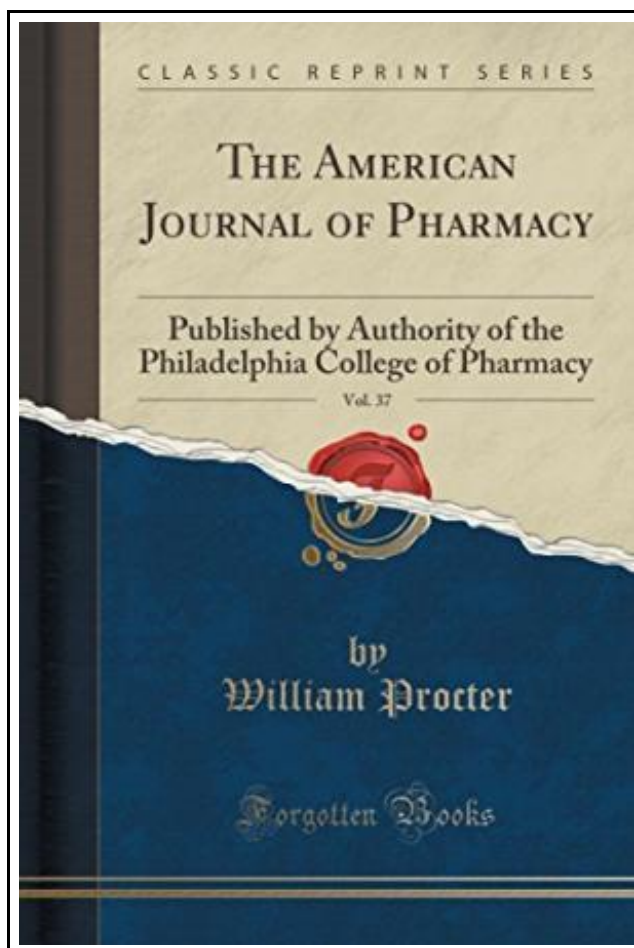


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Reviews

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Forgotten Books, United States, 2015. Paperback. Book Condition: New. 229 x 152 mm. Language: English . Brand New Book ***** Print on Demand *****.Excerpt from The American Journal of Pharmacy, Vol. 37: Published by Authority of the Philadelphia College of Pharmacy The writer, having observed the solution of citrate of bismuth, obtained by the addition of ammonia, to have an acid reaction, and also that an excess of that alkali occasioned a precipitate of oxide of bismuth, inferred the existence of a compound salt of bismuth and ammonia; this was subsequently confirmed. The following formulas, suggested by theory and numerous experiments, were finally adopted; Bismuthi Citras. Take of Subcarbonate of Bismuth a troyounce; Citrate of Potassa a troyounce and 120 grains; Nitric Acid a troyounce and a half; Distilled water a sufficient quantity. Dissolve the subcarbonate of bismuth in the nitric acid, and, when effervescence has ceased, dilute the solution with a fluid-ounce and a half of distilled water, gradually added. Dissolve the citrate of potassa in two pints of distilled water, and to this liquid add slowly, with constant stirring, the acid solution of bismuth. Permit the mixture to stand for several hours; then pour it on a moistened paper filter, and, when the liquid has nearly ceased to pass, cover the surface of the precipitate with distilled water. Repeat this operation until the washings no longer contain nitric acid. Allow the magma to drain, and dry it on bibulous paper, with a gentle heat. Bismuthi et Ammoni Citras. Take of Citrate of Bismuth, in powder, a convenient quantity; Stronger Water of Ammonia, Distilled Water, each, a sufficient quantity. Rub the citrate of bismuth with sufficient distilled water to reduce it to a uniform pasty consistence, and add, cautiously, with constant trituration, stronger water of ammonia, until a solution...



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