BSTT562 Project I

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1 Abstract

In this project, we want to reproduce the results presented in Table 3 Interlaboratory Data for Cadmium from Bhaumik and Gibbons [2005].

2 Introduction

A key characteristics of the data is its heteroscedasticity.

3 Data

This data is the experimental data fro cadmium from an interlaboratory study conducted by Ford Motor Company (J. Phillips, personal communication). These data were generated as part of a blind interlaboratory study of laboratories that hold Michigan State Drinking Water Certifications for the parameters tested. Samples were prepared by an independent source, randomized, and submitted on a weekly basis over a 5-week period. Cadmium was analyzed by inductively coupled plasma atomic emissions spectroscopy using EPA method 200.7. The data set used for this example comprised five replicates at each of three concentrations (0, 20 and $100 \ \mu g/L$) in each of the q=5 laboratories. Using the first replicate from the first three laboratories as the new measurement (i.e., q'=3), we would like to reproduce the results of point estimates, variances, confidence intervals and simulated confidence levels for the true concentrations in each of the three cases. The data is displayed as follows:

Table 1: Interlaboratory Data for Cadmium (ug / L)

Lab	Replication	0	20	100
1	1	-3.000	10.00	92.00
1	2	4.000	20.00	100.00
1	3	-4.000	17.20	97.80
1	4	3.000	24.00	100.00
1	5	3.100	19.10	109.00
2	1	-0.060	17.82	90.45
2	2	0.010	17.30	87.61

Lab	Replication	0	20	100
2	3	0.115	16.57	85.55
2	4	-0.055	17.36	89.92
2	5	0.340	18.12	90.07
3	1	-7.400	27.10	107.40
3	2	-2.100	19.40	108.10
3	3	-11.400	9.00	83.80
3	4	-11.100	10.50	81.90
3	5	-1.400	19.30	94.20
4	1	1.000	21.00	96.00
4	2	-2.126	16.05	90.65
4	3	0.523	16.08	89.39
4	4	-2.000	17.00	91.00
4	5	-0.551	15.49	85.87
5	1	0.000	18.00	91.00
5	2	0.000	19.00	101.00
5	3	0.000	19.00	102.00
5	4	-1.000	18.70	92.70
5	5	0.038	19.79	99.88

4 The Model

To measure the true concentration of an analyte, x, a simple and straightforward method is to propose a linear calibration model as:

$$y = \alpha + \beta x + e \tag{1}$$

with normal assumption on errors. However because our data set has heteroscedastic variances, this simple linear models fails to explain the increasing measurement variation with increasing analyte concentration, which is also commonly observed in analytic data. On the other hand, a log-linear model, for example, $y = xe^{\eta}$, where η is a normal random variable with mean 0 and standard deviation σ_{η} also fails to explain the near-constant measurement variation of y for low true concentration level (x) Rocke and Lorenzato [1995]. To fix these two problems at the same time, Bhaumik and Gibbons [2005] proposed a log-normal model that combines both types of errors:

$$y_{ijk} = \alpha_i + \beta_i x_j e^{\eta_{ijk}} + e_{ijk}, \tag{2}$$

where y_{ijk} is the kth measurement at the jth concentration level in the i th laboratory, $i = 1, 2, ..., q, j = 1, 2, ..., r, k = 1, 2, ..., N_{ij}, x_j$ is the true concentration at the jth level, and α_i and β_i are the calibration parameters for the ith laboratory.

In this model the η_{ijk} 's represent proportional error at higher true concentrations, and we assume that the distributions of the proportional errors remain the same for all laboratories.

Note that for this particular model, when x is 0 or near 0, the model is reduced to:

$$y_{ijk} = \alpha_i + \beta_j \times 0 \times e^{\eta_{ijk}} + e_{ijk}$$

$$y_{ijk} = \alpha_i + e_{ijk}$$
 (3)

5 Analysis

We choose the Method of Moments to estimate the model parameters. The Method of Moments is straightforward and easy to implement when observations with lower concentrations are available, which happens to be the case with our Interlaboratory Data for Cadmium.

Why do we choose the Method of Moments: A good property of MOM: the estimates obtained by MOM are asymptotically efficient.

5.1 The Parameter Estimation Procedures (Using Data Separation Technique)

5.1.1 (Partial) Estimation Using Low-Concentration Observations

We estimate the variance σ_e^2 of the additive errors e_{ijk} 's (of the kth measurement at the jth concentration level in the i th laboratory, $i = 1, 2, ..., q, j = 1, 2, ..., r, k = 1, 2, ..., N_{ij}$) that are present primarily at low-level concentrations and the calibration parameter α_i for the ith laboratory using low-concentration observations.

Assuming that observations corresponding to zero or near-zero are available. Let y_{i0k} be the kth measured concentration corresponding to the true low-level concentration from laboratory i and n_{i0} is the number of samples with true low-level concentrations submitted to laboratory i:

(1) Estimate σ_e^2 by the variance of the observations with zero or near-zero concentrations (y_{i0k}) by

$$\hat{\sigma}_e^2 = \frac{1}{q} \sum_{i=1}^q \left(\frac{\sum_{k=1}^{n_{i0}} (y_{i0k} - \bar{y}_{i0})^2}{n_{i0} - 1} \right) \tag{4}$$

(2) Estimate α_i from the observations with zero or near-zero concentrations (y_{i0k}) by

$$\hat{\alpha}_i = \frac{\sum_{k=1}^{n_{i0}} y_{i0k}}{n_{i0}} \tag{5}$$

5.1.2 (Partial) Estimation Using Higher-Concentration Observations

(1) Let
$$z_{ijk} = \frac{(y_{ijk} - \alpha_i)}{x_i}$$
 and $u_{ijk} = \frac{e_{ijk}}{x_j}$.

Using the estimate $\hat{\alpha}_i$ of α_i we compute the z_{ijk} 's by

$$y_{ijk} = \alpha_i + \beta_i x_j e^{\eta_{ijk}} + e_{ijk}$$

$$y_{ijk} - \alpha_i = \beta_i x_j e^{\eta_{ijk}} + e_{ijk}$$

$$\frac{y_{ijk} - \alpha_i}{x_j} = \beta_i e^{\eta_{ijk}} + \frac{e_{ijk}}{x_j} \text{ (given } x_j's \text{ have higher concentrations)}$$

$$z_{ijk} = \beta_i e^{\eta_{ijk}} + \mu_{ijk}$$

$$\frac{z_{ijk}}{\beta_i} = e^{\eta_{ijk}} + \frac{\mu_{ijk}}{\beta_i} \text{ (given } \beta_i \neq 0).$$
(6)

(2) Let

$$\gamma = E(e^{\eta}) = e^{\sigma_{\eta}^2/2}, \ \mu_{zi} = E(z_{ijk}), \ \sigma_{zi}^2 = \frac{\sum_{j=1}^r var(z_{ijk})}{r}, \ \sigma_{\mu}^2 = \frac{\sigma_e^2 \sum_{j=1}^r 1/x_j^2}{r}$$
(7)

Using equation (6) we can calculate

$$\frac{\mu_{zi}}{\beta_i} = \gamma \text{ and } \frac{\sigma_{zi}^2}{\beta_i^2} = (\gamma^4 - \gamma^2) + \frac{\sigma_{\mu}^2}{\beta_i^2}.$$
 (8)

(3) Replacing γ by $\frac{\mu_{zi}}{\beta_i}$ in the second part of equation (8) we obtain the following equation:

$$\beta_i = \sqrt{\frac{\mu_{zi}^4}{\sigma_{zi}^2 - \sigma_{\mu}^2 + \mu_{zi}^2}} \text{ and } \sigma_{\eta}^2 = \frac{2\sum_{i=1}^q \ln(\mu_{zi}/\beta_i)}{q}.$$
 (9)

To estimate β_i and σ_{η} from equation (9), we need estimates of μ_{zi} , σ_{zi}^2 , and σ_{μ}^2 (the error variance of the model (6)).

(4) Let n_{ij} be the number of observations with higher concentrations collected from the *i*th laboratory for concentration level *j* and $n_i = \sum_{j=1}^r \eta_{ij}$, then

$$\hat{\mu}_{z_{ij}} = \frac{\sum_{k=1}^{n_{ij}} z_{ijk}}{n_{ij}}, \ \hat{\mu}_{z_i} = \frac{\sum_{j=1}^{r} \hat{\mu}_{z_{ij}}}{r},$$

$$\hat{\sigma}_{z_i}^2 = \frac{\sum_{j=1}^{r} \sum_{k=1}^{\eta_{ij}} (z_{ijk} - \hat{\mu}_{z_{ij}})^2 / (\eta_{ij} - 1)}{r}, \ \hat{\sigma}_{\mu}^2 = \frac{\sum_{j=1}^{r} \hat{\sigma}_e^2 / x_j^2}{r}$$
(10)

(5) Therefore We can calculate the estimates of the variance of the proportional error at higher true concentrations σ_{η}^2 and the calibration parameter β_i for the *i*th laboratory as

$$\hat{\beta}_i = \sqrt{\frac{\hat{\mu}_{z_i}^4}{\hat{\sigma}_{z_i}^2 - \hat{\sigma}_{\mu}^2 + \hat{\mu}_{z_i}^2}} \text{ and } \hat{\sigma}_{\eta}^2 = \frac{2\sum_{i=1}^q \ln(\hat{\mu}_{z_i}/\hat{\beta}_i)}{q}$$
(11)

5.2 Point Estimation Of X

```
[1] 7.896
##
   [1] 0.8720 0.8682 1.1870 0.8877 0.9545
   [1] 0.9914 0.8865 1.0176 0.9121 0.9751
  [1] 0.9317 0.8774 1.1023 0.8999 0.9648
  [1] 0.0661450 0.0008652 0.1370575 0.0124709 0.0010388
   [1] 0.0037388 0.0004391 0.0155757 0.0013298 0.0025826
  [1] 0.9187 0.8829 1.0735 0.9018 0.9692
        0.6200 0.0700 -6.6800 -0.6308 -0.1924
  [1] 0.9187 0.8829 1.0735 0.9018 0.9692
##
             [,1]
                        [,2]
                                                                [,6]
                        "0.872"
                                 "0.8682" "1.187"
                                                                "0.95452"
## mu zij20
             "20 ug/L"
                                                    "0.88774"
## mu zij100 "100 ug/L" "0.9914" "0.88652" "1.0176" "0.912118" "0.975092"
  [1] 0.9317 0.8774 1.1023 0.8999 0.9648
  [1] 0.0349419 0.0006522 0.0763166 0.0069003 0.0018107
## [1] 7.896
  [1] 0.01026
  [1] 0.01102
  [1] 1.006
```

5.3 Point Estimation of X

Suppose that the same analyte with an unknown true concentration X is tested in q' independent laboratories, and that $Y_1, Y_2, ..., Y_{q'}$ are the corresponding new observations. We want to compute a point estimate of X using the available information from each of the q' laboratories and then combine them. Following model (2), we can obtain

$$E(Y_i) = \alpha_i + \beta_i X \gamma$$
,

and hence an estimate of X from the ith laboratory, denoted by \hat{X}_i , is

$$\hat{X}_{i} = \frac{Y_{i} - \hat{\alpha}_{i}}{\hat{\beta}_{i}\hat{\gamma}} \text{ and } \hat{X} = \sum_{i=1}^{q'} \frac{\hat{X}_{i}}{q'}, \tag{12}$$

where \hat{X} is the combined estimate of X. \hat{X}_i is asymptotically unbiased, and the asymptotic expression of the variance of \hat{X} is

$$var(\hat{X}) = E[var(\hat{X}/y)] + var[E[\hat{X}]/y]$$

$$= \frac{\sum_{i=1}^{q'} \sigma_e^2 / (\beta_i^2 \gamma^2) (1 + 1/n_{0i})}{q'^2} + \frac{X^2 (\gamma^2 - 1)}{q'},$$
(13)

where y is the vector of past observations, $var(\hat{X})/y$ is the conditional variance of \hat{X} given y, and $E(\hat{X}/y)$ is the conditional expectation of \hat{X} given y. The variance of \hat{X} in equation (13) depends on X and increases with increasing levels of concentrations.

5.4 Confidence Region For X

```
## variance
variance \leftarrow (sigma_e2 / (y1$beta_i^2 * gamma^2)*(1 + 1 / 5)
              + sigma e2 / (y2\$beta i^2 * gamma^2)*(1 + 1 / 5) +
                  sigma_e2 / (y2\$beta_i^2 * gamma^2)*(1 + 1 / 5)) / 3^2 +
  0^2 * (gamma^2 - 1) / 3
## confidence interval
y bar <- ( y1\$Real + y2\$Real + y3\$Real ) / 3
sigma alpha2 <- var(low$alpha i)</pre>
cll \leftarrow \max(0, y_{\text{bar}} - 1.96*\text{sqrt}((\text{sigma_e2} + \text{sigma_alpha2})/3))
clu <- y_bar + 1.96 * sqrt((sigma_e2 + sigma_alpha2) /3)</pre>
CL low <- c(cll, clu)
# 20 ug / L
## point estimate
y12 <- twenty[1,]
y22 <- twenty[6,]
y32 \leftarrow twenty[11,]
x12 \leftarrow x(y12\$Real, y12\$alpha i, y12\$beta i, gamma)
x22 \leftarrow x(y22\$Real, y22\$alpha i, y22\$beta i, gamma)
x32 \leftarrow x(y32\$Real, y32\$alpha_i, y32\$beta_i, gamma)
x hat2 <- (x12 + x22 + x32) / 3
x hat2
## [1] 20.48
## variance
variance2 <- (sigma_e2 / (y12$beta_i^2 * gamma^2)*(1 + 1 / 5)</pre>
               + sigma e2 / (y22\$beta i^2 * gamma^2)*(1 + 1 / 5) +
  sigma e2 / (y32\$beta_i^2 * gamma^2)*(1 + 1 / 5)) / 3^2 +
  20^2 * (gamma^2 - 1) / 3
## confidence interval
X1 <- 20
c1i <- beta i^2 * X1^2 * (gamma^4 - gamma^2) + sigma e2
c2i <- c1i / (beta_i^2*X1^2)
```

```
c3i \leftarrow log((1 + sqrt(1 + 4*c2i)) / 2)
# 100 ug / L
## point estimate
y13 <- hundred[1,]
y23 <- hundred[6,]
y33 <- hundred[11,]
x13 \leftarrow x(y13\$Real, y13\$alpha_i, y13\$beta_i, gamma)
x23 \leftarrow x(y23\$Real, y23\$alpha i, y23\$beta i, gamma)
x33 \leftarrow x(y33\$Real, y33\$alpha_i, y33\$beta_i, gamma)
x_hat3 <- (x13 + x23 + x33) / 3
x hat3
## [1] 102.1
## variance
variance3 <- (sigma_e2 / (y13$beta_i^2 * gamma^2)*(1 + 1 / 5)</pre>
               + sigma_e2 / (y23$beta_i^2 * gamma^2)*(1 + 1 / 5)
               + sigma e2 / (y33$beta i^2 * gamma^2)*(1 + 1 / 5)) / 3^2
        + 100^2 * (gamma^2 - 1) / 3
## [1] 36.95
## confidence interval
X2 < -100
c1i2 <- beta_i^2 * X2^2 * (gamma^4 - gamma^2) + sigma_e2
c2i2 <- c1i / (beta_i^2*X2^2)
c3i2 \leftarrow log((1 + sqrt(1 + 4*c2i)) / 2)
```

5.5 Variance of X

Variance for X is 3.905, 4.9507, 3.4728 for 0, 20, 100 $\mu g/L$ true concentration level, respectively.

5.6 Confidence Interval of *X*

Bhaumik and Gibbons [2005] proposed a method for constructing confidence regions, which is an approximation based on a normal or lognormal distribution.

Firstly, to construct a confidence region for low-level concentrations, let Y_{i0} be an observation collected from the *i*th laboratory with low-level true concentration. Define

$$\bar{Y}_0 = \sum_{i=1}^{q'} Y_{i0}/n_0.,$$

where n_0 is the total number of measurements for low-level true concentration from all q' laboratories. For a low-level true concentration X_0 , the $(1 - \alpha)100\%$ confidence region of X_0 is

$$(max(0, \bar{Y}_0 - z_{\alpha/2}\sqrt{(\hat{\sigma}_e^2 + \hat{\sigma}_\alpha^2)/n_0}), \bar{y}_0 + z_{\alpha/2}\sqrt{(\hat{\sigma}_e^2 + \hat{\sigma}_\alpha^2)/n_0}).$$

 $\hat{\sigma}_{\alpha}^2$ represents the variability of α_i across all q laboratories in the calibration sample.

Secondly, to construct a confidence region for larger X, we use the following lognormal approximation. Let

$$c_{1i} = var(Y_i) = \beta_i^2 X^2 (\gamma^4 - \gamma^2) + \sigma_e^2, c_{2i} = \frac{c_{1i}}{\beta_i^2 X^2}, and \ c_{3i} = ln(\frac{1 + \sqrt{1 + 4c_{2i}}}{2}).$$

Bhaumik and Gibbons [2005] also showed in a lemma that the approximate variance of $ln(\frac{Y-\alpha_i}{\beta_i X})$ is c_{3i} .

6 Table/Discussion

True Concentration	X_hat	Var(X)
0 ug/L	-1.577	3.905
20 ug/L	20.479	4.951
100 ug/L	102.137	3.473

References

Dulal K Bhaumik and Robert D Gibbons. Confidence regions for random-effects calibration curves with heteroscedastic errors. *Technometrics*, 47(2):223–231, may 2005.

David M. Rocke and Stefan Lorenzato. A two-component model for measurement error in analytical chemistry. *Technometrics*, 37(2):176–184, may 1995.