

**CONCORDIA UNIVERSITY**  
**FACULTY OF ENGINEERING AND COMPUTER SCIENCE**  
**DEPARTMENT OF MECHANICAL ENGINEERING**

**PROBLEM I [30 pts]**

A piston-cylinder device contains 0.12 kg of helium gas. The gas occupies a volume of  $0.5 \text{ m}^3$  and is at a temperature of  $27^\circ\text{C}$ . The gas undergoes a polytropic compression (with  $n=1.2$ ) until the pressure reaches 250 kPa.

- Determine the amount of work required.

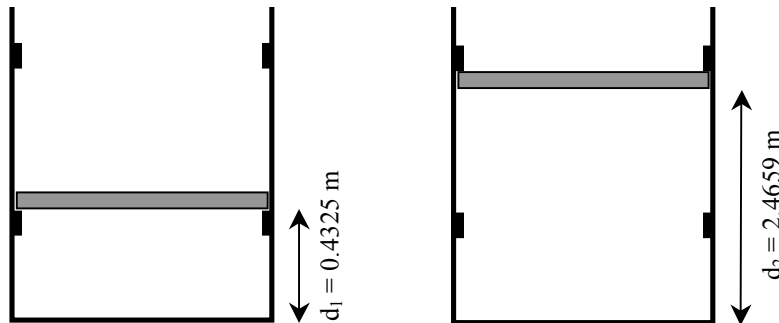
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For helium:  $R=2.0769 \text{ kJ/ kg K}$

**PROBLEM II [50 pts]**

A piston-cylinder device (Figure below) contains 4 kg of saturated water mixture at 100 kPa with a quality of 0.2. Heat is supplied until the final pressure reaches 300 kPa.

- Determine the amount of work required to raise the piston until its final position.
- Determine the amount of heat supplied during the process.


**PROBLEM III [20 pts]**

- What is the algebraic definition of specific enthalpy?
  - What is a quasi-static process?
  - Show that  $C_p - C_v = R$
  - What is the name of the state of matter discovered close to zero Kelvin?
- (a) Superconductivity; (b) Kelvin-Joule gas; (c) Bose-Einstein condensate