

1 Wave particle duality

Energy Quanta: (1)

$$u_v dv = \frac{8\pi v^2}{c^3} \cdot \frac{hv}{e^{\frac{hv}{K_b T}} - 1} \quad (2)$$

$$E = hv \quad (3)$$

Photoelectric Effect: (4)

$$\frac{1}{2} m_e v_k^2 = E_k = hv - W \quad (5)$$

$$\frac{1}{\lambda} = R_H \left(\frac{1}{n_1^2} - \frac{1}{n_2^2} \right) \quad (6)$$

$$R_H = 1.097373 \times 10^7 m^{-1} \quad n_1 = 1 \Rightarrow Lyman \quad (7)$$

$$n_1 = 2 \Rightarrow Balmer \quad (8)$$

$$n_1 = 3 \Rightarrow Paschen \quad (9)$$

$$n_1 = 4 \Rightarrow Brackett \quad (10)$$

$$n_1 = 15 \Rightarrow Pfund \quad (11)$$

$$n_2 > n_1 \quad (12)$$

De Broglie Relationship:

$$E^2 = p^2 c^2 + m^2 c^4 \quad (13)$$

$$\text{for zero rest mass: } E^2 = p^2 c^2 \rightarrow E = pc \quad (14)$$

$$hv = h \frac{c}{\lambda} = pc \quad (15)$$

$$\therefore p = \frac{h}{\lambda} \quad (16)$$

Electron diffraction (17)

$$E_K = eV - \frac{p^2}{2m} \quad (18)$$

$$p = \sqrt{2meV} \quad (19)$$

$$\lambda = \frac{h}{\sqrt{2meV}} \quad (20)$$

$$\lambda = 2d \sin(\theta) \quad (21)$$

$$\sin^2(\theta) = \frac{C}{V}, \text{ Where } C = \frac{h^2}{8med^2} \quad (22)$$

Bohr Model (23)

$$F_{centripetal} = F_{electric} \quad (24)$$

$$m \frac{v^2}{r} = \frac{e^2}{4\pi\epsilon_0 r^2} \quad (25)$$

$$\therefore v^2 = \frac{e^2}{4\pi\epsilon_0 mr} \quad (26)$$

$$I_{orbit} = n\lambda = 2\pi r \text{ Where } n \text{ is the quantisation condition} \quad (27)$$

$$\therefore v = \frac{hn}{2\pi mr} \quad (28)$$

$$L = mvr = n \frac{h}{2\pi} = n\hbar \quad (29)$$

$$\text{As such, allowed radii are described by the expression:} \quad (30)$$

$$r_n = \frac{h^2\epsilon_0 n^2}{\pi m e^2} = a_{Bohr} n^2 \quad (31)$$

$$r_1 = a_{Bohr} = \frac{h^2\epsilon_0}{\pi m e^2} = 5.3 \times 10^{-11} m \quad (32)$$

$$E_k = \frac{1}{2}mv^2 = \frac{e^2}{8\pi\epsilon_0 r} \quad (33)$$

$$E_T = E_k + E_c = \frac{e^2}{8\pi\epsilon_0 r} - \frac{e^2}{4\pi\epsilon_0 r^2} = -\frac{e^2}{8\pi\epsilon_0 r} \quad (34)$$

$$\text{Substiting in } r_n \text{ gives:} \quad (35)$$

$$E_T = -\left(\frac{me^4}{8\epsilon_0^2 h^2}\right) \cdot \frac{1}{n^2} = \frac{-13.6eV}{n^2}, \quad 1 \text{ eV} \approx 1.602 \times 10^{-19} J \quad (36)$$

$$\Delta E = E_i - E_f = \frac{me^4}{8\epsilon_0^2 h^2} \left(\frac{1}{n_f^2} - \frac{1}{n_i^2}\right) \quad (37)$$

$$\text{textsince : } E = hv = h \frac{c}{\lambda} \quad (38)$$

$$\frac{1}{\lambda} = \frac{me^4}{8\epsilon_0^2 h^3} \left(\frac{1}{n_f^2} - \frac{1}{n_i^2}\right) = R_H \left(\frac{1}{n_f^2} - \frac{1}{n_i^2}\right) \quad (39)$$