546.912 6956.326 6951.302 5.078 (00)(32) E 7205.494 0.811 (01)(30) F1 7546.912 7562.767 6966.013 6960.834 5.179 (00)(23) E 7224.569 7220.130 4419 (01)(30) F1 7562.927 7562.927 7562.927 7562.927 7562.927 7562.927 666.013 7562.927 | (01)(03) F2 | 6925.417 | 6924.907 | 0.510 | (00)(32) A1 | 7190.048 | 7185.795 | 4.253 | (02)(10) F1 | 7501.091 | 7501.022 | 690.0 |
| 6956.325 6949.996 5.229 (00)(32) F2 7205.049 738.90 (02)(10) AI 7540.011 7539.91 6956.380 6951.302 5.078 (00)(32) F2 7205.049 7200.832 4.217 (01)(30) F1 756.292 756.921 6956.310 6956.320 5.184 (00)(32) F2 720.625 4.200 (01)(30) F2 7570.195 756.927 756.927 6962.104 6956.920 5.184 (00)(32) F2 7226.625 4.200 (01)(30) F2 7579.44 756.924 7570.195 756.947 6978.71 600(32) A2 7226.625 4.200 (01)(30) F2 7579.44 757.94 7579.45 7579.44 7579.43 7579.44 7579.44 7579.44 7579.44 7579.43 7579.44 7579.43 7579.44 <td>(01)(03) E</td> <td>6927.081</td> <td>6926.541</td> <td>0.540</td> <td>(00)(32) F1</td> <td>7193.320</td> <td>7189.028</td> <td>4.292</td> <td>(02)(10) A2</td> <td>7533.954</td> <td>7533.941</td> <td>0.013</td> | (01)(03) E | 6927.081 | 6926.541 | 0.540 | (00)(32) F1 | 7193.320 | 7189.028 | 4.292 | (02)(10) A2 | 7533.954 | 7533.941 | 0.013 |
| 6956.380 6951.302 5.078 (100/32) E 7205.049 7200.832 4.217 (101/10) AI 7546.932 7546.932 7546.912 6957.418 6951.348 (957.418 6957.418 6951.014 (101/30) F1 754.935 7546.912 6957.416 6957.418 6957.418 696.013 696.013 696.014 675.749 7570.195 756.924 6966.013 6966.013 696.034 5.179 (00)(32) R2 723.4569 723.156 724.499 0.245 (01)(30) F2 7579.451 7579.452 7579.451 7579.451 7579.451 7579.451 7579.451 7579.451 7579.451 7579.451 7579.451 7579.451 7750.752.744 7750.751 7550.752 7283.254 | (00)(23) E | 6955.225 | 6949.996 | 5.229 | (00)(32) F2 | 7203.317 | 7199.467 | 3.850 | (02)(10) E | 7540.011 | 7539.991 | 0.020 |
| 6957.415 6951.979 5.436 (10)(21) F2 7223.755 722.944 0.811 (01)(30) F1 756.942 756.767 6962.104 6951.979 5.436 (10)(21) F2 7223.755 722.944 0.811 (01)(30) F2 7570.195 756.942 6966.104 6952.104 (01)(30) F2 724.394 7244.149 0.245 (01)(30) F2 7575.798 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.98 7575.79 7576.79 7575.79 7575.79 7575.79 7575.74 7775.75 7774.74 7774.74 7774.74 <td>(00)(23) A1</td> <td>6956.380</td> <td>6951.302</td> <td>5.078</td> <td>(00)(32) E</td> <td>7205.049</td> <td>7200.832</td> <td>4.217</td> <td>(02)(10) A1</td> <td>7546.935</td> <td>7546.912</td> <td>0.023</td> | (00)(23) A1 | 6956.380 | 6951.302 | 5.078 | (00)(32) E | 7205.049 | 7200.832 | 4.217 | (02)(10) A1 | 7546.935 | 7546.912 | 0.023 |
| 6962.104 6956.920 5.184 (00)(32) E 7226.625 4.200 (01)(30) F2 7570.195 7569.942 6966.013 696.834 5.179 (00)(32) A2 7236.865 4.200 (01)(30) F2 7570.195 7575.948 6986.013 696.013 698.818 6986.834 5.179 (00)(21) F2 7244.629 0.527 (01)(30) F2 7579.481 7579.135 698.818 6983.620 5.198 (10)(21) F2 7244.629 0.527 (00)(00) A1 742.638 763.736 691.835 6991.835 6991.835 6991.835 6901.837 (02)(01) F2 7245.840 0.042 (00)(50) B2 7579.431 7579.431 702.836 7024.749 0.33 (02)(01) F1 7286.337 7285.239 7285.239 7289.237 7600 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 7650.703 | (00)(23) F2 | 6957.415 | 6951.979 | 5.436 | (10)(21) F2 | 7223.755 | 7222.944 | 0.811 | (01)(30) F1 | 7562.952 | 7562.767 | 0.185 |
| 6966.013 6966.0834 5.179 (00)(32) A2 7234.569 7230.150 4419 (01)(30) F1 7575.788 7575.488 6986.013 6966.013 6966.013 6966.013 6966.013 6966.013 6966.013 6968.818 7124.149 0.245 (01)(30) F2 7579.451 75 | (00)(23) F1 | 6962.104 | 6956.920 | 5.184 | (00)(32) E | 7230.825 | 7226.625 | 4.200 | (01)(30) F2 | 7570.195 | 7569.942 | 0.253 |
| 6978,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,8,751 697,115 724,4,146 724,4,146 0.245 (00)(50) F2 7579,451 7579,135 764,055 764,055 764,055 769,138 769,111 764,059 760,004 760,001 762,041 762,043 762,043 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,650 7650,534 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7651,702 7650,703 760,003 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 7650,703 | (00)(23) F1 | 6966.013 | 6960.834 | 5.179 | (00)(32) A2 | 7234.569 | 7230.150 | 4.419 | (01)(30) F1 | 7575.798 | 7575.488 | 0.310 |
| 6988.818 6988.620 5.198 (10)(21) F1 7245.156 7244.629 0.527 (00)(06) A1 7642.658 765.784 6991.853 6991.115 0.738 (10)(21) F1 7245.156 7244.629 0.527 (00)(50) B 7647.223 7647.055 7020.853 7020.912 0.941 (02)(01) F1 7286.099 0.005 (00)(50) A1 7651.02 7651.526 7021.853 7024.749 0.837 (20)(10) F1 7286.099 0.062 (00)(50) A1 7651.02 7651.02 7025.586 7024.749 0.837 (20)(10) F1 7286.099 7289.24 7006 (00)(60) F2 7658.01 7651.05 7025.586 7024.749 0.837 (02)(01) F2 7287.29 7289.29 7289.20 7600(06) F2 7658.01 7659.03 7035.699 7034.00 7025.389 0.073 (00)(06) F2 7725.154 7736.04 7082.115 7082.115 7081.01 7328.089 7327.973 0.116 7000(06) F1 7736.04 | (00)(23) F2 | 6978.751 | 6973.629 | 5.122 | (02)(01) F2 | 7244.394 | 7244.149 | 0.245 | (01)(30) F2 | 7579.451 | 7579.135 | 0.316 |
| 6991.853 6991.115 0.738 (10)(21) F2 7266.403 0.428 (00)(50) E 7647.223 7447.055 7020.853 7004.032 5.326 (02)(01) E 7285.89 728.534 0.042 (00)(50) A1 7651.730 7650.537 7021.853 7024.032 5.326 (02)(01) F1 7286.099 0.062 (00)(60) A2 7650.537 7650.537 7025.86 7024.79 0.837 (20)(01) A1 7285.39 7287.225 0.065 00)(06) F2 7659.222 7659.034 7035.699 7034.960 0.739 (02)(01) F2 7289.39 7289.273 0.066 (00)(06) F2 7659.034 7659.034 7045.10 7046.377 0.713 (02)(01) F2 7289.295 0.073 (00)(06) F2 7659.034 7659.034 7086.115 7081.216 7057.359 0.653 (01)(21) F2 7328.089 7328.49 0.016 7774.69 7764.74 7756.934 7084.115 7084.235 7010.010 7328.089 7328.169 0.073 <td>(00)(23) F1</td> <td>6988.818</td> <td>6983.620</td> <td>5.198</td> <td>(10)(21) F1</td> <td>7245.156</td> <td>7244.629</td> <td>0.527</td> <td>(00)(06) A1</td> <td>7642.658</td> <td>7636.784</td> <td>5.874</td> | (00)(23) F1 | 6988.818 | 6983.620 | 5.198 | (10)(21) F1 | 7245.156 | 7244.629 | 0.527 | (00)(06) A1 | 7642.658 | 7636.784 | 5.874 |
| 7009.358 7004.032 5.326 (02)(01) E 7285.540 0.042 (00)(50) A2 7650.730 7650.557 7009.358 7004.032 5.326 (02)(01) F1 7286.099 0.062 (00)(50) A1 7651.702 7651.526 7025.886 7024.749 0.837 (02)(01) B1 7287.325 0.105 (00)(06) F2 7658.691 7651.526 7025.886 7024.749 0.837 (02)(01) B2 7287.39 7289.226 0.066 (00)(06) F2 7659.222 7659.034 7047.090 7046.377 0.713 (02)(01) F2 7289.99 7289.26 0.073 (00)(06) F2 7659.22 7659.03 7058.012 7057.359 0.653 (01)(21) F1 738.089 7328.40 0.178 (00)(06) F2 7755.94 7756.9 | (10)(12) E | 6991.853 | 6991.115 | 0.738 | (10)(21) F2 | 7266.831 | 7266.403 | 0.428 | (00)(50) E | 7647.223 | 7647.055 | 0.168 |
| 7021.853 7020.912 0.941 (02)(01) F1 7286.009 0.062 (00)(50) A1 7651.526 7025.886 7024.749 0.837 (20)(10) E 7287.225 0.105 (00)(06) F2 7658.691 7652.860 7035.699 7034.960 0.739 (02)(01) A1 7289.339 7289.273 0.066 (00)(06) F2 7658.691 7652.860 7047.09 7044.90 0.713 (02)(01) F2 7289.999 7289.276 7659.034 7765.214 7058.012 7058.012 7060.066 F1 7765.014 7765.214 7765.214 7088.012 7054.016 7725.143 7725.143 7765.934 7765.934 7765.934 7765.934 7776.604 7082.115 7081.781 0.347 (01)(21) F2 7328.48 7328.46 0.017 (00)(06) F1 776.73 776.243 7104.536 7104.099 0.437 (01)(21) F1 7339.241 7338.454 0.017 (00)(06) F2 776.75.79 776.76 7111.5247 7114.710 | (00)(23) F2 | 7009.358 | 7004.032 | 5.326 | (02)(01) E | 7285.582 | 7285.540 | 0.042 | (00)(50) A2 | 7650.730 | 7650.557 | 0.173 |
| 7025.386 7024.749 0.837 (20)(10) E 7287.325 0.105 (00)(66) F2 7658.691 7652.860 7035.699 7034.749 0.837 (20)(10) A1 7289.39 7289.273 0.066 (00)(60) E 7659.222 7659.034 7036.09 7044.377 (01)(21) F1 7318.898 7328.926 0.073 (00)(06) E 7659.34 7690.165 7080.505 7080.231 0.274 (01)(21) F2 7224.018 7323.840 0.178 (00)(06) F1 7756.934 7750.644 7080.505 7080.235 7081.889 7328.18 0.178 (00)(06) F1 7756.934 7750.644 7082.115 7082.115 7328.089 7327.973 0.116 (00)(06) F1 7756.934 7750.43 7094.235 7093.869 0.366 (02)(01) F1 7328.18 7328.169 0.079 (00)(06) F1 7763.753 7763.243 7104.536 7104.099 0.437 (01)(21) F1 7339.081 7340.063 0.265 00)(06) F1 7763.753 <td< td=""><td>(10)(12) F1</td><td>7021.853</td><td>7020.912</td><td>0.941</td><td>(02)(01) F1</td><td>7286.071</td><td>7286.009</td><td>0.062</td><td>(00)(50) A1</td><td>7651.702</td><td>7651.526</td><td>0.176</td></td<> | (10)(12) F1 | 7021.853 | 7020.912 | 0.941 | (02)(01) F1 | 7286.071 | 7286.009 | 0.062 | (00)(50) A1 | 7651.702 | 7651.526 | 0.176 |
| 7035.699 7034.960 0.739 (02)(01) A1 7289.339 7289.273 0.066 (00)(50) E 7659.222 7659.034 7047.090 7046.377 0.713 (02)(01) F2 7289.996 7289.926 0.073 (00)(06) E 7696.374 7690.165 7047.090 7046.377 0.713 (01)(21) F1 7318.898 7318.659 0.239 (00)(06) F2 7725.154 7718.794 7088.012 7087.321 0.0274 (01)(21) F2 7324.018 7327.349 0.116 (00)(06) F1 7756.934 7756.934 7082.115 7081.781 0.334 (02)(01) F2 7328.089 7327.349 0.116 (00)(06) F1 7756.934 7756.934 7094.235 7094.235 7091.000 F1 7328.089 7328.464 0.017 (00)(06) F1 7756.934 7757.243 7104.536 7104.536 7104.09 0.344 (01)(21) E 7339.241 7338.658 0.258 (00)(06) F1 7827.744 7111.757 7114.710 0.537 (01) | (10)(12) A1 | 7025.586 | 7024.749 | 0.837 | (20)(10) E | 7287.330 | 7287.225 | 0.105 | (00)(06) F2 | 7658.691 | 7652.860 | 5.831 |
| 7047.090 7046.377 0.713 (02)(01) F2 7289.996 7289.926 0.073 (00)(06) E 7696.374 7690.165 7047.090 7046.377 0.713 (01)(21) F1 7318.898 7318.659 0.239 (00)(06) F2 7725.154 7718.794 7088.012 7057.359 0.653 (01)(21) F2 7324.018 7323.840 0.178 (00)(06) F1 7756.934 7750.604 7082.115 7081.781 0.334 (01)(21) F2 7328.089 7327.1973 0.116 (00)(06) F1 7756.934 7756.243 7094.235 7093.869 0.366 (02)(01) F1 7328.148 7328.169 0.079 (00)(06) F1 774.669 7768.315 7104.536 7104.099 0.437 (01)(21) F1 7340.328 7340.646 0.253 (00)(06) F2 7824.863 7113.240 7112.724 0.476 (01)(21) F2 7340.346 0.253 (00)(06) F2 7861.814 7855.028 7118.843 7118.444 0.369 (01)(21) F2 7349.245 | (10)(12) F2 | 7035.699 | 7034.960 | 0.739 | (02)(01) A1 | 7289.339 | 7289.273 | 990.0 | (00)(50) E | 7659.222 | 7659.034 | 0.188 |
| 7088.012 7057.359 0.653 (01)(21) F1 7318.898 7318.659 0.239 (00)(06) F2 7725.154 7718.794 77080.505 7080.231 0.274 (01)(21) F2 7324.018 7323.840 0.178 (00)(06) F1 7756.934 7750.604 7785.315 7081.781 0.334 (02)(01) F2 7328.089 7327.973 0.116 (00)(06) E 7763.753 7757.243 7794.235 7093.869 0.366 (02)(01) F1 7328.248 7328.169 0.079 (00)(06) F1 7756.934 7756.934 7756.934 7756.934 7756.934 7757.243 7094.235 7093.869 0.366 (02)(01) F1 7328.248 7328.169 0.079 (00)(06) F1 7783.335 7824.863 7111.757 7111.433 0.324 (01)(21) F1 739.081 7338.658 0.265 (00)(06) F1 7833.334 7825.744 7855.028 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7344.844 0.531 (00)(06) F2 7870.586 7863.836 7121.894 7121.111 0.783 (01)(21) F1 7356.587 7356.241 0.346 0.465 (00)(06) F2 7874.747 7867.861 7128.876 7128.876 7128.876 7128.814 0.553 (01)(21) F2 7360.835 7360.83 | (10)(12) E | 7047.090 | 7046.377 | 0.713 | (02)(01) F2 | 7289.999 | 7289.926 | 0.073 | (00)(00) E | 7696.374 | 7690.165 | 6.209 |
| 7080.505 7080.231 0.274 (01)(21) F2 7324.018 7323.840 0.178 (00)(06) F1 7756.934 7750.604 7082.115 7081.781 0.334 (02)(01) F2 7328.089 7327.973 0.116 (00)(06) E 7763.753 7757.243 7094.235 7093.869 0.366 (02)(01) F1 7328.248 7328.169 0.079 (00)(06) F1 7774.669 7768.315 7104.536 7104.099 0.437 (01)(21) A1 7339.081 7338.464 0.617 (00)(06) F1 7832.315 7824.863 7111.757 7111.433 0.324 (01)(21) A1 739.081 7338.658 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F2 7340.328 7340.063 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7346.549 0.531 (00)(06) F2 7861.814 7855.028 7121.894 7121.111 0.783 (01)(21) A2 7345.375 7348.659 0.586 (00)(06) F2 7874.747 7867.861 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 7128.143 0.763 (01)(21) B 7357.761 7357.296 0.465 7128.441 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (10)(12) A2 | 7058.012 | 7057.359 | 0.653 | (01)(21) F1 | 7318.898 | 7318.659 | 0.239 | (00)(06) F2 | 7725.154 | 7718.794 | 6.360 |
| 7082.115 7081.781 0.334 (02)(01) F2 7328.089 7327.973 0.116 (00)(06) E 7763.753 7757.243 7094.235 7093.869 0.366 (02)(01) F1 7328.248 7328.169 0.079 (00)(06) A1 7774.669 7768.315 7104.536 7104.536 7104.099 0.437 (01)(21) A1 7339.081 7338.464 0.617 (00)(06) F1 7832.315 7824.863 7111.757 7111.433 0.324 (01)(21) E 7339.241 7338.658 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) F2 7861.814 7855.028 7121.894 7121.111 0.783 (01)(21) B 7349.245 7348.659 0.586 (00)(06) E 7874.747 7867.861 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 7128.441 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (20)(01) F2 | 7080.505 | 7080.231 | 0.274 | (01)(21) F2 | 7324.018 | 7323.840 | 0.178 | (00)(06) F1 | 7756.934 | 7750.604 | 6.330 |
| 7094.235 7093.869 0.366 (02)(01) F1 7328.248 7328.169 0.079 (00)(06) A1 7774.669 7768.315 7104.536 7104.099 0.437 (01)(21) A1 7339.081 7338.464 0.617 (00)(06) F1 7832.315 7824.863 7111.757 7111.433 0.324 (01)(21) E 7339.241 7338.658 0.583 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F2 7340.328 7340.063 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) F2 7835.779 7828.485 7118.843 7118.474 0.369 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 7121.894 7121.111 0.783 (01)(21) A1 7356.587 7356.241 0.346 0.465 0.465 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 7128.441 0.573 (01)(21) F2 7360.835 7360.836 7360.835 7360.836 7360.835 7360.835 7360.835 7360.835 7360.835 7360.835 7360.836 7360.835 7360.835 7360.835 7360.835 7360.835 7360.835 7360.836 | (01)(12) F1 | 7082.115 | 7081.781 | 0.334 | (02)(01) F2 | 7328.089 | 7327.973 | 0.116 | (00)(00) | 7763.753 | 7757.243 | 6.510 |
| 7104.536 7104.099 0.437 (01)(21) A1 7339.081 7338.464 0.617 (00)(06) F1 7832.315 7824.863 7111.757 7111.433 0.324 (01)(21) E 7339.241 7338.658 0.583 (00)(06) A1 7833.34 7825.744 7113.200 7112.724 0.476 (01)(21) F2 7340.328 7340.063 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) A2 7861.814 7855.028 7118.843 7118.474 0.369 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 7121.894 7121.111 0.783 (01)(21) A1 7356.587 7356.241 0.346 (00)(06) E 7874.747 7867.861 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 7128.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (01)(12) F2 | 7094.235 | 7093.869 | 0.366 | (02)(01) F1 | 7328.248 | 7328.169 | 0.079 | (00)(00) | 7774.669 | 7768.315 | 6.354 |
| 7111.757 7111.433 0.324 (01)(21) E 7339.241 7338.658 0.583 (00)(06) A1 7833.34 7825.744 7113.200 7112.724 0.476 (01)(21) F2 7340.328 7340.063 0.265 (00)(06) F2 7835.779 7828.485 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) A2 7861.814 7855.028 7118.843 7118.474 0.369 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 7121.894 7121.111 0.783 (01)(21) A1 7356.287 7356.241 0.346 (00)(06) E 7874.747 7867.861 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 7128.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (01)(12) E | 7104.536 | 7104.099 | 0.437 | (01)(21) A1 | 7339.081 | 7338.464 | 0.617 | (00)(06) F1 | 7832.315 | 7824.863 | 7.452 |
| 2 7113.200 7112.724 0.476 (01)(21) F2 7340.328 7340.063 0.265 (00)(06) F2 7835.779 7828.485 1 115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) A2 7861.814 7855.028 1 115.247 7114.710 0.537 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 1 1121.894 7121.111 0.783 (01)(21) B 7349.245 7348.659 0.586 (00)(06) E 7874.747 7867.861 1 128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 0.465 0.465 1 128.876 7128.113 0.763 (01)(21) B 7357.761 7357.296 0.465 0.396 | (01)(12) A2 | 7111.757 | 7111.433 | 0.324 | (01)(21) E | 7339.241 | 7338.658 | 0.583 | (00)(00) A1 | 7833.334 | 7825.744 | 7.590 |
| 1 7115.247 7114.710 0.537 (01)(21) F1 7340.921 7340.646 0.275 (00)(06) A2 7861.814 7855.028 2 7118.843 7118.474 0.369 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 3 7121.894 7121.111 0.783 (01)(21) E 7349.245 7348.659 0.586 (00)(06) E 7874.747 7867.861 3 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 5 7128.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (01)(12) F2 | 7113.200 | 7112.724 | 0.476 | (01)(21) F2 | 7340.328 | 7340.063 | 0.265 | (00)(06) F2 | 7835.779 | 7828.485 | 7.294 |
| 2 7118.843 7118.474 0.369 (01)(21) A2 7345.375 7344.844 0.531 (00)(06) F2 7870.586 7863.836 (118.842 7118.474 0.783 (01)(21) E 7349.245 7348.659 0.586 (00)(06) E 7874.747 7867.861 (118.856.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 (00)(06) E 7874.747 7867.861 (118.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 0.465 (0.465 (0.763 (01)(21) E 7357.761 7360.439 0.396 | (01)(12) F1 | 7115.247 | 7114.710 | 0.537 | (01)(21) F1 | 7340.921 | 7340.646 | 0.275 | (00)(06) A2 | 7861.814 | 7855.028 | 982.9 |
| 7121.894 7121.111 0.783 (01)(21) E 7349.245 7348.659 0.586 (00)(06) E 7874.747 7867.861 0.7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 0.346 0.465 0.763 (01)(21) E 7357.761 7357.296 0.465 0.465 0.573 (01)(21) F2 7360.835 7360.439 0.396 | (01)(12) F2 | 7118.843 | 7118.474 | 0.369 | (01)(21) A2 | 7345.375 | 7344.844 | 0.531 | (00)(06) F2 | 7870.586 | 7863.836 | 6.750 |
| 7128.653 7126.531 2.122 (01)(21) A1 7356.587 7356.241 7128.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 | (01)(12) A1 | 7121.894 | 7121.111 | 0.783 | (01)(21) E | 7349.245 | 7348.659 | 0.586 | (00)(00) E | 7874.747 | 7867.861 | 988.9 |
| (01)(21) E 7128.876 7128.113 0.763 (01)(21) E 7357.761 7357.296 F1 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 | (01)(12) F2 | 7128.653 | 7126.531 | 2.122 | (01)(21) A1 | 7356.587 | 7356.241 | 0.346 | | | | |
| F1 7129.014 7128.441 0.573 (01)(21) F2 7360.835 7360.439 | (01)(12) E | 7128.876 | 7128.113 | 0.763 | (01)(21) E | 7357.761 | 7357.296 | 0.465 | | | | |
| | | 7129.014 | 7128.441 | 0.573 | (01)(21) F2 | 7360.835 | 7360.439 | 0.396 | | | | |

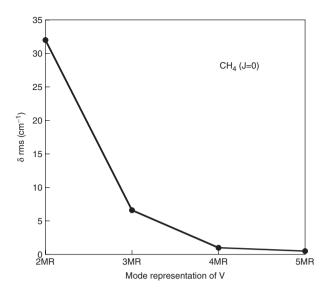


Figure 1. Root mean square difference between nMR MULTIMODE energies and benchmark ones [158] versus n for roughly 50 energies $4500 \,\mathrm{cm}^{-1}$ above the zero-point energy.

Table 4. Comparison of 5MR MULTIMODE rovibrational energies of CH_4 (cm⁻¹) for J=1 with benchmark results of [158].

| State | Benchmark | MULTIMODE |
|-------------|-----------|-----------|
| Zero point | 10.43 | 10.43 |
| (00)(01) A2 | 1312.41 | 1312.50 |
| (00)(01) F2 | 1317.25 | 1317.34 |
| (00)(01) F1 | 1326.73 | 1326.82 |
| (00)(01) E | 1327.03 | 1327.12 |
| (00)(10) F2 | 1543.79 | 1543.88 |
| (00)(10) F1 | 1543.91 | 1544.01 |
| (00)(02) F1 | 2600.02 | 2600.58 |
| (00)(02) F1 | 2615.61 | 2616.36 |
| (00)(02) E | 2622.05 | 2622.82 |
| (00)(02) F2 | 2626.64 | 2627.37 |
| (00)(02) A2 | 2634.83 | 2635.61 |
| (00)(02) F2 | 2641.58 | 2642.25 |
| (00)(02) F1 | 2644.03 | 2644.68 |
| (00)(11) E | 2835.80 | 2836.67 |
| (00)(11) F1 | 2837.85 | 2838.73 |
| (00)(11) F2 | 2843.53 | 2844.39 |
| (00)(11) A2 | 2848.87 | 2849.75 |
| (00)(11) F2 | 2856.65 | 2857.36 |
| (00)(11) E | 2858.02 | 2858.73 |
| (00)(11) F1 | 2860.39 | 2861.11 |
| (00)(11) A1 | 2862.34 | 2863.06 |
| (10)(00) F1 | 2924.16 | 2924.32 |
| (01)(00) A2 | 3022.81 | 3023.02 |
| (01)(00) F2 | 3023.36 | 3023.58 |
| (01)(00) F1 | 3024.50 | 3024.72 |
| (01)(00) E | 3024.56 | 3024.78 |
| (00)(20) F1 | 3074.26 | 3074.68 |
| (00)(20) F2 | 3075.80 | 3076.22 |
| (00)(20) F1 | 3075.82 | 3076.50 |

A sample of the energies and comparison with experiment is given in Table 5. As seen there is good agreement with experiment and the pattern of tunnelling splittings is well reproduced by the calculations.

8.2. Vibrational spectroscopy of the Zundel cation

The Zundel cation $(H_5O_2^+)$ is the smallest protonated water cluster in which an excess proton is shared between different water molecules. This species has been the target of much research effort during the last years owing to its key role as a limiting structure of the hydrated proton in bulk water. The development and improvement of new experimental techniques to measure accurate IR spectra of ionic species in the gas phase has led to an accurate characterisation of the IR spectrum of $H_5O_2^+$ [205–209]. Several theoretical studies have appeared in parallel to provide the necessary basis for the rationalisation and understanding of the measured spectra. In particular 4D simulations that included the position of the central proton (x, y, z) and the oxygen-oxygen distance (R)can be found in the literature [210]. Those simulations yielded a too high frequency for the fundamental excitation of the central proton along the O-O axis (asymmetric stretch v_{ap}), while the out-of-axis (x, y)motions of the proton presented too low frequencies, to the extent that the frequencies of these states presented an inverted ordering pattern. VCI calculations showed later that a full-dimensional treatment would be necessary to regain the correct ordering of the fundamental modes related to the central proton due to the strong coupling of these motions to displacements of the surrounding water molecules [208,211]. The IR spectrum of the Zundel cation recorded with the messenger atom technique features a doublet structure in the spectral region around 1000 cm⁻¹ with a splitting of about 120 cm⁻¹ [208]. The origin of the doublet remained an unsolved issue that could not be reproduced nor explained by means of VCI [208] nor by classical dynamics simulations [212,213], although there was accumulated evidence that the v_{ap} fundamental would be mostly responsible for the highest intensity line of the doublet. Moreover, the lowest frequency region of the spectrum below 800 cm⁻¹, where the fundamental states of the most anharmonic motions are found, had not yet been accessed experimentally. Other lines appearing between 1000 and 2000 cm⁻¹ had not been yet conclusively assigned and their origin was unclear. This inherently complex nature of $H_5O_2^+$ is caused by the floppy, anharmonic nature of the interatomic potential governing the motion of the cation, in which internal

Table 5. Comparison of experimental and MULTIMODE-RPH vibrational energies (cm $^{-1}$) and splittings (right section of the table) of CH₃OH. Left-hand entries are the non-degenerate A₁ and A₂ levels in the G₆ point group arising from the MM-RPH A' and A" levels from calculations in C_s symmetry, respectively. Right-hand entries are the relative energies of the corresponding degenerate E levels in the G₆ point group.

| Level | Symm. | Expt. | MM-RPH | Expt. | MM-RPH |
|--------------------|------------------|---------------------|--------|--------|--------|
| ν_8 | A_1 | 1033.8 | 1026.3 | 8.27 | 10.50 |
| ν_7 | A_1 | 1074.5 | 1079.9 | 4.61 | 7.62 |
| ν_6 | A_1 | 1335.8 | 1321.9 | / | 20.01 |
| ν_5 | A_1 | 1454.5 | 1446.8 | / | 8.72 |
| ν_4 | A_1 | 1486.4 | 1483.8 | / | -2.72 |
| ν_3 | A_1 | 2844.2 | 2839.5 | 9.07 | 9.09 |
| ν_2 | A_1 | 2999.0 | 2986.3 | -3.26 | -0.68 |
| ν_1 | A_1 | 3683.9 | 3675.0 | 6.30 | 6.90 |
| $2\nu_8$ | A_1 | 2054.7 | 2038.7 | 8.90 | 12.85 |
| $2v_7$ | A_1 | 2140.6 | 2152.5 | +2.29 | +5.67 |
| $2\nu_6$ | A_1 | 2632.0 | 2678.4 | / | -26.47 |
| $v_7 + v_8$ | A_1 | 2097.3 | 2097.6 | +4.65 | +6.36 |
| ν_{11} | A_2 | 1164.0 | 1156.5 | -7.43 | -5.42 |
| ν_{10} | A_2 | 1481.8 | 1475.1 | / | -4.24 |
| ν ₉ | A_2 | 2966.6 | 2961.5 | -5.48 | -3.43 |
| $\nu_8 + \nu_{12}$ | $\overline{A_2}$ | 1231.6 ^e | 1234.9 | -76.93 | -56.74 |
| $v_8 + v_{11}$ | A_2 | 2190.6 | 2176.7 | -7.70 | -3.19 |

rotation and long range angular motions take place at relatively low vibrational energies. In this context the study of the IR spectroscopy and dynamics of the Zundel cation was undertaken by the MCTDH method [175,214,215].

8.2.1. Set-up of the Hamiltonian operator

The simulation of floppy systems like the Zundel cation requires the use of a proper set of coordinates that is able to describe largely anharmonic, angular motions, e.g. torsions or internal rotations. To this end a set of polyspherical vectors is used to describe the configuration of the system. For a detailed derivation of the KEO in the set of polyspherical coordinates used for $H_5O_2^+$ see [215]. The set of polyspherical vectors that are used to describe the $H_5O_2^+$ cation are shown in Figure 2. This is a semi-rigid system (at least in the range of vibrational energies involved in the linear IR spectrum), i.e. no singularities in the kinetic energy exist for the configurations accessible to the system. This allows the use of standard one-dimensional DVRs (see Appendix in [215]). The 15 coordinates are defined as follows: the distance between the centres of mass of both water molecules (R), the position of the central proton with respect to the centre of mass of the water dimer (x, y, z), the Euler angles defining the relative orientation between the two water molecules

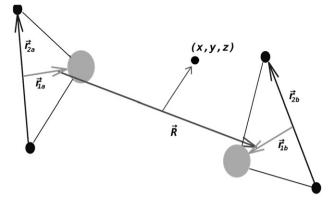


Figure 2. Jacobi description of the $H_5O_2^+$ system. The vector **R** connects the two centres of mass of the water monomers. The position of the central proton is measured from the centre of mass of the water dimer.

(waggings: γ_A , γ_B ; rockings: β_A , β_B ; internal relative rotation: α) and the Jacobi coordinates which account for the particular configuration of each water molecule $(R_{1(A,B)}, R_{2(A,B)}, \theta_{(A,B)})$, where R_{1x} is the distance between the oxygen atom and the centre of mass of the corresponding H₂ fragment, R_{2x} is the H–H distance and θ_x is the angle between these two vectors.

The generation of a potential energy operator (PEO), i.e. a working approximation to the full PES, is a rather challenging task of its own, in particular for problems as large as the present one (15D). A way to represent such high-dimensional surfaces is the so-called Cartesian reaction surface (CRS) approach, in which a small number of coordinates (typically two) are selected as defining the reaction surface while the potential for the rest of the coordinates is given as a second order Taylor expansion whose coefficients are dependent on the position of the coordinates in the first group. The CRS approach has the required product form, Equation (3), meaning that it is an efficient PES represention for MCTDH, and in fact it has been used in a number of MCTDH applications [216]. Since the approach is based on orthogonal, rectilinear coordinates, the KEO is simply diagonal. Such an approximation can be used if a large number of coordinates behave almost harmonic and they are only weakly coupled among each other and to the reaction coordinates. Such an approximation breaks down however for loosely bonded systems that feature relatively soft, anharmonic potentials. Protonated water clusters belong to this kind of system and the relatively straightforward CRS is therefore not applicable.

Another possibility to efficiently represent the 15D PES $V(\mathbf{Q})$ is to use a variant of the *n*-mode representation of the potential given by Equation (2).

In a similar way to the definition of the MCTDH ansatz. Equation (68), it is advantageous to work here with combined ('logical') coordinates instead of uncombined ('physical') coordinates when defining the PES expansion. The logical Q_{κ} coordinates, defined in Equation (69), refer then to composite coordinates of one or more physical coordinates. In this case we call the $V^{(n)}$ terms combined clusters (CC). The numerical advantage in the representation of the PES obtained when using combined coordinates instead of single coordinates is analysed in [215]. The definition of meaningful mode combinations should be possible in most cases. Highly correlated coordinates should be grouped together so that their correlation is taken into account already by the first order CCs. Such a definition would be based on chemical commonsense, e.g. coordinates belonging to the same chemical group or to the same molecule in a cluster are good candidates to be combined. The application of this scheme does not only lead to numerical advantage in the representation of the PES with respect to the uncombined scheme. Its application in MCTDH will be (much) more efficient if the mode combination used to define the PES is the same as that used in defining the wavefunction. In such a case the first order CCs $V^{(1)}(Q_i)$ are full product grids defined in the space of the Q_i combined coordinate. The higher order terms are given by potfit representations of the total product grids $V^{(n>1)}(Q_i,...)$. A potfit representation can be defined in terms of combined modes and in such a case the SPPs are product grids in the space of each Q_i . This strategy can efficiently be applied as long as the full product grid of the *n*th order cluster fits into memory. In practice this means around seven single coordinates, roughly corresponding to CCs of third order. As a final remark, the convergence of the *n*-mode representation is highly dependent on the set of coordinates used and the grouping of coordinates into particles. With our set of polyspherical coordinates, which correctly account for the fundamental motions of the cation in terms of rotations, torsions, etc., and a clever grouping of the coordinates, we obtain a relatively good potential already with a second-order expansion. Further details are given in the following section.

8.2.2. Representation of the potential energy surface for $H_5O_9^+$

The PEO for the ${\rm H}_5{\rm O}_2^+$ system is constructed using the full-dimensional PES developed by Huang *et al.* [217]. Such a PES is based on several ten thousands of CC energy calculations which are combined with a clever fitting algorithm based on polynomials that respect the total permutational symmetry of the system. In order

to construct an efficient representation of that PES for the $H_5O_2^+$ cation employing the approach described above one must start by defining the combined modes, also called particles, that shall be used. In the present case the following five multi-dimensional modes are selected: $Q_1 = [z, \alpha, x, y]$, $Q_2 = [R, u_{\beta_A}, u_{\beta_B}]$, $Q_3 = [\gamma_A, \gamma_B]$, $Q_4 = [R_{1A}, R_{2A}, u_{\theta_{1A}}]$ and $Q_5 = [R_{1B}, R_{2B}, u_{\theta_{1B}}]$ where $u_x = \cos x$. It is convenient that coordinates x, y and α are grouped together due to symmetry conserving reasons [215]. Modes Q_2 and Q_3 contain the rocking and wagging coordinates, respectively. Modes Q_4 and Q_5 contain the Jacobi coordinates representing the internal configuration of each water molecule. The specific form of the used PEO is given by

$$\tilde{V}(\mathbf{Q}) = V^{(0)} + \sum_{i=1}^{5} V_i^{(1)}(Q_i) + \sum_{i=1}^{4} \sum_{j=i+1}^{5} V_{ij}^{(2)}(Q_i, Q_j) + V_{z23}^{(3)}(z, Q_2, Q_3),$$
(93)

where the modes $Q_1 \cdots Q_5$ have been defined above. The $V^{(0)}$ term is the energy at the reference geometry. The $V_i^{(1)}$ terms are the intra-group potentials obtained by keeping the coordinates in other groups at the reference geometry, while the $V_{ij}^{(2)}$ terms account for the group–group correlations. The potential with up to second-order terms gives already a very reasonable description of the system. The $V_{223}^{(3)}$ term accounts for three-mode correlations between the displacement of the central proton, the distance between both water molecules and the angular wagging and rocking motions. To ensure the full symmetry of the PES, an average over several reference geometries was used. For details see [215].

The quality of the PES expansion can be assessed by monitoring the expectation values of the different terms of the *n*-mode representation with respect to the ground vibrational state $|\Psi_0\rangle$. These values are given in cm⁻¹ in Table 6. The sum of the first order $\langle \Psi_0 | V^{(1)}(Q_i) | \Psi_0 \rangle$ terms is close to 6800 cm⁻¹, half the ZPE, indicating that they carry the major weight in the description of the PES. The second-order clusters introduce the missing correlation between modes. They have expectation values one order of magnitude smaller than the first-order terms with one exception, the matrix element arising from the $V^{(2)}(Q_1, Q_2)$ potential. This can be easily understood by noting that modes Q_1 and Q_2 contain coordinates z and R, respectively. These two coordinates are strongly correlated and indeed they would be good candidates to be put in the same mode in an alternative modecombination scheme. The only third-order term that was introduced presents a rather marginal contribution to the potential energy of the system. These values

Table 6. Expectation value of the different terms of the potential expansion (central column) and square root of the expectation value of the potential squared (right column). All energies in cm⁻¹. The combined modes read: $Q_1 = [z, \alpha, x, y]$, $Q_2 = [R, u_{\beta_A}, u_{\beta_B}], \quad Q_3 = [\gamma_A, \gamma_B], \quad Q_4 = [R_{1A}, R_{2A}, u_{\theta_{1A}}]$ and $Q_5 = [R_{1B}, R_{2B}, u_{\theta_{1B}}], \quad u_x = \cos x, \quad \gamma$ refers to wagging and β refers to rocking motion. Q_4 and Q_5 contain the Jacobi coordinates of the water molecules.

| | $\langle \Psi_0 V \Psi_0 angle$ | $\langle \Psi_0 V^2 \Psi_0 \rangle^{1/2}$ |
|----------------------|--------------------------------------|---|
| $V^{(1)}(Q_1)$ | 1293.6 | 1807.7 |
| $V^{(1)}(Q_2)$ | 750.6 | 966.9 |
| $V^{(1)}(O_3)$ | 171.5 | 266.9 |
| $V^{(1)}(O_4)$ | 2293.2 | 3062.8 |
| $V^{(1)}(O_5)$ | 2293.1 | 3062.8 |
| $V^{(2)}(O_1, O_2)$ | -526.9 | 1037.2 |
| $V^{(2)}(O_1, O_3)$ | -78.8 | 290.2 |
| $V^{(2)}(O_1, O_4)$ | -27.5 | 231.8 |
| $V^{(2)}(O_1, O_4)$ | -27.4 | 231.7 |
| $V^{(2)}(Q_2,Q_3)$ | -10.5 | 37.6 |
| $V^{(2)}(O_2, O_4)$ | -24.7 | 117.5 |
| $V^{(2)}(O_2, O_5)$ | -24.7 | 117.9 |
| $V^{(2)}(O_3, O_4)$ | -18.8 | 180.9 |
| $V^{(2)}(O_3, O_5)$ | -18.8 | 180.9 |
| $V^{(2)}(O_4, O_5)$ | 1.2 | 9.9 |
| $V^{(3)}(z,Q_2,Q_3)$ | 1.0 | 50.4 |

prove that the PES representation used is of a good quality and rather well converged with respect to the reference PES, at least for the energy domain of interest. The square root of the expectation value of the potential squared is depicted in the third column. It is a measure of the dispersion around the expectation value and should also ideally vanish. The values indicate that the PES representation is good, albeit not yet fully converged.

8.2.3. Ground vibrational state and eigenstates in the low frequency domain

The ground vibrational state and some eigenstates of the vibrational Hamiltonian are computed using the improved relaxation algorithm presented above, Section 7.4. The comparison between the largest, converged MCTDH calculation and other reported results on the same PES is given in Table 7. As a reference we take the given diffusion Monte Carlo (DMC) result [218], which has an associated statistical uncertainty of 5 cm⁻¹. The most comprehensive calculations on the vibrational ground state based on a wavefunction approach prior to the MCTDH calculations were done by some of us [218] using the MULTIMODE program [76]. The VCI results, both using the single reference (SR) and reaction path (RP) variants are found in Table 7. These calculations use a normal-mode based Hamiltonian. They incorporate

Table 7. Comparison of the zero point energy (ZPE) of the $H_5O_2^+$ cation calculated by various approaches on the PES by Huang *et al.* [217]: diffusion Monte-Carlo (DMC), normal mode analysis (harmonic), vibrational CI single reference (VCI-SR) and reaction path (VCI-RP) as published in [218] and MCTDH results. Δ denotes the difference to the DMC result. The converged MCTDH result is obtained with 10,500,000 configurations. Compare with Table 8.

| Method | $ZPE (cm^{-1})$ | $\Delta \text{ (cm}^{-1}\text{)}$ |
|----------|-----------------|-----------------------------------|
| DMC | 12,393 | 0 |
| Harmonic | 12,635 | 242 |
| VCI-SR | 12,590 | 197 |
| VCI-RP | 12,497 | 104 |
| MCTDH | 12,376.3 | -16.7 |

Table 8. Comparison of the zero point energy (ZPE) of the $H_5O_2^+$ cation between different MCTDH calculations with ascending number of configurations. The Δ values are given with respect to the diffusion Monte Carlo result, $12,393 \, \text{cm}^{-1}$ [218].

| SPFs per Mode | N configs. | $ZPE (cm^{-1})$ | $\Delta \text{ (cm}^{-1}\text{)}$ |
|----------------------|------------|-----------------|-----------------------------------|
| (20, 20, 12, 6, 6) | 172,800 | 12,383.7 | -9.3 |
| (35, 25, 15, 8, 8) | 840,000 | 12,378.5 | -14.5 |
| (40, 40, 20, 8, 8) | 2048,000 | 12,377.8 | -15.2 |
| (60, 40, 20, 8, 8) | 3072,000 | 12,376.7 | -16.3 |
| (70, 50, 30, 10, 10) | 10,500,000 | 12,376.3 | -16.7 |

correlation between the different degrees of freedom due to the *n*-mode representation of the potential [76] and the use of a CI wavefunction. The best reported VCI result for the ZPE lies still 104 cm⁻¹ above the DMC value. It is worth mentioning that before switching to a Hamiltonian based on polyspherical coordinates a Hamiltonian expressed in rectilinear coordinates was tried together with MCTDH and the obtained results were similar to those obtained by VCI. The MCTDH converged result for the ZPE is given in Table 7. The obtained value for the ZPE is $12,376.3 \,\mathrm{cm}^{-1}$, $16.7 \,\mathrm{cm}^{-1}$ below the DMC value. Table 8 contains ZPE values obtained using an increasing number of configurations. According to these results the MCTDH reported values are assumed to be fully converged with respect to the number of configurations. The deviation from the DMC result must be attributed to the cluster expansion (n-mode representation) of the potential. Table 8 illustrates an interesting property of MCTDH, namely its early convergence. The difference in energy between the computation on the top and the most expensive one below is only about 7 cm⁻¹, yet the first calculation can be performed on a laptop while the most expensive one

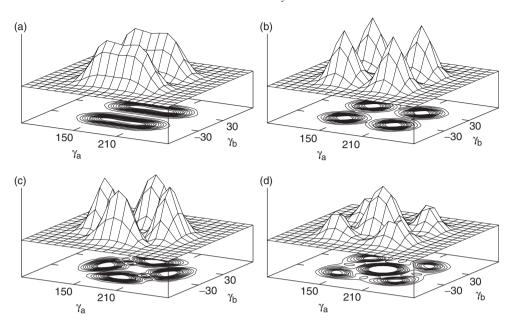


Figure 3. Reduced probability density on the wagging coordinates γ_A and γ_B of (a) the first-excited ($w_{1a,b}$) wagging-mode state and (b), (c), (d), excited states w_2 , w_3 and w_4 respectively, characterized by two quanta of excitation. In *ket* notation these four states can be represented as $w_{1a,b} = |10\rangle \pm |01\rangle$, $w_2 = |11\rangle$, $w_3 = |20\rangle - |02\rangle$ and $w_4 = |20\rangle + |02\rangle$, respectively, where the numbers represent the quanta of excitation in the wagging motion of each water molecule in a harmonic limit.

needed several days on a workstation. The variational optimality of the MCTDH equations, in particular the optimality of the SPFs, leads to qualitatively correct results as soon as there are enough SPFs to describe the physics of the problem at hand, while more accurate results can be obtained at the cost of increased computational resources.

The convergence depicted in Table 8, however, is unusually slow. We attribute this to a 'hole' in the PEO, i.e. a region where the potential is too low. Such holes are unfortunately rather common in *n*-mode representations. They appear at regions where almost all coordinates are far from their equilibrium position. Then many configurations are needed to enable the wavefunction to leak into the hole making the energy too low.

In the following we analyse the excited vibrational states related to the wagging motion due to the relevance of some of these modes in shaping the IR spectrum, as discussed in the next section. Further details on the internal rotation motion can be found in [215]. These states were computed by *improved relaxation*. Figure 3 depicts the probability density projected onto the wagging coordinates of the four lowest excited states related to the wagging motion. The shapes of these density plots make clear that the wagging motion takes place in a very anharmonic region of the PES. State $w_{1a,b}$, which is a doubly

degenerate state of an E irreducible representation, is responsible for the absorption at about 100 cm⁻¹ in the IR spectrum. In the case the system is considered in its full \mathcal{G}_{16} symmetry, also the rotational motion around α is involved in the state corresponding to such absorption [215]. However, in this text we consider the symmetry labels as corresponding to the more familiar D_{2d} point group. The energies of the next three wagging-mode states (w_2, w_3, w_4) are, respectively, 232, 374 and $422 \,\mathrm{cm}^{-1}$ and they are shown in Figures 3(b), (c) and (d), respectively. These three states correspond to two quanta of excitation in the wagging motions and they can be represented by kets $|11\rangle$, $(|20\rangle - |02\rangle)/2^{1/2}$ and $(|20\rangle + |02\rangle)/2^{1/2}$, respectively, where the $|ab\rangle$ notation signifies the quanta of excitation in the wagging motions of monomer A and B. These states have symmetries B_1 , B_2 and A_1 , respectively. In the harmonic limit these three states would be degenerate. The w₃ state has four probability-density maxima along the 2D space spanned by γ_A and γ_B . They correspond to geometries in which one of the water molecules adopts a planar geometry (H₂O character) and the other adopts a pyramidal geometry (H₃O⁺ character). This state transforms according to the B₂ symmetry representation, which is also the symmetry of the proton-transfer fundamental. State w_3 will play a major role due to its strong coupling to the protontransfer mode.

8.2.4. Infrared absorption spectrum

The IR spectrum calculated with MCTDH is shown in Figure 4. The MCTDH spectrum in the full range $0\text{--}4000\,\mathrm{cm^{-1}}$ is depicted in Figure 4 (left panel) and compared to the recent experimental results of [209] in Figure 4 (right panel). The dipole-moment operated ground state $\hat{\mu}|\Psi_0\rangle$ was propagated for 500 fs, yielding an autocorrelation of 1000 fs. The spectrum was calculated according to Equation (81) and the FWHM resolution of the spectrum is, according to Equation (84), about $30\,\mathrm{cm^{-1}}$.

Even if the computation of the IR spectrum in the time-dependent representation is feasible, a means of assigning the different lines to specific motions of the system is still required. Wavefunctions of excited states converged by *improved relaxation*, contain all the possible information on that specific state. The intensity of a given excited state $|\Psi_n\rangle$ is readily obtained by computing the dipole moment $|\langle \Psi_n|\hat{\mu}|\Psi_0\rangle|^2$. But even if an excited state of interest $|\Psi_n\rangle$ has been obtained it is difficult to directly inspect these mathematical objects due to their high dimensionality. Moreover, for the higher excited states we do not have $|\Psi_n\rangle$ at our disposal but only an autocorrelation function providing spectral lines. In both cases one can characterize the eigenstates by their overlaps with carefully chosen test states, i.e.

by the numbers $|\langle \Phi_{\text{test}} | \Psi_n \rangle|^2$. The following procedures are used.

- (1) Test states $|\Phi_{\text{test}}\rangle$ are generated:
- (a) by applying some operator \hat{O} to a previously converged eigenfunction,

$$|\Phi_{\text{test}}\rangle = N\hat{O}|\Psi_n\rangle,$$
 (94)

where N is a normalisation constant, e.g. $N\hat{z}|\Psi_0\rangle$ generates a test state which in essence differs from the ground state $|\Psi_0\rangle$ by a one quantum excitation in the proton-transfer coordinate z.

- (b) by forming Hartree products, where the SPFs are obtained through diagonalisation of mode-Hamiltonians \hat{h}_{κ} . The \hat{h}_{κ} are low-dimensional Hamiltonians obtained from the full Hamiltonian by freezing the remaining coordinates. Each \hat{h}_{κ} operates on the coordinate-space of a MCTDH-particle. Rather than using single Hartree products one may use linear combinations of a few products in order to satisfy a symmetry constraint.
- (2) The overlaps $|\langle \Phi_{\text{test}} | \Psi_n \rangle|^2$ are then computed:
- (a) by direct evaluation of the scalar product if $|\Psi_n\rangle$ is available.

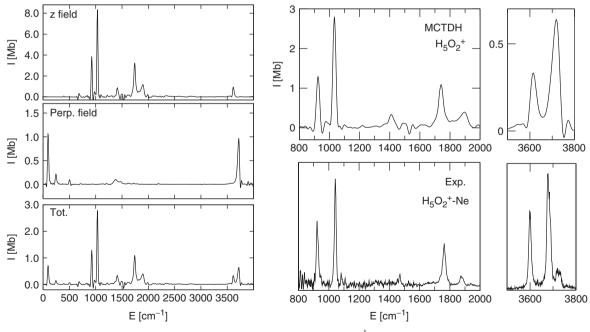


Figure 4. Left panel: Simulated MCTDH spectrum in the range 0–4000 cm $^{-1}$. Excitation in the z direction (top), perpendicular to the O–H–O axis (middle) and total spectrum, i.e. (1/3)z + (2/3)* perpendicular (bottom). Note the different scale of intensities in the perpendicular-component plot. Autocorrelation time $T=1000\,\mathrm{fs}$. Right panel: Comparison between the MCTDH spectrum (top) and the $\mathrm{H}_5\mathrm{O}_2^+$ -Ne spectrum of [208] (bottom). The intensity of the experimental spectrum is adjusted in each spectral region (800–2000 and 3500–3800 cm $^{-1}$) using the most intense peak of the MCTDH spectrum as a reference. Absorption for the MCTDH spectrum is given in absolute scale in mega-barns. (1Mb = 10^{-18} cm 2).

(b) by Fourier transform of the autocorrelation function $a(t) = \langle \Phi_{\text{test}} | \exp(-iHt) | \Phi_{\text{test}} \rangle$. The overlap is obtained via the formula [141]:

$$|\langle \Phi_{\text{test}} | \Psi_n \rangle|^2 = \frac{\pi}{2T} \operatorname{Re} \int_0^T \exp(\mathrm{i}E_n t) a(t) \cos\left(\frac{\pi t}{2T}\right) \mathrm{d}t. \tag{95}$$

8.2.5. Analysis of the middle spectral region

In the following we apply the described techniques to analyse the spectral region between 800 and $2000 \,\mathrm{cm}^{-1}$. This complicated spectral region is shaped by several couplings that involve collective motions of the whole cation. The complication arises from the floppiness and anharmonicity of $H_5O_2^+$. Its complete assignment could only be achieved by means of the MCTDH method [175,214] and it constitutes a nice example of the power of this approach to unravel complicated IR signatures. A detailed description of the OH-stretchings' region at about $3600 \,\mathrm{cm}^{-1}$ as well as a comprehensive list of all fundamentals and various overtones of all vibrational modes is found in [175].

The doublet centred at 1000 cm⁻¹ is the most characteristic feature of the IR spectrum of $H_5O_2^+$. It is depicted in Figure 4 (right, top). The highest energy line has been measured to be at 1047 cm⁻¹ while the low energy component appears at 928 cm⁻¹ [208]. These values slightly change to 1042 and 923 cm⁻¹ in the most recent measurements [209], respectively. There is accumulated evidence in the literature that the absorption of the proton-transfer fundamental occurs in the region of 1000 cm⁻¹ [208,211–213,217]. This band is the most intense band of the spectrum since the central proton motion along the z axis induces a large change in the dipole-moment of the cation. The low-energy component has been recently assigned by us [214]. The doublet is seen to arise from coupling between the proton-transfer motion, the low frequency water-wagging modes and the water-water stretching motion. In order to obtain a fundamental understanding of the low-energy $(|\Psi_d^l\rangle)$ and high energy $(|\Psi_d^h\rangle)$ components of the doublet, test states were constructed by operating with \hat{z} on the ground

state: $|\Phi_{1z}\rangle = \hat{z}|\Psi_0\rangle N$, where N is a normalisation constant, and by operating with $(\hat{R} - R_0)$ on the third excited wagging state w_3 : $|\Phi_{1R,w_3}\rangle = (\hat{R} - R_0)|\Psi_{w_3}\rangle N$. Note that $|\Phi_{1z}\rangle$ is characterized by one quantum of excitation in the proton-transfer coordinate whereas $|\Phi_{1R,w_3}\rangle$ by one quantum in the O-O stretch motion and two quanta in the wagging motion. These two test states were propagated and their auto- and crosscorrelation functions were used for FD analysis, which yielded an energy of $918\,\mathrm{cm}^{-1}$ for $|\Psi_\mathrm{d}^\mathrm{l}\rangle$ and an energy of $1033 \,\mathrm{cm}^{-1}$ for $|\Psi_{\rm d}^{\rm h}\rangle$. These energies are in good accordance with the peaks in Figure 4 which arise from the propagation of $|\Psi_{\mu,0}\rangle$. The spectral intensities were also obtained by FD analysis. The overlaps of the test states to the states making the doublet read: $|\langle \Phi_{1z} | \Psi_{\rm d}^{\rm l} \rangle|^2 = 0.09, \quad |\langle \Phi_{1R,w_3} | \Psi_{\rm d}^{\rm l} \rangle|^2 = 0.83 \quad {\rm and} \quad |\langle \Phi_{1z} | \Psi_{\rm d}^{\rm l} \rangle|^2 = 0.46, \quad |\langle \Phi_{1R,w_3} | \Psi_{\rm d}^{\rm l} \rangle|^2 = 0.10. \; {\rm One \; should \; take}$ into account that these numbers depend on the exact definition of the test states, which is not unique. However, they provide a clear picture of the nature of the doublet: the low-energy band has the largest contribution from the combination of the symmetric stretch and the third excited wagging (see Figure 3(c)), whereas the second largest is the proton-transfer motion. For the high-energy band the importance of these two contributions is reversed. Thus, the doublet may be regarded as a Fermi-resonance between two zero-order states which are characterized by $(1R, w_3)$ and (1z) excitations, respectively. The reason why the third wagging excitation plays an important role in the proton-transfer doublet is understood by inspecting Figures 3(c) and 5. The probability density of state w_3 has four maxima, each of which corresponds to a bent conformation of H₂O-H⁺ (H₃O⁺ character) for one of the waters, and a planar conformation (H₂O character) where a lone-pair H₂O orbital forms a hydrogen bond with the central proton. When the proton oscillates between the two waters, the two conformations exchange their characters accordingly. Thus, the asymmetric wagging mode $(w_3, 374 \text{cm}^{-1})$ combines with the water-water stretch motion $(1R, 550 \,\mathrm{cm}^{-1})$ [175]) to reach an energy close to the natural absorption frequency of the proton transfer, making

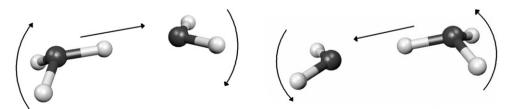


Figure 5. Schematic representation of the two most important coupled motions responsible for the doublet peak at 1000 cm⁻¹.

these motions coupled. The two states of the doublet transform according to the B_2 irreducible representation of D_{2d} and hence couple to the z-component of the dipole operator.

The region between the proton-transfer doublet and 2000 cm⁻¹ features couplings related to the PT and O-O stretch motions. The MCTDH spectrum reported in Figure 4 presents three main absorptions in this range, located at 1411, 1741 and 1898 cm⁻¹ and we could assign them to 1z1R, bu and 1z2R excitations, respectively [175]. Here bu stands for bending ungerade, i.e. one water opens while the other closes. This water bending state deserves particular consideration. It couples very strongly to the proton transfer motion and, in fact, its strong brightness is due to this coupling. In a fashion similar to the coupling to the w_3 wagging motion, as the proton approaches one water molecule the equilibrium value of the H-O-H angle shifts to a larger value because this water molecule acquires more H₃O⁺ character. Conversely, the water molecule at a larger distance of the central proton acquires H₂O character and the angle H-O-H shifts to lower values. That is, the concerted motion is similar as depicted in Figure 5, but with the wagging motion replaced by the bending motion. This coupling shifts the proton-transfer line down and the bending line up by at least 120 cm⁻¹. This estimate was obtained by suppressing the correlation between proton-transfer and bending by setting the number of SPFs to one, for alternatively the proton-transfer mode and the water modes. The frequency of the ungerade bending when decoupled from the proton-transfer motion resembles that of the bare-water bending mode. Thus the shift to a higher frequency is almost entirely due to coupling with the proton-transfer mode.

The reported calculations on $H_5O_2^+$ were in excellent agreement to the experimental measurements of [208] and [209] on the messenger-predissociation spectrum of $H_5O_2^+$ ·Ne. Such a remarkable consistency between experiment and theory along the whole spectral range represents, on the one hand, a validation of the underlying potential energy surface of Huang *et al.* [217] and of the mode-based cluster expansion of the potential used in the quantum-dynamical simulations, but is also a clear indication that a suitable set of coordinates was selected to tackle the problem [215].

The fact that the reported simulations were successful in obtaining accurate results for a system of the dimensionality of the protonated water dimer is to be attributed in great part to the MCTDH algorithm, in which not only the expansion coefficients, but also the SPFs, are variationally optimal. For a 15-dimensional system the use of only four basis

functions per degree of freedom represents of the order of 10⁹ configurations. The largest calculations reported here consist of about 10⁷ configurations, while already good results are obtained by using as few as 10⁵ configurations (Table 7). Such an *early* convergence of the MCTDH method becomes crucial as high-dimensional problems are attempted.

9. Conclusions

We reviewed three current approaches to variational solutions to the computation of vibrational energies of polyatomic molecules. Together they span a range from 'numerically exact' and general in both the kinetic energy operator and potential to more approximate methods that are more feasible to apply to larger molecules. This 'division of labour' is quite typical of the field and is likely to continue in the future.

The numerically exact approach is based on general, so-called polyspherical coordinates. The near exponential scaling of the linear algebra computatations in a direct-product representation of the wave function are avoided by using a combination of multidimensional contractions and an iterative eigensolver. The KEO in polyspherical coordinates (Equation (27)) is general but somewhat more complicated than the Watson normal coordinate KEO. A general polyspherical code that could be used for molecules of different sizes has not been written. Product basis iterative methods have been applied to numerous tetraatomic molecules, however, here we reviewed the application of the contracted basis iterative combination to CH₄ using a realistic full-dimensional potential energy surface. The method is also being applied to the highly fluxional cation CH₅ again using a full-dimensional PES.

The two other methods approach the vibrational energies of larger molecules quite differently. One is based on obtaining eigenvalues and eigenfunctions of the Watson Hamiltonian in rectilinear normal modes using Vibrational Self-Consistent Field/Virtual State Configuration Interaction methods. In this approach the key element that permits extension to larger molecules is the *n*-mode representation of the potential. While this representation can be made exact, its power is in providing an accurate representation of the potential (and also the small-in-magnitude vibrational angular momentum terms) with values of n less than or equal to six. This leads to a 'divide-and-conquer' approach to the evaluation of matrix elements of the Hamiltonian. The limitation of normal modes to describe semi-rigid molecules was noted; however, the power of the approach was illustrated for (semi-rigid) CH₄ where agreement with benchmark calculations was shown to be excellent. The restriction to semi-rigid molecules can be overcome to an extent by the 'Reaction Path Hamiltonian' version of the normal-mode/*n*-mode representation approach. This power of this approach was illustrated by an application to CH₃OH which has one large amplitude mode, the floppy periodic torsional motion.

The third approach is distinguished from the two others because it is a time-dependent one. It uses the powerful Multiconfiguration Time-Dependent Hartree method together with a separable product form of the Hamiltonian. This combination permits applications to molecules with six and seven atoms using curvilinear coordinates. The power of the method was illustrated for the floppy 15 degree-of-freedom $H_5O_2^+$ cation, where the IR spectrum was calculated. As usual with time-dependent approaches the resolution of the spectrum is limited by the length of the propagation time, which cannot be arbitrarily large. In the case of $H_5O_2^+$ the resolution is of the order of $10-30 \,\mathrm{cm}^{-1}$. Nevertheless this was a 'triumphant' calculation that was made possible by using a variant of the *n*-mode representation of the potential instead of the separable product form. This illustrates how various ideas can cross-fertilize overall progress in the field.

Progress in this field has been remarkable. We are now at the stage where we can compute and understand the dynamics of many important molecules. The techniques we have outlined and others related to them can also be used to study the dynamics of dissociating, colliding and reacting molecules. Looking ahead a bit we predict that there will be many interesting applications of these approaches to calculation of vibrational energies of five-, six-, seven-atom polyatomic molecules. With advances in computational power extensions to 10–12 atom molecules are very likely in the near future.

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