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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$  3. An aluminum strip is immersed in a solution of silver nitrate.

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### **THE ULTIMATE CHEMICAL EQUATIONS HANDBOOK ANSWERS CHAPTER 1**

The Ultimate Chemical Equations Handbook by George R. Hauge, Jane D. Smith, June 2001, Flinn Scientific edition, Paperback in English - Student edition

### **The Ultimate Chemical Equations Handbook (June 2001 ...**

The Ultimate Chemical Equations Handbook, Student Edition (2nd Edition) by Jane D. Smith, George R. Hague Paperback, 100 Pages, Published 2011: ISBN-10: 1-933709-30-8 / 1933709308 ISBN-13: 978-1-933709-30-7 / 9781933709307: handbook containing equations and procedures for chemical equations.

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Re-typed from The Ultimate Chemical Equations Handbook by Hague and Smith 7. Free elements, no matter how complex the molecule, have an oxidation number (valence or charge) equal to zero. OJ. Fe. T~ He < /1 1 ~ ~ L~td. 8. The following are diatomic or polyatomic elements in nature which must be committed to memory.

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The basic rules and table from page 13 in the Ultimate Chemical Equations Handbook. Learn with flashcards, games, and more — for free.

### **Oxidation Numbers of Representative Element Cations and ...**

Hague, G., & Smith, J. (2001). The Ultimate Chemical Equations Handbook . Batavia, IL: Flinn Scientific, Inc. Correlating questions (ii) from: Rodney Schmitz (2007) Moore County Schools, NC For

each of the following eight reactions, in part (i) write a BALANCED equation and in part (ii) answer the question about the reaction.

### Key to Reaction Prediction -1

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