

BSTT562 Project I

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1 Abstract

In this project, we want to reproduce the results presented in Table 3 Interlaboratory Data for Cadmium from Bhaumik and Gibbons [2005]. # Introduction

A key characteristics of the data is its heteroscedasticity.

2 Data

This data is the experimental data fro cadmium from an interlaboratory study conducted by Ford Motor Company (J. Phillips, personal communication). These data were generated as part of a blind interlaboratory study of laboratories that hold Michigan State Drinking Water Certifications for the parameters tested. Samples were prepared by an independent source, randomized, and submitted on a weekly basis over a 5-week period. Cadmium was analyzed by inductively coupled plasma atomic emissions spectroscopy using EPA method 200.7. The data set used for this example comprised five replicates at each of three concentrations (0, 20 and 100 $\mu g/L$) in each of the $q = 5$ laboratories. Using the first replicate from the first three laboratories as the new measurement (i.e., $q' = 3$), we would like to reproduce the results of point estimates, variances, confidence intervals and simulated confidence levels for the true concentrations in each of the three cases. The data is displayed as follows:

Table 1: Interlaboratory Data for Cadmium (ug / L)

Lab	Replication	0	20	100
1	1	-3.000	10.000	92.000
1	2	4.000	20.000	100.000
1	3	-4.000	17.200	97.800
1	4	3.000	24.000	100.000
1	5	3.100	19.100	109.000
2	1	-0.060	17.815	90.455
2	2	0.010	17.305	87.610
2	3	0.115	16.570	85.550
2	4	-0.055	17.360	89.925
2	5	0.340	18.120	90.070
3	1	-7.400	27.100	107.400

Lab	Replication	0	20	100
3	2	-2.100	19.400	108.100
3	3	-11.400	9.000	83.800
3	4	-11.100	10.500	81.900
3	5	-1.400	19.300	94.200
4	1	1.000	21.000	96.000
4	2	-2.126	16.049	90.650
4	3	0.523	16.082	89.388
4	4	-2.000	17.000	91.000
4	5	-0.551	15.489	85.867
5	1	0.000	18.000	91.000
5	2	0.000	19.000	101.000
5	3	0.000	19.000	102.000
5	4	-1.000	18.700	92.700
5	5	0.038	19.790	99.884

3 Method

To measure the true concentration of an analyte, x , the traditional way is to propose a simple linear calibration model:

$$y = \alpha + \beta x + e \quad (1)$$

with normal assumption on errors. But as the data is heterogeneous, this models fails to explain the increasing measurement variation with increasing analyte concentration commonly observed in analytic data. To overcome this drawback of this simple model, we propose a log-normal model:

$$y = xe^{\eta} \quad (2)$$

4 Analysis

We choose the Method of Moments to estimate the model parameters. The Method of Moments is straightforward and easy to implement when observations with lower concentrations are available, which happens to be the case with our Interlaboratory Data for Cadmium.

Why do we choose the Method of Moments: A good property of MOM: the estimates obtained by MOM are asymptotically efficient.

Data Separation Technique

4.1 The model

$$y_{ijk} = \alpha_i + \beta_j x_j e^{\eta_{jik}} + e_{jik} \quad (3)$$

Note that for this particular model, when x is 0 or near 0, the model is reduced to:

$$y_{ijk} = \alpha_i + \beta_j \times 0 \times e^{\eta_{jik}} + e_{jik} \quad (4)$$

$$y_{ijk} = \alpha_i + e_{jik} \quad (5)$$

4.2 The Estimation Procedures

4.2.1 (Partial) Estimation Using Low-Concentration Observations

We estimate the variance σ_e^2 of the additive errors e_{ijk} 's (of the k th measurement at the j th concentration level in the i th laboratory, $i = 1, 2, \dots, q, j = 1, 2, \dots, r, k = 1, 2, \dots, N_{ij}$) that are present primarily at low-level concentrations and the calibration parameter α_i for the i th laboratory using low-concentration observations.

Assuming that observations corresponding to zero or near-zero are available. Let y_{i0k} be the k th measured concentration corresponding to the true low-level concentration from laboratory i and n_{i0} is the number of samples with true low-level concentrations submitted to laboratory i :

- (1) Estimate σ_e^2 by the variance of the observations with zero or near-zero concentrations (y_{i0k}) by

$$\hat{\sigma}_e^2 = \frac{1}{q} \sum_{i=1}^q \left(\frac{\sum_{k=1}^{n_{i0}} (y_{i0k} - \bar{y}_{i0})^2}{n_{i0} - 1} \right) \quad (6)$$

- (2) Estimate α_i from the observations with zero or near-zero concentrations (y_{i0k}) by

$$\hat{\alpha}_i = \frac{\sum_{k=1}^{n_{i0}} y_{i0k}}{n_{i0}} \quad (7)$$

4.2.2 (Partial) Estimation Using Higher-Concentration Observations

- (1) Let $z_{ijk} = \frac{(y_{ijk} - \alpha_i)}{x_j}$ and $u_{ijk} = \frac{e_{ijk}}{x_j}$. Using the estimate $\hat{\alpha}_i$ of α_i we compute the z_{ijk} 's by

$$\frac{z_{ijk}}{\beta_i} = e^{\eta_{ijk}} + \frac{u_{ijk}}{\beta_i}.$$

- (2) Let

$$\gamma = E(e^\eta) = e^{\sigma_\eta^2/2}, \mu_{zi} = E(z_{ijk}), \sigma_{zi}^2 = \frac{\sum_{j=1}^r \text{var}(z_{ijk})}{r}, \sigma_\mu^2 = \frac{\sigma_e^2 \sum_{j=1}^r 1/x_j^2}{r}$$

5 Table/Discussion

References

Dulal K Bhaumik and Robert D Gibbons. Confidence regions for random-effects calibration curves with heteroscedastic errors. *Technometrics*, 47(2):223–231, may 2005.