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An acid dissociation constant,  $K_a$ , (also known as acidity constant, or acid-ionization constant) is a quantitative measure of the strength of an acid in solution. It is the equilibrium constant for a chemical reaction known as dissociation in the context of acid-base reactions.  $K_a = \frac{[A^-][H^+]}{[HA]}$ . The chemical species  $HA$ ,  $A^-$ , and  $H^+$  are said to be in equilibrium when their concentrations ...

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