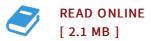




A laboratory manual of inorganic chemistry

By John Bernard Ekeley

RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 44 pages. Dimensions: 9.7in. x 7.4in. x 0.1in. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1912 Excerpt: . . . and Subnitrate. H 173. Examine a crystal of bismuth nitrate. Add about twenty parts of water to the crystal, and boil the mixture. What is the precipitate () Bismuth Trisulphide. 174. Pass hydrogen sulphide into an acid solution of bismuth nitrate. Filter off the precipitate and try its solubility in acids and in alkaline sulphides. CARBON Hlf 176-179 Experiment 133. Physical and Chemical Properties. Examine the different forms of carbon that may be found in the laboratory, and make a table showing their physical and chemical properties so far as you can learn them through experiment or reading. See Exp. 14, (6). METHANE If 180 Experiment 134. (a) Powder 2 g. of fused sodium acetate and 2 g. of sodium hydroxide. Mix these with 8 g. of powdered quick lime (or 10 g. of powdered soda lime may be used with 2 g. of sodium acetate)....



Reviews

A top quality publication along with the font used was intriguing to read. I really could comprehended everything using this written e ebook. Its been designed in an remarkably straightforward way and it is only after i finished reading through this publication by which basically altered me, modify the way i believe.

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