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___ 2. the mass of a mole of any element or compound ___ 3. the volume occupied by a mole of any gas at STP Match each item with the correct statement below. a. representative particle d. percent composition b. mole e. standard temperature and pressure c. Avogadro's number f. empirical formula ___ 4.

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The Mole Unit. The mole is a convenient unit for expressing the relative amounts of substances in chemical reactions. The burning of carbon in oxygen can be written with the 2 oxygen atoms bonded in a single O₂ molecule: The mole is the most common unit used to express the quantity of a chemical substance.

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Calculate the number of moles that contain 4.50 10²⁴ atoms of zinc (Zn). 1. You are given the

number of atoms of zinc and must find the equivalent number of moles. If you compare 4.50 10²⁴ atoms Zn with 6.02 10²³, the number of atoms in one mole, you can predict that the answer should be less than 10 moles.

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