## Latent Heats of Fusion and Vaporization at Standard Pressure

Substance	Melting point (°C)	$L_f$ (J/kg)	Boiling point (°C)	L <sub>v</sub> (J/kg)
nitrogen	-209.97	$2.55 \times 10^4$	-195.81	$2.01\times10^5$
oxygen	-218.79	$1.38 \times 10^4$	-182.97	$2.13 \times 10^{5}$
ethyl alcohol	-114	$1.04 \times 10^5$	78	$8.54 \times 10^5$
water	0.00	$3.33 \times 10^5$	100.00	$2.26 \times 10^{6}$
lead	327.3	$2.45 \times 10^4$	1745	$8.70 \times 10^{5}$
aluminum	660.4	$3.97 \times 10^5$	2467	$1.14 \times 10^{7}$