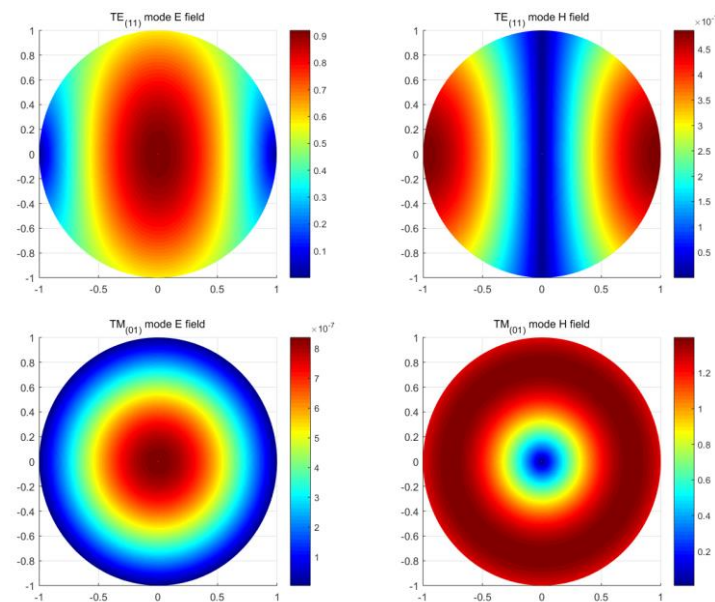
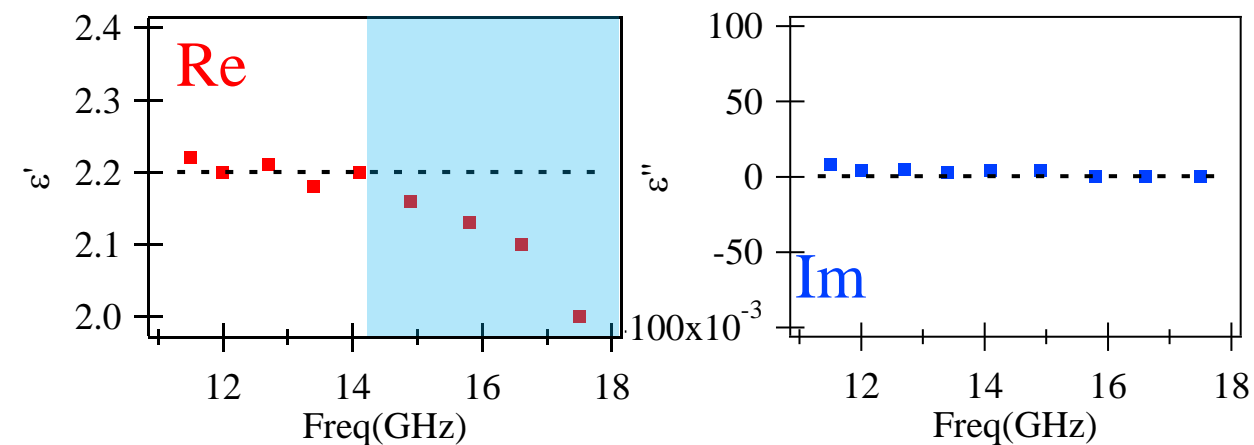


Dielectric constant of chemical water solutions

Yutong Zhao

Jan 21st 2019

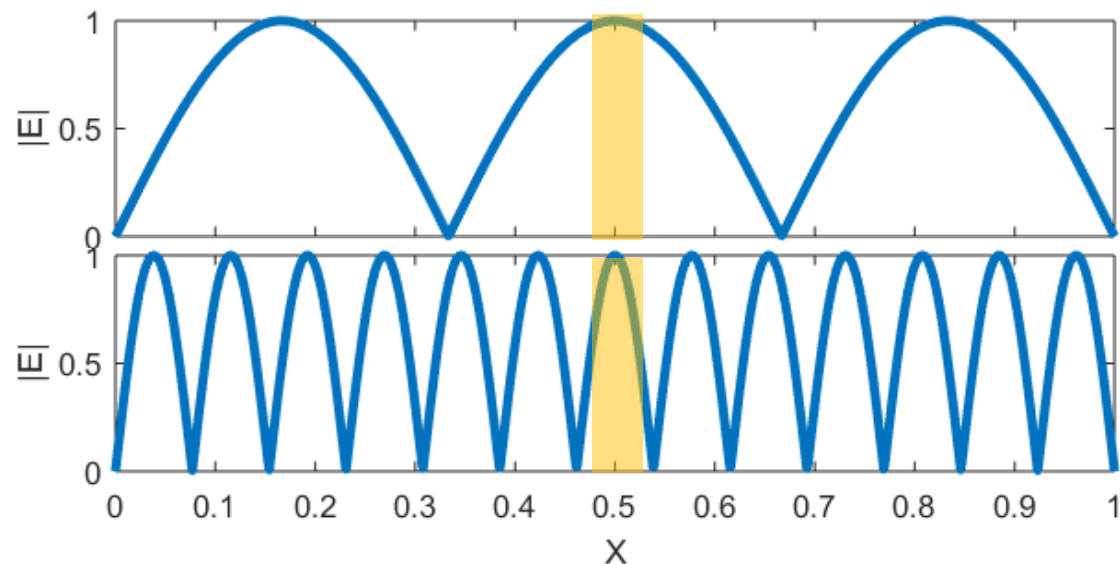
Resonant method: Yao's 1D cavity



$$(f_c)_{TE_{11}} = 10.92 \text{ GHz}$$

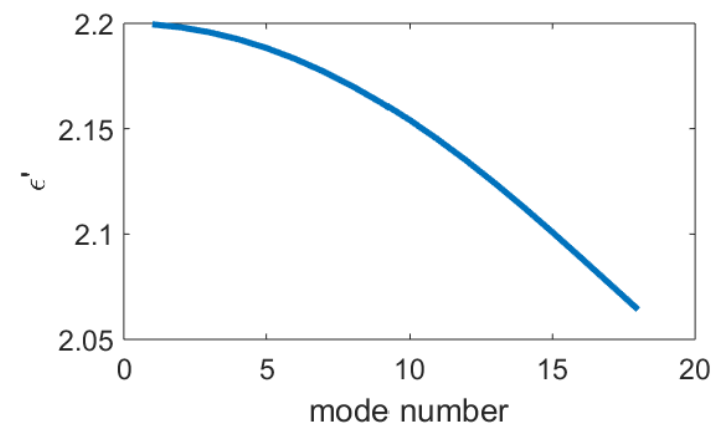
$$(f_c)_{TM_{01}} = 14.26 \text{ GHz}$$

The shift may be caused by the existence of TM_{01} mode



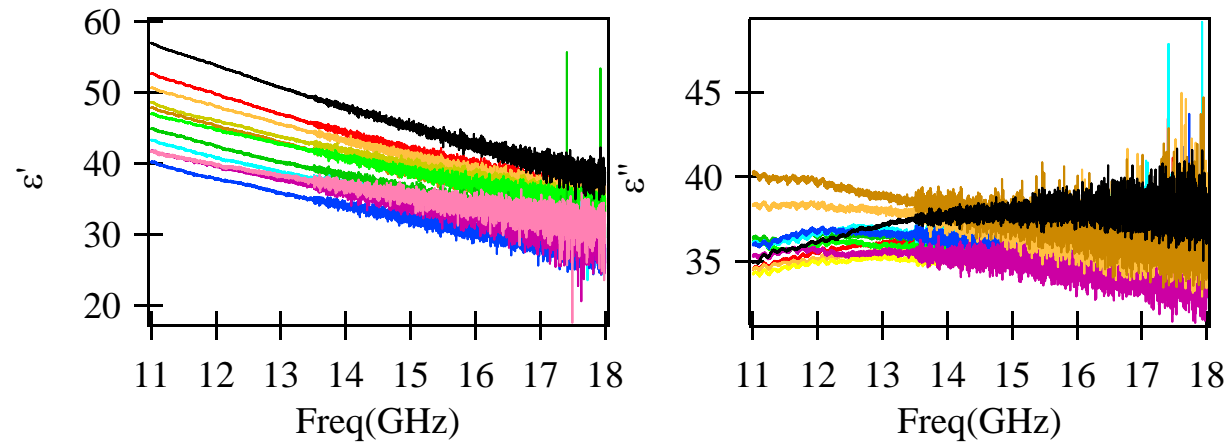
$$\frac{\delta f}{f_0} = \frac{\iiint \delta \epsilon |E|^2}{\iiint \epsilon_0 |E|^2}$$

$$|E| = \sin\left(\frac{\pi x}{L}\right)$$

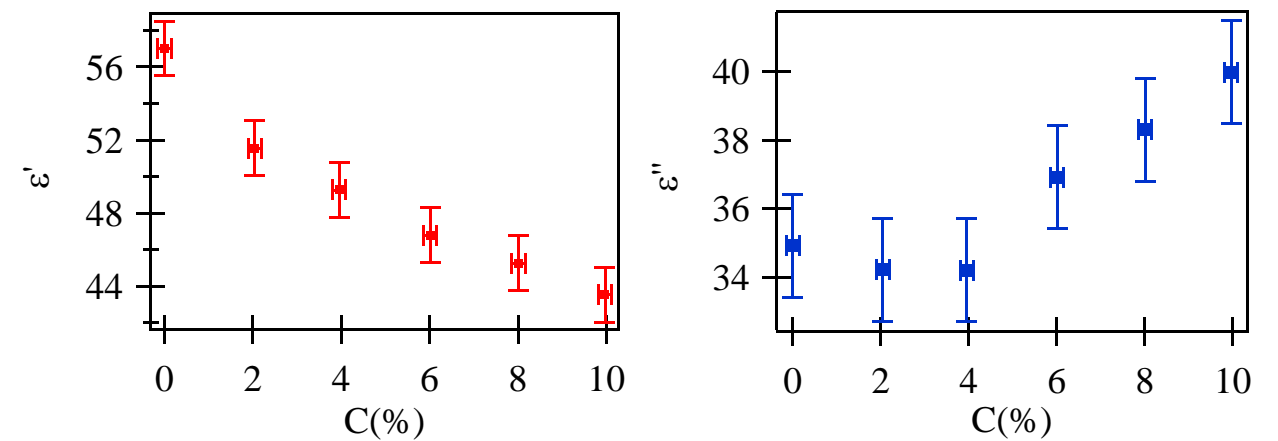
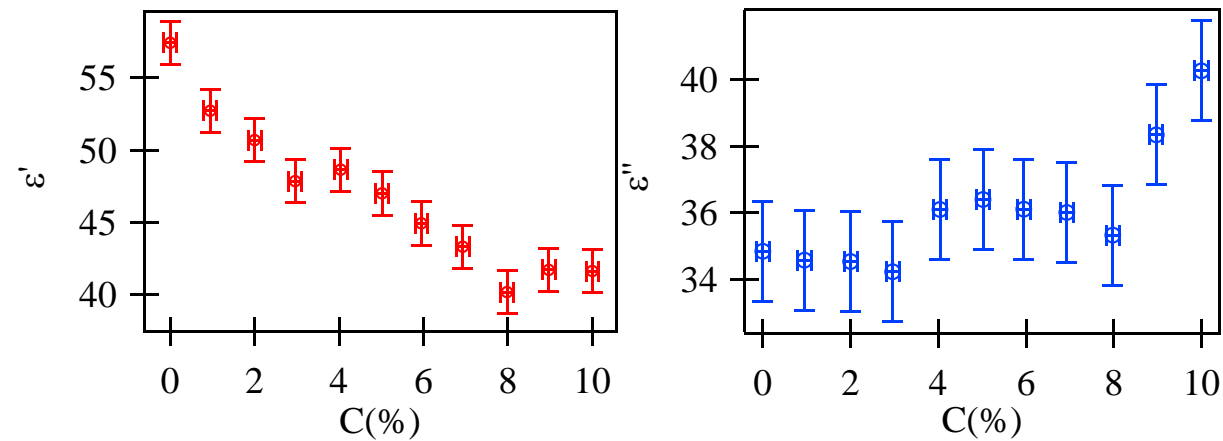
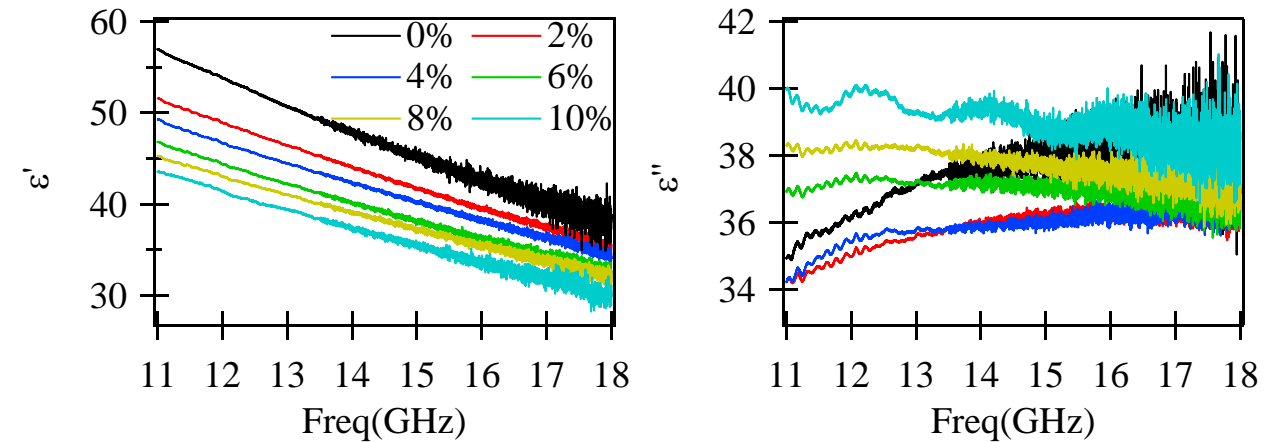


Dielectric constant of solutions

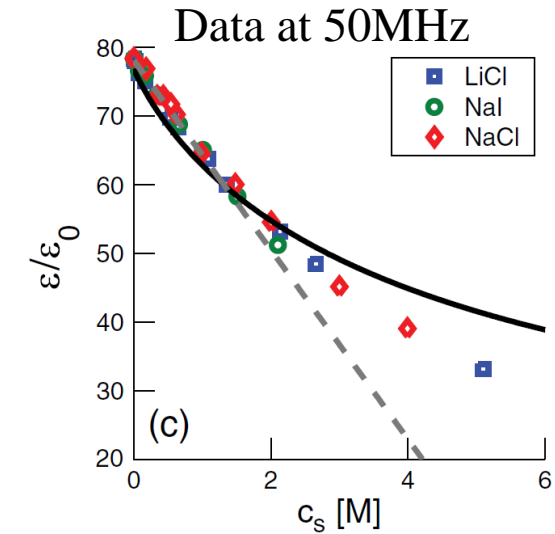
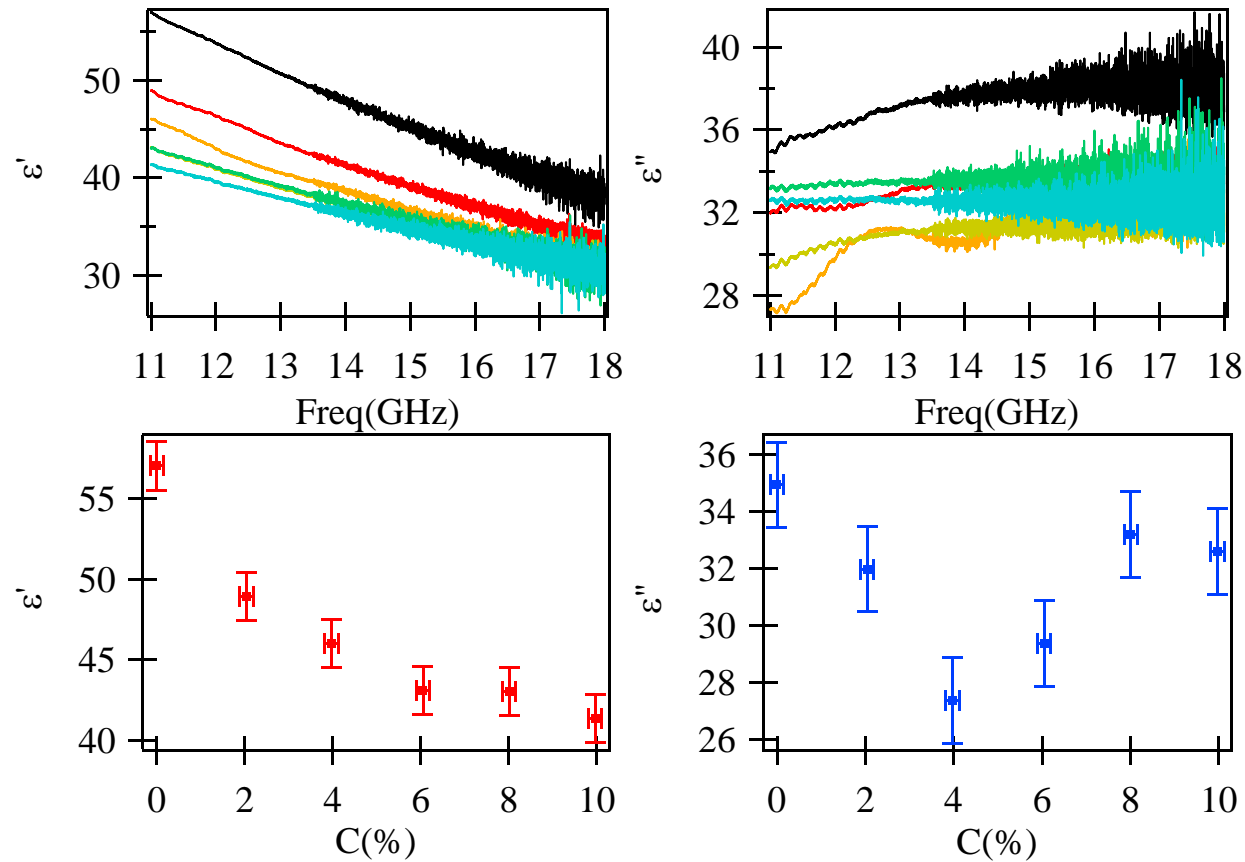
NaCl solution (1mL)



NH₄NO₃ solution (1mL)



NaNO₃ solution (1mL)



Summary:

Real part: decrease with concentration.

Imaginary part: Complicated behavior

Next step:

Collect the data of

The rest three chemical solutions.

KNO₃, KClO₃, sugar

Possible application on industry

- Composites innovation center
 - Composites Material Testing (dielectric measurement)
 - Sensor design (near field imaging)
- Magnetic insight
 - Magnetic particle imaging. (microwave imaging on magnetic material)
- MITACS
 - Project/funding application