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Triorthobrombenzylamine Chloplatinate, (C6H4BrCH2), NH2PtCl, made by adding chlorplatinic acid to an ethereal solution of the amine, gave the following result on analysis: -- 0.39708 gr. of the salt gave 0.0540 gr. Pt. Calculated for (C, H, Br)2NH2PtCl, . Found. Platinum 13.51 13.60 A whitish yellow, barely crystalline precipitate, insoluble in water and ether, and very slightly, if at all, soluble in alcohol. On boiling the tertiary amine with dilute alcohol and hydrochloric acid, it dissolved, and on cooling deposited white spheres made up of radiated needles; but an analysis of the substance gave an amount of chlorine too low for...



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