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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$  3. An aluminum strip is immersed in a solution of silver nitrate.

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### **Oxidation Numbers of Representative Element Cations and ...**

Hague, G., & Smith, J. (2001). The Ultimate Chemical Equations Handbook . Batavia, IL: Flinn Scientific, Inc. Correlating questions (ii) from: Rodney Schmitz (2007) Moore County Schools, NC For each of the following eight reactions, in part (i) write a BALANCED equation and in part (ii) answer the question about the reaction.

### **Key to Reaction Prediction -1**

View Test Prep - DescriptiveChem from CHEM 232 at University of Illinois, Urbana Champaign. 94 The Ultimate Chemical Equations Handbook Some symbols are derived from non—English words. i.e.,

### **DescriptiveChem - 94 The Ultimate Chemical Equations ...**

The Ultimate Chemical Equations Greek, or German names. Ag Sn w (ferrum) (natrium) (argentum) (stannum) (plumbum) (wolfram) iron sodium silver tin lead tungsten (cuprum) (kalium) (hydrargyrum) (stibium) (aurium) copper potassium mercury antimony gold Symbols are used as a sort of shorthand in writing the names of elements.

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