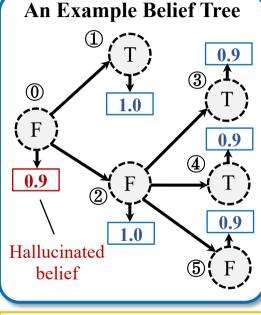
① At the top of Mount Everest, where the atmospheric pressure is much lower than at sea level, the freezing point of water is as low as -40°C.
 An Example Belief Tree
 ① The top of Mount Everest has lower



atmospheric pressure compared to sea level.

2) The freezing point of water can be as low as -40 °C at the top of Mount Everest.

- 3 Lower pressure can decrease the boiling point of water, not the freezing point4 The low pressure does not alter the
- determine its freezing point.

 (5) The freezing point of water can be as low as 2°C at the top of Mount Everest.

fundamental chemical properties of water that

-2°C at the top of Mount Everest.

Hidden variable (Truth value)

e) 0.9 Observed variable (confidence)