# **Relation between Statistical Mechanics and Thermodynamics**

- $\langle E 
  angle_{NVT} = k_B T^2 rac{\partial \ln Z}{\partial T}|_{NV} \hat{=} U(N,V,T)$
- $k_B T rac{\partial \ln Z}{\partial V}|_{N,T} = p$
- $-k_B T \frac{\partial \ln Z}{\partial N}|_{V,T} = \mu$
- $ullet k_B T rac{\partial \ln Z}{\partial T}|_{NV} + k_B \ln Z = S$

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## Lecture 3

The partition function contains all information regarding a given thermodynamic state.

$$Z = \sum_i e^{-eta E_i}$$

- Sum over all quantum states (including degeneracy).
- Z=Z(N,V,T) (infinite heat bath)

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## Computing the trace in CM for a continuous system

- Let  $H=\sum_{i=1}^Nrac{p_i^2}{2m}+V(r_1,...,r_N)$  .
- $ullet \ Z_{NVT}^{CM} = rac{1}{h^{3N}N!} \int dp^{3N} dr^{3N} e^{-eta H(\{p_i\},\{r_i\})}.$

## **Normalisation**

- h: Historically introduced to make dimensions work, but its origin is in QM (Gibbs + Boltzmann couldn't find this).
- **N!**: Gibbs factor, corrects for particles to be indistinguishable. (This is not fully correct since Bose-Einstein/Fermi-Dirac statistics lead to other prefactors in different thermal limits).

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$$Z_{NVT}^{CM}=rac{1}{N!h^{3N}}\int dp^{3N}\exp[-eta\sum_irac{p_i^2}{2m}]\cdot\int dr^{3N}\exp[-eta V(r_1,...r_N)]$$

#### **Ideal Gas partition function**

- An ideal gas consists of point-like, non-interacting particles confined in a fixed volume.
- $H = \sum_{i} \frac{p_i^2}{2m_i}$  (assume same m)
- $ullet = \overline{Z} = rac{1}{N!h^{3N}} \int dp^{3N} \exp[-eta \sum_i rac{p_i^2}{2m}] \int_V d^{3N} r = rac{V^N}{N!h^{3N}} \int dp^{3N} \exp[-eta \sum_i rac{p_i^2}{2m}]$
- The momentum integral can be separated:  $\int d^{3N}p \exp[-\beta \sum_i \frac{p_i^2}{2m}] = \prod_{i=1}^N \int \exp[-\beta \frac{p^2}{2m}] d^3p = \prod_{i=1}^N 4\pi \int_0^\infty dp \ p^2 \exp[-\beta p^2/2m].$
- ullet Using the standard integral:  $\int_0^\infty x^2 \exp(-ax^2) dx = rac{\pi^{1/2}}{4a^{3/2}}.$

## Page 3

•  $J=-k_BT\ln Z_{\mu VT}=ho V$  (Massieu) for the grand canonical ensemble.

## **Evaluating the Partition Function**

- $Z=\sum_i e^{-eta E_i}$
- More correctly,  $Z=tr(e^{-eta\hat{H}})$  (true for both classical and quantum mechanics).

# Computing the trace in QM

- $tr(\hat{ heta}) = \sum_i \langle \psi_i | \hat{ heta} | \psi_i 
  angle$
- Also,  $\hat{H}|\psi_i
  angle=E_i|\psi_i
  angle\Rightarrow e^{-eta\hat{H}}|\psi_i
  angle=e^{-eta E_i}|\psi_i
  angle.$
- ullet (Exercise: show that this is true). This allows recovery of  $Z=\sum_i e^{-eta E_i}.$

## Page 2

- $-k_BT\ln Z=F(N,V,T)$  (Helmholtz free energy)
- Exercise: prove these.
- It holds that F = U TS.
- $dF = -SdT p dV + \mu dN$
- Knowing Z as a function of N, V, T defines the thermodynamic state of the system exactly.
- Since  $Z = Z(\{E_i(N,V)\},T)$ , knowing all quantum states yields the thermodynamics.

## **Note: Thermodynamic Potentials**

- $F=-k_BT\ln Z_{NVT}$  (Helmholtz) for the canonical ensemble.
- $G=-k_BT\ln Z_{NPT}$  (Gibbs) for the isothermal-isobaric ensemble.
- $S = +k_BT \ln Z_{NVE}$  (entropy) for the microcanonical ensemble.

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- $ullet \ \Rightarrow Z = rac{V^N}{N!} (rac{1}{\lambda_T})^{3N}$
- Where  $\lambda_T=rac{h}{[2\pi mk_BT]^{1/2}}$  is the thermal de Broglie wavelength.

# Motivating Monte Carlo Statistical Sampling 🎲

- How to compute  $Z=tr(e^{-\beta\hat{H}})$ ?
- It is often not feasible to determine all eigenstates of  $\hat{H}$  or to integrate analytically over all of phase space.
- Not all of phase space is relevant. Many regions are exponentially suppressed and thus don't contribute to Z.

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The probability of a state is given by  $\frac{\exp(-eta H)}{Z}$  .

# **Importance Sampling**

- The expectation value of an observable heta is  $\langle heta 
  angle = \int heta(r,p) P(r,p) dr^{3N} dp^{3N} pprox rac{1}{M} \sum_{i=1}^M heta(r_i,p_i).$
- This approximation holds if the M configurations  $(r_i,p_i)$  are drawn from the distribution P(r,p).
- $\langle heta 
  angle = \lim_{M o \infty} rac{1}{M} \sum_{i=1}^M heta(r_i, p_i)$
- ullet Recall  $P(r,p)=rac{\exp[-eta H(r,p)]}{Z}$  , where  $Z=\int \exp(-eta H)drdp$  .
- Note that (a)  $\exp(-eta H(r,p))$  is relatively easy to compute, but (b) Z is difficult to compute.

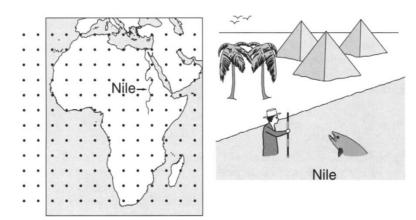
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**Key point in Metropolis Monte Carlo**: It allows us to draw samples from a probability distribution  $P(\underline{x})$  when we only know the distribution up to a constant, i.e.,  $h(\underline{x})=cP(\underline{x})$ .

- It relies on the fact that the ratio of probabilities can be computed without knowing the constant:  $\frac{h(x_1)}{h(x_2)} = \frac{p(x_1)}{p(x_2)}$ .
- **Downside**: The generated samples are no longer independent.

Figure 3.1: Measuring the depth of the Nile: a comparison of conventional quadrature (left), with the Metropolis scheme (right).

- The left image shows a map of Africa and the Middle East with a uniform grid of sample points laid over it.
- The right image is a cartoon depicting a more targeted sampling method, with a person sampling in the Nile river itself, near pyramids.



**Figure 3.1:** Measuring the depth of the Nile: a comparison of conventional quadrature (left), with the Metropolis scheme (right).