PROBLEM #1:

-- CAA: Clean Air Act

- TDS: Total Dissolved Solid

-- NOAEL: No Observable Adverse Effects Level

-- IPCC: Intergovernmental Panel on Climate Change

- MCL: Maximum Contaminant Level

-- PCB:

-- T.: Temperature of the Earth in degrees K

- USDOE: United State Department of Energy

PROBLEM #2:

- a) Name two major contaminants for ground waters and their sources?
- Petroleum compounds (organic chemicals)- Leaking underground storage tanks
- · Nitrates Agricultural fertilizers
- b) Give two benefits of designing a product to fit into a life cycle?
- · It minimizes adverse impacts of wastes.
- · It reduces the use of natural resources to make new products.
- c) Name some major effects of oil pollution?

PROBLEM #3:

a) What is a pathogen?

Are disease-causing agents such as bacteria, viruses, protozoa and parasitic worms. These microorganisms are commonly found in the intestines of infected people or animals, and they are excreted in the feces that enter sewer systems or fall onto the ground. The can cause human illnesses such as typhoid, cholera, diarrhea, gastrointestinal, respiratory, and skin diseases.

b) Define MCL for drinking water

Maximum contaminant level (MCL). An MCL of 1 mg/L is equivalent to one unit of contaminant per million units of water on a mass basis.

 c) Give the quantities for as many standards as you can (international, Us or Canadian) for drinking water.

PROBLEM #4:

Among the many toxic metals the quantitative analysis on mercury is one of the most widely studied. Give me some reasons why mercury has received attention in scientific studies?

Mercury can damage the central nervous system and the brain. Exposure to any of these metals can be fatal at sufficiently high levels. Mercury bio accumulates in the blood. Mercury tends to bioaccumulate in tissues of fish and other organisms. Humans can then be affected through the consumption of fish.

PROBLEM #5:

a) What is the difference between Potency Factor and Chronic Daily Intake for a carcinogenic chemical? The potency factor represents the incremental lifetime cancer risk corresponding to a chronic daily intake of lmg/kg-da of a particular chemical. Chronic daily intake (CDI) is the average daily dose of a chemical over the lifetime of an individual.

b) Explain some of the uncertainties in assessing risk for noncarcinogens?

<u>The reference dose</u> is a key parameter used in risk assessments to characterize the safe dose of a noncarcinogenic chemical.

<u>Hazard quotient (HQ)</u> is the metric used in risk assessments to compare an actual dose of a chemical to the reference dose. Is defined as the ratio of the average daily dose (ADD) of a chemical divided by the reference dose.

PROBLEM #6

- a) What are the 3 major factors affecting increase in CO, emissions?
- · Population Growth
- · GDP per Capita
- · Energy intensity

b) What is bioaccumulation and give an example

<u>Bioaccumulation</u> is the accumulation of substances, such as pesticides, or other organic chemicals in an organism. Mercury is an example of a toxic metal that tends to bioaccumulate in tissues of fish and other organisms.

PROBLEM #7

a) What is Hazard Quotient?

<u>Hazard quotient (HQ)</u> is the metric used in risk assessments to compare an actual dose of a chemical to the reference dose. Is defined as the ratio of the average daily dose (ADD) of a chemical divided by the reference dose. HQ = ADD / RfD

b) Name 4 steps of risk assessment

Assessment of hazards, The development of dose-response relationships, Exposure assessment

Risk characterization

PROBLEM #8

a) Describe	what are the	scientific	results of	the	research	done	on
the ice	core section a	t the Vost	ok. Antar	ctica	test stat	ion?	

b) Describe some of the stabilization scenarios developed by the intergovernmental panel on climate change (IPCC)

PROBLEM #9

The chronic daily intake of a carcinogenic chemical is equal to the value of 3 ug/da from the oral total dose and 16 ug/da for the adult male weighting 70 kg. Using the potency factor values from table 14.3, find his incremental lifetime cancer risk if the chemical of concern is a) arsenic, b) Trichloroethylene (TCE). How do these risks compare to the EPA guideline of 1 x 10⁻⁶? Give your answers as a ratio of multiple (such as 150 times greater, or smaller, than 10⁻⁶)

PROBLEM #10

A well has 2.0 mg/L of zinc, 2.5 mg/L formaldehyde and 70 ug/L of chloroform. Would there be any concern about carcinogenic health effects of using this water for drinking purposes?

PROBLEM #11

A contaminated site has 100 mg/kg of chloroform. Is the cancer risk low enough for the site to be used as a playground for children according to the EPA guideline of 1x10⁻⁶? Assume that a child would use it 4 hours/day, 350 days/year for 12 years.

PROBLEM #12

- a) What are the 3 key factors that influence the environmental change?
- · Housing and industrial development
- · Agriculture
- · Emissions of chemical substances to land, air and water
- b) The population of a city is currently 1 million people. Using a constant annual growth rate, what is the percent increase in population after 10 years with an annual growth rate of 5 percent?

Po= 1,000,000 r= 0.05 t= 10 years P= Po (1+r) **à** P= 1,000,000 (1 + 0.05) P=1,628,894.63 ((1,628,894.63-1,000,000) / 1,000,000) * 100 = 62.89%

BONUS

a) What is "bad ozone" and its source?

Air pollutant ozone found at ground level. Sources: Automobiles, power plants, factories.

b) What is eutrophication and what is its cause?

Eutrophication is the over-enrichment of water by nutrients such as nitrogen and phosphorus in lakes, rivers and streams.