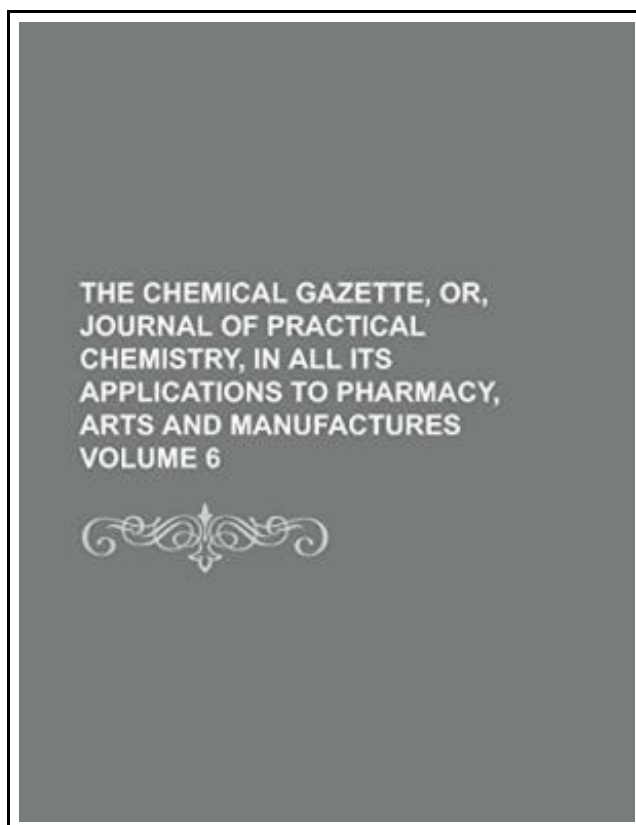


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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1848 edition. Excerpt: .if carbonic acid is absorbed by sulphuric acid, it neither adds to its weight, nor will it when in solution form carbonate of lime, both of which seem quite incomprehensible. It may be as well to add, that the experiment was performed with a delicate balance made by Oertling of London, and the carbonic acid was set free from carbonate of lime by dilute hydrochloric acid, so that no absorption could have taken place in the evolution flask. I am, Sir, Your most obedient Servant, 35 Michael-s Place, Brompton. Reginald J. Morley. Mode of detecting Neutral Carbonated Alkali in Bicarbonate. Chcvallier recommends the following method: --Starch-sugar is added to the aqueous solution of the bicarbonate under examination, and the mixture heated; if any neutral carbonate is present, the mixture turns yellow or brown--Liebig-s Annalen, Dec. 1847. THE CHEMICAL GAZETTE. No. CXXXVII--July 1, 1848. SCIENTIFIC AND MEDICINAL CHEMISTRY. On the Nitrates of Mercury. By C. Gerhardt. It is well known that hyponitric vapours instantly attack metallic mercury, and convert it into a salt. Several chemists conclude, from the nitrogen substitutions in organic substances, that the hypo-nitric vapours behave under these circumstances as a simple body, combining directly with the mercury according to the equation $\text{NO}_3 \text{] Hg } = \text{NOS Hg} = \text{NO, HgO}$. The experiment can never have been made, for I have assured myself that the reaction is a totally different one. When hyponitric vapours are passed into a balloon containing mercury and immersed in...



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