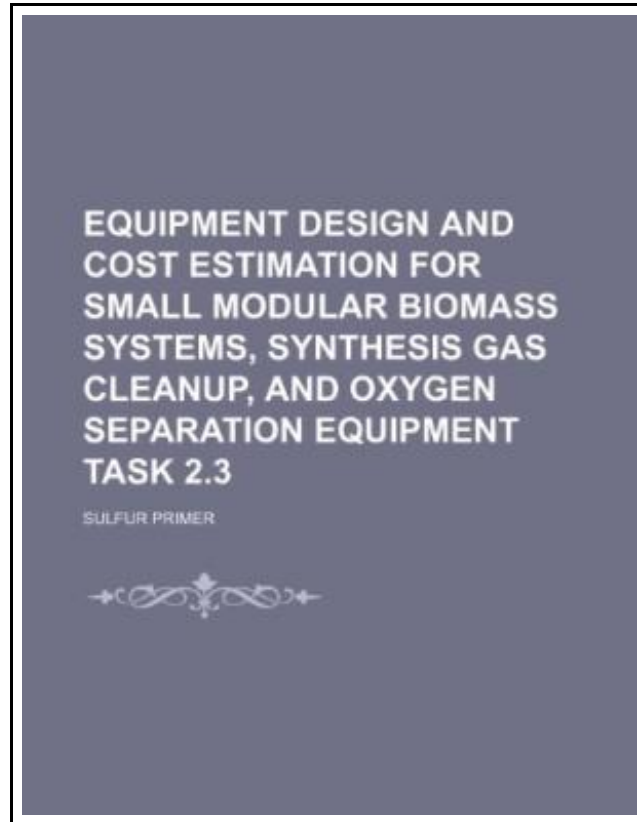


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Reviews

Extremely helpful to all of group of individuals. It really is loaded with knowledge and wisdom Its been designed in an extremely basic way and is particularly simply after i finished reading through this ebook where actually altered me, affect the way i believe.

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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 24 pages. Original publisher: Golden, Colo. : National Renewable Energy Laboratory, 2006 OCLC Number: (OCoLC)177061629 Subject: Desulfurization. Excerpt: . . . 4. Hydrocarbons in physical solvents have caused significant foaming and operational problems in the past, especially with Selexol. While some steps can be taken to minimize this, it can become a major operational issue. 5. It is unclear due to limited data how removing large quantities of tars and oils will impact the ability of the physical solvent to remove H₂S. As the physical solvent becomes more loaded (with preferential solubility to aromatic compounds over H₂S and CO), greater recycle rates and larger vessels would likely be necessary. SECTION 4: CATALYTIC ABSORBENTS Catalytic absorbents, such as zinc oxide (ZnO) or copper oxide (CuO), are often used as a polishing step for sulfur removal in gas streams where the incoming sulfur level is 10 to 20 ppm or lower. For hydrocarbon and IGCC applications, catalytic absorbents are usually not a primary sulfur removal process. They are typically applied downstream of either a chemical or physical solvent to assure the sulfur level entering other process equipment, such as chemical synthesis reactors, is very low. Since ZnO is the technology most commonly deployed, it will be focused upon here. Technology Description and Rationale for Use H₂S in a gas stream reacts with the ZnO absorbent in a fixed bed reactor, where it is converted to zinc sulfide. According to technology providers, sulfur removal 12 down to 20 to 50 ppb is attainable with ZnO. Since the sulfur specifications for alcohols and FT liquids are 1 ppm in the syngas stream, ZnO could be used to assist in...



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