请证明理想气体状态方程的 $\alpha=\delta=1$, 定义: $d\ln\rho=\alpha d\ln p-\delta d\ln T$

理想气体状态方程: $P = \frac{n}{V}RT = \frac{m}{MV}RT = \frac{\rho}{M}RT$

$$\rho = \frac{M}{R} \frac{P}{T}$$

$$\ln \rho = \ln \frac{M}{R} + \ln P - \ln T$$

$$d \ln \rho = d \ln p - d \ln T$$

$$\alpha = 1, \delta = 1$$