Equations

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1 Variables

Temperature (K)	T
Heat Capacity $(\frac{J}{K})$	
Heat Energy (J)	C
Internal Energy (J)	Q
Volume (m ³)	E
, ,	V
Mass (Kg)	m
Density $\left(\frac{\text{Kg}}{\text{m}^3}\right)$	ho
Mass-specific heat capacity $\left(\frac{J}{kg\cdot K}\right)$	
	c_m
Thickness (m)	L
Area (m ²)	A
Thermal Conductivity (K-Value $\frac{W}{m \cdot K})$	
Th T. T	K
Thermal Transmittance (U-Value m	U
	-

Thermal Resistance (R-Value
$$\frac{m^2 \cdot K}{W})$$

R

Thermal Conductance
$$(\frac{W}{K})$$

UA

2 Formulas

Definition of Mass-specific Heat capacity

$$c_m = \frac{C}{m}$$

Definition of Heat capacity

$$C = \frac{Q}{\Delta T}$$

Definition of Density

$$\rho = \frac{m}{V}$$

Definition of Internal energy

$$E = c_m \cdot T$$

Definition of Thermal Transmittance

$$U = \frac{1}{R} = \frac{K}{L}$$

Definition of Thermal Conductance

$$UA = U \cdot A$$

3 Simulation

Rate of Energy Transmission

$$\dot{Q}_{a\to b} = T_a \cdot U A_{a\leftrightarrow b}$$

$$\dot{Q}_{b\to a} = T_b \cdot U A_{a \leftrightarrow b}$$

Energy Transmitted

$$Q_{a\to b} = \dot{Q}_{a\to b} \cdot \Delta t$$

$$Q_{b\to a} = \dot{Q}_{b\to a} \cdot \Delta t$$

New Total Energy

$$E_a = E_a + Q_{b \to a} - Q_{a \to b}$$

$$E_b = E_b + Q_{a \to b} - Q_{b \to a}$$

New Temperature

$$T_a = \frac{E_a}{C_a}$$

$$T_b = \frac{E_b}{C_b}$$

4 Matrix

Substitution into base variables

$$T_{a-new} = \frac{\left(C_a \cdot T_a - \left(U A_{a \leftrightarrow b} \cdot T_a \cdot \Delta t\right) + \left(U A_{a \leftrightarrow b} \cdot T_b \cdot \Delta t\right)\right)}{C_a}$$

$$T_{b-new} = \frac{\left(C_b \cdot T_b + \left(U A_{a \leftrightarrow b} \cdot T_a \cdot \Delta t\right) - \left(U A_{a \leftrightarrow b} \cdot T_b \cdot \Delta t\right)\right)}{C_b}$$

Turning into matrix format

$$T_{a-new} = T_a(1 - UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_a}) + T_b \cdot (UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_a})$$

$$T_{b-new} = T_a \cdot (UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_b}) + T_b(1 - (UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_b}))$$

Matrix Equation

$$\begin{bmatrix} T_{a-new} \\ T_{b-new} \end{bmatrix} = \begin{bmatrix} 1 - UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_a} & UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_a} \\ UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_b} & 1 - UA_{a \leftrightarrow b} \cdot \Delta t \cdot \frac{1}{C_b} \end{bmatrix} \cdot \begin{bmatrix} T_a \\ T_b \end{bmatrix}$$

$$\vec{T}_{new} = M \cdot \vec{T}$$

5 Time Step Matrix (M)

$$M = C \cdot (M_{adjacency} + E_{lost}) \cdot \Delta t + I$$

Identity Matrix (I)

$$\begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix}$$

Weighted Adjacency Matrix/Rate of Energy Transmitted into the mass $(M_{adjacency})$

$$\begin{bmatrix} 0 & UA_{a\leftrightarrow b} \\ UA_{a\leftrightarrow b} & 0 \end{bmatrix}$$

Loosing Energy Matrix/Rate of Energy Transmitted out of the mass (E_{lost})

$$\begin{bmatrix} -UA_{a\leftrightarrow b} & 0 \\ 0 & -UA_{a\leftrightarrow b} \end{bmatrix} = \begin{pmatrix} M_{adjacency} \cdot \begin{bmatrix} -1 \\ -1 \end{bmatrix} \end{pmatrix}.diag$$

Thermal Capacity Matrix (C)

$$\begin{bmatrix} C_a & 0 \\ 0 & C_b \end{bmatrix} = \begin{bmatrix} C_a \\ C_b \end{bmatrix} .diag$$

6 Superfast simulation

Some ways of simulating the change in heat area over 1024 timesteps with a 10*10 matrix:

Both these ways take very long and are inefficient so to make it super fast I calculate it like this:

$$\vec{T}_{new} = \left(\left(\left(M \cdot M \right) \cdot \left(M \cdot M \right) \right) \cdot \left(\left(M \cdot M \right) \cdot \left(M \cdot M \right) \right) \right) \cdot \vec{T}$$

simplified it looks like this:

 $\vec{T}_{new} = M^{(2^3)} \cdot \vec{T}$ this way takes 10,000 multiplications

$$\vec{T}_{new} = (M \cdot M))))))) \cdot \vec{T} = M^{(2^3)} \cdot \vec{T}$$

$$\dot{Q}(energy~flow) = \frac{\Delta T}{R(thermal~resistance)}$$
 is analogous to $I = \frac{V}{R}$