Quantum Mechanics Fall 2019

Yuxuan Zhang, XJTU, 2160909016

Important equations and constants

Schrödinger equation

$$\begin{split} i\overline{h}\frac{\partial}{\partial t}\Psi(x,t) &= -\frac{\overline{h}^2}{2m}\frac{\partial^2}{\partial x^2}\Psi(x,t) + V\Psi\\ \overline{h}\frac{\partial}{\partial t}\Psi &= H\Psi \end{split}$$

Hamiltonian operator

$$H = -\frac{\overline{h}^2}{2m}\nabla^2 + V$$

Week 1

Get into the quntum world

Notes on Lesson 1 Monday Sep 2

Notes on Lesson 2 Tuesday Sep 3

Bohr-Sommerfeld Quantization (Old quantum theory)

$$\oint p \, dq = nh$$

While p is the momentum of a particle, which is a function of position of particle q. This equation constrains the movement of a particle in a specific potential field V(q).

Notes on Lesson 3 Friday Sep 6

Born Statistical Interpretation of Wave Function

The wave function $\Psi(x,t)$ discribes the "Amplitude" of the probability finding the paticle at coordinate(x,t). While the probability density equals amplitude squared.

That is, the probability finding the particle among $x \in [a, b]$ at time t_0 is discribed as:

$$P(x \in [a, b], t = t_0) = \int_a^b |\psi|^2 dx = \int_a^b \psi^* \psi dx$$

Or, to say, the *Probability Density* can be written as:

$$p(x,t) = |\psi(x,t)|^2 = \psi^*(x,t)\psi(x,t)$$

Homework

[week1 hw1]

- Problem 1 [Thinking][Optional] A Brief History of Clues
- Problem 2 Caculate wave length of an electromn after a 1000V potential field
- Problem 3 Derivation of Klein-Golden equation

[week1 hw2]

- Problem 4 Griffiths 1.2
- (a) Find the standard deviation of the distribution in Example 1.1.
- (b) What is the probability that a photograph, selected at ramdom, would show a distance x more than one standard deviation away from the average?

• Problem 5 - Griffiths 1.4

At time t = 0 a particle is represented by the wave function

$$\Psi(x,0) = \begin{cases} A\frac{x}{a}, & if \ 0 \le x \le a, \\ A\frac{(b-x)}{(b-a)}, & if \ a \le x \le b, \\ 0, & otherwise, \end{cases}$$

where A, a and b are constants.

- (a) Normalize Ψ (that is, find A in terms of a and b)
- (b) Sketch $\Psi(x,0)$, as a function of x.
- (c) Where is the particle most likely to be found, at t = 0?.
- (d) What is the probability of finding the particle to the left of a? Check your result in the limiting cases b=a and b=2a.
- (e) What is the expectation value of x?

Week 2

Play with Schrödinger equation

Notes on Lesson 1 Monday Sep 9

Measurements and Operators

Starting with the most basic measurable value: expectation value of position $\langle x \rangle$

$$\langle x \rangle = \int x \, \rho(x) \, dx = \int \psi^* \, x \, \psi \, dx$$

Then consider $\langle x \rangle$ change with time:

$$\frac{d}{dt} < x > = \frac{d}{dt} \int \Psi^* x \, \Psi \, dx \tag{1}$$

$$= \int x \frac{\partial}{\partial t} |\Psi|^2 dx \tag{2}$$

$$= \int x \left(\frac{\partial \Psi^*}{\partial t} \Psi + \frac{\partial \Psi}{\partial t} \Psi^* \right) dx \tag{3}$$

$$SchrdingerEq \Rightarrow i\overline{h} \frac{\partial}{\partial t} \Psi = -\frac{\overline{h}^2}{2m} \frac{\partial}{\partial x^2} \Psi + V \Psi$$
 (4)

$$-i\overline{h}\,\frac{\partial}{\partial t}\Psi^* = -\frac{\overline{h}^2}{2m}\,\frac{\partial}{\partial x^2}\Psi^* + V\Psi^* \tag{5}$$

Conclude to
$$\Rightarrow \frac{\partial \Psi}{\partial t} \cdot \Psi^* = +\frac{i\overline{h}}{2m} \frac{\partial}{\partial x^2} \Psi \cdot \Psi^* + \frac{V}{i\overline{h}} \Psi \cdot \Psi^*$$
 (6)

$$\frac{\partial \Psi^*}{\partial t} \cdot \Psi = -\frac{i\overline{h}}{2m} \frac{\partial}{\partial x^2} \Psi^* \cdot \Psi - \frac{V}{i\overline{h}} \Psi^* \cdot \Psi \tag{7}$$

$$= \frac{i\overline{h}}{2m} \int x \cdot \left(\Psi^* \frac{\partial^2}{\partial x^2} \Psi - \Psi \frac{\partial^2}{\partial x^2} \Psi^* \right) dx \tag{8}$$

$$Mathtricks \Rightarrow (u\ddot{v} - v\ddot{u}) = \frac{d}{dx}(u\dot{v} - v\dot{u})$$
 (9)

$$= \frac{i\overline{h}}{2m} \int x \cdot \frac{\partial}{\partial x} \left(\Psi^* \frac{\partial}{\partial x} \Psi - \Psi \frac{\partial}{\partial x} \Psi^* \right) dx \tag{10}$$

(11)

(17)

Using integral by part $\Rightarrow u \cdot dv = uv - v \cdot du$

$$let \Rightarrow u = x \tag{12}$$

$$v = \Psi^* \frac{\partial}{\partial x} \Psi - \Psi \frac{\partial}{\partial x} \Psi^* \tag{13}$$

$$We \ get \Rightarrow$$
 (14)

$$x \cdot d\left(\Psi^* \frac{\partial}{\partial x} \Psi - \Psi \frac{\partial}{\partial x} \Psi^*\right) = u \cdot dv = wv - v \cdot du = -\left(\Psi^* \frac{\partial}{\partial x} \Psi - \Psi \frac{\partial}{\partial x} \Psi^*\right) dx \tag{15}$$

$$= -\frac{i\hbar}{2m} \int \left(\Psi^* \frac{\partial}{\partial x} \Psi - \Psi \frac{\partial}{\partial x} \Psi^* \right) dx \tag{16}$$

Using integral by part again $\Rightarrow u \cdot dv - v \cdot du = -uv + 2u \cdot dv$

$$let \Rightarrow u = \Psi^* \tag{18}$$

$$v = \Psi \tag{19}$$

$$We \ get \Rightarrow \Psi^* \partial \Psi - \Psi \partial \Psi^* = u \cdot dv - v \cdot du = \Psi + 2v \cdot du = 2 \Psi^* \partial \Psi = 2 \Psi^* \frac{\partial}{\partial x} \Psi dx$$
 (20)

$$= -\frac{i\overline{h}}{2m} \int 2 \cdot \Psi^* \frac{\partial}{\partial x} \Psi \, dx \tag{21}$$

$$= \int \Psi^* \left(-\frac{i\overline{h}}{m} \frac{\partial}{\partial x} \right) \Psi \, dx \tag{22}$$

Simply using
$$p = mv \Rightarrow = m \frac{d < x >}{dt}$$

$$= \int \Psi^* \left(-i \overline{h} \, \frac{\partial}{\partial x} \right) \Psi \, dx = \int \Psi^* \left(\frac{\overline{h}}{i} \cdot \frac{\partial}{\partial x} \right) \Psi \, dx$$