



# CHEME 5440/7770: Prelim 1 Q2

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## 2. a) Formulate a three micro-state model for PEK activity

- **State  $s = 0$ :** no effector+no substrate (base state, no activity)
- **State  $s = 1$ :** no effector+substrate (the data shows low activity)
- **State  $s = 2$ :** effector+substrate (activity)

take the form

$$\hat{r}_j = r_j v(\dots)_j$$

The probability of each microstate is given by

$$p_i = \frac{1}{Z} \times f_i \exp(-\beta \epsilon_i) \quad i = 0, 1, 2, \dots, S$$

where

$$W_i = \exp(-\beta \epsilon_i)$$

$$Z = \sum_{s=0}^S f_i \exp(-\beta \epsilon_i)$$

which gives:

$$p_i = \frac{f_i \exp(-\beta \epsilon_i)}{\sum_{s=0}^S f_i \exp(-\beta \epsilon_i)} \quad i = 0, 1, 2, \dots, S$$

Given these microstates, we know that enzyme  $E$  can catalyze its reaction in microstate  $s = 1$  and  $s = 2$ , thus:

$$v(\dots)_j = \frac{f_1 \exp(-\beta \epsilon_1) + f_2 \exp(-\beta \epsilon_2)}{\sum_{s=0}^2 f_s \exp(-\beta \epsilon_s)}$$

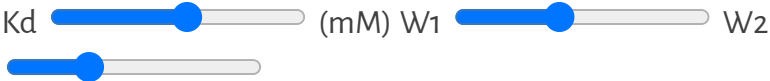
let  $j = 1$

$$r_1 = k_{cat} E_1 \left( \frac{F6P}{K_{K6P} + F6P} \right) \left( \frac{ATP}{K_{ATP} + ATP} \right)$$

$$\hat{r}_1 = r_1 v(\dots)_1$$

## 2. b) Estimate the parameters by using the dataset in Table 1

From definition we know  $\epsilon_0 = 0$ , then  $W_0 = 1$  and we also know  $f_0$  and  $f_1$  are both set to 1. So I estimate  $\epsilon_1$ ,  $\epsilon_2$ , the binding constant  $K_d$  and an order parameter  $n$  to get  $W_1$ ,  $W_2$  and  $f_2$  to match the dataset in table 1.



(1.54, 0.046, 260.0)

• [DSM\\_parameters](#)

```
v = [0.04397705544933078, 0.05389273057560039, 0.08243820770
```

```

• begin
•   # get Effector - A
•   A = [0:0.01:1;]
•   v = A
•   for i in eachindex(A)
•       Kd = DSM_parameters[1]
•       W0 = 1
•       W1 = DSM_parameters[2] # state 1 (E bound to S,
•           but no A)
•       W2 = DSM_parameters[3] # state 2 (E bound to A)
•
•       # setup system -
•       R = 8.314           # units: J/mol-K
•       T = 273.15 + 25.0   # units: K
•       β = 1/R*T
•
•       # setup binding parameters for state 2 -
•       n = 2.0
•
•       # compute the state-specific factor-
•       f0 = 1.0 # state 0
•       f1 = 1.0 # state 1
•       f2 = ((A[i]/Kd)^(n))/(1+(A[i]/Kd)^(n)) # state 2
•
•       # compute the v variable -
•       microstate_0 = f0.*W0
•       microstate_1 = f1.*W1
•       microstate_2 = f2.*W2
•
•       Z = microstate_0 + microstate_1 + microstate_2
•       p1 = (1/Z)*microstate_1
•       p2 = (1/Z)*microstate_2
•       v[i] = p1 + p2
•   end
•   # show -
•   with_terminal() do
•       println("v = $(v)")
•   end
• end

```

Show the  $\bar{r}$  calculated when the concentration of the effector is the same as the ones that are shown in table 1. See if they are matched to each other.

```
r_bar[1,5,9,18,40,99,100] = 3.05991612706268,12.607458955719
```

```

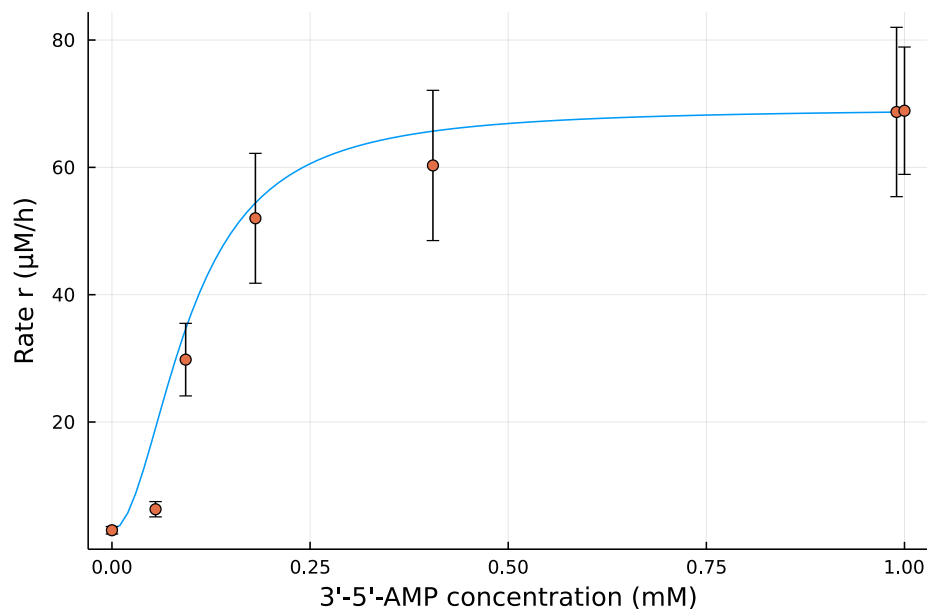
• begin
• # Set up some parameters
•   S1 = 0.1 # units: mM -- concentration for F6P
•   S2 = 2.3 # units: mM -- concentration for ATP
•   E = 0.12 # units: μM
•   K_F6P = 0.11 # units: mM
•   K_ATP = 0.42 # units: mM
•   kcat = 0.4 # units: s^-1
•
• # calculate the rate
•   r_bar = (kcat*E)*(S1/(S1+K_F6P))*
•           (S2/(S2+K_ATP))*v*3600
•
• # show -
•   with_terminal() do
•     println("r_bar[1,5,9,18,40,99,100] =
•           $(r_bar[1]),$(r_bar[5]),$(r_bar[9]),$(r_bar[18]),$(r_bar[40]),$(r_bar[99]),$(r_bar[100])")
•   end
end

```

so the final results I choose are  $k_d = 1.54$ ,  $W_1 = 0.046$ ,  $W_2 = 260.0$

## 2. c) Plot the converted data with errorbars

from the image we can see the proposed model describes the data well except for the second one



```

• begin
•
•   # 3'-5'-AMP concentration -
•   x = 0:0.01:1
•   conc = [0, 0.055, 0.093, 0.181, 0.405, 0.990, 1.0]
•   rate = [3, 6.3, 29.8, 52.0, 60.3, 68.7, 68.9]
•   std = [0.59, 1.2, 5.7, 10.2, 11.8, 13.3, 10.0]
•
•   # overall rate -
•   y = r_bar
•
•   # plot -
•   plot(x,y,label="r")
•   plot!(conc,rate,yerror=std,seriestype = :scatter,
•         legend =false)
•   xlabel!("3'-5'-AMP concentration (mM)",fontsize=18)
•   ylabel!("Rate r (μM/h)",fontsize=18)
•
• end

```

### 3. a) convert the <n> values in Table 2

```
[0.26125, 0.28875, 0.56375, 0.92125, 1.1825, 1.27875, 1.27875]
```

```

• begin
•   n_array = [19;21;41;67;86;93;93]; #units: nM from
•   the assume(ii)
•   mc = 2 * 10^(-13) # units:g
•   Vc = 2.75 # units: μm^3
•
•   n_array_new = n_array * Vc / mc * 10^(-15)
•
• end

```

```
• md"""  
• ##### 3. b)  
• """
```

```
• begin  
•   # import some packages -  
•   using PlutoUI  
•   using PrettyTables  
•   using LinearAlgebra  
•   using Plots  
• end
```