

Field Ionization and Field Emission with Intense, Single-cycle THz Pulses

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by

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Abstract

Nuclear-polarized ^3He targets have been widely used in electron-scattering experiments in Thomas Jefferson National Accelerator Facility (JLAB) since mid 1990s. It is of great importance to produce large amounts of ^3He gas with high polarization.

The latest experiments run in JLAB prior to the 12GeV upgrade have been using cells polarized with Spin-Exchange Optical Pumping (SEOP). These cells were made of the GE180 glass and use a two-chambered design. The top chamber, known as the pumping chamber, is where ^3He is polarized through SEOP. The bottom chamber, known as the target chamber, is where electron scattering occurs. Great effort has been made in our lab to develop this generation of cells. Alkali-hybrid SEOP together with narrowband laser diode arrays have increased the ^3He polarization from 37% to 65%. Among other things, we also carefully studied an additional spin relaxation mechanism that limits the maximum achievable ^3He polarization.

The 12GeV upgrade makes the future experiments much more demanding in terms of target cell performance. One challenge it brings is the high relaxation due to electron beam. We have designed and tested a new style cell that uses convection instead of diffusion to increase the rate at which the polarization in the target chamber is being replenished by gas from pumping chamber. We have obtained over 50% polarization with controllable convection speed so far.

An additional problem that comes with higher beam current is that the glass end windows of traditional design are not likely to survive the experiments. Our group started exploring the option of using metal end windows from a decade ago. The first problem to solve is to find out the correct material and the proper technique

to incorporate metal without introducing significant spin relaxation and still being able to hold high pressure gas (12 atm) inside. This is a brand new technique that may have a profound impact of future cell designs once fully developed. Although no metal end windows have been tested so far, multiple glass cells with different kinds metal tubes (much larger in area compared to the end windows that will be used in JLAB experiments) attached were examined and were enough to convince us the extra spin relaxation is not likely to cause significant problems. The metals tubes were connected to Pyrex glass with knife-edge (housekeeper) seals and stayed intact through high pressure tests. After exploring options such as pure copper, gold coated copper, titanium, stainless steel, gold coated titanium, we have established that electroplating gold on copper substrate yields the best result so far. Further tests are planned before attaching metal end windows to GE180 glass and using them in electron-scattering experiments.

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Chapter 1

Spin-Exchange Optical Pumping

1.1 Overview

Spin-polarized noble gases have been widely used for various purposes [8, 7, 48, 45].

In JLAB (Thomas Jefferson National Accelerator Facility), polarized ^3He is used as target for electron-scattering experiments. This is because a ^3He nucleus has a pair of protons with paired spins and a single neutron that contributes the most of the nuclear spin. In MRI, polarized ^3He has seen uses such as detecting structural damage in the lungs.

There are generally two ways of polarizing ^3He : Metastability-Exchange Optical Pumping (MEOP) [26] and Spin-Exchange Optical Pumping (SEOP) [25, 40, 43]. Our group focuses on SEOP as MEOP polarizes gas at relatively low pressure (~ 1 torr), thus further compression is required to produce target cells of several atms that meet the need of JLab experiments.

In SEOP, alkali metal is polarized by circularly polarized laser light tuned to the

D₁ transition of the particular alkali species used. ³He obtains polarization from alkali metal through spin-exchange process [16]. With the combination of hybrid alkali mixtures (typically Rb and K) and spectrally narrowed lasers [34, 11], more than 80% polarizations have been produced.

1.2 Optical pumping

1.2.1 Rb for SEOP

In optical pumping, Rb is often the alkali metal chosen to be optically pumped by circularly polarized laser light. The angular momentum of polarized photons are transferred to the valence electrons of Rb atoms [61]. Sometimes hybrid mixtures of Rb and K are used together, in which case Rb is still the alkali metal that is directly pumped by laser light while K serves as an efficient medium to transfer the polarization from Rb to ³He. Because the spin destruction rates are much lower for lighter alkali metals, K-³He spin-exchange process is a lot more efficient than that of Rb-³He. Rb is still selected as the alkali metal to be pumped directly because of the relative ease of acquiring laser at the Rb D₁ line wavelength and the wide separation between D₁ (794.7nm) and D₂ line (780nm).

The Rb melting point is at 39.5°C, so it's easy to achieve enough Rb vapor without having to drive the oven temperature too high. In our lab, depending on if the cell contains pure Rb or Rb/K mixture, the oven temperature can be between 85°C to as high as 255°C. Perhaps the most used oven temperature for hybrid cells is 235°C which has empirically been a good temperature to produce Rb/K mixture vapor,

while around 170°C is typically enough for pure Rb.

1.2.2 Vapor Pressure Curves

When the alkali metal is heated to above its melting point, a small amount of alkali metal atoms evaporate. The equilibrium vapor pressure is temperature dependent:

$$P = 10^{5.006+\alpha+\beta/T} \text{ Pa} \quad (1.1)$$

where α and β are listed in Table 1.2.2 [18].

	Patassium	Rubidium
α	4.402	4.312
β	-4453	-4040

Thus the number density is

$$[A] = \frac{10^{5.006+\alpha+\beta/T}}{k_B T} \quad (1.2)$$

The number density curves of pure Rb and K vapor are shown in Fig. 1.1.

1.2.3 Energy Levels of Alkali Metal in External Magnetic Field

The Hamiltonian for ground state ($L=0$) alkali metal atoms in external magnetic field is:

$$\mathbf{H} = A\mathbf{I} \cdot \mathbf{S} - g_e \mu_B S_z B_z - g_N \mu_N I_z B_z \quad (1.3)$$

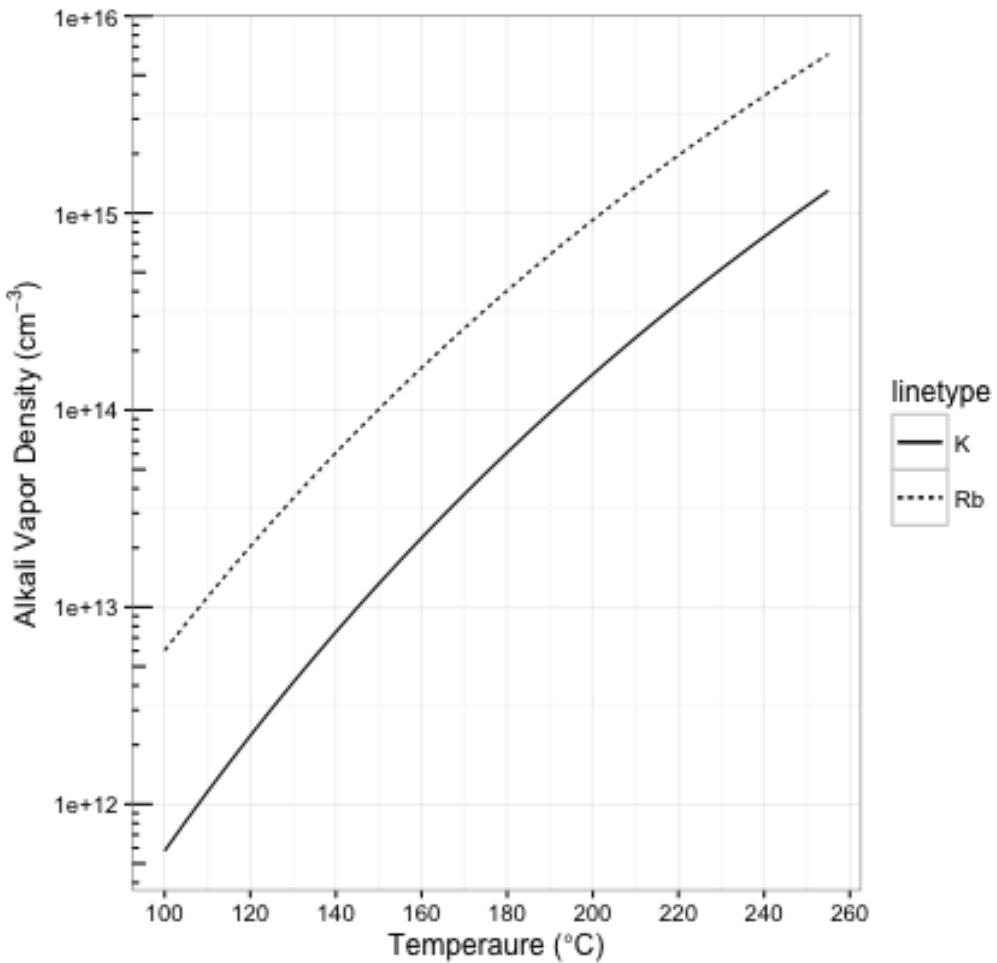


Figure 1.1: Rb And K Number Density Curves

The first term $A\mathbf{I} \cdot \mathbf{S}$ describes the coupling of the nuclear spin \mathbf{I} with the electron spin \mathbf{S} and is key to spin exchange, where A is the isotropic magnetic-dipole coupling coefficient [9]. The resulting energy levels from the first term are referred to as hyperfine structure. The second and third terms describe the Zeeman splitting due to the presence of an external magnetic field. $\mu_B = 9.274 \times 10^{-24} \text{ J/T}$ and $\mu_N = 5.051 \times 10^{-27} \text{ J/T}$ are the Bohr and nuclear magnetons. $g_e \approx 2$ and $g_N \approx 5.59$ are the electronic and nuclear Lande g-factors.

The linear relationship between energy levels and magnetic field only holds for weak magnetic fields which applies to our lab where \sim 13 Gauss is used most of the time. When the Zeeman splitting grows comparable to the hyperfine energy difference one would have to take into account the quantum mixing of the states, the result is described by Breit-Rabi Formula. With \sim 13 Gauss, the hyperfine term dominates the total Hamiltonian. ^{85}Rb and ^{87}Rb are both present in the alkali metal used for SEOP, the energy levels of ^{87}Rb are shown in Fig. 1.2.

1.2.4 Optical Pumping Process Overview

For simplicity, the following discussion will ignore the nuclear spins for now. The inclusion of nuclear spins will increase the number of energy states but the optical pumping mechanism remains the same. In our typical setup, circularly polarized laser light is tuned to the D1 line of Rb and excites valence electrons of Rb from $5\text{S}_{1/2}$ state to $5\text{P}_{1/2}$ state as shown in Fig. 1.3[61].

Left-circularly polarized light is assumed in the figure. Conservation of angular momentum requires $\Delta m = +1$ as the figure shows. Through collisions with other Rb atoms, excited electrons will mix and evenly distribute between the two $2\text{P}_{1/2}$ states. Electrons then decay to the two ground states with equal probabilities. The selection rule for the decay process is $\Delta m = 0$ or ± 1 . Even though both ground states receive electrons from the decay, only the $m = -1/2$ state absorbs the circular polarized photons and is being depleted, so atoms are in effect pumped to the $m = +1/2$ state. When we consider Rb with nuclear spins, both $5\text{S}_{1/2}$ and $5\text{P}_{1/2}$ states are split into more energy levels, but the net effect is still that the ground state with

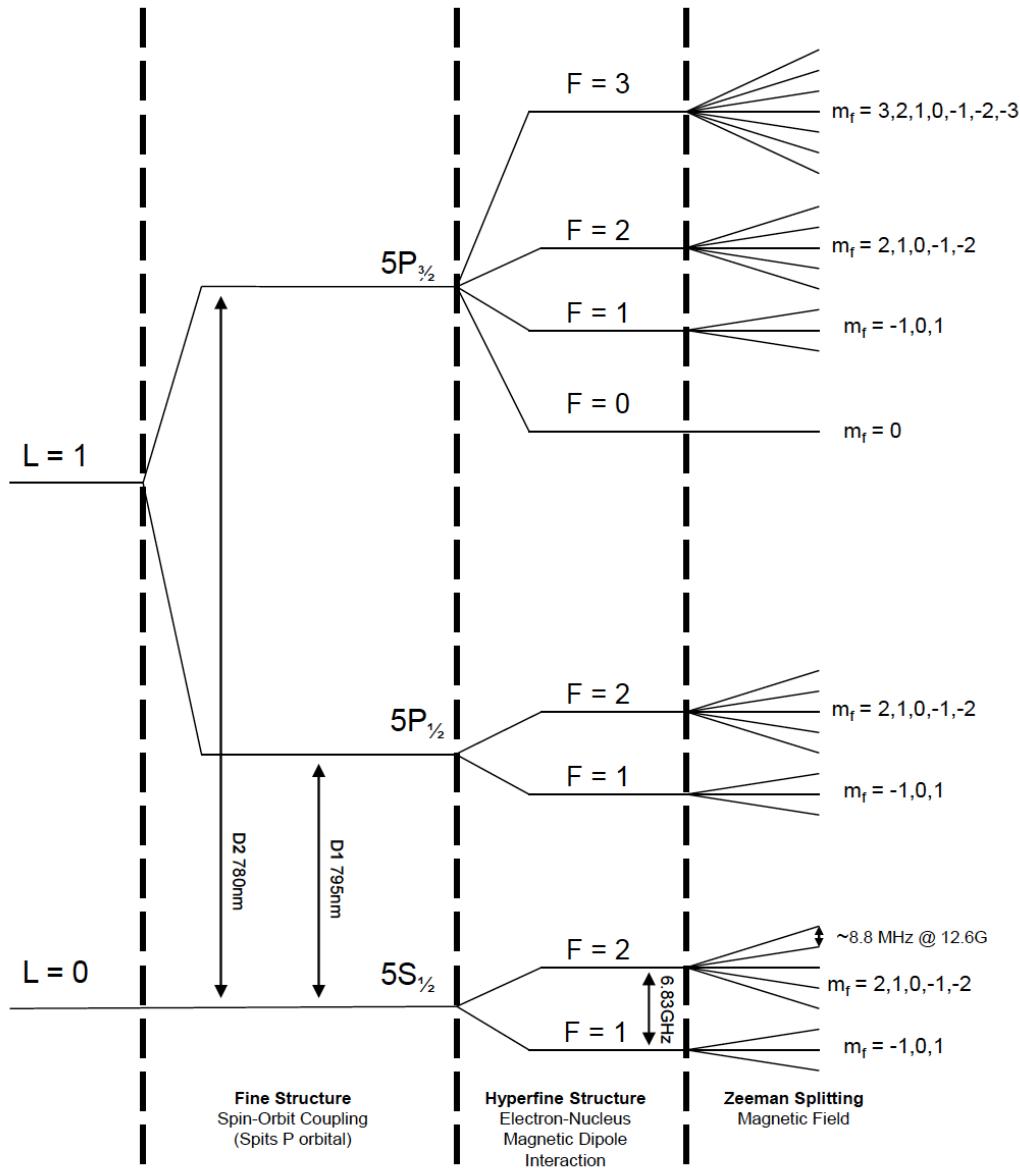


Figure 1.2: Level Diagram of ^{87}Rb . The splittings are not to scale.

Adapted from Dolph's PhD thesis.

highest m accumulate atoms over time.

When the excited electrons decay back to the ground state, they emit unpolarized photons with angular momentum in random directions which can depolarize the gas.

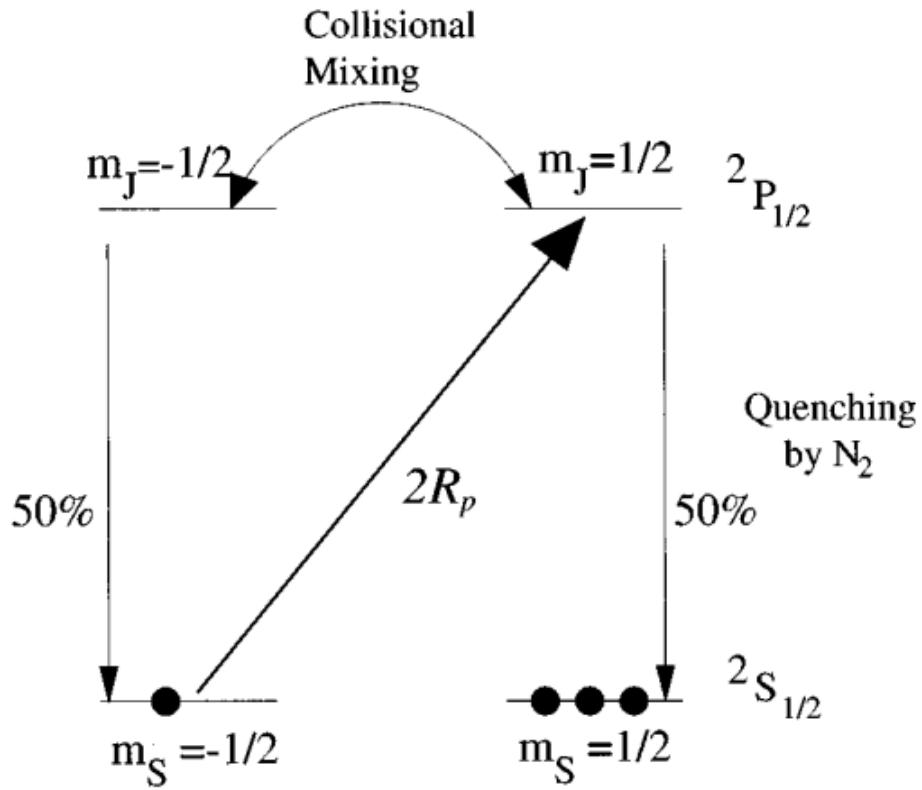


Figure 1.3: The interaction of alkali-metal atoms with left-circularly (σ^+) polarized light. (from Ref. [61])

A small amount of N₂ gas is added into the cell (typically around 0.1 Amagats) to non-radiatively quench the excited electrons as N₂ molecules can absorb the released energy of spontaneous decays into their rotational and vibrational modes of oscillation. With an appropriate amount of N₂[59], the photon-emitting decays can be reduced to less than 3%.

1.2.5 Optical Pumping Rate

The optical pumping rate at position \vec{r} can be described by

Table 1.1: Pressure broadening of Rb D₁ lines by ³He, ⁴He and N₂. The broadening and shifting density coefficients are listed. The 4th and 6th columns are the temperature dependence for He and N₂, respectively. All coefficients are given for 353 K, values for different temperatures can be calculated with the temperature dependence.

	⁴ He	³ He	Temp. depen.	N ₂	Temp. depen.
D ₁ full width (GHz/amg)	18.0±0.2	18.7±0.3	T ^{0.05±0.05}	17.8±0.3	T ^{0.3}
D ₁ line shift (GHz/amg)	4.3±0.1	5.64±0.15	T ^{1.1±0.1}	-8.25±0.15	T ^{0.3}

$$R = \int \Phi(\nu, \vec{r}) \sigma(\nu) d\nu \quad (1.4)$$

where $\Phi(\nu, \vec{r})$ is the position dependent photon spectral flux density and $\sigma(\nu)$ is the photon absorption cross section. The cross section has a natural Lorentzian lineshape which is broadened by the Doppler effect and pressure broadening. The pressure broadening effect dominates the lineshape as our cells normally have densities well above one amagat. The collisions of Rb with ³He and N₂ cause the broadening as well as a slight frequency shift of the D₁ line. The coefficients of pressure broadening for ³He, ⁴He and N₂ are listed in Table 1.1, and can be used to calculate the broadened line width and the shifted line center.

$\sigma(\nu)$ follows the sum rule[42]:

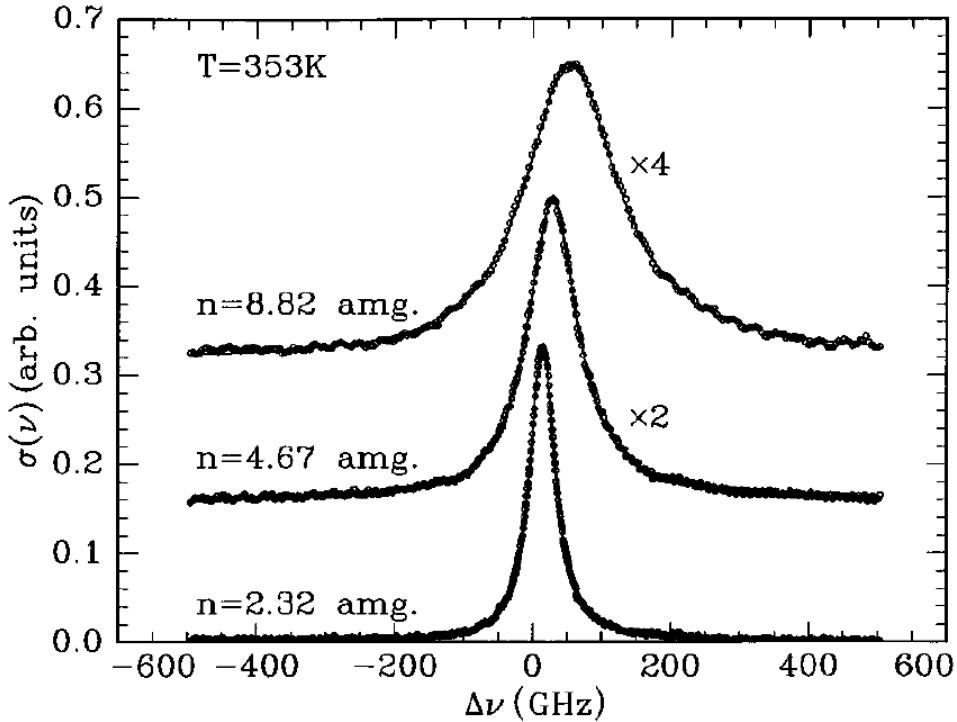


Figure 1.4: Absorption cross section for Rb D_1 line in the presence of three different densities of ${}^3\text{He}$. (from Ref. [50])

$$\int \sigma(\nu) d\nu = \pi r_0 c f \quad (1.5)$$

where $r_0 = 2.82 \times 10^{-13}$ cm is the classical electron radius and $f = 0.337$ [40] is the transition oscillator strength. Since our lineshape is dominated by pressure broadening, the photon absorption cross section is well approximated by a Lorentzian lineshape:

$$\sigma(\nu) = f r_e c \frac{\frac{\Gamma_A}{2}}{(\nu - \nu_0)^2 + (\frac{\Gamma_A}{2})^2} \quad (1.6)$$

where Γ_A is the pressure dependent FWHM, $\Gamma_A \approx 0.04\text{nm/amg. } [{}^3\text{He}]$. At the front

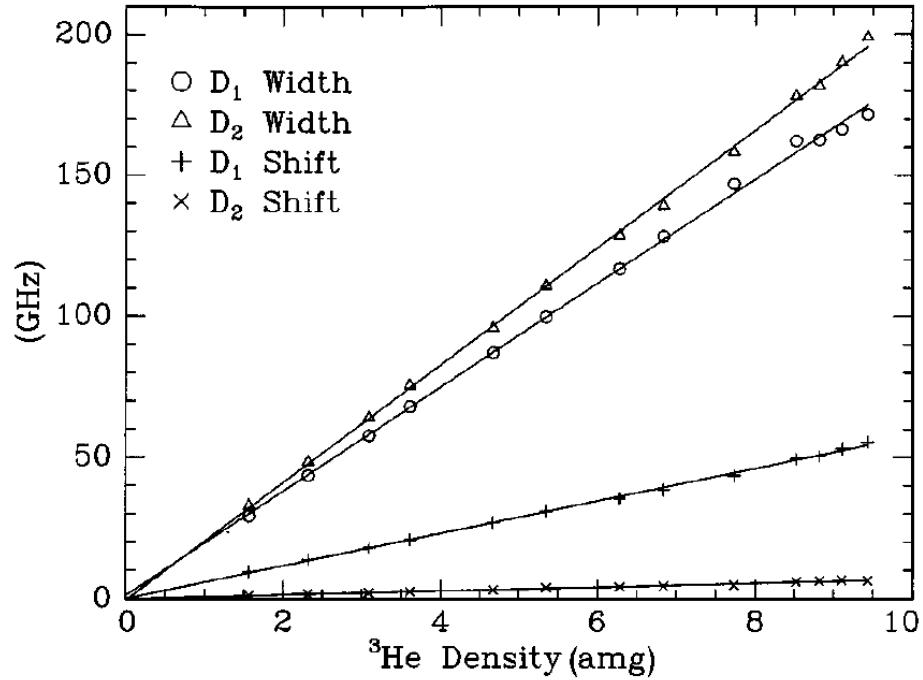


Figure 1.5: The shift and the broadening due to presence of ${}^3\text{He}$ for Rb D_1 and D_2 lines. (from Ref. [50])

of the cell, the photon spectral flux density is reasonably described as the product of a Gaussian spatial distribution and a Gaussian spectrum.

$$\phi(\nu, \vec{r}) = \phi_0(\vec{r})G(\nu) \quad (1.7)$$

with

$$\phi_0(\vec{r}) = \frac{P}{h\nu} \frac{2}{\omega^2 \pi} e^{2r^2/\omega^2} \quad (1.8a)$$

$$G(\nu) = \frac{1}{\sqrt{2\pi}\sigma_l} e^{-(\nu-\nu_l)^2/2\sigma_l^2} \quad (1.8b)$$

where P is the laser power; ω is the beam waist; σ_l is the Gaussian width of the laser and ν_l is the central laser frequency.

1.2.6 Polarization Time Evolution

^3He nuclei have an intrinsic nuclear spin of $1/2$, thus it is relatively simple to explain the math of polarization accumulation with ^3He as the example. One typically defines the polarization as the asymmetry between $+1/2$ state and $-1/2$ state:

$$P = \frac{\rho_{+1/2} - \rho_{-1/2}}{\rho_{+1/2} + \rho_{-1/2}} = \rho_{+1/2} - \rho_{-1/2} \quad (1.9)$$

where $\rho_{\pm 1/2}$ is the ensemble population in the $\pm 1/2$ state.

While it might not be true for Rb, we are only treating the time evolution of polarization of Rb electrons the same as ^3He for our purpose with the following the equation:

$$\frac{dP}{dt} = \gamma(1 - P) - \Gamma \cdot P \quad (1.10)$$

where γ is the polarization rate and Γ is the depolarization rate due to all other processes. The solution of this differential equation has the simple form of:

$$P(t) = Ce^{-(\gamma+\Gamma)t} + \frac{\gamma}{\gamma + \Gamma} \quad (1.11)$$

The saturated polarization is defined as the value of P in the limit $t \rightarrow \infty$:

$$P_\infty = \frac{\gamma}{\gamma + \Gamma} \quad (1.12)$$

The initial polarization is defined as the value of P at $t = 0$:

$$P_0 = C + \frac{\gamma}{\gamma + \Gamma} = C + P_\infty \quad (1.13)$$

Thus, $P(t)$ can be expressed as:

$$P(t) = (P_0 - P_\infty)e^{-(\gamma+\Gamma)t} + P_\infty \quad (1.14)$$

In the case of polarizing Rb with a pump laser, γ is the pumping rate R and Γ is the Rb spin relaxation rate Γ_{Rb} . There is typically a small angle θ between the pump laser and the holding field even though great effort has been made to minimize the angle. Thus $P(t)$ can be rewritten as:

$$P(t) = P_0 e^{-(R+\Gamma_{Rb})t} + P_{laser} \cos \theta \frac{R}{R + \Gamma_{Rb}} (1 - e^{-(R+\Gamma_{Rb})t}) \quad (1.15)$$

where θ is called the skew angle, P_{laser} is the circular polarization of the pump laser which is normally above 99.5%. Rb close to the front side of the cell can reach above 97% (depends on the laser power and other factors) on the order of 100's of microseconds. As the laser propagates through the cell, power is attenuated by Rb vapor. Therefore Rb polarization at the back side of the cell is typically lower than that at the front side. One way to minimize this problem is to shine pump laser light from both sides of the cell to achieve higher overall Rb polarization and ^3He polarization.

Spins are thermally polarized in the presence of a magnetic field even without being actively polarized. The probability for a spin to be in state s is:

$$\text{Prob.} = \frac{e^{-E_s/k_B T}}{\sum_{si} e^{-E_{si}/k_B T}} \quad (1.16)$$

where E_s is the energy of the state, k_B is the Boltzmann constant and T is the temperature. Using the thermal distribution, under typical operating conditions,

3He polarization is 10^{-9} and Rb polarization is 10^{-5} . Both are negligible without active pumping.

1.2.7 Rb Spin Destruction Rate

There are two main mechanisms of Rb depolarization: the binary collisions with Rb, 3He and N_2 , and the formation and breakup of van der Waals molecules, the second mechanism is negligible for 3He cells [61]. The Rb spin destruction rate can then be expressed as

$$\Gamma_{Rb} = k_{Rb-Rb}[Rb] + k_{Rb-{}^3He}[{}^3He] + k_{Rb-N_2}[N_2] \quad (1.17)$$

where k_{Rb-X} is the spin destruction rate constant and $[X]$ is the density of X. The constants are listed as follows [60, 51]:

$$k_{Rb-{}^3He}(T) = 55.9(9) \left(\frac{T}{473.15K} \right)^{3.31(12)} \text{Hz/amg} \quad (1.18a)$$

$$k_{Rb-N_2}(T) = 290(30) \left(\frac{T}{473.15K} \right)^{2.0(25)} \text{Hz/amg} \quad (1.18b)$$

$$k_{Rb-Rb} = 4.813(48) \times 10^{-13} \text{Hz} \cdot \text{cm}^3 \quad (1.18c)$$

For a pure Rb cell at 170°C with the following densities in the pumping chamber:

$$[{}^3He] \approx 8.0 \text{amg} \quad (1.19a)$$

$$[N_2] \approx 0.08 \text{amg} \quad (1.19b)$$

$$[Rb] \approx 6.0 \times 10^{14} \text{cm}^{-3} \quad (1.19c)$$

The approximate spin destruction rates due to various gases are:

$$\Gamma_{Rb-{}^3He} \approx 360 \text{ Hz} \quad (1.20a)$$

$$\Gamma_{Rb-N_2} \approx 20 \text{ Hz} \quad (1.20b)$$

$$\Gamma_{Rb-Rb} \approx 289 \text{ Hz} \quad (1.20c)$$

The total spin destruction rate is 669 Hz. By contrast, the average optical pumping rate at the front of the cell with 20 W and 1.5 cm beam radius narrowband laser light is often of 100s kHz.

1.3 Spin Exchange

Following equation 1.15, the time evolution of 3He polarization can be expressed as:

$$P_{{}^3He}(t) = P_0 e^{-(\gamma_{se} + \Gamma)t} + P_{Rb} \frac{\gamma_{se}}{\gamma_{se} + \Gamma} (1 - e^{-(\gamma_{se} + \Gamma)t}) \quad (1.21)$$

The saturation polarization is

$$P_\infty = P_{Rb} \frac{\gamma_{se}}{\gamma_{se} + \Gamma} \quad (1.22)$$

where γ_{se} is the spin exchange rate between 3He and Rb, and Γ is the spin relaxation rate.

1.3.1 Spin-Dependent Interactions

The key process in spin-exchange optical pumping is collisional transfer of polarization between optically pumped alkali-metal atoms and the nuclei of the noble gas atoms.

As in Fig. 1.6, the transfer of angular momentum occurs either while the atoms are bound in van der Waals molecules or in simple binary collisions. The first mechanism is only important for heavy noble gas nuclei such as X3, while binary collisions is the dominating mechanism for ${}^3\text{He}$. The time scale for binary collisions is on the order of 10^{-12} sec, the collision can induce both $\Delta F = \pm 1$ and $\Delta F = 0$ transitions between hyperfine sublevels. For heavier noble gases like ${}^{129}\text{Xe}$ at pressure of a few tens of Torr, the contributions of van der Waals molecules can greatly exceed that of binary collisions. At several atm which is the typical operating pressure for SEOP, the time scale of van der Waals molecules is greatly limited by collision so that the binary collisions dominate [61].

Spin-dependent interactions produce the spin transfer and relaxation. For SEOP, spin-rotation interaction between \vec{S} and the rotational angular momentum \vec{N} of the atom pair formed by Rb and noble gas atom, and the isotropic hyperfine interaction between \vec{S} and the noble-gas nuclear spin \vec{I}_b dominate the spin-exchange process:

$$V_1(\vec{R}) = \gamma(R)\vec{N} \cdot \vec{S} + A(R)\vec{I}_b \cdot \vec{S} \quad (1.23)$$

The spin-rotation interaction is caused by the magnetic fields from relative motion of the charges of the colliding atoms, and the isotropic hyperfine interaction originates from the Fermi-contact magnetic fields produced by two nuclei. The spin-rotation interaction produces relaxation of the alkali-metal electron-spins, while the isotropic hyperfine interaction transfers angular momentum back and forth between the alkali-metal electron spins and the noble-gas nuclear spins.

An alkali-metal atom and a noble-gas atom interact via both a large spin-independent

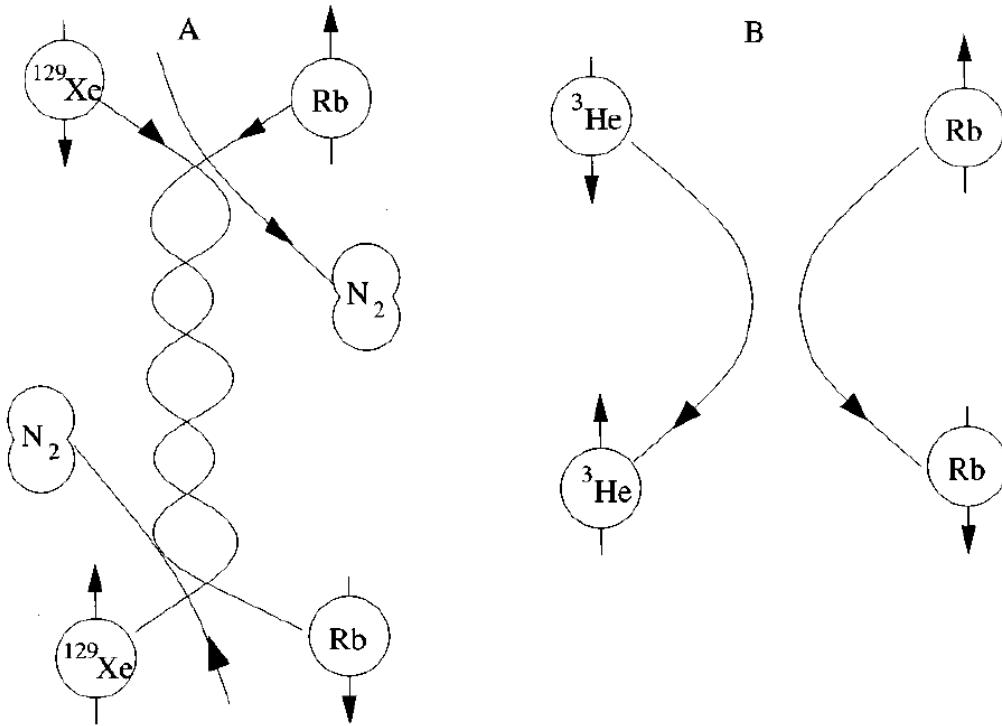


Figure 1.6: A. Formation and breakup of alkali-metal/noble-gas van der Waals molecule. B. Binary collision between an alkali-metal atom and a noble-gas atom. (from Ref. [61])

interaction $V_0(R)$ and a small spin-dependent interaction $V_1(R)$. At the high operating temperatures, V_0 determines classical collision trajectories, while V_1 acts as a small perturbation. We'll focus on V_1 below since it is responsible for spin exchange.

Including a few more terms that were neglected in Eq. 1.23, the spin-dependent interaction $V_1(R)$ can be expressed as [61]:

$$\begin{aligned}
V_1(\vec{R}) = & \gamma(R) \vec{N} \cdot \vec{S} + \sum_k A_k(R) \vec{I}_k \cdot \vec{S} \\
& + \sum_k B_k(R) \vec{I}_k \cdot (3\vec{R}\vec{R} - 1) \cdot \vec{S} \\
& + \sum_k C_k(R) \vec{I}_k \cdot (3\vec{R}\vec{R} - 1) \cdot \vec{I}_k
\end{aligned} \tag{1.24}$$

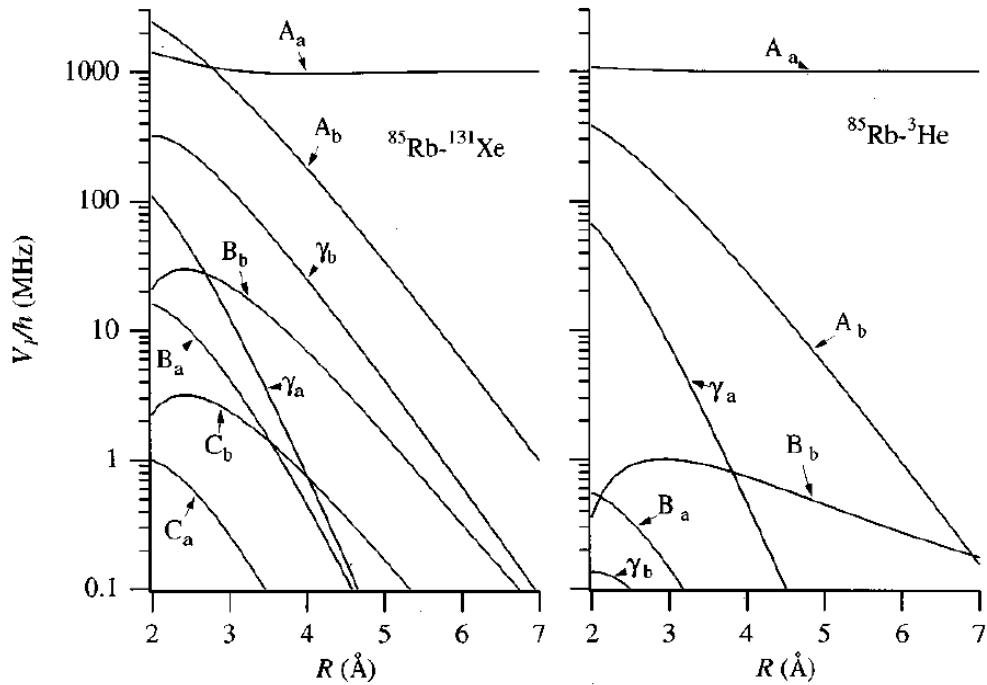


Figure 1.7: Strengths of various spin-dependent interactions as functions of separation (from Ref. [61])

where \vec{I}_a and \vec{I}_b are the nuclear spins of the atomic pair, where a stands for alkali metal atom and b stands for noble gas atom. γ is the coefficient of the spin-rotation interaction, while A_k , B_k , C_k are the coefficients for isotropic magnetic-

dipole hyperfine interactions, anisotropic magnetic-dipole hyperfine interactions, and electric quadrupole interactions, respectively. A_a greatly exceed other coefficients as the separations between atoms increase.

The isotropic hyperfine interactions come from the Fermi-contact magnetic fields of the two nuclei. A_b is the term responsible for spin exchange:

$$A_b(R) = \frac{8\pi g_s \mu_B \mu_b}{3I_b} |\eta \phi_0(R)|^2 \quad (1.25)$$

where η is the enhancement factor which equals to the ratio of the perturbed wave function at the noble gas nucleus to that without the noble gas atom. The isotropic hyperfine interaction also introduces a frequency shift of the magnetic resonance lines for alkali-metal and noble gas atoms. The frequency shift is characterized by another enhancement factor κ which is the ratio of the actual shift of the alkali metal electron lines due to the presence of polarized noble gas nuclei to what would be produced by the bulk magnetization of polarized noble gas. The shift is used in the technique Electron Paramagnetic Resonance (EPR) to calculate the polarization of noble gas nuclei.

The isotropic magnetic-dipole coupling polarizes the noble gas nuclei parallel to the electron spin polarization, while the anisotropic magnetic-dipole coupling polarizes in the opposite direction. Even though the anisotropic magnetic-dipole coupling is weak compared with the isotropic interaction, it might be an important factor for the “X factor” that will be discussed later.

1.3.2 Spin Exchange Rate

The spin exchange rate due to binary collisions is:

$$\gamma_{se} = \langle \sigma_{se} v \rangle [Rb] = k_{se} [Rb] \quad (1.26)$$

where $k_{se} = \langle \sigma_{se} v \rangle$ is the velocity-averaged spin exchange rate constant. k_{se} for spin exchange between ^3He and Rb is [14]:

$$k_{se}^{^3\text{He}-\text{Rb}} = (6.7 \pm 0.7) \times 10^{-20} \text{ cm}^3/\text{s} \quad (1.27)$$

At 170°C which is a typical temperature that we run tests with,

$$[Rb] = 2.60 \times 10^{14} \text{ cm}^{-3} \quad (1.28)$$

Thus for a simple spherical cell with pure Rb,

$$\frac{1}{\gamma_{se}} \approx 15.9 \text{ hrs} \quad (1.29)$$

1.4 ^3He Spinup and Relaxation

Similar to the optical pumping process of Rb, ^3He polarization can be described by

$$P_{^3\text{He}}(t) = P_0^{^3\text{He}} e^{-(\gamma_{se} + \Gamma)t} + P_\infty^{^3\text{He}} (1 - e^{-(\gamma_{se} + \Gamma)t}) \quad (1.30)$$

where the saturation polarization is

$$P_\infty^{^3\text{He}} = P_\infty^{\text{Rb}} \frac{\gamma_{se}}{\gamma_{se} + \Gamma} \quad (1.31)$$

And Γ is the total relaxation rate of ^3He nucleus spin polarization due to all processes except for spin exchange,

$$\Gamma = \Gamma_{dipolar} + \Gamma_{inhomogeneity} + \Gamma_{wall} \quad (1.32)$$

When a target cell are used in electron scattering experiments where an electron beam goes through part of the cell, an additional relaxation rate due to the beam Γ_{beam} should also be included.

The coupling of nuclear spin to orbital angular momentum creates an intrinsic ^3He relaxation rate that depends on density. At room temperature (23°C), the dipolar relaxation rate is [43]

$$\frac{1}{\Gamma_{dipolar}} = \frac{[{}^3\text{He}]}{744} \text{hr}^{-1} \quad (1.33)$$

where $[{}^3\text{He}]$ is the ${}^3\text{He}$ density in amagats. Assuming the cell density is 8 amg, the relaxation rate is $1/93 \text{ hr}^{-1}$. In addition, there is an additional intrinsic relaxation due to the spin-rotation interaction. This mechanism dominates the relaxation for ^{129}Xe but is small for ${}^3\text{He}$.

The relaxation rate due to field inhomogeneities is [32, 52, 21]

$$\Gamma_{inhomogeneity} = D \frac{|\nabla B_x|^2 + |\nabla B_y|^2}{B_0^2} \quad (1.34)$$

where D is the ${}^3\text{He}$ diffusion constant, ∇B_x and ∇B_y are the transverse magnetic field inhomogeneities, B_0 is the holding field along z-axis. Under operating conditions, assuming the pressure is around 12 atm and field is 12.6 G, $D \approx 0.16 \text{cm}^2/\text{s}$ and the field inhomogeneities are 10mG/cm , the relaxation rate is $1/1400 \text{ hr}^{-1}$.

Wall relaxation is typically the dominant relaxation mechanism for cells in our lab. This mechanism depends on the properties of the inner surface of glass. Most of the target cells are constructed with reblown General Electric Type 180 (GE-180) glass. This aluminosilicate glass is highly impermeable to ^3He . The wall relaxation is believed to be associated to several different mechanisms, such as paramagnetic impurities in the glass and microfissures in the surface that could trap ^3He atoms. It has been found reblowing the glass can help lower the wall relaxation rate because it reduces the number of microfissures. The wall relaxation is not well understood, but it is believed to scale with the surface-to-volume ratio:

$$\Gamma_{wall} = \rho S/V \quad (1.35)$$

where ρ is called relaxivity.

1.5 X Factor

In 2006, Babcock *et al.* reported evidence of a previously unrecognized spin relaxation mechanism, and named it X factor. This mechanism appears to be temperature dependent and roughly proportional to alkali density. The X factor limits the maximally achievable ^3He polarization even with infinite laser power. The saturation polarization is

$$P_\infty^{^3\text{He}} = P_\infty^{Rb} \frac{\gamma_{se}}{\gamma_{se}(1 + X) + \Gamma} \quad (1.36)$$

In the presence of infinite laser power where $\gamma_{se} \gg \Gamma$, the saturation polarization becomes

$$P_{\infty}^{^3He} = P_{\infty}^{Rb} \frac{1}{1+X} \quad (1.37)$$

The X factor will be discussed in greater detail in chapter 4.

Chapter 2

^3He Polarimetry

2.1 Overview

Traditional pure glass target cells are studied mainly using Adiabatic Fast Passage (AFP) [6] Nuclear Magnetic Resonance (NMR) and Electron Paramagnetic Resonance (EPR). AFP is a technique that allows us to monitor a signal that is directly proportional to the ^3He polarization, which serves as a means to gain knowledge of properties of cell including the time it takes to polarize it and the relaxation rates of its polarization. The EPR technique utilizes the fact that polarized ^3He produces frequency shift of the magnetic resonance lines of alkali metal to measure the ^3He polarization. When AFP and EPR are combined, we can calculate the calibration constant between an AFP signal and the corresponding ^3He polarization.

A significant focus of my studies is on exploring cells that incorporate metal. Unfortunately, AFP is not suitable for studying these cells as it requires exposing the entirety of the cell to a Radio Frequency (RF) magnetic field in an attempt to flip all

spins in the cell. The RF field would induce an Eddy current induced in the metal body that significantly affects the resulting signal. For these cells, Pulsed Nuclear Magnetic Resonance (PNMR) has proven to be very useful. PNMR only applies a pulsed RF field to a small selected part of the cell which makes it relatively easy to prevent metal from distorting the signal. However, the spins tipped by applying the pulse lose their transverse component (which depends on the "tip angle"), we typically allow some time for this portion of gas to diffuse out of the region being monitored before we can the next measurement on a fresh sample of the gas. The highest rate at which PNMR measurements can be taken is limited by this requirement.

This chapter introduces the three techniques mentioned above and how they're used for our studies.

2.2 Adiabatic Fast Passage

2.2.1 Nuclear Magnetic Resonance

The energy of a magnetic moment in an external field is

$$E = -\vec{\mu} \cdot \vec{B}_0 = -\mu_z B_0 \quad (2.1)$$

where $\vec{\mu}$ is the magnetic moment, for a spin-1/2 nuclei, the energy is

$$E = -\gamma B_0 \hbar / 2 \quad (2.2)$$

where γ is the gyromagnetic ratio, $\gamma/2\pi \approx 3.2434 \text{kHz/Gauss}$. When a oscillating magnetic field with the frequency $\omega = \gamma B_0$ is present, transitions between the +1/2

and $-1/2$ states are induced. This frequency is called Larmor frequency. When a nucleus is placed in an external magnetic field that is not aligned with its magnetic moment, it will precess at the Larmor frequency.

2.2.2 The Rotating Coordinate System

2.2.2.1 Classical Formulation

For a nucleus in an external field \vec{B} with $\gamma\hbar\vec{I}$ as its nuclear angular momentum, the equation of motion in a stationary coordinate system is [46]

$$\hbar \frac{d\vec{I}}{dt} = \gamma\hbar\vec{I} \times \vec{B} \quad (2.3)$$

Let $\frac{\partial}{\partial t}$ represent the derivative with respect to a coordinate system that rotates with angular velocity $\vec{\omega}$,

$$\frac{d\vec{I}}{dt} = \frac{\partial \vec{I}}{\partial t} + \vec{\omega} \times \vec{I} \quad (2.4)$$

Substitute Eq. 2.4 into Eq. 2.3, \vec{I} in the rotating frame satisfies the equation of motion

$$\hbar \frac{\partial \vec{I}}{\partial t} = \gamma\hbar\vec{I} \times (\vec{B} + \vec{\omega}/\gamma) = \gamma\hbar\vec{I} \times \vec{B}_{eff} \quad (2.5)$$

where \vec{B}_{eff} is the effective field in the rotating frame

$$\vec{B}_{eff} = \vec{B} + \vec{\omega}/\gamma \quad (2.6)$$

Thus, the effective field experienced by an observer in the rotating frame is simply the external field \vec{B} plus an additional field $\vec{\omega}/\gamma$.

If we apply this result to rotating magnetic fields, we will get the core idea of performing an Adiabatic Fast Passage (AFP) measurement. Assuming a constant field \vec{B} and another field \vec{B}_1 perpendicular to \vec{B} that is rotating with angular velocity $-\omega$. In the rotating frame that rotates with \vec{B}_1 , both aforementioned fields are just constant and the effective field in the rotating frame is

$$B_{eff}\vec{z} = (B - \omega/\gamma)\vec{z} + B_1\vec{x}' \quad (2.7)$$

where \vec{x}' is the direction that \vec{B}_1 is in. When on resonance ($B = \omega/\gamma$), the effective field is perpendicular to the constant field \vec{B} .

2.2.2.2 Quantum Mechanical Formulation

The above conclusion can be easily reached with quantum mechanics [46]. The Shrödinger equation for a magnetic moment in an external field is

$$i\hbar\dot{\psi} = \mathcal{H}\psi = -\gamma\hbar\vec{I}\cdot\vec{B}\psi \quad (2.8)$$

Let ψ and \vec{B} be the wave function and magnetic field in a stationary frame and ψ_r and \vec{B}_r be the same quantities in a rotating frame with angular velocity $\vec{\omega}$. Using the rotation operator in quantum mechanics,

$$\psi = e^{-i\vec{\omega}\cdot\vec{I}t}\psi_r \quad (2.9a)$$

$$\vec{I}\cdot\vec{B}_r = e^{i\vec{\omega}\cdot\vec{I}t}\vec{I}\cdot\vec{B}e^{-i\vec{\omega}\cdot\vec{I}t} \quad (2.9b)$$

Substituting 2.9 into Eq.2.8, the Shrödinger equation in the rotating frame is obtained

$$i\hbar\dot{\psi}_r = -\gamma\hbar\vec{I} \cdot (\vec{B}_r + \vec{\omega}/\gamma)\psi_r = -\gamma\hbar\vec{I} \cdot \vec{B}_{eff}\psi_r \quad (2.10)$$

The same effective field in the rotating frame is reached as that from the classical derivation.

2.2.3 Adiabatic Fast Passage

Adiabatic Fast Passage (AFP) NMR is used to measure the ^3He polarization. In an AFP measurement, with the assistance of a oscillating radiofrequency (RF) field, the spins follow the effective field in a rotating frame (as discussed in more detail below) and are flipped 180 degrees to the opposite direction and then flipped back, producing two peaks in signal when they're perpendicular to the holding field and the pick up coils.

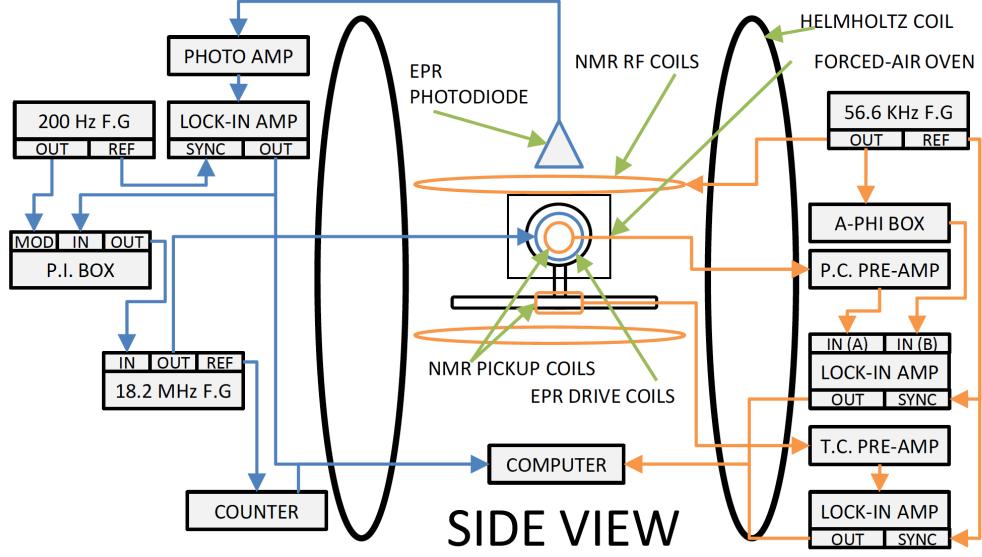


Figure 2.1: EPR (left) and AFP (right) setup. Adapted from Dolph's PhD thesis.

The flipping process can be achieved by either sweeping the main holding field or sweeping the RF frequency so that the longitudinal component of effective field in the rotating frame goes through zero. AFP measurements in our lab are typically done by sweeping the holding field while keeping the RF frequency constant. The RF coils produce a RF field of magnitude $2B_1$ perpendicular to the main holding field B . The oscillating field has a frequency of ω and can be decomposed into two counter-rotating components with the same amplitude B_1 . Only the component rotating in direction able to give a resonance in Eq. 2.7 has an important effect. In this frame, the effective field is

$$B_{eff} \vec{z} = (B - \omega/\gamma) \vec{z} + B_1 \vec{x}' \quad (2.11)$$

as discussed above. The other rotating component that rotates in the opposite direction does not affect the spins. In an AFP measurement, the holding field starts from a value lower than ω/γ ($\omega/\gamma - B \gg B_1$), so that the effective field is almost aligned with the holding field and the spins. The holding field is then swept at a constant rate through resonance to a value greater than ω/γ . The sweeping rate is of great importance. The sweep needs to be slow enough so that the nuclear spins can follow the effective field

$$\frac{\dot{B}}{B_1} \ll \omega \quad (2.12)$$

Sweep that satisfies this condition is considered as adiabatic.

Sweep rate cannot be too slow either, because the relaxation rate of the spins are faster near the resonance especially with a small effective field B_1 . The relaxation rate of ^3He in the rotating frame at resonance is

$$\frac{1}{T_{1r}} = D \frac{|\nabla B_z|^2}{B_1^2} \quad (2.13)$$

where D is the ^3He self-diffusion constant. In order to keep the AFP loss low, it's important for the time scale that the spins stay close to resonance to be much shorter than $1/T_{1r}$:

$$D \frac{|\nabla B_z|^2}{B_1^2} \ll \frac{\dot{B}}{B_1} \quad (2.14)$$

Typically, the field is swept from 12.6 Gauss to 20.4 Gauss in 6s, thus

$$\dot{B} = 1.3G/s \quad (2.15a)$$

$$B_1 \approx 100mG \quad (2.15b)$$

$$f = 56.6kHz \quad (2.15c)$$

$$D \approx 0.16cm^2/s \quad (2.15d)$$

$$|\nabla B_z| \approx 10mG/cm \quad (2.15e)$$

$$(2.15f)$$

With these operating conditions,

$$D \frac{|\nabla B_z|^2}{B_1^2} \approx 1.6mHz \quad (2.16a)$$

$$\frac{\dot{B}}{B_1} \approx 13Hz \quad (2.16b)$$

$$w \approx 356kHz \quad (2.16c)$$

The AFP conditions are clearly well satisfied. Fig.2.2 shows the evolution of effective field in the rotating frame during an AFP measurement.

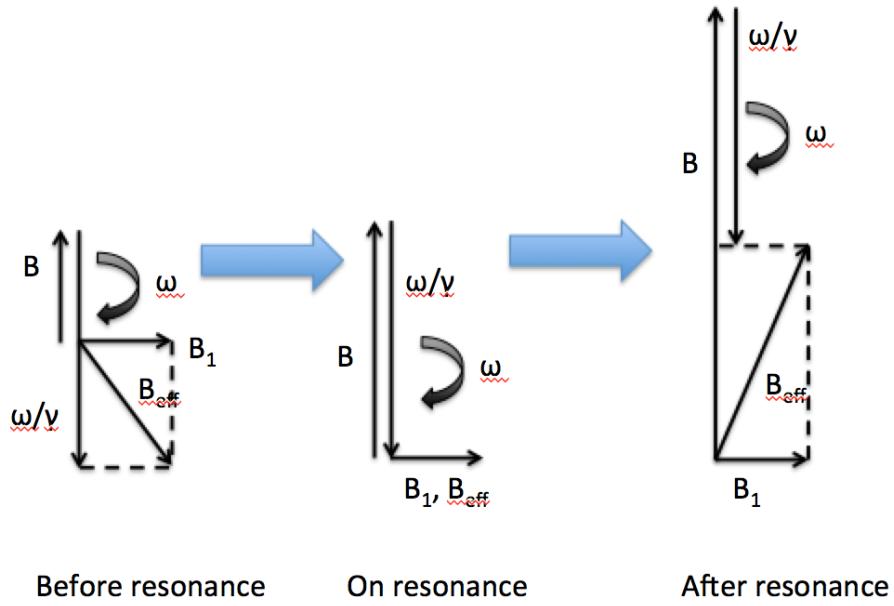


Figure 2.2: Effective field in the rotating frame during an Adiabatic Fast Passage measurement. The ${}^3\text{He}$ spins follow the direction of the effective field. B_1 is exaggerated to show different components of effective field clearly.

The pick up coils are placed close to the cell and perpendicular to the holding field and RF field. As the ${}^3\text{He}$ spins precess along the holding field, the transverse component of the spins will induce an electromotive force (EMF) that is directly proportional to the amplitude of the component in the pick up coils. The signal can be written as:

$$S = A\omega \sin \alpha(t) = A\omega \frac{B_1}{\sqrt{B_1^2 + (B(t) - \omega/\gamma)^2}} \quad (2.17)$$

where A is a constant that accounts for the cell and coils geometry, the cell magnetization and the electronics factors that affect the size of signal; ω is the RF frequency; α is the angle between the effective field and the holding field in the rotating frame; $B(t)$ is the holding field as a function of time. The signal reaches peak value when $B(t) = \omega/\gamma$. Fig.2.3 shows the result of a typical AFP measurement.

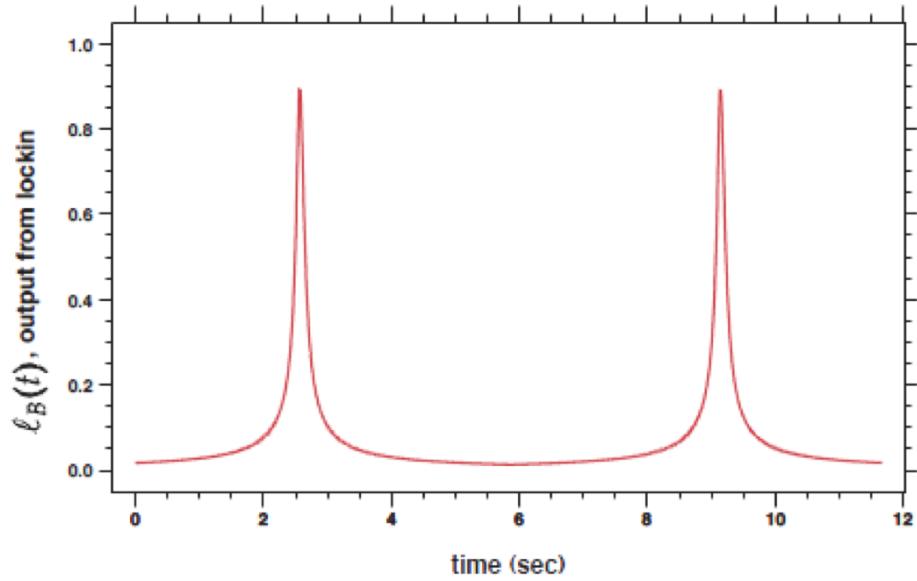


Figure 2.3: A typical AFP signal. y axis is in arbitrary unit.

2.2.4 AFP Loss

The longitudinal spin relaxation rate due to static field inhomogeneities is [32, 52, 21]

$$\frac{1}{T_1} = D \frac{|\nabla B_x|^2 + |\nabla B_y|^2}{B_0^2} \quad (2.18)$$

where D is the diffusion constant for the polarized spins, and is inversely proportional to the gas pressure. B_0 is the mean magnetic field along z axis. B_x and B_y are the x and y components of the magnetic field. However, when performing AFP measurement, the spins are exposed to a small oscillating RF field, the spin relaxation can be greatly accelerated under magnetic resonance conditions [22],

$$\frac{1}{T_{r1}} = \frac{8R^4}{175D} |\nabla\Omega_z|^2 \sum_n \frac{175}{4(\chi_{1n}^2 - 2)(\chi_{1n}^4 + r^2 + r^2s^2)(1 + s^2)} \quad (2.19)$$

where R is the cell radius, D is the diffusion constant, Ω_z is the Larmor frequency of the holding field, $r = \frac{\omega_r R^2}{D}$, $s = \frac{\Omega_0 - \omega}{\omega_r}$, the numbers χ_{1n} are the zeros of the derivatives of the spherical Bessel functions

$$\frac{d}{dx} j_1(x_{1n}) = 0 \text{ for } n = 1, 2, 3\dots \quad (2.20)$$

Since $r^2 \gg \chi_{1n}^4$, and $\sum_n \frac{1}{\chi_{1n}^2 - 2} = \frac{1}{2}$ [21],

$$\frac{1}{T_{r1}} = \frac{R^4 |\nabla\Omega_z|^2}{r^2(1 + s^2)^2 D} = \frac{|\nabla B_z|^2 D}{B^2(1 + s^2)^2} \quad (2.21)$$

If P_0 is the polarization before AFP, the polarization P after a single AFP flip is given by

$$P = P_0 e^{-\int \Gamma_{r1} dt} = P_0 e^{-\int \frac{1}{T_{r1}} dt} \quad (2.22)$$

Given the field sweep starts from 12.6G, ends at 20.4G, the RF frequency is 56.6kHz, the sweep time is 6s and B_1 is 100mG, we can safely approximate the integral by

$$\int_{-\infty}^{\infty} \frac{1}{T_{r1}} dt = \frac{\pi D |\nabla B_z|^2}{2B_1 \partial B_1 / \partial t} \quad (2.23)$$

which is the fractional loss due to a single AFP flip.

To better understand AFP loss, we performed a study where we took AFP measurements at various different field gradients to study the relation between AFP loss and inhomogeneities. The gradients were produced by Maxwell-style transverse gradient coils and increased from 0 to a little under 160 mG/cm. At each set gradient, we take one AFP to look at the difference between the two peaks to determine the loss due to a single flip. Fig 2.4 shows AFP losses collected from experiments and theoretical predictions. They agree mostly within the error bar.

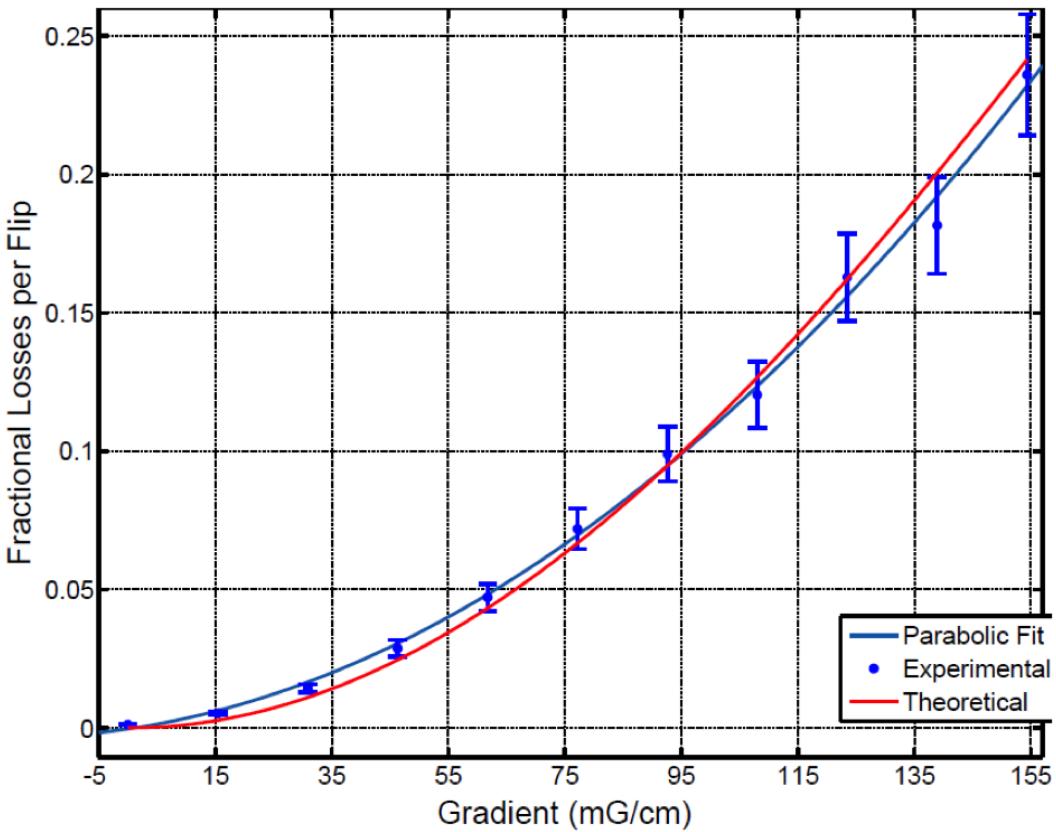


Figure 2.4: Fractional AFP loss (single flip) as a function of field gradient.

2.3 Electron Paramagnetic Resonance

2.3.1 Overview

Electron Paramagnetic Resonance (EPR) is a important technique for measuring the frequency shift of alkali metal Zeeman resonance due to the effective magnetic field produced by polarized ^3He gas. The EPR shift is largely caused by the Fermi-contact interaction $\alpha \mathbf{K} \cdot \mathbf{S}$ between the nuclear spin \mathbf{K} of the noble gas nucleus of magnetic moment μ_K and the electron spin \mathbf{S} of the alkali metal atom [12]. The magnetic

field created by the bulk magnetization of the ${}^3\text{He}$ gas also contributes directly to a relatively small part of the shift (roughly 1/6 for K). The total measured shift is therefore written as the expected Zeeman interaction with the field produced by the polarized ${}^3\text{He}$ multiplied by an enhancement factor κ_0 . The enhancement effect comes from overlapping of alkali metal electrons and ${}^3\text{He}$ nuclei during binary collisions, thus κ_0 is different for each alkali metal species and slightly temperature dependent.

During the process of optical pumping, the Rb atoms are excited to the $5\text{P}_{\frac{1}{2}}$ state by the pump laser. The majority of these atoms are quenched non-radiatively to the ground state by N_2 . While at $5\text{P}_{\frac{1}{2}}$ state, Rb atoms can also be excited to the $5\text{P}_{\frac{3}{2}}$ state through collisions with other Rb atoms. A small fraction of the excited atoms ($5\text{P}_{\frac{1}{2}}$ and $5\text{P}_{\frac{3}{2}}$) decay by emitting either a D_1 photon or D_2 photon. The intensity of fluorescence is proportional to the excited Rb atoms, thus is higher when the Rb polarization is low so more Rb atoms can absorb laser and jump to the excited state. We typically induce Zeeman transitions with a RF coil to lower alkali polarization and detect D_2 photons with a photodiode behind a D_2 filter. The highest amount of D_2 photons is detected when the RF frequency is exactly equal to the Zeeman transition frequency.

2.3.2 The Breit-Rabi Equation

The Zeeman energy levels of ground state ($L = 0$) can be described with the Breit-Rabi equation[17]

$$E_{F=I\pm 1/2, m_F} = -\frac{\hbar \Delta \nu_{hfs}}{2(2I+1)} - \mu_N g_I B m_F \pm \frac{\hbar \Delta \nu_{hfs}}{2} \sqrt{1 + \frac{4m_F x}{2I+1} + x^2} \quad (2.24)$$

where

$$x = (g_I\mu_N - g_s\mu_B) \frac{B}{h\Delta\nu_{hfs}} \quad (2.25)$$

B is the magnetic field, $\Delta\nu_{hfs}$ is the hyperfine splitting frequency, I is the nuclear spin, g_I and g_s are the g factors of nuclear and electron spin, μ_N and μ_B are the nuclear and Bohr magneton, respectively.

The Zeeman transition frequency of $m_F \rightarrow m_F - 1$ is

$$\begin{aligned} \nu_{m_F \rightarrow m_{F-1}} &= \frac{E_{F,m_F} - E_{F,m_{F-1}}}{h} \\ &= -\frac{g_I\mu_N B}{h} \pm \frac{\Delta\nu_{hfs}}{2} \left(\sqrt{1 + \frac{4m_F}{2I+1}x + x^2} - \sqrt{1 + \frac{4m_F-1}{2I+1}x + x^2} \right) \end{aligned} \quad (2.26)$$

The second term is much greater than the first term under our operating conditions, so the sign of the frequency $\nu_{m_F \rightarrow m_{F-1}}$ depends on the second term only. If we focus on the top hyperfine manifold in the energy level graph, the transition frequency is

$$\nu_{m_F \rightarrow m_{F-1}} = -\frac{g_I\mu_N B}{h} + \frac{\Delta\nu_{hfs}}{2} \left(\sqrt{1 + \frac{4m_F}{2I+1}x + x^2} - \sqrt{1 + \frac{4m_F-1}{2I+1}x + x^2} \right) \quad (2.27)$$

2.3.3 Shift of Zeeman Frequency

Under our operating condition, the size of Zeeman splitting is much less than hyperfine splitting, which makes x a small number. The Taylor expansion of Eq. 2.27 is

$$\begin{aligned}\nu_{m_F \rightarrow m_{F-1}} = & -\frac{g_I \mu_N B}{h} \\ & + \frac{\Delta \nu_{hfs}}{2} \left(\frac{2x}{2I+1} - \frac{2(2m_F-1)x^2}{(2I+1)^2} + \frac{(-(2I+1)^2 + 4 - 12m_F + 12m_F^2)x^3}{(2I+1)^3} + \dots \right)\end{aligned}\quad (2.28)$$

with the approximation

$$g_s \mu_B \gg g_I \mu_N \quad (2.29a)$$

$$x \approx -\frac{g_s \mu_B B}{h \Delta \nu_{hfs}} \quad (2.29b)$$

then to the lowest order approximation, the shift of $\nu_{m_F \rightarrow m_{F-1}}$ due to a small effective field ΔB ($\Delta B \ll B$) from polarized ${}^3\text{He}$ is

$$\begin{aligned}\Delta \nu_{m_F \rightarrow m_{F-1}} = & -\frac{g_s \mu_B}{h(2I+1)} \Delta B \left[1 + 2(2m-1) \frac{g_s \mu_B B}{h \Delta \nu_{hfs} (2I+1)} \right. \\ & \left. + 6 \left(-\frac{(2I+1)^2}{4} + 1 - 3m + 3m^2 \right) \left(\frac{g_s \mu_B B}{h \Delta \nu_{hfs} (2I+1)} \right)^2 + \dots \right]\end{aligned}\quad (2.30)$$

Usually the pumping chamber is spherical, the magnetic field produced inside a uniformly magnetized sphere is [39]

$$\Delta \mathbf{B} = \frac{2}{3} \mu_0 \mathbf{M} \quad (2.31)$$

where μ_0 is the vacuum permeability, \mathbf{M} is the magnetization of ${}^3\text{He}$,

$$\mathbf{M} = \boldsymbol{\mu}_K [\text{He}] P \quad (2.32)$$

where μ_K is the magnetic moment of ${}^3\text{He}$, $[\text{He}]$ is its density, and P its polarization. As we mentioned before, as a result of the Fermi-contact interaction $\alpha \mathbf{K} \cdot \mathbf{S}$ between the nuclear spin \mathbf{K} of the noble gas nucleus and the electron spin \mathbf{S} of the alkali metal atom, the effective magnetic field from the polarized ${}^3\text{He}$ gas is enhanced by a factor of κ_0 [49]:

$$\Delta \mathbf{B} = \frac{2}{3} \kappa_0 \mu_0 \boldsymbol{\mu}_K [\text{He}] P \quad (2.33)$$

The enhancement factor κ_0 was measured by Romalis *et al.* in 1998 with an error of 1.5% [49]

$$\kappa_0^{Rb-{}^3He} = 4.52 + 0.00934[T(\text{ }^\circ\text{C})] \quad (2.34)$$

then it was measured by Babcock *et al.* in 2005 [12]

$$\kappa_0^{Rb} = 6.39 + 0.00914[T - 200(\text{ }^\circ\text{C})] \quad (2.35a)$$

$$\kappa_0^K = 5.99 + 0.0086[T - 200(\text{ }^\circ\text{C})] \quad (2.35b)$$

$$\kappa_0^{Na} = 4.84 + 0.00914[T - 200(\text{ }^\circ\text{C})] \quad (2.35c)$$

The two results agree within the error. Thus we can calculate ${}^3\text{He}$ polarization with the EPR frequency shift.

2.3.4 Experimental Methods

2.3.4.1 Overview

Under normal operating conditions, hybrid cells with mixture of Rb and K are used. The vapor density of K is an order of magnitude higher than that of Rb, we typically induce the $m_F = 2 \rightarrow m_F = 1$ (assuming the angular momentum of laser photons is +1) ^{39}K transition, which lowers the K polarization. Rb-K spin-exchange rate is fast enough that Rb is depolarized almost instantly. This allows more Rb atoms to absorb laser and be excited to the $5\text{P}_{\frac{1}{2}}$ state which in turn produces more D_2 fluorescence. The D_2 fluorescence is at maximum intensity when the RF frequency is on resonance for the Zeeman transition.

We first locate the frequency with a frequency-modulated (FM) sweep, and set the RF frequency to the found value. The RF is locked to the frequency (which is slightly changing) that induces maximum D_2 light with a proportional-integral feedback circuit (P.I. box). This frequency is referred to as EPR frequency and is measured with a counter. To separate the frequency-shifting effect of polarized ^3He from other sources that may affect the transition frequency, we flip the ^3He magnetization by performing a RF frequency sweep. A frequency sweep is chosen rather than a holding field sweep to keep external magnetic field constant, thus reducing factors that affect Zeeman splitting size. By comparing the frequency measured before and after the flip, together with the real temperature inside the pumping chamber, we can calculate the ^3He polarization. We typically take AFP measurements right before and after the relatively quick EPR measurement, so that a calibration constant that translates AFP signal size to ^3He polarization can be calculated.

2.3.4.2 Locating Zeeman Transition Frequency

The P.I. box only works well in locking the EPR frequency to the $m_F = 2 \rightarrow m_F = 1$ K transition when the EPR frequency is close to the transition. Thus the first step in EPR measurements is to locate the Zeeman transition. A frequency-modulated (FM) sweep is performed through a range that covers the Zeeman transition, the range is known from experience or calculation and the P.I. box remains off during the sweep.

The RF frequency is generated by the 18.2MHz voltage-controlled oscillator (VCO). The D₂ fluorescence is detected with the photodiode and recorded during the sweep. The RF is frequency-modulated by a 200Hz signal, the VCO output at any moment during the sweep can be described as:

$$V_{FM}(t) = V_{C0} \sin(2\pi[f_c + D_f \sin(2\pi f_m t + \phi_m)]t + \phi_c) \quad (2.36)$$

where V_{C0} is the amplitude of the sweeping RF frequency (carrier), f_c is the RF frequency that is being swept through a set range, D_f is the peak frequency deviation, f_m is the modulating frequency (200Hz in our case), ϕ_m and ϕ_c are the phase of the modulation frequency and carrier frequency, respectively. Thus the RF frequency is

$$f_{FM}(t) = f_c(t) + D_f \sin(2\pi f_m t + \phi_m) \quad (2.37)$$

where $f_c(t)$ emphasizes the RF frequency is sweeping over time.

The D₂ light intensity can be described with a Lorentzian function:

$$I(f(t)) = \frac{I_0}{(f_{FM}(t) - f_0)^2 + \Gamma^2} \quad (2.38)$$

where f_0 is the Zeeman transition frequency, Γ is the line width. Keeping the first order term of the Taylor expansion of Eq. 2.38, the D_2 light intensity is

$$I(f(t)) = I(f_c(t)) + \frac{\partial I}{\partial f} \Big|_{f=f_c(t)} D_f \sin(2\pi f_m t + \phi_m) \quad (2.39)$$

A lock-in amplifier is used to select only the f_m term and reduce the noise, the signal picked by the lock-in is

$$s(t) = \frac{\partial I}{\partial f} \Big|_{f=f_c(t)} D_f \sin(2\pi f_m t + \phi_m) \quad (2.40)$$

which is the derivative of the Lorentzian function multiplied by a sine function. The FM sweep line crosses zero when the RF frequency is equal to the Zeeman transition frequency (peak of the Lorentzian function), which produces the maximum D_2 light intensity. The region between the lowest and highest points of the derivative line is fitted to a line, and the zero-crossing point of the line is used as the Zeeman transition frequency. Fig. 2.5 shows an FM sweep.

2.3.4.3 EPR Spin Flip Process

After the transition frequency is located, the VCO frequency is first set to it and then stays locked with a proportional-integral feedback circuit (P.I. box). The circuit is shown in Fig. 2.6.

The output of the lock-in amplifier is sent to the input of the P.I. box, and the output of the P.I. box controls the input of the VCO. If the effective magnetic field drifts away from resonance, the P.I. box would receive a non-zero input, and attempt to change the frequency of the VCO output so that its input is zero again. Therefore,

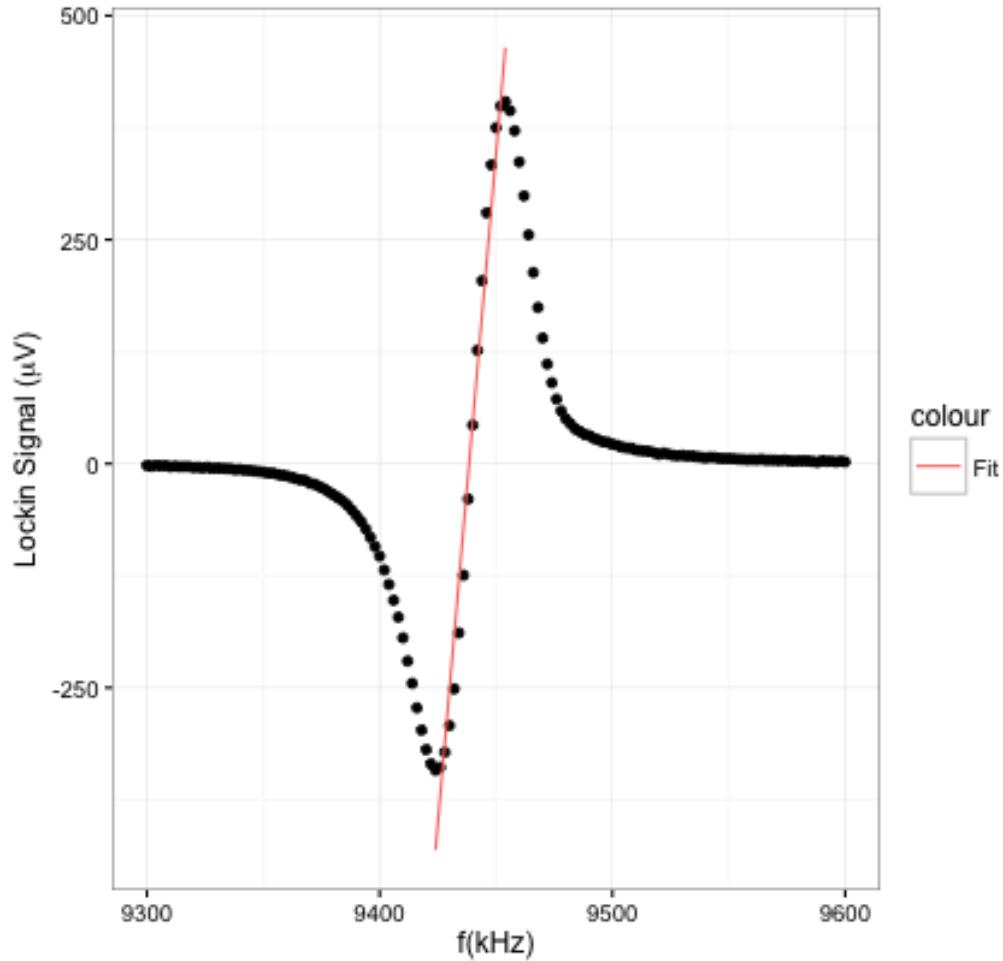


Figure 2.5: A typical FM sweep on a hybrid cell. The central region between the minimum and maximum is fitted to a line. The zero crossing point corresponds to the Zeeman transition frequency.

the VCO output always matches the transition frequency.

Because the EPR frequency is also affected by sources other than the polarized ^3He such as the holding field and earth field, we flip the ^3He spins by sweeping the frequency while keeping the holding field unchanged. The contribution from the flipped spins has a opposite sign while other factors still contribute in the same

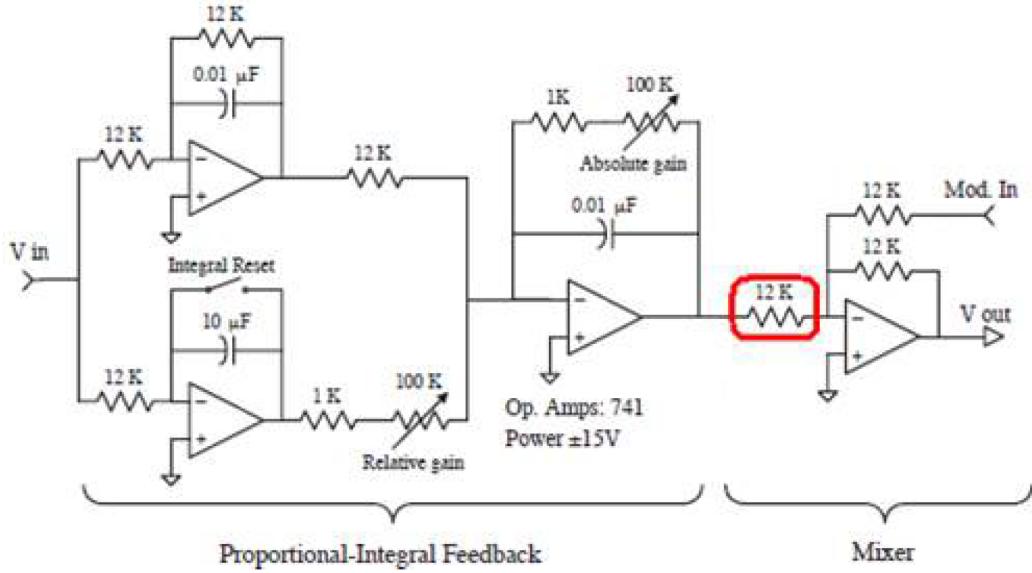


Figure 2.6: The same P.I. circuit that was first used by Romalis in our lab. The drawing was then corrected by Peter Dolph.[5]

way which allows us to extract the change of Zeeman transition frequency due to polarized ${}^3\text{He}$ and calculate the polarization. We typically let the cell polarization reach saturation before performing EPR measurements. AFP measurements are taken right before and after the EPR measurements for calculating the calibration constant (the ratio between polarization and AFP signal size). Fig. 2.7 shows a typical EPR spin flip process.

Under normal operating conditions for a double-chambered cell, the pumping chamber is heated to around 170°C or 235°C depending on if the cell is hybrid, the target chamber and transfer tube remain at room temperature. The temperature difference causes differences in gas densities and affects AFP signal size. Temperature controller of the oven only maintains the surface temperature of the pumping chamber at set temperature, but the gas inside the pumping chamber is always hotter due to

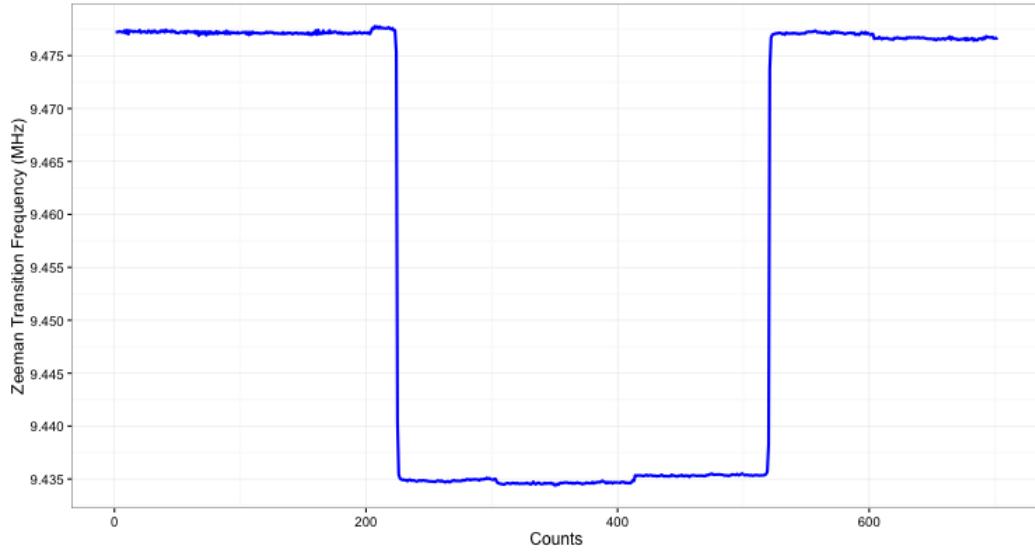


Figure 2.7: An EPR measurement for a hybrid cell at 235°C.

The spins are flipped around 200 mark, and flipped back around 500 mark.

absorption of laser energy. The enhancement factor κ_0 is also slightly temperature dependent which may be underestimated by $\sim 4\%$ when using the surface temperature as the gas temperature. Dolph described a method called temperature test to extract gas temperature inside the pumping chamber in detail in his thesis [5]. The idea is to take AFP measurements when the laser is blocked and unblocked multiple times, assuming the change of gas densities due to absorption of laser is the only reason for the difference in signal size (aside from AFP loss which is compensated through fitting) and the gas temperature when laser is blocked is the same as that measured by RTDs (resistance temperature detectors) on the exterior of the pumping chamber, one can calculate the inside temperature when laser is unblocked.

2.4 Pulsed Nuclear Magnetic Resonance

Adiabatic Fast Passage has been the main technique used in our lab for measuring ^3He polarization through detecting precessing spins with detection coils (we refer to them as pickup coils). In an AFP measurement, all ^3He spins are flipped by sweeping the holding field while applying a RF field. In more recent studies, we have been exploring the possibility of replacing conventional glass windows with metal end windows in response to the 12 GeV upgrade of JLab. Because of the lack of studies on spin relaxation of polarized ^3He on metal surfaces, various test cells made with large metal parts as well as glass parts are being studied in our lab. The inclusion of metal parts immediately renders AFP almost useless in the situation because of effects such as Eddy current, thus we have been performing Pulsed Nuclear Magnetic Resonance (PNMR) on metal cells.

2.4.1 The Rotating Coordinate System

In a PNMR measurement, a short pulse of RF frequency is applied to a small fraction of ^3He gas. The RF frequency is tuned to be on resonance at the Larmor frequency of the holding field. As discussed before with AFP, in the rotating coordinate system, there will be an effective field due to rotation that exactly cancels the holding field which we assume to be in the z direction. Thus the z component of the effective field is zero and there is a non-zero constant transverse component which we will call B_1 . The nuclear spins will precess along B_1 and end up at an angle away from z axis:

$$\alpha = \gamma B_1 \Delta t \quad (2.41)$$

where α is the angle (tip angle), γ is the gyromagnetic ratio, and Δt is the RF pulse duration.

2.4.2 Free Induction Decay

At the end of the RF pulse, the tipped spins will have a transverse component equal to the magnetization multiplied by $\sin \alpha$. The spins continue to precess along the holding field and the transverse component will induce a signal in the pickup coils.

In addition to precession, the spins are affected by two types of relaxation processes. The first type is called the spin-lattice relaxation, it describes the rate at which the longitudinal component of magnetization approaches the thermodynamic equilibrium value. It is characterized by the spin-lattice relaxation time constant T_1 . The rate of change of the longitudinal component is

$$\dot{M}_z = -(M_z - M_0)/T_1 \quad (2.42)$$

where M_0 is the thermodynamic equilibrium magnetization. Solving the differential equation gives

$$M_z(t) = M_0 - [M_0 - M_z(0)] e^{-t/T_1} \quad (2.43)$$

The name spin-lattice relaxation refers to the process in which the spins transfer energy to surrounding, thereby restoring their equilibrium state.

The second relaxation process is the transverse relaxation, which is also referred to as the T_2 relaxation and spin-spin relaxation. The transverse component of magnetization decays because variations in the local magnetic field cause different moments

to precess at different rates. This is the T_2 process. Normally, the dominating relaxation effect however, is another dephasing process due to inhomogeneities in the holding field over the volume of the cell.

The measured relaxation rate of the tipped spins is the result of all these effects combined:

$$\frac{1}{T_2^*} = \frac{1}{T_2} + \gamma\Delta B_0 \quad (2.44)$$

where ΔB_0 is the variation in the holding field. $\gamma\Delta B_0$, the dominant term, is a spread in Larmor frequencies $\Delta\omega_0$, which causes spin-spin dephasing in a characteristic time of $1/\Delta\omega_0$.

The time evolution of the nuclear magnetization \mathbf{M} is described by the Bloch equations [15]:

$$\frac{\partial M_x(t)}{\partial t} = \gamma (\mathbf{M}(t) \times \mathbf{B}(t))_x - \frac{M_x(t)}{T_2^*} \quad (2.45a)$$

$$\frac{\partial M_y(t)}{\partial t} = \gamma (\mathbf{M}(t) \times \mathbf{B}(t))_y - \frac{M_y(t)}{T_2^*} \quad (2.45b)$$

$$\frac{\partial M_z(t)}{\partial t} = \gamma (\mathbf{M}(t) \times \mathbf{B}(t))_z - \frac{M_z(t)}{T_1} \quad (2.45c)$$

where γ is the gyromagnetic ratio and the cross products are the precession terms, the last terms in each equation represent the decaying and dephasing of each component. The precessing spin magnetization generates a signal in the pickup coils that decays with time. This is called free induction decay, the induced signal is of the shape:

$$V(t) = A\omega_0 \sin \alpha \sin(\omega_0 t + \phi) e^{-t/T_2^*} \quad (2.46)$$

where A is just a constant, ω_0 is the Larmor frequency for the holding field, α is the tip angle, T_2^* is the measured decay time constant. For our metal test cells, depending on the location of the pickup coils and the field setup, T_2^* varies between several milliseconds to more than 300 milliseconds.

2.4.3 Experimental Methods

Our PNMR setup is shown in Fig. 4.10. The Labview program on the computer controls the timing of a gate signal that is fired from the first function generator. The gate signal is fed to the back of the second function generator and triggers it to produce a short pulse. The second function generator sends out RF pulse with pre-set amplitude, duration and frequency only when the gate signal is of voltage higher than the threshold. The frequency of the RF pulse is carefully tuned to be at the Larmor frequency of the holding field.

The pulse is sent from the function generator to a coil wrapped directly on a small portion of the cell. The spins contained in the coil are exposed to the pulse and tipped by an angle which depends on the amplitude and the duration of the pulse. In the rotating frame, the effective field B_1 causes the spins to precess around it (as discussed before), the precession frequency is γB_1 so the angle the spins rotate by (tip angle) is

$$\alpha = \gamma B_1 \Delta t \quad (2.47)$$

where γ is the gyromagnetic ratio, the effective field B_1 is directly proportional to the amplitude of the RF pulse, Δt is the duration of the pulse. Ideally, a 90° tip angle

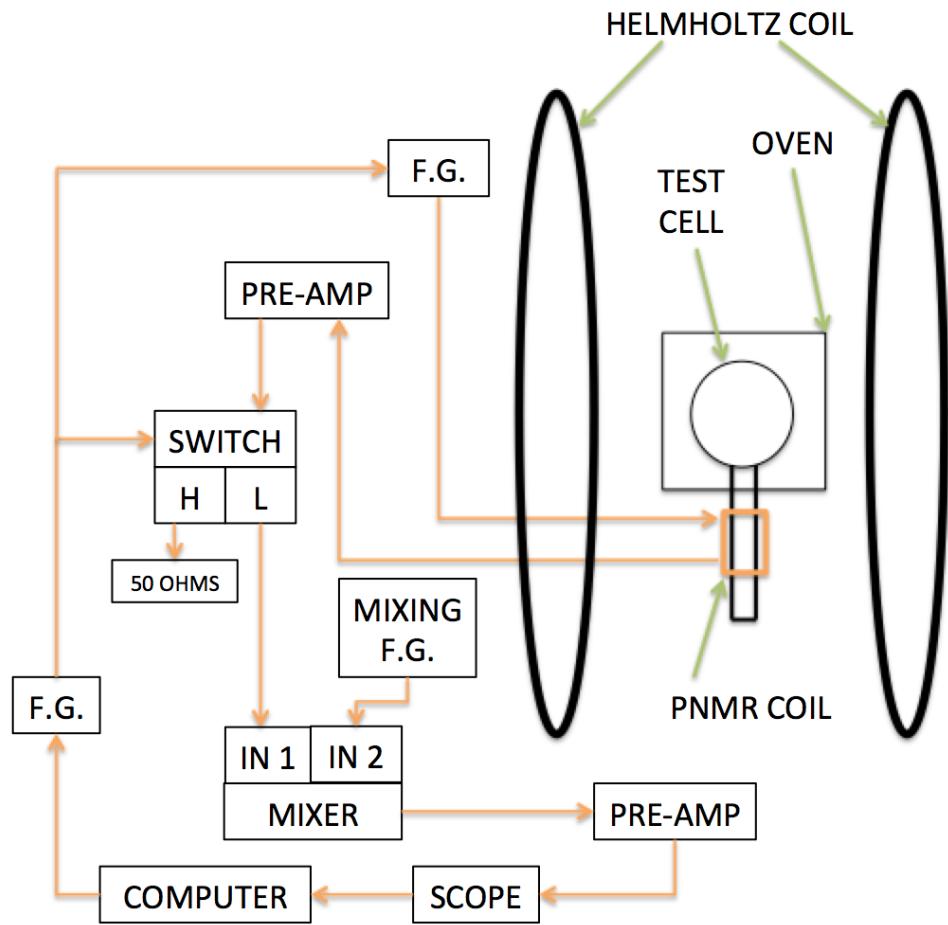


Figure 2.8: PNMR setup.

would result in the maximum signal, but it has not been the case for us most of the time because of field inhomogeneities. The coils are normally wrapped on the transfer tube of the cell which is off the center of the holding field and exposed to greater inhomogeneities. The details of how we measure the test cells will be discussed in later chapters. Relatively large inhomogeneities cause the spins to precess at different rates, and the dephasing becomes more significant with longer pulse duration and larger tip angle, which leads to non-optimal signal. A typical tip angle for us would

be between 30° and 45° .

After the spins are tipped away from z axis, they precess around the holding field and induce a signal in the detection coil. The signal is amplified by a low noise pre-amplifier first and then goes through an isolation switch. The switch only lets signal pass when the controlling gate voltage is low, thus stops the RF pulse from coming back through the detection circuit. The signal is at the Larmor frequency, and is mixed with another frequency after the switch. The mixing frequency is only slightly different from the Larmor frequency, the output of the mixer has both the sum of the two frequencies and the difference. A second pre-amplifier is used to select and amplify the lower of the two frequencies while filtering out high frequency noises. The final output is displayed on a oscilloscope and collected by the Labview program on the computer. Fig. 4.11 shows a PNMR measurement with around 150 ms decay time constant.

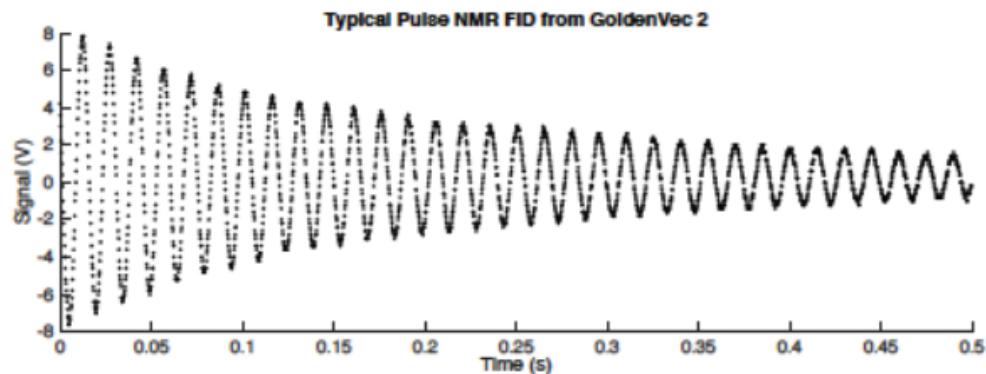


Figure 2.9: A PNMR signal taken with gold coated test cell.

The tip angle can be measured with a short sequence of FID signals. Theoretically

the tip angle can be calculated with Eq. 2.47. But because of inhomogeneities and other factors, the calculation serves as only an estimate, it is often more accurate and convenient to measure the tip angle directly. We take several PNMR measurements in quick succession with the same RF pulse settings. After every pulse, the transverse component of the spins quickly decay and dephase, leaving only the longitudinal component which is equal to $\cos \alpha$ times the original magnetization. The intervals between measurements are short enough so that T_1 can be safely ignored. The series of measurements also need to be performed on the same portion of the gas (i.e. the same group of spins tipped by the first pulse), thus it is important to know that the self-diffusion of ^3He is significantly slower than the sampling rate. The self-diffusion coefficient of ^3He at 300K is [47]

$$D = \frac{1440(80)\text{torr}}{P} \text{cm}^2/\text{s} \quad (2.48)$$

which is roughly $1.89 \text{ cm}^2/\text{s}$ at 760 torr (the test cells normally contain around 1 atm of ^3He). The diffusion length is described by

$$l = 2\sqrt{Dt} \quad (2.49)$$

Thus in one second, the gas will move around 2.75 cm through self-diffusion. For this reason, we only take 2 or 3 PNMR measurements to calculate the tip angle. As additional measurements would have given enough time for the tipped spins from the first PNMR and the surrounding spins to mix.

Since only the longitudinal component of the tipped spins are preserved, the amplitude of the i_{th} PNMR is

$$V_i = V_0 \cos^{i-1} \alpha \quad (2.50)$$

where V_0 is the induced signal in the first PNMR. We can then use this equation to calculate the tip angle α .

Chapter 3

Development of Hybrid Targets

3.1 Overview

In this chapter, I present the development of high-performance polarized ^3He targets for use in electron scattering experiments that utilize the technique of alkali-hybrid spin-exchange optical pumping [54]. Data of 24 separate target cells are presented, each of these cells was constructed while preparing for one of four experiments at Jefferson Laboratory (JLAB). The results document dramatic improvement in the performance of polarized ^3He targets. I focus on the data analysis work in this chapter since most of the data had already been taken by the time I joined the group. Other details are described by Jaideep Singh [54]. With the wide range of data, we successfully determined the so-called X-factors that quantify an as-yet poorly understood spin-relaxation mechanism that limits the maximum achievable ^3He polarization to well under 100%. The data collected also served as a measurement of the K- ^3He spin-exchange rate coefficient $k_{se}^K = (7.46 \pm 0.62) \times 10^{-20} \text{ cm}^3/\text{s}$ over the

temperature range 503 K to 563 K.

3.2 Development of Hybrid Targets

Spin-exchange optical pumping (SEOP) is a two step process in which an alkali-metal vapor is polarized with optical pumping which subsequently polarizes noble-gas nuclei via spin-exchange collisions[61]. A pure Rb vapor was used to polarize ^3He prior to the development of hybrid cells. However, it was found that K is far more efficient than Rb at transferring its polarization to ^3He nuclei. Hybrid mixtures of Rb and K were used more and more frequently to improve the efficiency of the polarization process [63, 14]. In alkali-hybrid spin-exchange optical pumping (AHSEOP), the Rb vapor is polarized by circularly polarized laser, but the polarization of Rb valence electrons is then rapidly shared with the K [11]. The rate at which Rb and K exchange polarization is so fast that their polarizations can be thought of as being equal. If the alkali-hybrid mixture contains significantly more K than Rb with appropriate ratio, the spin-exchange efficiency is greatly improved so that the rate at which ^3He is polarized is increased significantly for a given amount of laser power.

The second factor that proved to have improved target cells performance greatly was the use of spectrally-narrowed diode lasers [23]. We were able to achieve higher alkali polarization with the aid of these lasers, which in turn reduced the required laser power. The origins of the improved cell performance are twofold. Firstly, these narrowband lasers have spectral profiles closely match the Rb D₁ absorption line shapes, which results in higher optical pumping rates and hence higher alkali polarizations. Secondly, it allows us to use higher alkali densities (which increases spin-exchange

rates) without sacrificing alkali polarization.

The data collected over the years include ^3He polarization achieved under different operating conditions, the time constants of polarization process, the geometric properties of the target cells, and cell fill information such as pressure and ratio of K to Rb in hybrid mixtures, the time constants of spin relaxation process. In roughly half the cells, the alkali polarization and alkali density were also measured with Faraday rotation techniques. The results contain several thousand hours worth of data and provide valuable information for future cell development.

Two figures of merit (FOMs) are plotted in Fig. 3.1, both of which are relevant in evaluating the performance of a polarized ^3He target. The one on the left axis is the effective luminosity $\mathcal{L}^{eff} = \mathcal{L}P_{He}^2$, where \mathcal{L} is the luminosity for a fixed-target experiment (the product of beam current, target density, and interaction length) and P_{He} is the ^3He polarization. The luminosity \mathcal{L} represents the number of scattering opportunities per unit time per unit area, while P_{He}^2 accounts for the reduction in statistical error of some polarization-dependant asymmetry. The FOM on the right axis is used to quantify the potential effective luminosity of a target. The definition is $\mathcal{L}^N = \mathcal{N}\Gamma_s P_{He}^2$, where \mathcal{N} is the total number of ^3He atoms in the target, Γ_s is the rate at which polarization builds up. The target cell Antoinette is the first one with such high value of \mathcal{L}^N , which indicates this cell could tolerate higher luminosities than previously achieved. The high potential further demonstrates the importance of the development of the new convection style target cell [29]. With even higher luminosities in electron scattering experiments, significantly faster gas transfer becomes quite necessary to reduce the polarization gradient between the pumping chamber of target chamber.

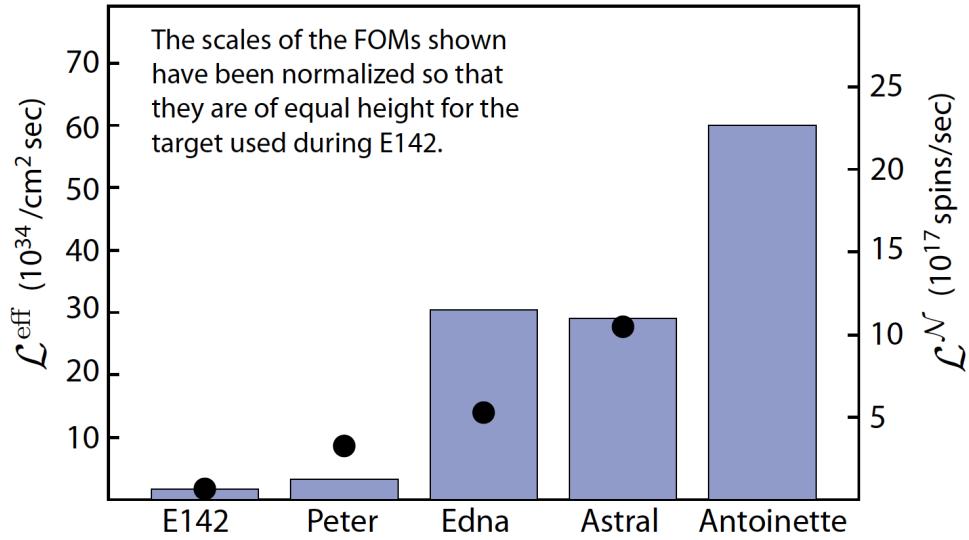


Figure 3.1: Shown are two figures of merit (FOM) for targets built for the indicated experiments. The circles (left axis) indicate the luminosity weighted by the square of polarization. The bars (right axis) represent the total number of spins being polarized per second weighted by the square of polarization. While the right FOM is an indication of the potential of the polarization technique, the left FOM indicates performance achieved during an experiment. The scales have been normalized so that the two FOMs have the same height for the cell marked E142

3.2.1 Experimental Methods

3.2.1.1 The ${}^3\text{He}$ Targets

Chapter 2 has already described single-chambered cell polarization dynamics to some extent as it is a simpler model for introducing spin-exchange optical pumping. The ${}^3\text{He}$ target cells JLab uses for electron scattering experiments usually include two chambers, a pumping chamber (PC), which is placed in an oven and pumped by

circularly polarized laser, and a target chamber (TC) that the electron beam passes through. Fig.3.2 shows a typical cell.

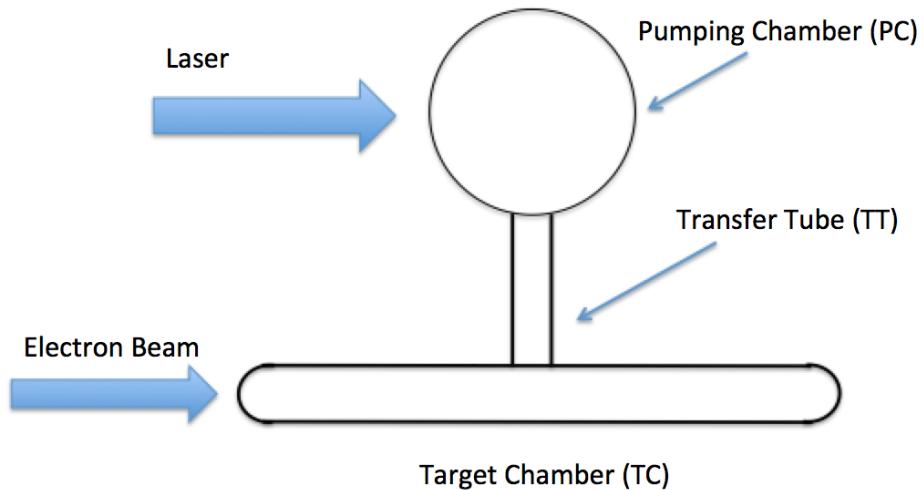


Figure 3.2: A target cell. The dimensions of different parts of the cell are not to scale.

After baking the cell to remove moisture and other contaminants, mixtures of Rb and K are chased into the cell. Once the cell has been pumped with a diffusion pump for about a week, we can fill the cell with N₂ and ³He.

The ³He density is of great importance for characterizing the target cells. One way to determine the ³He density is through measurements during the cell-filling process. A carefully calibrated volume, together with pressure and temperature measurements gives the volume of different spaces in the gas system (the system that is used to pump the cell and fill it with N₂ and ³He) and the cell. By comparing the amount of ³He

left in the system, the amount that went into the cell is obtained. The volume of the cell can be measured by determining its buoyancy force in water. The ^3He density is determined to within about 1% with the method.

Another method used quite often for determining the ^3He density is through measurements of the pressure broadening of the D_1 and D_2 absorption lines with a scannable single-frequency laser [50]. This measurement also provides the value of D , which is the ratio of K vapor density to Rb vapor density. Although the value of D is for the temperature at which the measurement is performed, its value for operating condition can be inferred with alkali-metal vapor pressure curves. D is also measured with the Faraday rotation technique in many cases, and the two methods agree with each other quite well. The fill densities and geometric properties of the aforementioned 24 cells are shown in Table 3.1.

3.2.1.2 Target Cell Polarization Dynamics

As previously mentioned, AFP is used to monitor the polarization of ^3He [6]. An exact value of polarization remains to be calibrated with EPR, but the signal size is directly proportional to the polarization, thus is an indication of how the polarization changes relatively. Two processes that are monitored with AFP are spinups and spindowns. The details of the dynamics has been discussed in detail by Dolph *et al.* [29].

The process of ^3He gaining polarization through spin-exchange with Rb that is being constantly pumped by circularly polarized laser is called a spinup. A typical example of a spinup is shown in Fig. 3.3.

EXP	Cell	Total Volume(cc)	PC Volume(cc)	ill F Density(amg)	TC length(cm)
saGDH	Proteus	235.9	90.8	6.88	34.3
	Peter	208.6	111.3	8.80	39.4
	Penelope	204.3	102.2	8.93	39.7
	Powell	213.3	111.6	8.95	40.5
	Prasch	257.7	114.5	6.94	35.3
GEN	Al	168.4	90.2	8.91	38.4
	Barbara	386.2	306.8	7.60	38.7
	Gloria	378.2	298.8	7.40	38.4
	Anna	386.8	303.7	8.09	38.7
	Dexter	181.4	99.3	9.95	38.7
	Edna	378.3	290.3	7.47	38.7
	Dolly	378.3	293.5	7.42	38.7
	Simone	219.5	118.6	8.17	37.9
	Sosa	388.8	304.7	7.96	38.7
Transversity and d_2^n	Boris	246.1	166.1	8.08	38.4
	Samantha	259.0	176.9	7.97	38.4
	Alex	278.3	193.9	7.73	39.1
	Moss	269.8	184.7	7.92	38.7
	Tigger	271.7	186.9	7.81	38.7
	Astral	251.4	164.9	8.18	38.4
	Stephanie	244.3	164.9	8.10	38.5
	Brady	249.9	169.3	7.88	38.4
	Maureen	268.5	177.4	7.63	39.8
	Antoinette	437.8	351.8	6.57	40.3

Table 3.1: The table contains the names, total and pumping chamber volumes, fill densities and target chamber lengths of the 24 target cells. The fill densities are the average of the results from gas system measurements and pressure broadening measurements.

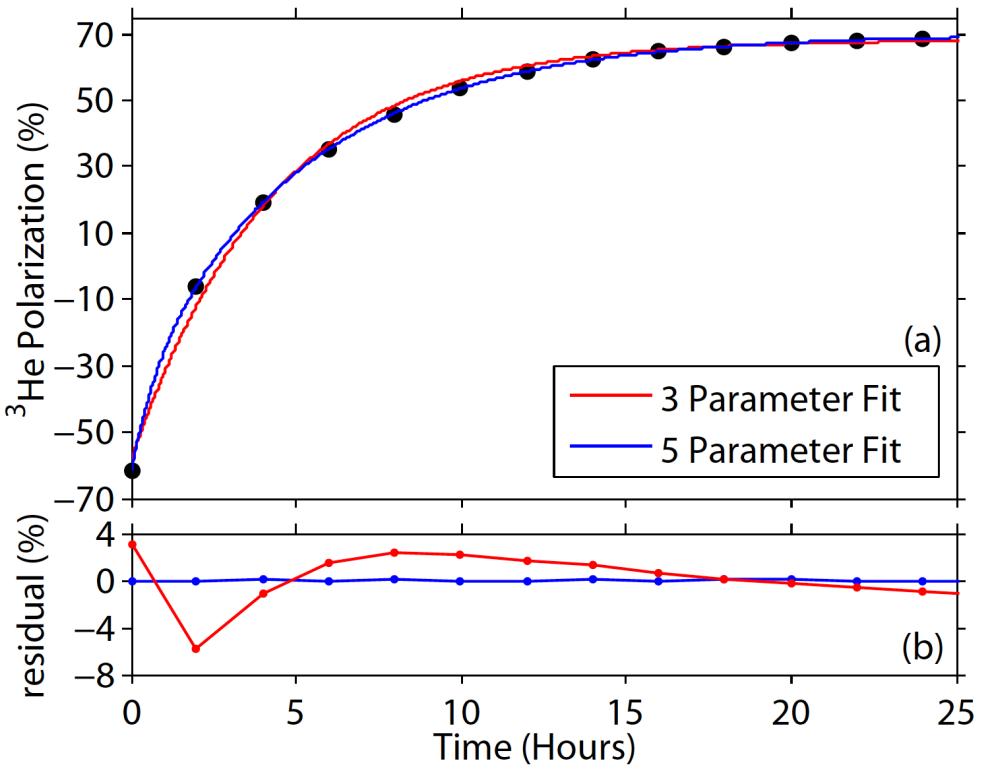


Figure 3.3: (a) Shown is a spinup of the target Brady. The spinup data has been fit with a 3-parameter and a 5-parameter formalism. (b) The residuals of the two fits. The error for 3-parameter fit is larger because it does not account for diffusion between two chambers.

The equation that describes spinups of single-chambered cell is:

$$P(t) = (P^0 - P^\infty)e^{-\Gamma_{sc}t} + P^\infty \quad (3.1)$$

where P^∞ is the saturation polarization, P^0 is the initial polarization, $\Gamma_{sc} = \gamma_{se}(1 + X) + \Gamma$ is the spin up rate of the buildup of polarization. The subscript “sc” here stands for “single chamber” to differ from the spinup rate of double-chambered cell.

γ_{se} is the spin-exchange rate, X is the X factor that limits the maximal achievable polarization, which will be discussed in more detail later in the chapter. Γ is the spin relaxation rate. When using this equation to fit spinup, there are only three parameters, hence the name 3-parameter fit. The saturation polarization is given by:

$$P^\infty = \frac{\langle P_A \rangle \gamma_{se}}{\Gamma_{sc}} = \frac{\langle P_A \rangle \gamma_{se}}{\gamma_{se}(1 + X) + \Gamma} \quad (3.2)$$

where $\langle P_A \rangle$ is the polarization of the alkali vapor averaged over the cell.

The following derivation will only focus on double-chambered cell. The polarization accumulation rate can be described by

$$\frac{dP_{pc}}{dt} = \Gamma_{se}(P_A - P_{pc}) - \Gamma_{pc}P_{pc} - d_{pc}(P_{pc} - P_{tc}) \quad (3.3a)$$

$$\frac{dP_{tc}}{dt} = -\Gamma_{tc}P_{tc} + d_{tc}(P_{pc} - P_{tc}) \quad (3.3b)$$

where $P_{pc}(P_{tc})$ is the ${}^3\text{He}$ polarization in PC (TC); γ_{se} is the spin-exchange rate in PC; $\Gamma_{pc}(\Gamma_{tc})$ is the relaxation rate o ${}^3\text{He}$ polarization in PC (TC); $d_{pc}(d_{tc})$ is the probability for a nucleus to leave PC (TC) and enter TC (PC). The transfer rates d_{pc} and d_{tc} are related by:

$$f_{pc}d_{pc} = f_{tc}d_{tc} \quad (3.4)$$

where $f_{pc}(f_{tc})$ is the fraction of atoms in PC (TC). The solutions to Eq.3.3 are

$$P_{pc}(t) = C_{pc}e^{-\Gamma_f t} + (P_{pc}^0 - P_{pc}^\infty - C_{pc})e^{-\Gamma_s t} + P_{pc}^\infty \quad (3.5a)$$

$$P_{tc}(t) = C_{tc}e^{-\Gamma_f t} + (P_{tc}^0 - P_{tc}^\infty - C_{tc})e^{-\Gamma_s t} + P_{tc}^\infty \quad (3.5b)$$

where $P_{pc}^0(P_{tc}^0)$ is the initial polarization in the pumping (target) chamber, $P_{pc}^\infty(P_{tc}^\infty)$ is the saturation polarization in the pumping (target) chamber. The "slow" time constant Γ_s is mostly determined by the volume averaged spin-exchange rate, which is given by

$$\Gamma_s = \langle \gamma_{se} \rangle (1 + X) + \langle \Gamma \rangle - \delta\Gamma \quad (3.6)$$

where $\langle \gamma_{se} \rangle = f_{pc}\gamma_{se}$ is the cell averaged spin-exchange rate, $\langle \Gamma \rangle$ is the cell averaged spin relaxation rate, as the rate might be different for pumping chamber and target chamber. The time independent quantities $\delta\Gamma$, Γ_f , C_{pc} and C_{tc} are functions of geometry, the various rates and initial conditions, which were discussed in Ref. [29]. The quantity $\delta\Gamma$ contains corrections due to the finite speed at which polarization moves between the two chambers. The size of $\delta\Gamma$ is usually no more than 10% of the size of Γ_s in our studies, and never more than 15%.

Again, the name 5-parameter fit comes from the fact that there are 5 parameters in each of the two equations. It's interesting to note the time evolution of ${}^3\text{He}$ polarization for double-chambered cells has a new time constant: the "fast" time constant γ_f that is dominated by the diffusion rates d_{pc} and d_{tc} when diffusion is relatively fast. In the fast-transfer limit, double-chambered solution reduces to single-chambered solution.

The other interesting point is the relation between the saturation polarization in PC and TC

$$P_{tc}^\infty = \frac{P_{pc}^\infty}{1 + \frac{\Gamma_{tc}}{d_{tc}}} \quad (3.7)$$

In the fast-transfer limit where $d_{tc} \gg \Gamma_{tc}$, $P_{tc}^\infty = P_{pc}^\infty$.

3.2.1.3 Initial Spinup

As shown in Fig.3.4, the early behavior of spinup with zero polarization for the pumping chamber and the target chamber are quite different. The initial part of the pumping chamber is almost linear but the target chamber shows a curved initial part. By performing a Taylor expansion on Eq. 3.5 we obtain the initial part of the spinup for both chambers [29]:

$$P_{pc}(t) = \gamma_{se} P_A t - \frac{1}{2} \gamma_{se} P_A (\gamma_{se} + \Gamma_{pc} + d_{pc}) t^2 \quad (3.8a)$$

$$P_{tc}(t) = \frac{1}{2} \gamma_{se} P_A d_{tc} t^2 \quad (3.8b)$$

where γ_{se} is the spin-exchange rate in the pumping chamber and P_A is the alkali polarization. The dominant term in $P_{pc}(t)$ is the linear term while the shape of $P_{tc}(t)$ is quadratic.

The slope of the linear shape of initial spinup of the pumping chamber gives access to the product $P_A \gamma_{se}$ and fitting the initial spinup of the target chamber to a quadratic function provides the product $\gamma_{se} P_A d_{tc}$. The alkali polarization P_A can be measured with a technique named Faraday rotation, we then gain knowledge of the spin-exchange rate γ_{se} and the diffusion rate d_{tc} . The slope of the polarization buildup in the pumping chamber is often written as $m_{pc} = P_A \gamma_{se}$.

The spin relaxation rate is also of great importance for characterizing target cells. The relaxation rates in the pumping chamber and the target chamber are different due to geometric and other properties. The cell-average relaxation rate can then be

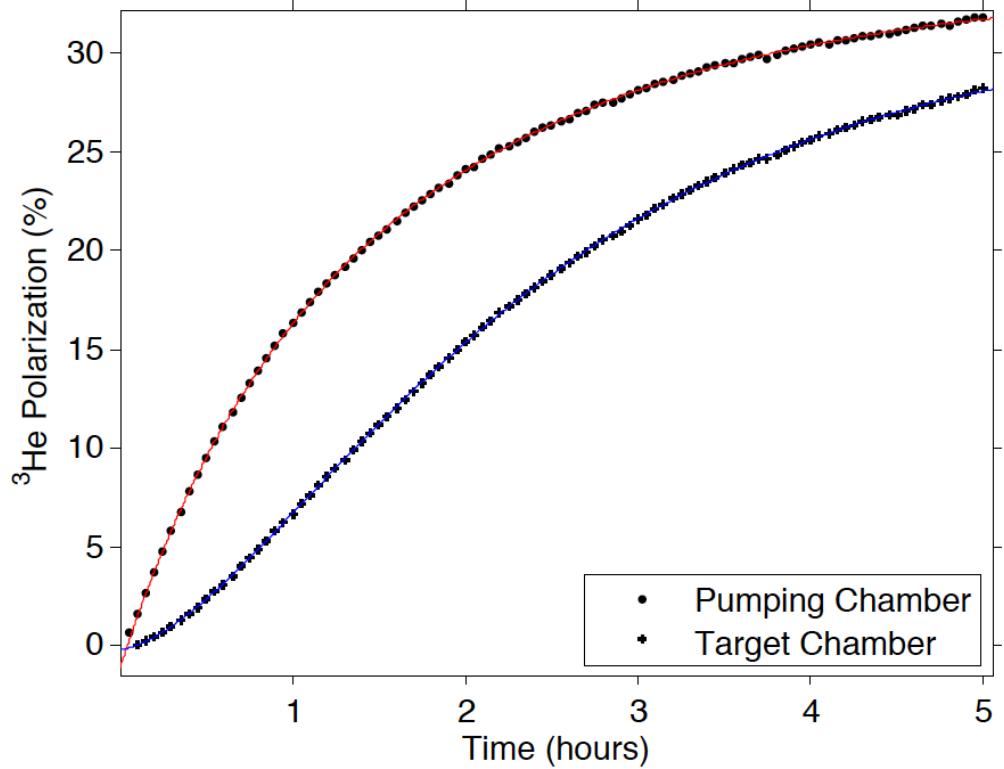


Figure 3.4: ${}^3\text{He}$ polarization as a function of time for both the pumping chamber and the target chamber. The top curve is the pumping chamber and the bottom curve is the target chamber. Data was taken at a fast pace so there would be enough points to demonstrate the initial behavior.

written as

$$\langle \Gamma \rangle = f_{pc} \Gamma_{pc} + f_{tc} \Gamma_{tc} \quad (3.9)$$

where f_{pc} (f_{tc}) is the fraction of atoms in PC (TC); Γ_{pc} (Γ_{tc}) is the average relaxation rate in PC (TC). When the cell is being pumped by laser, the pumping chamber is heated with hot air to create alkali vapor while the target chamber remains at

the room temperature. Difference in temperature further complicates difference in relaxation rates between the two chambers. However, when trying to measure the life time (the inverse of the relaxation rate) of the cell, we typically keep the entire cell at room temperature and perform a “spindown” measurement.

During a spindown, the cell starts with some polarization (normally as high as possible so we can obtain a more complete curve), and relaxes on its own while we take AFP measurements at a certain rate. Typically, the interval between measurements is anywhere between 30 mins and 2 hrs, depending on the lifetime of the cell. The rule of thumb is to take AFP frequently enough so the spindown curve has sufficient data points while not too often so the polarization relaxes too fast due to AFP losses. The ^3He polarization relaxation can be described by

$$P(t) = P_0 e^{-t/\tau_{true}} \quad (3.10)$$

The true lifetime τ_{true} of the cell without relaxation due to AFP loss can be measured with two methods: the first is to take 5 AFP measurements consecutively with very short interval (normally around 3 minutes), the second is to perform several spindown measurements, each with a different interval.

In the first method, because the lifetime of the cell is much longer than 3 minutes, we can safely attribute all losses to AFP measurements and extract the loss due to a single AFP $loss_{afp}$. The data values can then be corrected with the equation

$$S_i^{corrected} = S_i^{raw} / (1 - loss_{afp})^{i-1} \quad (3.11)$$

where $S_i^{corrected}$ is the corrected signal, S_i^{raw} is the raw signal, i represents it is the

*i*th measurement in the spindown, $loss_{afp}$ is the loss due to a single measurement. Fitting the corrected values to Eq. 3.10 gives the true lifetime τ_{true} .

A simple example for the second method would be to perform one spindown with one-hour interval and another spindown with two-hour interval, the relaxation rates in these two spindowns are

$$\frac{1}{\tau_{1hr}} = \frac{1}{\tau_{true}} + \Gamma_{AFP_1hr} \quad (3.12a)$$

$$\frac{1}{\tau_{2hr}} = \frac{1}{\tau_{true}} + \Gamma_{AFP_2hr} \quad (3.12b)$$

$$\Gamma_{AFP_1hr} = 2 \times \Gamma_{AFP_2hr} \quad (3.12c)$$

where τ_{1hr} and τ_{2hr} are the lifetimes measured with taking AFP every 1 hour and every 2 hours, τ_{true} is the true lifetime of the cell without interference from measurements, Γ_{AFP_1hr} (Γ_{AFP_2hr}) is the relaxation rate due to taking measurements every 1hr (2hr). We can then solve for τ_{true} .

3.3 The K- 3 He Spin-Exchange Rate Constant

As mentioned in the initial spinup section, the polarization in the pumping chamber at the beginning of accumulation process if started with completely unpolarized state can be described by

$$P_{pc} = \gamma_{se} \langle P_A \rangle (t - t_0) + b(t - t_0)^2 = m_{pc}t + bt^2 + c \quad (3.13)$$

where m_{pc} is the slope of the linear term. Typically, in the first 20 to 30 minutes, the spinup behaves so linearly that the effect of quadratic term is negligible.

During these initial spinups, an AFP measurement was taken every 3 minutes. The AFP losses were carefully accounted for when calculating m_{pc} . The ${}^3\text{He}$ spins were flipped to the opposite direction during every AFP measurement for a short period of time while still receiving polarization in the original direction. Care was taken to account for the time during which spins were “anti-aligned”. We refer to the slope collected from initial spinups as m_{pc}^s , to differentiate it from the same quantity measured with the Faraday rotation technique. I will briefly introduce Faraday rotation, the details were described thoroughly by Dolph. [5].

The Faraday rotation technique, as the name implies, is the observation of Faraday rotation using a linearly polarized probe laser. Faraday rotation refers to the change in the orientation of the polarization axis when linearly polarized light passes through a polarized alkali vapor. It is sufficient to consider only the alkali-metal atom’s D₁ and D₂ lines for our case, the Faraday rotation angle ϕ_r is:

$$\phi_r(\nu) = \left(\frac{r_e c}{6} \right) P_A \cos(\theta) [Rb] l \{ F_{Rb}(\nu) + D F_K(\nu) \} \quad (3.14)$$

where r_e is the classical electron radius, c is the speed of light in vacuum, l is the path length through the vapor, D is the ratio of the K to Rb vapor number densities, $F_A(\nu)$ is the frequency dependence of alkali species A.

During a Faraday rotation measurement, ϕ_r was measured at several probe wavelengths and fit to the Eq. 3.14. We were able to obtain the quantities $P_A[Rb]l$ and D from the fit. However, in order to extract [Rb], it is necessary to measure the path length l and P_A . Alkali polarization was measured by measuring the Faraday rotation angle while inducing Zeeman transition. It is worth noting this only gave

line-averaged polarization as only information on the path of the probe laser was collected. The volume-averaged alkali polarization can be obtained by applying small corrections from our simulation.

With knowledge of alkali densities, the spin-exchange rate is:

$$\gamma_{se} = k_{se}^{Rb}[Rb] + k_{se}^K[K] \quad (3.15)$$

where $k_{se}^{Rb}(k_{se}^K)$ is the spin-exchange rate constant between ${}^3\text{He}$ and Rb (K). The m_{pc} calculated with this manner is referred as m_{pc}^F , since this quantity was computed with Faraday rotation data.

The values of m_{pc}^F and m_{pc}^s are expected to be the same if they are measured against the same cell under identical conditions. The spin-exchange rate constants k_{se}^{Rb} and k_{se}^K are required to calculate m_{pc}^F . k_{se}^{Rb} has been measured and reported in literature multiple times. k_{se}^K on the other hand, is not as well-known. The value of k_{se}^{Rb} we used was the combined result from Baranga *et al.* [14] and from Chann *et al.* [24]:

$$k_{se}^{Rb} = (6.79 \pm 0.14) \times 10^{-20} \text{ cm}^3/\text{s} \quad (3.16)$$

The ratio of m_{pc}^F/m_{pc}^s is plotted in Fig. ???. The two measurements with the Rb only cell Sosa are shown with solid circles. To calculate the ratio for the rest of the measurements, a value for k_{se}^K is needed. Babcock reported this number to be $(5.5 \pm 0.4) \times 10^{-20} \text{ cm}^3/\text{s}$ in his thesis [3]. However, the resulting ratio was significantly lower than unity in all but one of the 6 measurements, as shown with open diamonds in the figure. We fit our data so the ratio m_{pc}^F/m_{pc}^s is equal to one while treating k_{se}^K

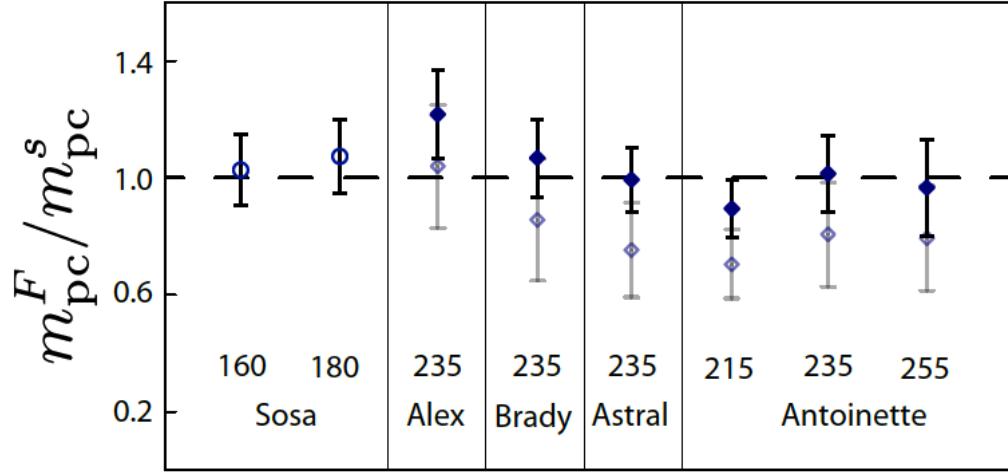


Figure 3.5: Plotted is the ratio m_{pc}^F / m_{pc}^s for eight separate measurements. The numbers above the cell names are the oven set temperatures at which the measurements were made.

as a free parameter. The results are shown with solid diamonds. Our fitted k_{se}^K value is

$$k_{se}^K = (7.46 \pm 0.62) \times 10^{-20} \text{ cm}^3/\text{s} \quad (3.17)$$

The reason why our result is significantly higher than Babcock's is unclear. One possibility may be temperature dependence as the temperatures under which our measurements were made were higher than that of Babcock's. We decided to use our own result of k_{se}^K to measure the X factor because it was measured under similar operating conditions to our other measurements, and our own result improves internal consistency of our data.

3.4 The X Factor

Prior to the introduction of the X factor, it was believed that ${}^3\text{He}$ polarization could approach unity with sufficiently high alkali vapor densities and laser power such that P_A is nearly 100% and $\gamma_{se} \gg \langle \Gamma \rangle$. However, the ${}^3\text{He}$ polarization during our studies had shown differently. The fact that it was never close to 100% even with high laser power and alkali densities could be explained by the X factor.

As mentioned earlier, Babcock *et al.* reported a previously unrecognized spin relaxation mechanism in his paper [10]. This mechanism appears to be roughly proportional to the spin-exchange rate γ_{se} , so it cannot be overcome by increasing the alkali density or laser power. The maximal achievable ${}^3\text{He}$ polarization can be expressed as

$$\lim_{\gamma_{se} \rightarrow \infty} = \lim_{\gamma_{se} \rightarrow \infty} \frac{\langle P_A \rangle \langle \gamma_{se} \rangle}{\langle \gamma_{se} \rangle (1 + X) + \langle \Gamma \rangle} = \frac{P_A}{1 + X} \quad (3.18)$$

The combination of alkali-hybrid and narrowband laser has made it much easier to achieve higher spin-exchange rates γ_{se} . Thus X factor is playing an increasingly more significant role in limiting the equilibrium ${}^3\text{He}$ polarization, which makes it an important subject of study.

Unlike many other properties of the cell that could be measured directly, X factor is a derived quantity. While characterizing our target cells, we collected enough data to determine X factor in several different ways. We were able to compare these values and combine them into weighted averages. We also looked at possible temperature dependence using X values obtained at different temperatures. The data used is presented in Table 3.2.

EXP	Cell	I_0 W/cm ²	T_{pc}^{set} °C	P_{He}^{∞}	Γ_s^{-1} hrs	$\langle \Gamma \rangle^{-1}$ hrs	$\langle P_A \rangle$	P_{line}^A	D_{fr}	D_{pb}	[Rb] _{fr} 10 ¹⁴ /cm ³	ΔT_{Rb} °C	ΔT_{He} °C	X	
saGDH	Proteus	3B	3.8	180	0.46	27	74	-	-	0	0	-	-	-	
	Priapus	3B	3.8	180	0.44	21	56	-	-	0	0	-	-	-	
	Penelope	3B	3.8	180	0.39	18	46	-	-	0	0	-	-	-	
	Powell	3B	3.8	180	0.38	13	25	-	-	0	0	-	-	-	
	Prasch	3B	3.8	180	0.33	13	33	-	-	0	0	-	-	-	
GEN	Al	2.5B	3.2	235	0.53(03)	7.86(05)	27.42(1.37)	-	-	-	4.53(25)	-	-	-	
		5B	6.1	235	0.54(03)	6.73(18)	27.42(1.37)	-	-	-	4.53(25)	-	-	-	
	Barbara	2.5B	1.6	235	0.37(02)	5.5(08)	42.95(2.15)	-	-	-	4.80(25)	-	-	-	
		5B	3.1	235	0.57(03)	4.76(63)	42.95(2.15)	-	-	-	4.80(25)	-	-	-	
	Gloria	3B	1.7	235	0.60(03)	6.13(04)	38.29(1.91)	-	-	-	7.20(40)	-	-	-	
	Anna	1B	0.6	235	0.33(02)	5.60(34)	11.38(57)	-	-	-	9.64(57)	-	-	-	
		1.5B	1.0	235	0.39(02)	5.37(08)	11.38(57)	-	-	-	9.64(57)	-	-	-	
	Dexter	1.5B	1.5	235	0.47(02)	7.58(17)	18.45(92)	-	-	-	-	-	-	-	
		5B	6.1	235	0.49(02)	6.63(12)	18.45(92)	-	-	-	-	-	-	-	
	Edna	3B	2.4	235	0.56(03)	5.71(02)	27.42(1.37)	-	-	-	3.63(20)	-	-	-	
Transversity	Dolly	3B	1.0	235	0.43(02)	6.16(03)	35.24(1.76)	-	-	-	20(1.3)	-	-	-	
		1N1B	1.4	235	0.62(03)	5.79(07)	35.24(1.76)	-	-	-	20(1.3)	-	17(10)	-	
	Simone	2N1B	3.8	215	0.31(01)	14.08(06)	22.87(1.14)	0.947(020)	0.91(05)	10.66(54)	8.89(45)	0.20(02)	-7(3)	-	-0.04(12)*
		2N1B	3.8	240	0.48(02)	6.89(20)	22.87(1.14)	-	-	-	9.76(49)	-	-	-	
		2N1B	3.8	255	0.58(02)	6.45(10)	22.98(1.14)	0.929(023)	0.92(05)	12.48(83)	10.3(52)	0.90(09)	-4(5)	-	0.11(06)*
d ₂ ⁿ	Sosa	2N1B	1.9	160	0.57(02)	16.69(09)	73.68(3.68)	0.966(020)	1.00(03)	0	0	1.97(13)	4(1)	30(7)	0.24(06) [†]
		2N1B	1.9	170	0.61(03)	11.67(04)	73.68(3.68)	0.964(020)	0.98(03)	0	0	3.00(33)	3(3)	38(14)	0.27(06)*
		2N1B	1.9	180	0.55(02)	8.79(09)	73.68(3.68)	0.954(022)	0.97(03)	0	0	4.30(27)	1(2)	47(7)	0.43(06) [†]
		2N1B	1.9	190	0.40(02)	6.39(22)	73.68(3.68)	0.854(075)	0.82(03)	0	0	5.69(63)	-2(3)	48(20)	0.58(12)*
		2N1B	1.9	200	0.26(01)	5.04(17)	73.68(3.68)	-	-	0	0	-	-	43(18)	-
	Boris	3B	1.8	235	0.42(02)	6.25(04)	23.74(1.19)	0.871(050)	0.79(07)	1.96(18)	2.45(23)	2.19(34)	-8(7)	-	0.26(10)*
	Samantha	3B	1.8	235	0.50(02)	6.30(13)	36.51(1.83)	-	-	-	4.34(23)	-	-	-	-
Transversity & d ₂ ⁿ	Alex	3N	2.6	235	0.68(03)	4.62(03)	22.13(1.11)	0.956(020)	0.99(03)	4.37(10)	4.34(23)	1.80(10)	7(2)	21(10)	0.12(05)*
	Moss	2N1B	2.6	235	0.59(03)	4.81(02)	32.96(1.65)	0.942(042)	0.99(03)	1.37(08)	1.19(07)	4.08(36)	0(4)	42(10)	0.34(06) [†]
	Tigger	1N1B	1.8	235	0.62(03)	5.35(04)	33.00(1.65)	-	0.95(09)	-	2.40(13)	-	-	29(8)	-
	Astral	1N1B	1.8	235	0.51(02)	4.89(05)	12.62(63)	-	0.95(09)	-	-	-	-	23(9)	-
	Stephanie	2N1B	2.6	235	0.69(03)	6.57(12)	48.90(2.45)	0.954(020)	0.99(03)	7.09(55)	6.21(56)	0.97(09)	3(5)	25(4)	0.17(05) [†]
		3N	2.6	235	0.63(03)	4.55(09)	48.35(2.42)	0.929(114)	0.99(03)	1.39(11)	1.50(10)	5.08(58)	7(5)	54(6)	0.31(08)*
	Brady	1N	0.9	235	0.62(03)	4.82(1.08)	33.50(1.68)	-	0.95(03)	-	2.36(24)	-	-	14(9)	-
		2N	1.8	235	0.68(03)	5.52(70)	33.50(1.68)	-	0.99(03)	-	2.36(24)	-	-	25(8)	-
		3N	2.6	235	0.70(03)	5.30(01)	33.50(1.68)	0.956(021)	0.99(03)	2.60(20)	2.36(24)	2.86(30)	6(5)	39(9)	0.14(05) [†]
	Maureen	3N	2.6	235	0.66(03)	5.42(12)	29.21(1.46)	-	0.97(09)	-	4.42(55)	-	-	32(12)	-
Antoinette	3N	1.7	215	0.49(02)	6.63(37)	20.93(1.05)	0.958(020)	0.99(03)	2.85(13)	-	0.96(07)	0(3)	16(8)	0.28(08) [†]	
		3N	1.7	235	0.61(03)	4.18(10)	20.93(1.05)	0.936(043)	0.99(03)	3.32(27)	-	1.83(20)	0(5)	20(10)	0.24(07) [†]
		3N	1.7	255	0.41(02)	2.66(11)	20.93(1.05)	0.776(099)	0.93(10)	3.57(23)	-	2.88(39)	-5(6)	33(9)	0.55(13) [†]

Table 3.2: Cell Performance for three sets of experiments: saGDH (top), GEN (middle), and Transversity & d_2^n (bottom). Within each experiment grouping, data is grouped by type of laser used (B = Broadband, N = Narrowband). I_0 is the nominal incident laser intensity at the center of the pumping chamber. T_{pc}^{set} is the oven set temperature. P_{pc}^{∞} is the equilibrium polarization in the pumping chamber and Γ_s is the slow time constant extracted from the five parameter fit to the polarization build up curve. Γ_c is the cell-averaged room temperature spin relaxation rate. $\langle P_A \rangle / P_A^l$ is the volume averaged to line averaged alkali polarizaiton ratio determined from the optical pumping simulation. P_A^l is the measured line averaged alkali polarization. D_{fr} & D_{pb} are the K to Rb density ratios determined from Faraday rotation and pressure broadening measurements. [Rb]_{fr} is the Rb number density measured from Faraday rotation. ΔT_{He} is the temperature of Rb inferred from the number density relative to the oven set temperature. ΔT_{He} is the temperature of ^3He inferred from temperature tests relative to the oven set temperature. X is the best combined value for the X-factor. * indicates X was measured using only spinup, alkali polarization, and Faraday rotation data. [†] indicates X was also measured using the early-time behavior of the spinup.

We calculated the X factors in several different ways, all of which required knowledge of the cell-averaged spin relaxation rate $\langle \Gamma \rangle$ at operating temperatures. However, the spindown measurements we performed only gave us the spin relaxation rate $\langle \Gamma \rangle_c$ at room temperature. We made the assumption that the difference between $\langle \Gamma \rangle$ and $\langle \Gamma \rangle_c$ is purely due to the change of cell-averaged ${}^3\text{He}$ - ${}^3\text{He}$ dipolar spin relaxation rate, and the relaxation rate due to collisions with the walls is the same for the two chambers and does not change at different temperature. The correction to the relaxation rate is given by

$$\langle \Gamma \rangle = \langle \Gamma \rangle_c - [n_0 - f_{pc}n_{pc}/f^d(t_{pc}) - f_{tc}n_{tc}/f^d(t_{tc})]/\tau^d \quad (3.19)$$

where n_0 is the ${}^3\text{He}$ fill density, $n_{pc(tc)}$ is the ${}^3\text{He}$ density in the pumping (target) chamber, $f_{pc(tc)}$ is the fraction of ${}^3\text{He}$ atoms in the pumping (target) chamber, $t_{pc} = T_{pc}/(296.15K)$, $t_{tc} = (313.15K)/(296.15K)$, $\tau^d = 744\text{hrs}\cdot\text{amg}$ [44], $f^d(t)$ is a function that parameterizes the temperature dependence of the dipolar relaxation [4]. $\langle \Gamma \rangle$ for is typically only a few percent less than $\langle \Gamma \rangle_c$ for us. In addition, whenever the quantity $(\Gamma_s - \langle \Gamma \rangle)$ is used, a small correction to account for the AFP losses is added.

Our methods of extracting X require using some form of the equation

$$\langle \gamma_{se} \rangle = \frac{\Gamma_s - \langle \Gamma \rangle + \delta\Gamma}{1 + X} \quad (3.20)$$

One method Babcock used and we applied on a small number of cells is called “hot relaxation method” [10]. We plot $\langle \gamma_{se} \rangle$ as a function of $\Gamma_s - \langle \Gamma \rangle + \delta\Gamma$, the slope of a linear fit to the data is expected to be $1/(1+X)$. Three such fits are shown in Fig. 3.6, all of which were constrained to go through the origin. Two of the three X

values were significantly different than zero. The X factor of Simone is too close to zero when taken the error into account for us to make a strong statement.

The small correction $\delta\Gamma$ came into play because of the double-chambered design. It can be approximated by

$$\delta\Gamma \approx f_{pc}f_{tc}(d_{pc} + d_{tc})u^2 + \text{higher order terms} \quad (3.21)$$

where $d_{pc(tc)}$ is the probability per unit time that a ${}^3\text{He}$ atom will exit the pumping (target) chamber. The quantity u is

$$u = \frac{\gamma_{se}(1 + X) + \Gamma_{pc} - \Gamma_{tc}}{d_{pc} + d_{tc}} \quad (3.22)$$

where $\Gamma_{pc(tc)}$ is the spin relaxation rate in the pumping (target) chamber. What makes the determination of X tricky is that we need to know its value before we can calculate u which is a prerequisite for X. However, because $\delta\Gamma$ is such a small correction typically being 10% of less of the size of Γ_s , we solve this problem in an iterative manner. X is initially assumed to be 0 and plugged into Eq. 3.22, which in the end lead to a different value of X that is closer to its real value. After iterating this process a few times, X quickly converges to a stable value.

However, the hot relaxation method assumes the temperature dependence of the X factor is identical to the temperature dependence of γ_{se} by combining data taken at different temperature into one value of X. It is also time consuming to perform measurements shown in Fig. 3.6. For these two reasons, the hot relaxation method was only applied to a small number of cells.

In order to measure X factor at a single temperature and explore its temperature

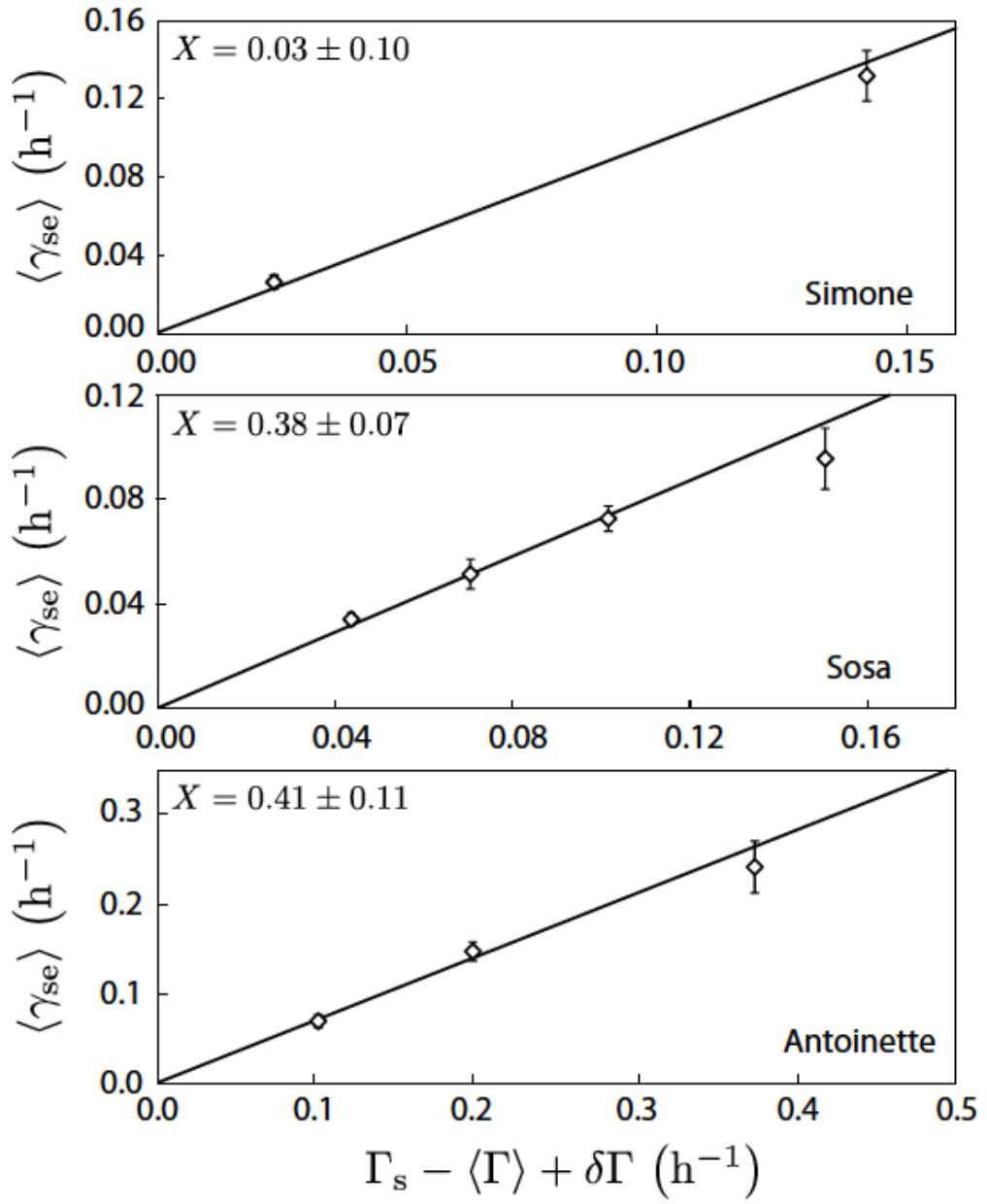


Figure 3.6: The cell-averaged spin-exchange rate $\langle \gamma_{se} \rangle$ is calculated using data from Faraday rotation and the spin-exchange constants k_{se}^{Rb} and k_{se}^K . The three linear fits shown here are constrained to go through zero. The errors quoted in values of X factor include the uncertainty in our determination of k_{se}^K .

dependence, we used 4 four other methods to determine the value of X factor. All but the second method described below are based on the "polarization method" from Ref. [10]. The difference is that we used early time measurements of the initial spinups for $P_A \langle \gamma_{se} \rangle$ for the third and fourth methods. And the second method is basically a single point measurement of the hot relaxation method described above.

We label the results from the four single temperature methods as X_1 , X_2 , X_3 and X_4 respectively. The most straightforward method requires measurements of $\langle P_A \rangle$, P_{pc}^∞ and Γ_s . The equilibrium ^3He polarization can be rewritten as

$$P_{pc}^\infty = \frac{\langle P_A \rangle \langle \gamma_{se} \rangle}{\Gamma_s + \delta\Gamma - \delta\Gamma'} \quad (3.23)$$

where $\delta\Gamma' = f_{tc}\Gamma_{tc}^2/(\Gamma_{tc} + d_{tc})$. Substitute Eq. 3.20 into Eq. 3.23, X_1 is

$$X_1 = \frac{\langle P_A \rangle}{P_{pc}^\infty} \left(\frac{\Gamma_s - \langle \Gamma \rangle + \delta\Gamma}{\Gamma_s + \delta\Gamma - \delta\Gamma'} \right) - 1 \quad (3.24)$$

Just as what we did for the hot relaxation method, here $\delta\Gamma$ is calculated in the same iterative manner. X is first taken as 0 and the iteration continued until X converged to a stable value.

For the second method, we first solve Eq. 3.20 for X:

$$X = \frac{\Gamma_s - \langle \Gamma \rangle + \delta\Gamma}{\langle \gamma_{se} \rangle} - 1 \quad (3.25)$$

then we substitute Eq. 3.15 into the equation above:

$$X_2 = \frac{\Gamma_s - \langle \Gamma \rangle + \delta\Gamma}{f_{pc}k_{se}^{Rb}[Rb](1 + D')} - 1 \quad (3.26)$$

Again $\delta\Gamma$ is calculated iteratively. We chose to use our value of k_{se}^K for better consistency with the rest of our data.

The third method is very similar to the second, but we determine $\langle\gamma_{se}\rangle$ with data from initial spinups:

$$\langle\gamma_{se}\rangle = f_{pc}m_{pc}^s/\langle P_A \rangle \quad (3.27)$$

Substitute the above equation into Eq. 3.25, we get

$$X_3 = \langle P_A \rangle \frac{\Gamma_s - \langle \Gamma \rangle + \delta\Gamma}{f_{pc}m_{pc}^s} - 1 \quad (3.28)$$

Note the quantity k_{se}^K used for X_2 was obtained by fitting the ratio m_{pc}^F/m_{pc}^s to 1, thus for any hybrid cell, X_2 and X_3 are highly correlated. However, for pure Rb cell such as Sosa, these two methods are independent.

The fourth method is similar to the first one. It starts from Eq. 3.23, what makes it different from the first method is that it treats $\langle\gamma_{se}\rangle$ as a known quantity and expresses Γ_s with it using Eq. 3.20, while the first method did it in the exact opposite way. Then the cell-averaged spin-exchange rate is evaluated with

$$\langle\gamma_{se}\rangle = f_{pc}m_{pc}^s/\langle P_A \rangle \quad (3.29)$$

Thus X_4 is

$$X_4 = \frac{P_A}{P_{pc}^\infty} - \frac{\langle P_A \rangle (\langle \Gamma \rangle - \delta\Gamma')}{f_{pc}m_{pc}^s} - 1 \quad (3.30)$$

The computed X factors are shown in Table 3.3. The different values of X are quite consistent with each other. It is worth noting that even though X_1 is completely

Cell	T(°C)	X_1	X_2	X_3	X_4	X_{12}/X_{1234}
Simone	215	-0.02(12)	-0.10(14)	-	-	-0.04(12)
	255	0.13(08)	0.08(09)	-	-	0.11(06)
Sosa	160	0.22(07)	0.28(09)	0.32(15)	0.18(09)	0.24(06) [†]
	170	0.24(07)	0.37(15)	-	-	0.27(06)
	180	0.45(08)	0.40(09)	0.50(17)	0.45(09)	0.43(06) [†]
	190	0.59(16)	0.57(17)	-	-	0.58(12)
Boris	235	0.21(14)	0.31(14)	-	-	0.26(10)
Sam.	235	0.08(06)	0.22(09)	-	-	0.12(05)
Alex	235	0.34(09)	0.35(09)	0.63(20)	0.29(10)	0.34(06) [†]
Astral	235	0.15(07)	0.22(10)	0.20(14)	0.14(07)	0.17(05) [†]
Steph.	235	0.31(17)	0.31(10)	-	-	0.31(08)
Brady	235	0.13(07)	0.15(09)	0.23(14)	0.11(07)	0.14(05) [†]
Antoinette	215	0.27(09)	0.44(17)	0.30(19)	0.25(11)	0.28(08) [†]
	235	0.20(09)	0.34(12)	0.36(17)	0.15(09)	0.24(07) [†]
	255	0.55(26)	0.54(16)	0.50(30)	0.56(26)	0.55(13) [†]

Table 3.3: Shown are the values of the X factor at the indicated over set temperatures.

The last column is a weighted average of results from either the first two methods or all four methods. A [†] indicates combined values computed with all 4 methods.

independent of m_{pc} and k_{se}^K , it is still quite consistent with the rest of the X values. The last column in the table is a weighted average of either X_1 and X_2 (X_{12}) or all four X values (X_{1234}).

To the best of our knowledge, there has not been a dedicated study of the X factors and their temperature dependence, with a large number of cells using measurements of the alkali polarization since the Babcock's *et al.* original work [10]. Our results are an independent evidence of the existence of the X factor. According to our study, the X factor limits ${}^3\text{He}$ polarization to 62-88% for the range of temperature we operate within.

Since we have evaluated X at multiple temperatures for Simone, Sosa and Antoinette, we can explore the temperature dependence of X factor. The combined value of X is plotted against temperature in Fig. ?? for the three cells. The figure seems to indicate a systematic variation with temperature.

If we assume a linear dependence of X with temperature, the fitted slope for Sosa with 4 points is $(0.012 \pm 0.002)/{}^\circ\text{C}$, which is six-sigma away from zero. The slope for the three points of Antoinette is $(0.007 \pm 0.005)/{}^\circ\text{C}$, which is slightly above one sigma from zero. Simone has only two points available to us, the second point seems to be right around the edge of the error bar on the first point. Although we cannot make a strong statement, but it is still likely that there is also a trend here. We do not consider this result conclusive, but they still suggest the existence of temperature dependence of X.

We considered the possibility that the temperature dependence was introduced by our own value of the spin-exchange constant k_{se}^K . After considering both our value and that from Babcock, we found that the trend exists with both, but our value provides

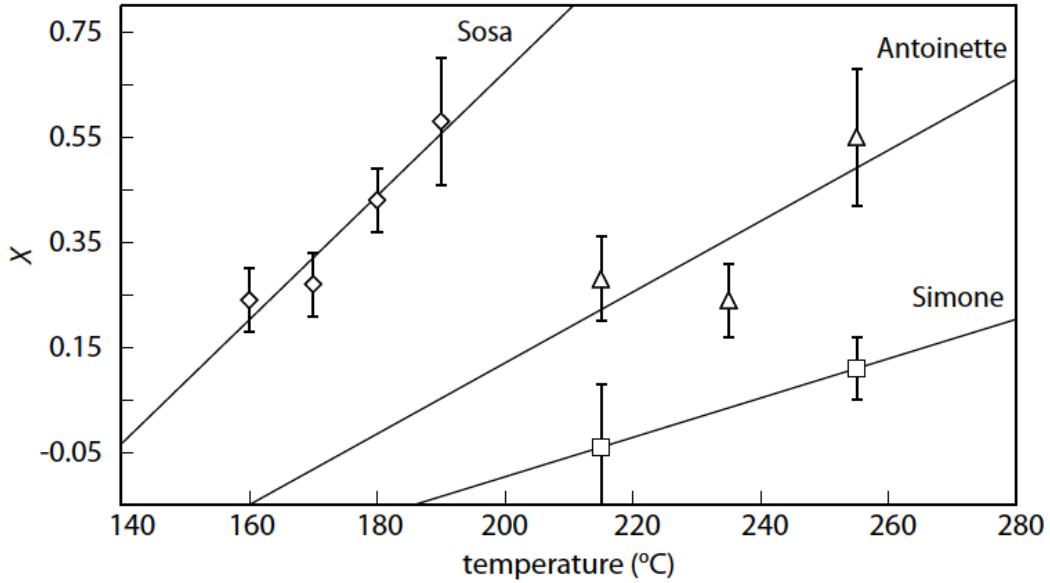


Figure 3.7: Shown are the combined values for X factor (either X_{12} or X_{1234} depending on the availability of data) versus temperature for the cell Sosa, Simone and Antoinette.

better self consistent. Another possibility is that the temperature dependence of k_{se}^K is the source of temperature dependence of X. However, if that is the case, we should be able to observe different behaviors in what the 4 different methods produced. From what is shown by Table 3.3, the different methods produced similar results within errors, thus k_{se}^K is not likely to be the cause of temperature dependence of X.

Another possibility was the temperature dependent contribution from anisotropic spin exchange [62]. However, even though Calculations from Tscherbul *et al.* [58] indicate the anisotropic spin exchange contribution to the X factor has a temperature dependence, the contribution is very small in our operating temperature range.

We also considered whether other systematic effects could result in such temperature dependence. For example, the distribution of the gas between pumping chamber

and target chamber changes at different temperatures. We then concluded that the uncertainties in the gas distribution had only negligible effects on the observed temperature dependence of X. After analyzing our data, we were unable to identify a systematic effect as a plausible explanation, our results suggest further study of X and its temperature dependence may be of great importance.

Chapter 4

Development of Cells with Metal End Windows

4.1 Overview

Electron scattering experiments at JLab have traditionally used ${}^3\text{He}$ targets made of glass due to the compatibility with spin-exchange optical pumping, the capability to be shaped into desired geometries through glass blowing, and the excellent nuclear spin relaxation properties. The borosilicate glass Pyrex had been the glass of choice prior to the discovery that the much less permeable aluminosilicate glass generally provides longer spin-relaxation time. The most recent targets were thus composed entirely of the aluminosilicate glass GE180.

After the 12 GeV upgrade, JLab plans to run experiments with much higher electron beam currents. The maximum current used before the upgrade was $15\mu\text{A}$, while future experiments will be run at up to $60\mu\text{A}$. We believe an all-glass target

cell might survive long enough for an experiment with $30\mu\text{A}$, but it is unlikely to survive at $60\mu\text{A}$. A natural solution would be to replace the thin glass window (where electron beam enters and exits the target cell) with a material with higher strength and good spin-relaxation properties.

Deninger *et al.* [53] from the Mainz group showed relaxation time of various metal surfaces: Mg (6 h), Al (6 h), Zn(12 h) etc. Gold caught our attention in particular, because it has a relatively long relaxation time of 20 h. This relaxation time was measured with the entire glass surface coated with gold, thus the area of gold surface was much larger than what will be needed for target windows. In addition, while the coating process made sure of the chemical purity, it did not make effort in ensuring the microscopic smoothness, which means the surface area was further increased. Therefore one would expect a even longer relaxation time with a smooth and smaller metallic surface. In light of this, our group have tested 19 cells with various geometries and materials, most of which incorporated a OFHC (oxygen-free high thermal conductivity) copper tube with gold coating. OFHC copper was chosen as the substrate in most cases because of its structural strengths and the familiarity with manufacturers. Towards the end of my work, we achieved a 15.6 h relaxation time with a Pyrex cell that had a 5" long by 1" gold coated copper tube attached horizontally. By extrapolating the relaxation rate due to gold surface from this result, we believe the relaxation rate introduced by small metal windows in a target cell will be less than $1/135 \text{ hr}^{-1}$. To the best of our knowledge, our group was the first to have proved the potential of incorporating metal to target cells in the presence of alkali vapor.

4.2 Wall Relaxation of ${}^3\text{He}$

4.2.1 Relaxation on Glass Surfaces

Fitzsimmons and Walters have studied surface-induced spin-lattice relaxation times as a function of temperature for ${}^3\text{He}$ gas in glass containers [30]. There are mainly two categories of wall relaxation mechanisms: ${}^3\text{He}$ adsorption on the glass surface and the permeation of ${}^3\text{He}$ into glass. The latter mechanism can be greatly reduced by using impermeable aluminosilicate glass such as GE180.

Timsit and Daniels [1] then studied surface relaxation on a great number of common materials and presented a phenomenological model to describe the relaxation processes. For permeable glasses, the relaxation is determined by absorption of gas in the surface layer of the glass and by the paramagnetic impurity contents of the glass. The surface adsorption of ${}^3\text{He}$ near paramagnetic sites on the walls also contributes to the nuclear relaxation. Relaxation due to absorption for permeable glasses will be discussed first below.

The diffusion coefficient D of a noble gas in a glass can be calculated with the following equation:

$$D = D_0 e^{-Q_d/kT} \quad (4.1)$$

where Q_d is the activation energy for diffusion and D_0 is a constant. The diffusion coefficient can also be expressed with the mean diffusion jump distance of ${}^3\text{He}$ atom in the glass $\langle \Delta r \rangle$ as:

$$D = \frac{\langle \Delta r \rangle^2}{6\tau} \quad (4.2)$$

with τ being the mean time between diffusion jumps

$$\tau = \tau_0 e^{E_{dif}/kT} \quad (4.3)$$

where $\tau_0 = \langle \Delta r \rangle^2 / 6D_0$.

Let n_g be the number of atoms dissolved in the surface layer of mean thickness $\langle \Delta r \rangle$, the rate at which ${}^3\text{He}$ atoms enter and leave the surface layer of the glass is then $n_g/6\tau$ [1]. n_g should be proportional to the solubility S of ${}^3\text{He}$ in the glass, so for a spherical cell

$$n_g = \frac{6NkT\langle \Delta r \rangle S}{d} \quad (4.4)$$

where d is the diameter of the cell, N is the total number of free ${}^3\text{He}$ atoms.

For most trapping sites in the glass, the intrinsic relaxation time T_i is longer than τ which is the time it takes for ${}^3\text{He}$ to leave the $\langle \Delta r \rangle$ layer. However, T_i for a paramagnetic site is shorter than τ , thus will completely relax the nuclear spin of a ${}^3\text{He}$ atom. The relaxation time of ${}^3\text{He}$ in permeable glass cells is controlled by absorption of the atoms in the surface layer at paramagnetic sites. The average nuclear relaxation time of a ${}^3\text{He}$ trapped in the glass close to a Fe^{3+} ion (one common type of paramagnetic impurity in glass) is [6]:

$$\frac{1}{T_i} \approx \frac{3}{5} \frac{\mu_{He}^2 \mu_B^2 g^2}{\hbar^2 b^6} \frac{T_{Fe}}{1 + \omega_0^2 T_{Fe}^2} \quad (4.5)$$

where μ_{He} is the nuclear dipole moment of ${}^3\text{He}$, μ_B is the Bohr magneton, g is the g factor of the Fe^{3+} (${}^6S_{5/2}$) ion, and b is the distance between the spins. Taking b as 1 \AA and g as 5.9 [?], T_i is $\sim 10^{-11}$ s, which is 10 times smaller than the shortest τ .

Even a small amount of paramagnetic impurities among the trapping sites in the glass can provide dominating contribution on the ${}^3\text{He}$ spin relaxation. Assuming

during the random walk of ${}^3\text{He}$ atom in the glass, there are on average β atoms in its vicinity, and atom fraction of paramagnetic impurities is $N_{impurity}$, the relaxation time due to absorption is if $T_i \ll \tau$:

$$T_{ab} = \frac{6N\tau}{\beta N_{impurity} n_g} \quad (4.6)$$

For impermeable glasses such as GE180, the relaxation rate due to absorption into the glass walls is typically negligible. The dominating relaxation mechanism here is adsorption of ${}^3\text{He}$ on the glass wall in vicinity of a paramagnetic site.

The sticking time τ_s is given by Frenkel's Law [31]:

$$\tau_s = \tau_{s0} e^{E_{ad}/kT} \quad (4.7)$$

where τ_{s0} is on the order of 10^{-13} s for most solid surfaces [31], E_{ad} is the adsorption energy. At room temperature, we have $\tau_s \sim \tau_{s0} \sim 10^{-13}$ s. The number of atoms hitting the wall per unit time and unit area is given by $\frac{1}{4}n\bar{v}$, where n is the number density of ${}^3\text{He}$ gas and \bar{v} is the mean velocity. For a spherical cell with diameter d , the number of atoms adsorbed on the wall is

$$n'_g = \frac{3N\bar{v}\tau_s}{2d} \quad (4.8)$$

The intrinsic relaxation time T'_i of a ${}^3\text{He}$ near a paramagnetic site on the wall is much longer than the sticking time τ_s . The average number of collisions required to depolarize ${}^3\text{He}$ is T'_i/τ_s , thus the relaxation time due to adsorption is

$$T_{ad} = \frac{NT'_i}{N_{impurity}n'_g} \quad (4.9)$$

The total relaxation rate is the sum of that due to adsorption and absorption (for permeable glasses):

$$\frac{1}{T_{wall}} = \frac{1}{T_{ad}} + \frac{1}{T_{ab}} \quad (4.10)$$

Substitute Eq. 4.3, 4.4 into Eq. 4.6 and Eq. 4.7, 4.8 into Eq. 4.9, the wall relaxation rate can be written as

$$\frac{1}{T_{wall}} = \frac{\beta N_{impurity} k T \langle \Delta r \rangle S}{d\tau_0} e^{-E_{dif}/kT} + \frac{3N_{impurity} \bar{v} \tau_{s0}}{2dT'_i} e^{E_{ad}/kT} \quad (4.11)$$

According to the above equation, both the diffusion-induced and the surface-induced relaxation rates are proportional to the surface-volume ratio of the cell, *i.e.* to the inverse of the cell diameter d . Thus it is useful to describe the surface relaxation properties with a physical quantity ρ , which is commonly referred to as the "relaxivity". The relaxivity is independent of cell geometry and is related to the wall relaxation rate $1/T_{wall}$, the surface to volume ratio A/V by the following equation:

$$1/T_{wall} = \rho A/V \quad (4.12)$$

Fitzsimmons *et al.* [30] found by using impermeable aluminosilicate glass, the relaxation due to absorption can be greatly reduced thus increasing the total relaxation time. Heil *et al.* reported [35] glass cells that were internally coated with metallic films provided even longer relaxation time. Ce was one of the metals that greatly reduced wall relaxation rate as it blocks ^3He atoms from diffusing into the glass walls and it also has a low adsorption energy which leads to very short sticking time. For SEOP (Spin-Exchange Optical Pumping), we automatically profit from the Rb film which covers the inner surface of the pumping chamber. Similarly, another way to eliminate relaxation due to absorption is to coat the inner surface with sol-gel [37]. It is a mixture of aluminum nitrate nonahydrate $Al(NO_3)_3 \cdot 9H_2O$, ethanol and deionized water. The sol-gel coating also improves relaxation time by blocking ^3He atoms from diffusing into the glass.

4.2.2 Relaxation on Metal Surfaces

Since ${}^3\text{He}$ gas does not diffuse into metals [13], the relaxation is purely due to surface effects. However, the surface-induced relaxation on metal surfaces is not yet well understood. There are two additional relaxation mechanisms with metal surfaces compared to those for glass. First mechanism is the depolarization of ${}^3\text{He}$ nuclei near the metal surface by oscillating magnetic field from eddy currents induced by the movement of these same nuclear dipoles. Smythe [56] derived the magnetic field generated by eddy current induced in a metal sheet by a moving magnetic dipole using the method of images. Timsit *et al.* [1] estimated the relaxation due to eddy currents using the same method:

$$\frac{1}{T_m} = \frac{3 \times 10^{-4}}{\pi} \frac{\mu_{He}^4}{\bar{v} r_0^4 \hbar^2} \frac{A_p}{d^3} \quad (4.13)$$

where \bar{v} is the mean velocity of atoms, r_0 is the closest distance of the nucleus to the metal surface, d is the cell diameter, A_p is the area. According to Timsit *et al.* [1], T_m is on the order of 10^{16} s if $r_0 = 1\text{\AA}$ and $A_p = 9.2\text{cm}^2$. Even though the area of the cell is usually much larger in our study and other parameters may take different values, but it should still remain true that the relaxation rate of ${}^3\text{He}$ due to eddy currents can be safely neglected.

The second additional mechanism that contributes for relaxation of ${}^3\text{He}$ adsorbed on the metal surfaces is the Korringa scattering. Nuclear spin relaxes when the nucleus interacts with conduction electrons in the metal, where an electron flips its spin and the spin state of the nucleus changes. The Pauli exclusion principle states that the interaction is only allowed when the final state the electron jumps to is initially unoccupied. Thus according to Fermi statistics, only electrons with energy

close (approximately within kT) to the Fermi energy level can participate in such interactions. Slichter [55] has derived the Korringa relaxation rate in metal to be:

$$\frac{1}{T_K} = \frac{(4\pi)^6}{9h^7} (g_s \mu_B \frac{\mu_K}{K})^2 \eta^4 m^3 \epsilon_F k T \quad (4.14)$$

To calculate the total Korringa relaxation rate due to metal surface, one needs to consider the overlapping of wave functions of the conduction electrons and the nuclei of adsorbed ${}^3\text{He}$ atoms in the vicinity of the metal surface. Nelson [2] derived the total surface Korringa relaxation as:

$$\frac{1}{T} = \frac{1}{T_K} \frac{S}{V} \int_0^\infty f(l)^2 e^{-U(l)/kT} dl \quad (4.15)$$

where S and V are the surface area of the metal and total volume of the cell, respectively, $U(l)$ is the atom-surface potential with the edge of the metal taken as $l = 0$, $f(l)$ is the fractional density of conduction electrons outside the surface. Nelson further used the work function of the metal to estimate $f(l)$, and assumed a van der Waals form for $U(l)$. His calculations suggest that the Korringa relaxation times for ${}^3\text{He}$ on various metal surfaces should be thousands of years, including the metal of interest for our studies, gold.

If eddy currents and Korringa relaxation are the only surface relaxation mechanisms, we would have very long lifetime for our test metal cells. Unfortunately, our series of studies have shown relaxation times to be between a few hours and 16 hours most of the time, for which both aforementioned relaxation mechanisms can be safely neglected. Although the dominating relaxation mechanism is still not well understood and significantly faster than those we have better understanding about, the results of our studies are still promising, as they suggest the extra relaxation rate due to use of

metal end windows should still be small enough and allow future experiments to run at $60\mu\text{A}$ electron beam current.

4.3 Test Cell Fabrication

4.3.1 Overview

In order to study the relaxation rate due to metal surface, we have created and tested 19 cells so far, most of which contained metal tubes. The metal tubes are 5" in length and have 1" outer diameter, 3" of glass tubes are attached to both ends via Houskeeper seal. The total area of the metal surface is 101.3 cm^2 , and is far larger than what will be for metal end windows. This large geometry is chosen to intentionally increase the relaxation rate for studying. This particular design of metal tube is also favored for the convenience of manufacturing, as we were expecting a large number of test cells to be studied and any design that could save time would potentially lead to a much shorter test time before finding the final design.

The making of these test cells usually includes five steps (except control cells with pure glass and cells without gold coating):

1. Larson Electronic Glass [33] provides us a glass–metal–glass tube that uses Houskeeper seal to bond glass and metal together.
2. The machine shop in the physics department of University of Virginia then mechanically polishes the inside of the metal tube.
3. Able[57] Electropolishing electropolishes the metal tube.
4. Epner[38] Technology Inc. electroplates the metal tube with gold.

5. Mike Souza at Princeton University, our glass blower, resizes the glass from stock material and makes the final cell with our desired dimensions and attaches the metal tube to the glass cell. The resizing is for removing micro-fissures from the glass surface and reducing wall relaxation rates [20].

Each step will be described below.

4.3.2 Glass-Metal Seal

Larson Electronic Glass made the glass-metal-glass tubes for us, Fig. 4.1 is a picture of one of the tubes used in our tests. The glass-metal seal used by Larson is known as the Houskeeper seal [36]. The outside of the copper tube is machined down to a knife edge and is wetted with glass. The edge of copper is usually heated before covering with glass. The heating process creates a thin layer of crimson-color copper oxide as can be seen in Fig. 4.1. Houskeeper seals were originally used for vacuum, we checked each seal down to the $10^{-10} mbar \cdot l/s$ level. To make sure these seals will survive the high pressure JLab experiments, we also tested the integrity of them under up to 20 atm for extended period of time.

The copper used in our tests is OFHC (oxygen-free high thermal conductivity) copper. In the earlier stage of our tests, OFHC copper was attached to the Pyrex glass, in which case a direct connection could be made. For the later tests where we were moving closer to the final goal of using metal end windows with the impermeable aluminosilicate glass GE180, a transition glass between OFHC copper and GE180 had to be used. The coefficient of expansion for GE180 glass is not compatible for making a direct seal with OFHC copper, thus Corning 7052 Kovar sealing glass was used as the



Figure 4.1: Shown is a glass–metal–glass seal. The metal tube is 5" long by 1" outer diameter. The glass is wetted onto the knife-edge of copper on both ends.



Figure 4.2: Glass-metal seals survived pressure higher than 20 atm.

transition. The two materials connected by a seal should have similar coefficients of expansion, the 7052 Kovar sealing glass serves as an intermediate material to bridge the gap between OFHC copper and GE180. The other type of metal used in our glass–metal seals was titanium, for which only Pyrex was used.

4.3.3 Mechanical Polishing and Electropolishing

The mechanical polishing is done by our machine shop in the department. A wire brush attached to a lathe was placed inside the tube while the lathe spun. This first-step polishing produced a relatively smooth surface in preparation for the following electropolishing process.

After the tubes were mechanically polished by the machine shop, they were sent to Able for electropolishing. The tube serves as the cathode, whihc is immersed in a temperature-controlled bath of electrolyte and connected to the positive terminal of a DC power supply, while the cathode is the attached to the negative terminal. During electropolishing, the polarized film is subjected to combined effects of gassing (oxygen) that occurs with electrochemical metal removal, saturation of the surface with dissolved metal and the agitation and temperature of the electrolyte. Metal on the surface is oxidized and dissolved in the electrolyte, the microscopic high points on the surface dissolve faster than the rate of attack on the other parts of the surface, which provides a smoothing effect. As a result, the electropolishing process removes a thin layer of metal (about $20 \mu m$ for our tubes), leaving a microscopically smooth and featureless surface. By contrast, even a fine mechanically polished surface will still show smears and other directionally oriented patterns or effects[27]. Fig. 4.3 shows a

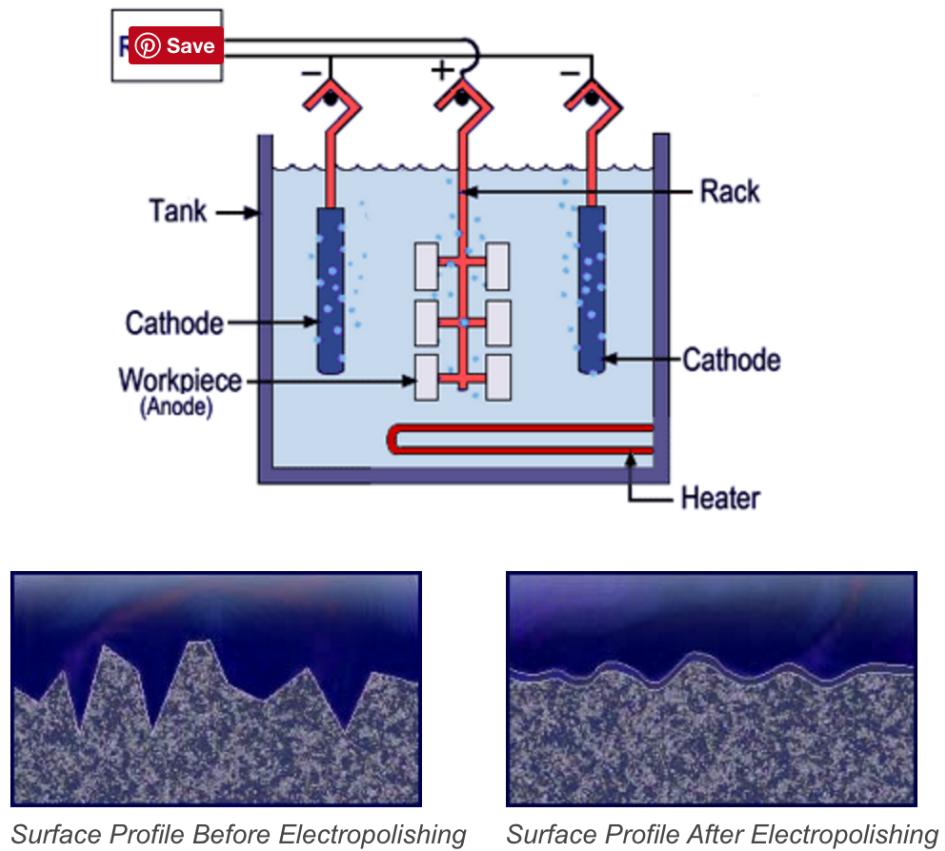


Figure 4.3: Electropolishing [27]

diagram of the polishing process and its smoothing effect.

4.3.4 Electroplating

As stated earlier, because of the good lifetime reported by Deninger *et al.*[28], gold was plated on the inner surface of the OFHC copper and titanium tubes. Epner Technology Inc. handled the electroplating for us. Electroplating is the reverse process of electropolishing. When electric current passes through the electrolyte, the electrolyte splits up and some of the desired metal atoms it contains are deposited in a thin layer on top of the electrodes. Nickel and chromium are two common undercoatings used

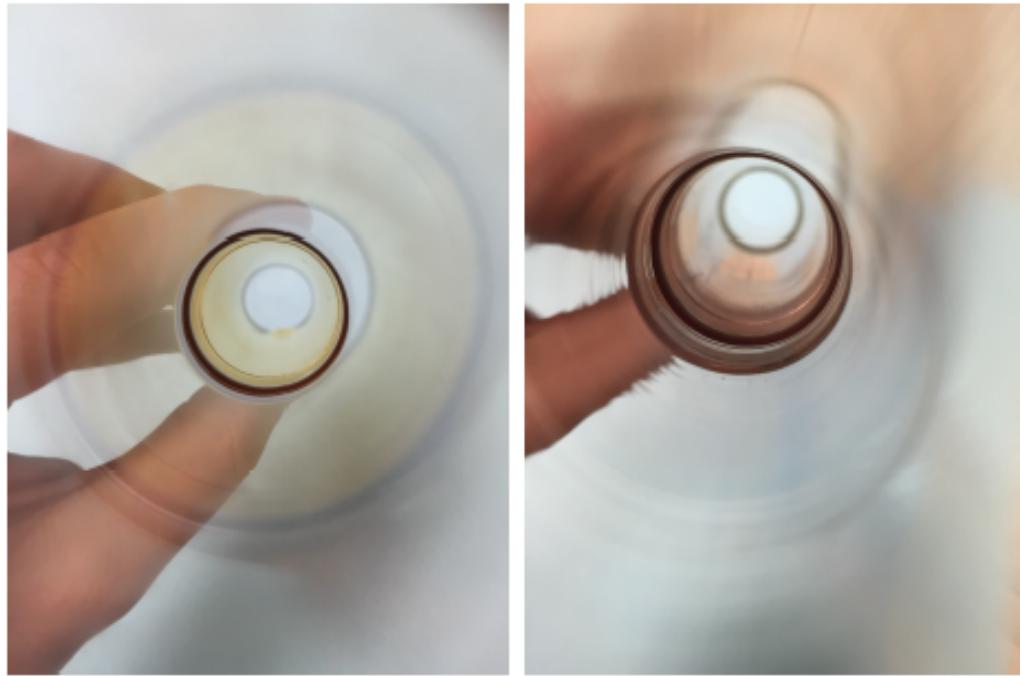


Figure 4.4: Shown left is the inner surface of a gold coated OFHC copper tube. Shown right is a OFHC copper tube without coating.

for electroplating, however, they are both ferromagnetic and cannot be used for us as they would introduce additional spin relaxations. As a result, Epner used a copper strike to improve the durability of gold coating. A $5 \mu m$ layer of gold was subsequently electroplated on the inner surface of the tubes. Fig.4.4 shows a comparison of a OFHC copper tube with gold-coating and a tube without any coating.

4.3.5 Final Assembly of the Cell

After the tubes were coated, shipped back to us and leak checked one more time, they were cleaned with our ultrasonic cleaner. Fig. 4.5 shows the setup of the cleaning process. We cleaned the impurities on the surfaces of the tubes with ethanol,

deionized water and methanol for thirty minutes each. As shown in the figure, a beaker containing the chemical solution and the tubes were placed in water bath inside the ultrasonic cleaner. Because of the dimensions of the cleaner and the beaker, we cleaned one end of the tubes first then flipped them to clean the other end before switching solution.



Figure 4.5: Ultrasonic cleaner with 3 tubes being cleaned.

The test tubes were then shipped to our glassblower Mike Souza at Princeton University. All glass was reblown to the right size to reduce micro-fissures which would lead to high relaxation rate. The test tube was spliced to transfer tube and pumping chamber. A string (see Fig. 4.6) which would be used for cell filling was also made at Princeton. Traditionally, a pure glass cell would be placed entirely in a oven for annealing. A cell made with GE180 would go through a five-minute ramping time

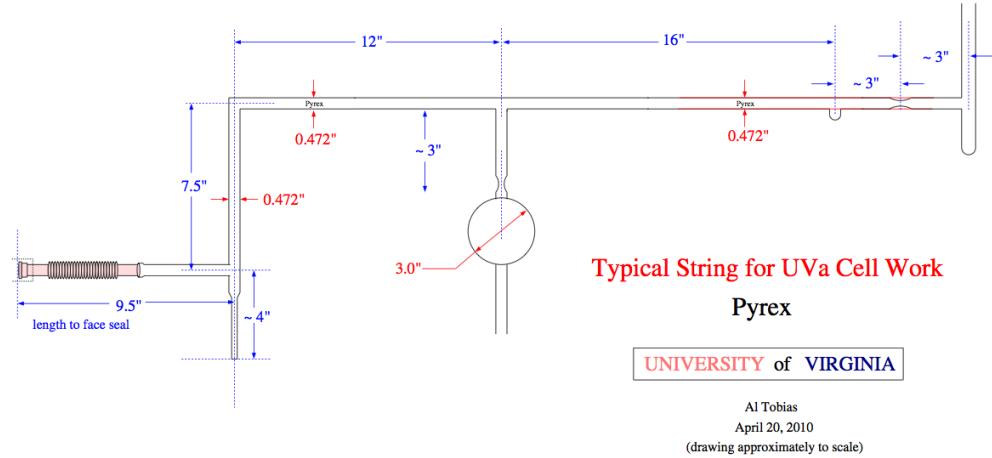


Figure 4.6: Shown is the design of a typical string for our test cells.

to 780°C , stay at 780°C for five minutes and slowly cool down to room temperature for at least 5 hours [41]. A Pyrex cell would be annealed in the exact same way except the highest temperature is 565°C . However, most our test cells could not have been annealed in the same way because of the glass-metal seal. Had we expose the seal to high temperature for long period of time, gold atoms might have migrated into the metal substrate and the seal might even break. Thus the metal tubes were attached to the rest of the glass parts after the pure glass parts were annealed.

4.4 Cell Fill Procedure

The details of cell fill was described thoroughly by Matyas [41], I will briefly cover the process for the sake of completeness.

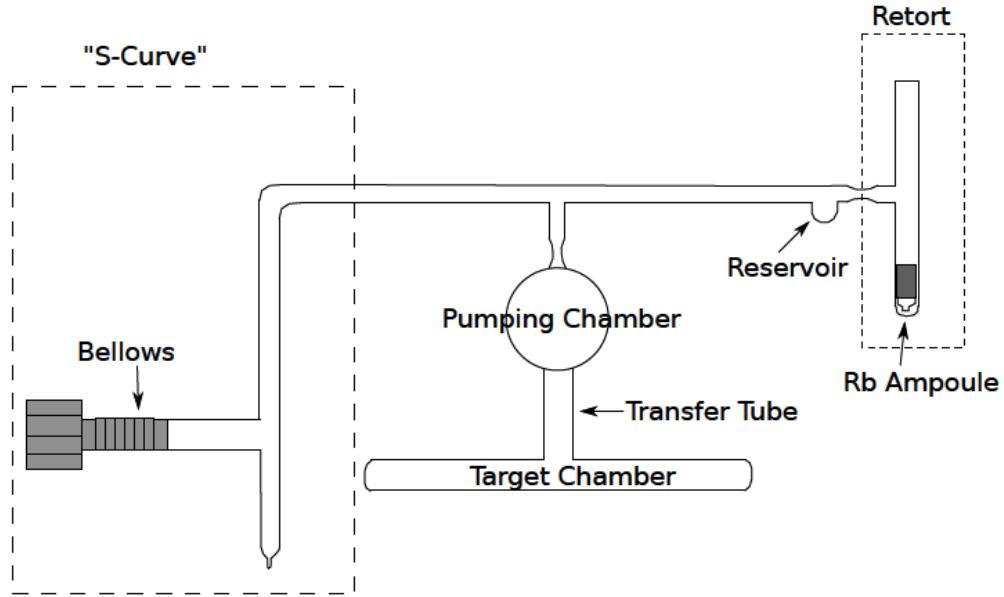


Figure 4.7: A diagram of a Pyrex string with a cell and a retort attached while connected to the gas system through the bellows. Adapted from Matyas [41].

4.4.1 Cell Fill Preparation

Although the actual cell fill work only took less than a day, the preparation that led to the fill usually took 10-15 days. The string, cell and the retort were spliced together and attached to our homemade gas system through the bellows. A pre-scored ampoule of alkali metal was dropped from the top of the retort. Fig. 4.7 is a diagram of the string, cell and retort all connected together.

The entire system was first rough-pumped down with a mechanical pump, and then kept being pumped by a diffusion pump to even higher vacuum. After the system was pumped for a few days and alkali metal was added in the system, we would keep pumping the system with the diffusion pump for roughly a week. To prevent the hot

oil in the diffusion pump from going up into the gas system and the cell, a cold trap above the diffusion pump was filled every day with liquid nitrogen. "Flamebakes" were also performed 2-3 times each day during the one-week pumping period. A methane-oxygen torch was used to gently bake on the glass in the flamebakes. The bake should start from the retort which is the farthest away from the diffusion pump, and move slowly towards the bellows, so the impurities chased off the inner surfaces can be "swept" towards the pump leaving as few impurities behind as possible. The alkali metal was typically melted in the retort on the second or third flamebake and should be melted during each of the remaining flamebakes. On the day before the fill, alkali metal was melted and chased into the pumping chamber.

4.4.2 Cell Fill

The first thing on the fill day was to pump all portions of the gas system to vacuum and selectively back fill some parts with appropriate gases (either N₂ or ³He) to minimize outgassing. The gas filled into the cell should always be cleaned during the path to the cell. Earlier test cells were filled with the noble gas purifier while the later cells were done with a homemade cryogenic trap.

The homemade cryogenic trap consisted of a copper tubing placed inside a box-shaped Dewar. The Dewar was filled with liquid nitrogen when filling N₂ and liquid helium when filling ³He, so impurities in the gas were frozen in the copper tubing. Temperature inside the dewar was monitored with two silicon diodes to determine whether copper tubing was fully submerged in the cold liquid.

The volume of the cell was also determined during the fill process. A Baratron

pressure gauge was used to measure a calibrated volume (CV) of 992.9 cc. 300 Torr of gas was filled into the calibrated volume, then the valve on CV was closed and any gas outside of CV was pumped away. Next the gas kept in CV was let out into the fill gap between CV and the string, then into the cell while monitoring pressure with the Baratron gauge, thus the volume of the cell could be calculated with ideal gas law.

Around 70 Torr of nitrogen was put into the cell before filling ^3He . To prevent nitrogen from escaping the cell while filling ^3He , the string valve was kept closed until ^3He pressure in fill gap rose well above 70 Torr. A total gas pressure of just under 760 Torr (1 atm) was reached. This target pressure was chosen because when the connection between the cell and the string was melted, the atmospheric pressure would collapse it and seal the cell for us. All cells except Kappa1 contained pure rubidium, while Kappa1 contained 5:1 alkali mixture of potassium to rubidium.

4.5 Experimental Setup and Procedure

All cells incorporated metal were tested with Pulse NMR as it only affects a small region of the cell and minimizes influences from eddy current in the metal tubes. Kappa1 was a simple spheric and pure GE180 cell, which was built to rule out the possibility that the melt of GE180 our cells had been made of was bad. Because of its lack of metal and lack of convenient places to wrap pickup coils on, Adiabatic Fast Passage was used to test Kappa1. Both PNMR and AFP were discussed in a general way in chapter 3, I will only add to the discussions with more specific experimental setups for this study. Spindown measurements were the main means

of studying relaxation properties, I will also discuss how we used spindowns with different sampling rate to extract PNMR losses and lifetime corrected for such losses.

4.5.1 Pickup Coils

In a AFP measurement, pickup coils are placed next to the side windows of the oven. However, the same setup proved to be more difficult for us to receive high-quality PNMR signals. To keep the pickup coils from the heat of the oven, we manually wrap a solenoid coil on the transfer tube of test cells where it is approximately 2" below the bottom of the oven. These coils were made with 40-50 turns of AWG 20 copper wire. Because of the off-center positions of the pickup coils, inhomogeneities were significant enough to affect FID signals, gradient coils were used to cancel such inhomogeneities.

4.5.2 Gradient Coils

Inhomogeneities were canceled to the best we could empirically with three sets of gradient coils. Each set of gradients coils consists of two oppositely wound coils separated by a distance d , this particular type of coils are referred to as "Maxwell coils". The setup is very similar to that of Helmholtz coils, except the opposite direction of currents and the larger optimum separation $d = \sqrt{3}R$, where R is the radius of the coils. The opposite direction is to cancel out the magnetic field at the center, and the optimum separation makes the first four leading terms in the Taylor expansion of the inhomogeneities zero[19]. Fig.4.8 shows the coil orientations. The z axis is defined to be aligned with the direction of the holding field while x and y axes

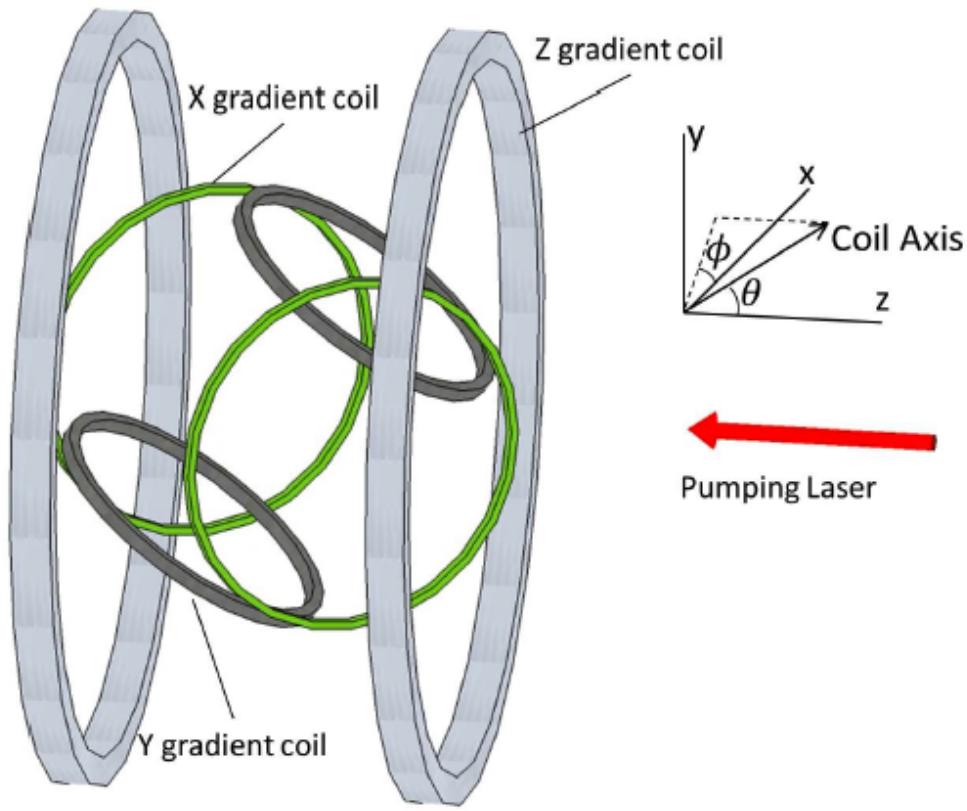


Figure 4.8: Diagram of the coils. Adapted from Zheng [64].

are in the transverse plane. The direction of coil axis is then defined by the angle θ and ϕ , where θ is with respect to the z axis and ϕ is the azimuthal angle in the x-y plane.

The magnetic field at the center is given by[21]

$$\begin{aligned}\nabla \mathbf{B}(\theta, \phi) &= \begin{bmatrix} \partial B_x / \partial x & \partial B_y / \partial x & \partial B_z / \partial x \\ \partial B_x / \partial y & \partial B_y / \partial y & \partial B_z / \partial y \\ \partial B_x / \partial z & \partial B_y / \partial z & \partial B_z / \partial z \end{bmatrix} \\ &= 3\kappa I \begin{bmatrix} \sin^2 \theta \cos^2 \phi - \frac{1}{3} & \sin^2 \theta \sin \phi \cos \phi & \sin \theta \cos \theta \cos \phi \\ \sin^2 \theta \sin \phi \cos \phi & \sin^2 \theta \sin^2 \phi - \frac{1}{3} & \sin \theta \cos \theta \sin \phi \\ \sin \theta \cos \theta \cos \phi & \sin \theta \cos \theta \sin \phi & \cos^2 \theta - \frac{1}{3} \end{bmatrix} \quad (4.16)\end{aligned}$$

where the calibration constant κ is:

$$\kappa = \frac{3\pi n R^2 d/2}{5(d^2/4 + R^2)^{5/2}} G \text{ cm A}^{-1} \quad (4.17)$$

The most important terms in Eq.4.16 are those related to B_z : $\partial B_z / \partial x$, $\partial B_z / \partial y$ and $\partial B_z / \partial z$. The orientations of the gradient coils can be chosen such that each of the three sets of coils controls one of the aforementioned terms. This "magic angle" θ_m is given by

$$\theta_m = \cos^{-1} 1/\sqrt{3} = 54.7^\circ \quad (4.18)$$

For $\theta = \theta_m$ and $\phi = 0$ the gradient tensor is

$$\nabla \mathbf{B}(\theta_m, 0) = \kappa I \begin{bmatrix} 1 & 0 & \sqrt{2} \\ 0 & -1 & 0 \\ \sqrt{2} & 0 & 0 \end{bmatrix} \quad (4.19)$$

For $\theta = \theta_m$ and $\phi = \pi/2$ we have

$$\nabla \mathbf{B}(\theta_m, 0) = \kappa I \begin{bmatrix} -1 & 0 & 0 \\ 0 & 1 & \sqrt{2} \\ 0 & \sqrt{2} & 0 \end{bmatrix} \quad (4.20)$$

Finally, for the z gradient coil we have

$$\nabla \mathbf{B}(\theta_m, 0) = \kappa I \begin{bmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & 2 \end{bmatrix} \quad (4.21)$$

Our gradient coils were built by Zheng [64]. The separations do not follow the optimum condition $d = \sqrt{3}R$ due to spatial limitations. The dimensions of the gradient coils are shown in Table 4.5.2.

	turns	radius	separation
x	42	33 cm	64 cm
y	100	28 cm	56 cm
z	8	66 cm	66 cm

4.5.3 Laser Setup

In earlier stage of the study, narrow-band Commet laser from Newport was used. Each Commet laser provides roughly 20 W power, the optical fibers were combined through a combiner which requires water cooling if multiple lasers are used at the same time. In later studies, single Raytum laser was used instead as it can provide much higher power. Both Commet laser and Raytum laser have around 0.2 nm FWHM (full width at half maximum).

Fig. 4.9 shows a diagram of the optics used for SEOP (spin-exchange optical pumping). After the combiner, laser was focused by two lenses L1 and L2 such that the power spread enough to not damage the polarizing cube and was focused to an appropriate size at the cell position. The polarizing cube separated the beam into two beams with orthogonal linear polarizations. The beam with polarization parallel to this plane went through the cube, and the other beam with polarization perpendicular to the plane was reflected. The reflected beam went through a QWP (quarter wave plate), reflected at a mirror, went through the QWP one more time. Its polarization was changed to be parallel to this plane at this point and went through the cube. This beam is referred to as the "main beam" as it is ideally aligned with the holding field. The beam that went through the cube the first time was sent towards the oven by a mirror. Because of the small angle relative to the main beam, this is called the "skew beam". Both the main beam and the skew beam went through QWP before arriving at the oven window and were turned to circular polarization in the same direction.

4.5.4 PNMR Losses and Corrected Lifetime

As described in chapter 3, there are generally two methods for us to determine polarization losses due to measurements: one is to take several measurements quickly enough that all depolarization can be attributed to measurement losses, the second method is to take multiple spindown measurements each with a different sampling rate. PNMR measurements cause a fraction of longitudinal polarization to be lost, two adjacent measurements require a long enough separation so the depolarization

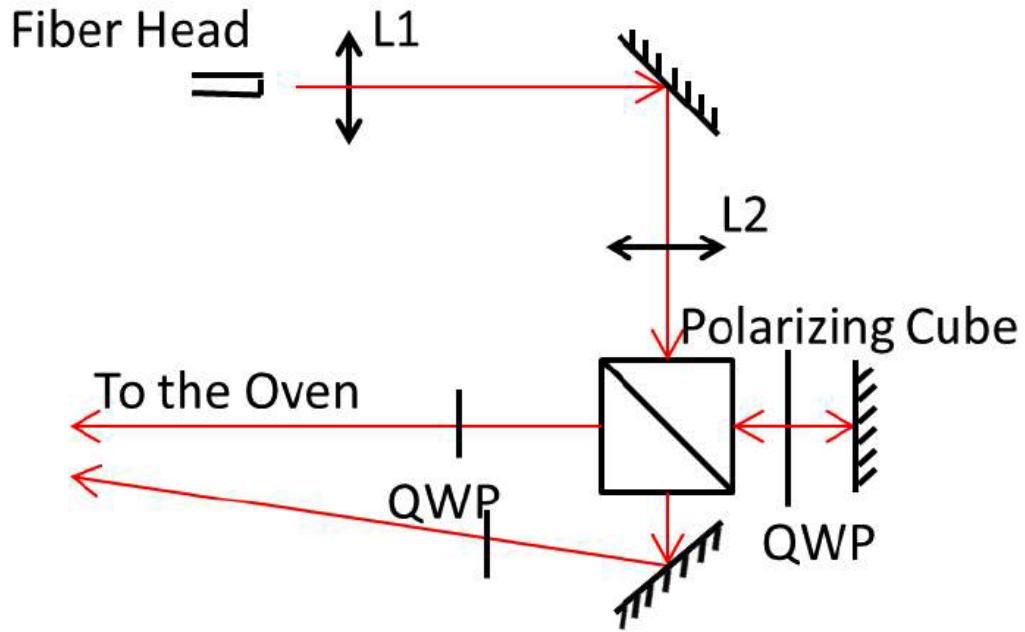


Figure 4.9: Optics for spin-exchange optical pumping. Adopted from Zheng [64].

effect can be fully dispersed. For this reason, the first method mentioned above is not an option for PNMR. We have applied the second method to most of the 19 test cells that had lifetime of at least a few hours.

A spindown is typically a series of measurements of FID (Free Induction Decay) signals. Obtaining a good FID signal generally requires tuning the holding magnetic field and the gradient settings to reduce inhomogeneities.

In a PNMR measurement, it is important that the RF frequency is the correct Larmor frequency of the holding field. The RF frequency was set to be at 56.6 kHz, it was often the case the holding field would start being off-resonance at the beginning. As Fig. 4.10 shows, the final recorded signal was the absolute value of the difference between the signal frequency and the mixing frequency, which does not provide information about whether the frequency of the signal (which was proportional to the

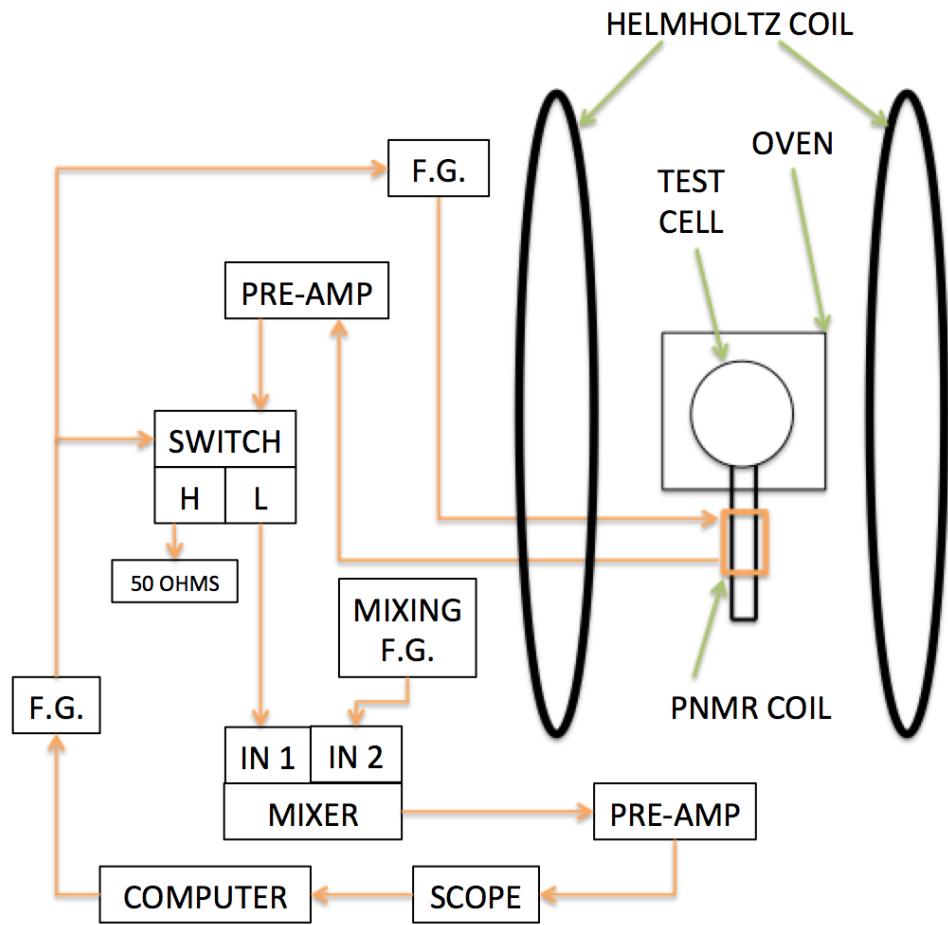


Figure 4.10: PNMR setup.

holding field) was above (or below) the mixing frequency. By intentionally changing the mixing frequency by an appropriate amount, we were able to tell whether the recorded frequency on the oscilloscope should be added to or subtracted from the mixing frequency to give us the signal frequency. The holding field was proportional to the current which was monitored with a shunt resistor, thus the next step was to use the proportionality to calculate the destination shunt resistor value for the correct field. However, this was further complicated by non-zero fields produced by gradient

coils as the pickup coil location was not at the center of the coils. The calculation of destination shunt resistor value typically brought us closer to the resonance field.

The settings of the gradient coils also required tweaking in most cases. In order to reach longer transverse relaxation time T_2 , the inhomogeneities should be low. Because the pickup coils were often at least 3-4 inches below the center of the fields, gradient coils were very useful in decreasing the inhomogeneities. The location of the pickup coils were in the same x-y plane (x, y and z axes were defined the same way as in Fig.4.8) with the center of the fields, so it was expected either $\partial B_z / \partial x$ or $\partial B_z / \partial y$ was initially the dominant term in field inhomogeneities. So x or y gradient coils should provide the most significant improvements at the beginning, while it also produced non-zero partial derivatives in other directions that needed to be canceled with other gradient coils later and non-zero field that required further field tuning. The adjusting of the holding field and the gradients settings were an iterative process for achieving long T_2 and high-quality FID signals. Fig. 4.11 shows a typical FID signal with approximately 150 ms T_2 .

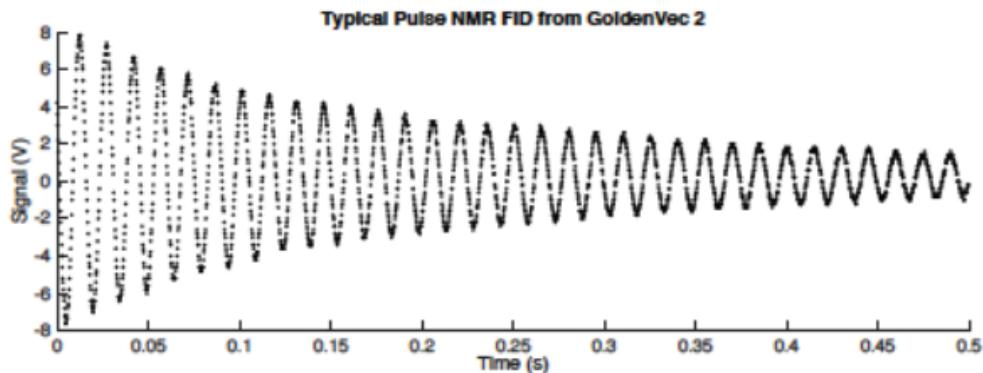


Figure 4.11: A PNMR signal taken with gold coated test cell.

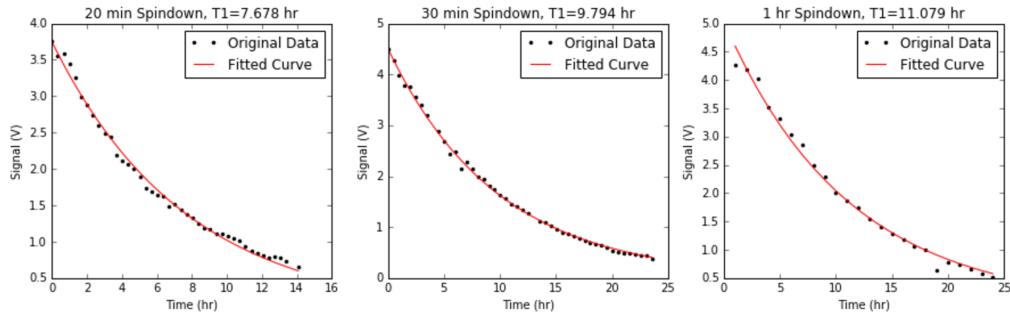


Figure 4.12: 3 spindowns of the cell GoldenVec1 each with a different sampling rate.

Once the optimal settings were found, multiple spindowns were taken. Each spindown was a series of PNMR measurements set apart with fixed time interval. Multiple values of intervals were used for different spindown measurements, each of the intervals was typically repeated at least twice. Fig. 4.12 shows 3 spindowns for the test cell GoldenVec1 which had a horizontal metal tube and similar configuration to that of a convection style cell. The time interval used for the 3 spindowns were 20 min, 30 min and 60 min, respectively. Using the additional relaxation rate due to taking PNMR once every hour as the base rate Γ_{PNMR} , the measured relaxation rate for taking n PNMR every hour can be related to the true lifetime (without relaxation from PNMR) by:

$$\frac{1}{T_{measured}} = \frac{1}{T_{true}} + n \cdot \Gamma_{PNMR} \quad (4.22)$$

in the relation of $\frac{1}{T_{measured}}$ versus n , $\frac{1}{T_{true}}$ is the intercept, and the Γ_{PNMR} is the slope.

Fig. 4.13 shows the linear fit between the sampling rates (n) as x and the inverse of measured lifetimes as y . The intercept is 0.0676 hr^{-1} , thus the true lifetime without

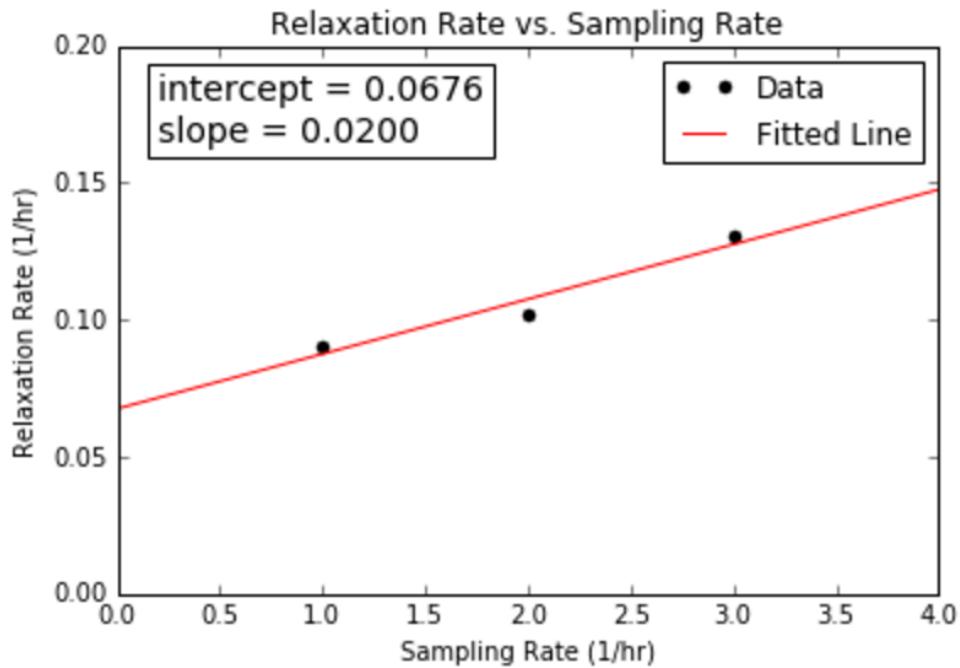


Figure 4.13: A linear fit to extract lifetime corrected for relaxation due to PNMR losses.

any loss due to taking PNMR measurements is $1/0.0676 = 14.8 \text{ hr}$. The slope is 0.02 hr^{-1} , which means the relaxation rate due to taking one PNMR per hour is $1/50 \text{ hr}^{-1}$. The percentage loss of polarization due to a single PNMR measurement can be calculated as:

$$1 - e^{-t \cdot \Gamma_{PNMR}} = 1 - e^{-1 \times \frac{1}{50}} = 0.0198 = 2\% \quad (4.23)$$



Figure 4.14: A diagram of target cell with metal end windows.

4.6 Relaxation Measurement Results and Discussion

The end goal of this series of studies is to replace the traditional glass windows on target chamber with metal end windows as Fig. 4.14 shows. Although no metal end windows have been produced yet, 19 test cells with various configurations and compositions were studied to explore the feasibility of incorporating metallic components into glass cells. We have gone through different stages of studies, from each we made new discoveries and stepped closer to the end goal. At this point, we have demonstrated that gold coated OFHC copper is strong enough to survive the high gas pressure of electron scattering experiments, only brings a relatively small amount of relaxation rates to a convection style target cell made with GE180 under SEOP conditions. The following discussion will be separated into different stages that each revealed parts of the mystery and generally follow chronological order (except cells

Cell Name	Fill Type	Geometry	Glass	Metal	Max Lifetime (hr)	Fill Date
Tyrion	NGP	Sphere	GE180	Gold on glass	1.21	6/18/09
Gold Maiden1	NGP	Flange	Pyrex	Gold on Copper	2.14	6/18/10
Gold Maiden2	NGP	Flange	Pyrex	Gold on Copper	None	8/14/10
Gold Maiden3	NGP	Flange	Pyrex	Gold on Copper	6.49	11/11/10
Goldfinger	NGP	Vertical	Pyrex	Gold on Copper	3.59	4/28/13
Cupid	NGP	Vertical	Pyrex	Bare Copper	3.13	6/15/13
Goldeneye	NGP	Vertical with Valve	Pyrex	Gold on Copper	13.94	10/2/13
GoldRush	NGP	Vertical	Pyrex	Gold on Copper	14.81 [†]	11/8/13
Pyrah	NGP	Vertical	Pyrex	None	26.52 [†]	2/1/14
GoldenVec	NGP	Horizontal	Pyrex	Gold on Copper	10.6	10/18/14
TitanVec	NGP	Horizontal	Pyrex	Gold on Titanium	0.52	12/15/14
GoldenVec2	Cryogenic	Horizontal	Pyrex	Gold on Copper	15.6	2/14/15
Titan	NGP	Vertical	Pyrex	Bare Titanium	None	3/11/15
GoldenVec180	Cryogenic	Horizontal	GE180	Gold on Copper	4.43	6/17/15
GolderVec360	Cryogenic	Horizontal	GE180	Gold on Copper	3.01	7/11/15
Tweety	Cryogenic	Vertical	Pyrex	Canary Glass	22.7	9/22/15
Sylvester	Cryogenic	Horizontal	GE180	Canary Glass	6.39	11/20/15
Kappa1	Cryogenic	Sphere	GE180	None	72.17	2/6/16
Goldfinger180	Cryogenic	Vertical	GE180	Gold on Copper	12.4 [†]	5/19/16

Table 4.1: Shown are the fill information, design and maximum measured lifetime of the test cells. Fill type is the method used to clean the gas. [†] indicates the maximum lifetime was obtained at an elevated position. Although canary glass is not metal, it is listed in the column of metal for Tweety and Sylvester for the sake of keeping the structure of the table simple.

made with titanium will be covered at the end). The cell preparation information and results of measurements are listed in Table 4.1.

4.6.1 Gold Coated Spherical Cell

Our very first attempt at incorporating metal was a spherical GE180 cell with gold coated on the interior surface. The measured lifetime was only 1.21 hr, which was attributed to impurities in the cell. Although this attempt was made before I joined the group as were the spool pieces that will be mentioned below, they are still included for the sake of completeness of this study.



Figure 4.15: A picture of Gold Maiden, generally referred to as the "spool piece".

4.6.2 Gold Coated Spool Pieces

The first cell with a gold coated (by Epner) OFHC copper tube was named as "Gold Maiden" and made in 2010. Fig. 4.15 is a picture of the cell which was referred to as the "spool piece". The bottom end of the cell was capped off, while the other end was attached to a glass flange. The flange was then connected to the transfer tube of the cell. The seal was made with o-rings.

The tests of Gold Maiden was divided into three stages. Initially it had a lifetime of 2.14 hr before discoloration of Rb was observed. The discoloration is typically a sign of Rb reacting away with air, which indicated either a leak or significant amount of outgassing from the o-rings. To separate Rb from possible outgassing, the pumping chamber was pulled off via glassblowing. Measurements of the separated pumping chamber showed a lifetime of 109.2 hr after correcting for AFP losses. The rest of the

cell which contained the transfer tube and the metal tube was measured again with PNMR, and a lifetime of 6.49 hr was recorded. The spool piece encouraged further investigation into gold coated OFHC copper that could be attached to glass without o-rings.

4.6.3 Vertical Cells

Since o-rings were the primary suspects of the short lifetime for the spool piece, a method to seal copper directly to the glass was highly desirable. We started using the Houskeeper seal, and five cells were made at this stage: Goldfinger, Cupid, Goldrush, Goldeneye and Pyrah. All 5 cells were designed to have the same dimensions. Fig. 4.16 shows the design drawing and a picture of the cell Goldfinger. Goldrush was of the same design. The only difference for Cupid was that the OFHC copper was not coated. Pyrah was a pure Pyrex control cell.

Goldfinger had a maximum lifetime of 3.59 hr at the beginning of the test. However, subsequent tests showed a somewhat obvious degradation. Towards the end of testing Goldfinger, the lifetime was at a minimum of 2.36 hr. Cupid had a similar maximum lifetime of 3.13 hr initially. A even more significant degradation of lifetime was observed with Cupid. The last measurement we were able to perform displayed only 0.27 hr. It is worth noting a leak on Goldfinger was discovered during the filling process. Although the leak was fixed with Celvaseal Leak Sealant, either the sealant or the leak itself might have contributed to the short lifetime of Goldfinger and its degradation. Rb is also known to form amalgam with copper, which could be the reason for the observed behavior of Cupid. Fig. 4.17 displays the lifetime degradation

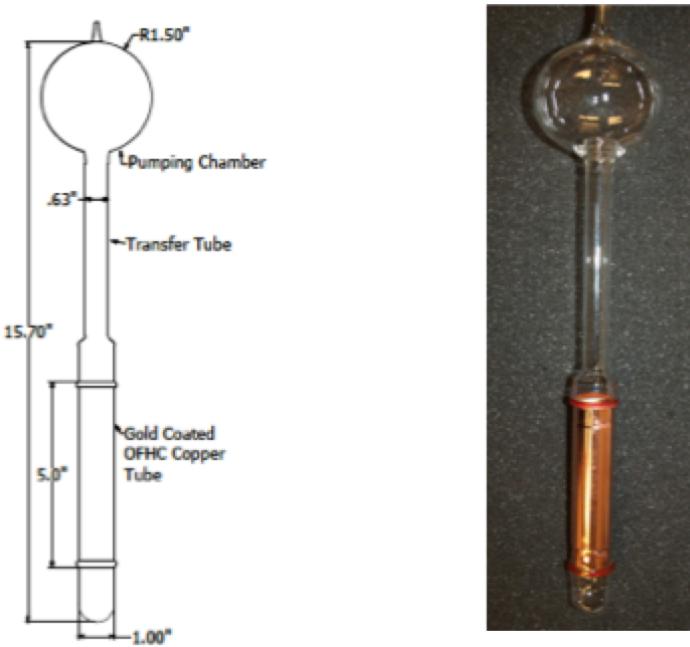


Figure 4.16: Design and picture of Goldfinger.

during our tests.

During the tests of Goldfinger and Cupid, it was suspected that Rb reacted with metal surfaces and produced detrimental effects to lifetime. A gold coated OFHC cell with valve was designed and produced to isolate Rb vapor from the metal tube, as shown in Fig. 4.18. GoldenEye was only polarized after the valve was closed, thus Rb vapor should never reach the metal tube. Spindown measurements were taken at room temperature both with the valve closed and open. The lifetime was measured to be 13.94 hr with the valve closed, and it was only 4.09 hr with the valve open. The large difference of lifetime indicated the part of the cell below the valve introduced very significant relaxation rate. It is also worth noting the inclusion of the valve increased the total length of the cell substantially, which exposed the bottom part of

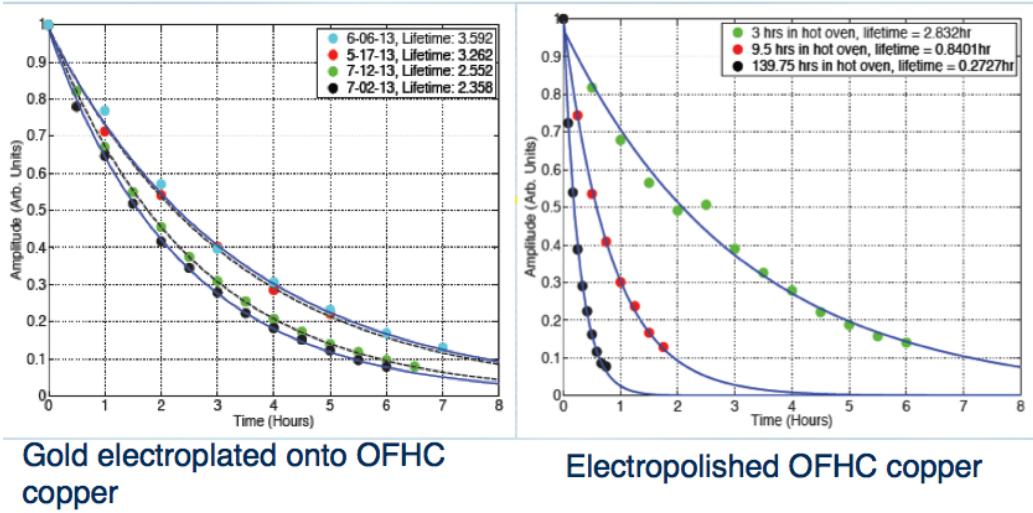


Figure 4.17: The observed degradation of lifetime for Goldfinger (left) and Cupid (right). Shown in each of the two plots were several spindowns at different stage during the tests. The initial amplitude of the spindowns were scaled to 1 for better comparison of lifetime.

the cell to high field inhomogeneities that could lead to high relaxation rate.

Due to the leak found in Goldfinger, it was necessary to produce another test cell with the same design, which was named as “Goldrush”. Fig. 4.19 display four spindowns taken before elevating the cell. All four measurements have similar results with the maximum lifetime being 12.1 hr. This was the first metal cell we produced that had more than 10 hr lifetime.

A calculation of the magnetic field inhomogeneities was also done around the time. As seen in Fig. 4.20, the relaxation time due to inhomogeneities at the center of the pumping chamber was about 300-700 hr. However, the bottom of cell was exposed to surprisingly large inhomogeneities which would result in a relaxation time of 1-5 hr. This provided incentive for measurements of lifetime while raising the cell to more



Figure 4.18: A picture of GoldenEye, the only test cell made with a valve.

homogeneous region. Due to spatial constraints, the cell was only lifted by 10 cm, which led to an increase in lifetime to 14.81 hr, the longest lifetime we had measured at the time with metal cells.

To understand the additional relaxation rates introduced by metal surfaces, a control cell, Pyrah, was made with pure Pyrex was produced. The control cell had a lifetime of 19.71 hr before elevation, and 26.52 hr after being lifted by 10 cm. Since Pyrah had the same dimensions as Goldrush, a direct comparison suggested the additional relaxation rate due to the metal tube was around $1/25\text{ hr}$.

4.6.4 Horizontal Cells

The increase of lifetime after elevating Goldrush and Pyrah motivated us to make more compact cell designs. This resulted in a design very similar to that of a con-

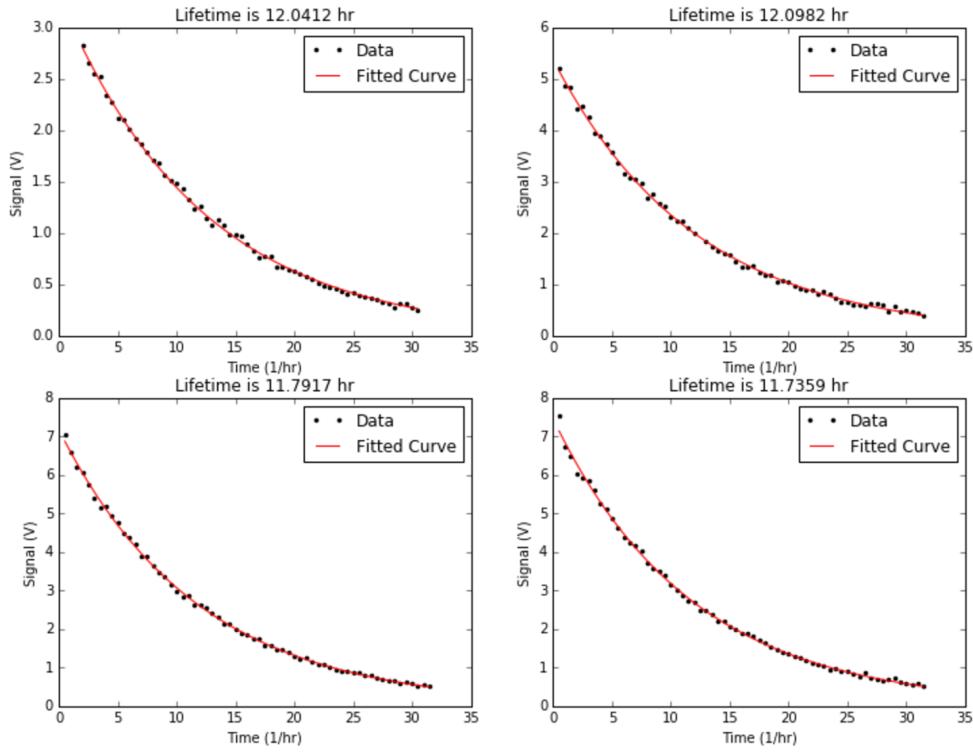


Figure 4.19: Four spindowns of Goldrush before elevating the cell. All four measurements display similar lifetime with no obvious sign of degradation.

vection style cell with the metal tube placed horizontally. The overall vertical length was shortened from 15.75" to about 10". The dimensions of this style of cells can be seen in Fig. 4.21.

The first horizontal cell was GoldenVec. It had a lifetime of 10.6 hr, which was shorter than we had hoped for. Up to this point, all test cells were filled using the noble gas purifier in our lab. A convection cell filled at the time that used the cryogenic trap to condense the impurities in the gas turned out to have much greater lifetime compared to the previous convection cell of the same design but used the noble gas purifier. In light of this, GoldenVec2 was created with the same design as

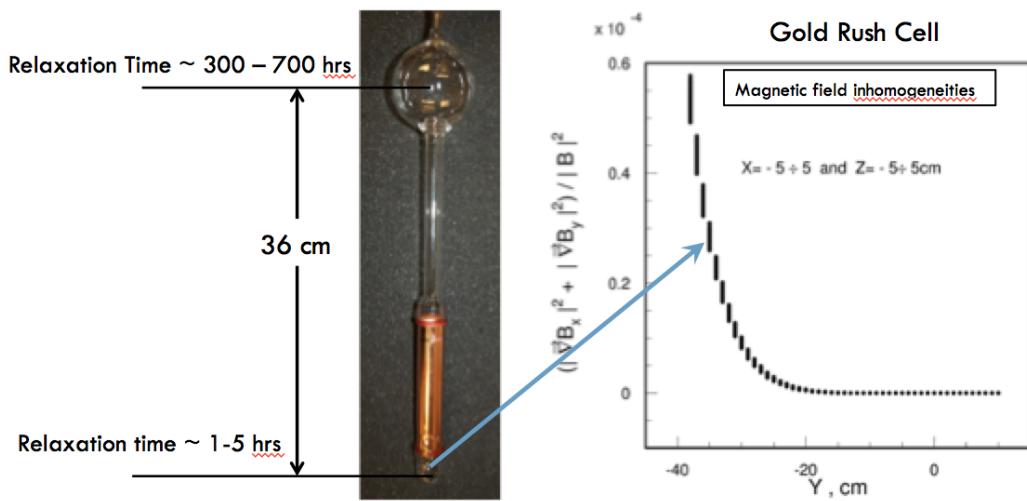


Figure 4.20: Shown on the right is the inhomogeneities vs. vertical distance from the center of the field. Shown on the left is the cell Goldrush with relaxation time due to field inhomogeneities as displayed on the right.

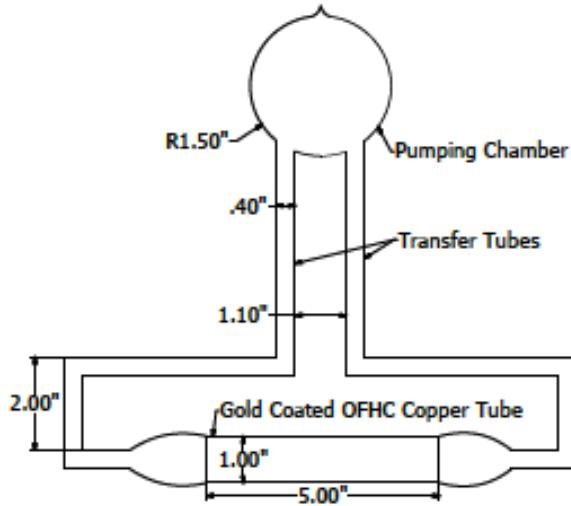


Figure 4.21: Design of the horizontal cell GoldenVec.

that of GoldenVec and the cryogenic trap. GoldenVec2's lifetime was measured to be 15.6 hr, the longest lifetime of metal cells we have obtained to date. All test cells

from then on were filled using the same cryogenic trap.

4.6.5 GE180 Cells

It was widely acknowledged that the impermeable aluminosilicate glass GE180 had less associated wall relaxation rates than that of the borosilicate glass Pyrex. Since GoldenVec2 and Goldrush had demonstrated good enough lifetime, it was time we started exploring GE180 cells with metal tubes. Two GE180 test cells, GoldenVec180 and GoldenVec360, were made with the design as seen in Fig. 4.21. Because of the difficulty in bonding GE180 to OFHC copper, Corning 7052 was used as the transition glass. The measured lifetime of GoldenVec180 and GoldenVec360 were 4.43 hr and 3.01 hr, respectively. Since GE180 glass normally provide longer relaxation time, we made additional cells to test different hypothetical interpretation of the short lifetime.

The first suspect occurred to us was the transition glass Corning 7052, since it had never been tested in our group. As we had experience with an alternative transition glass, canary glass (uranium glass), we built two test cells with 5" long by 1" OD tubes of the canary glass in place of copper tubes. The first was a replicate of the vertical cell Goldfinger and named as Tweety. The second was a replicate of the horizontal cell GoldenVec and named as Sylvester.

Tweety had a lifetime of 22.7 hr, which was quite close to the lifetime of the control cell Pyrah, 26.52 hr. A shorter lifetime was expected as uranium is paramagnetic, however, the result of Tweety proved the expectation was incorrect and canary glass could in fact be used to build target cell.

Results of Sylvester was also unexpected as the lifetime was only 6.39 hr even in the

absence of metal. This lifetime was shorter than that of GoldenVec and GoldenVec2.

At this point, we narrowed down the causes of such short lifetime to the last two:

1. The melt of the GE180 glass had too much impurities.
2. Insufficient annealing resulted in more remaining micro-fissures in the glass.

The first possibility is easy to understand and could be easily tested with a simple spherical cell built from the same melt of GE180. Kappa1 was produced for this exact purpose. Hybrid Mixture of Rb and K was used instead of Rb because Kappa1 was also intended for future measurements of the physical quantity κ_0 which require hybrid mixtures. Kappa1 was found to have a lifetime of 72 hr, so the cause of high relaxation in previous GE180 cells was not the melt itself.

As mentioned in previous sections, to keep the glass-metal seals intact, all glass parts not directly attached to metal tubes were placed in a oven for annealing. Larger amount of micro-fissures could still exist in the glass directly attached to metal. This could potentially lead to the short lifetime we observed with the GE180 cells up to this point. To test the hypothesis, Goldfinger180 was built following the design of Goldfinger. However, the glass directly attached to the metal tube that did not go into the oven was intentionally made shorter by our glassblower. During the fill process, a leak at the Houskeeper seal had to be fixed with two coats of an epoxy sealant Stycast 1266. The lifetime was measured to be 10.4 hr before elevation, and 12.4 hr after elevating in the same manner as Goldrush and Pyrah. The result both demonstrated the feasibility of using gold coated OFHC copper with GE180 and further confirmed the significance of glass annealing.

4.6.6 Titanium Tubes

Although gold coated OFHC copper was the main focus of the study due to the ease of manufacturing, titanium was also investigated for its stronger tensile strength and lower atomic number. However, the results of the two titanium cells were not satisfactory. A test cell with bare titanium tube, Titan, had lifetime too short to be measured. The other cell TitanVec with gold coated titanium tube had lifetime of only 0.52 hr. Because of the difficulty in coating gold onto titanium surface, parts of gold coating peeled off the surface during the cleaning process with our ultrasonic cleaner. ^3He was exposed to part of the titanium surface underneath, which could be the reason for the bad relaxation properties. As a result, titanium was not investigated further.

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Appendix A

Appendix title

This is Appendix A.

You can have additional appendices too (*e.g.*, `apdxb.tex`, `apdxc.tex`, *etc.*). If you don't need any appendices, delete the appendix related lines from `thesis.tex` and the file names from `Makefile`.