

Organic Materials for Energy and Optoelectronics

Overview of Organic Semiconductors

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February 2, 2022

Outline

- Introduction: real world examples
- What is π -conjugated system
- Examples and Applications
- Structure, Functional properties, Challenges
- Research at Skoltech
- Resources: references, links, further reading

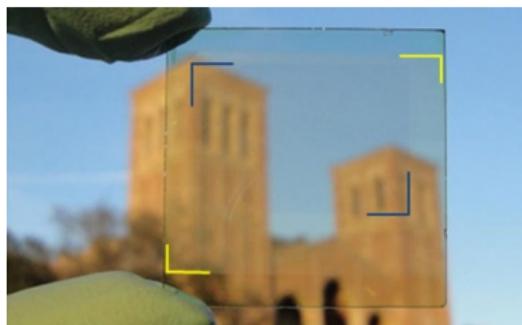
Real world examples: OLEDs

Commercially most successful application of organic semiconductors, \$30 billion in 2019

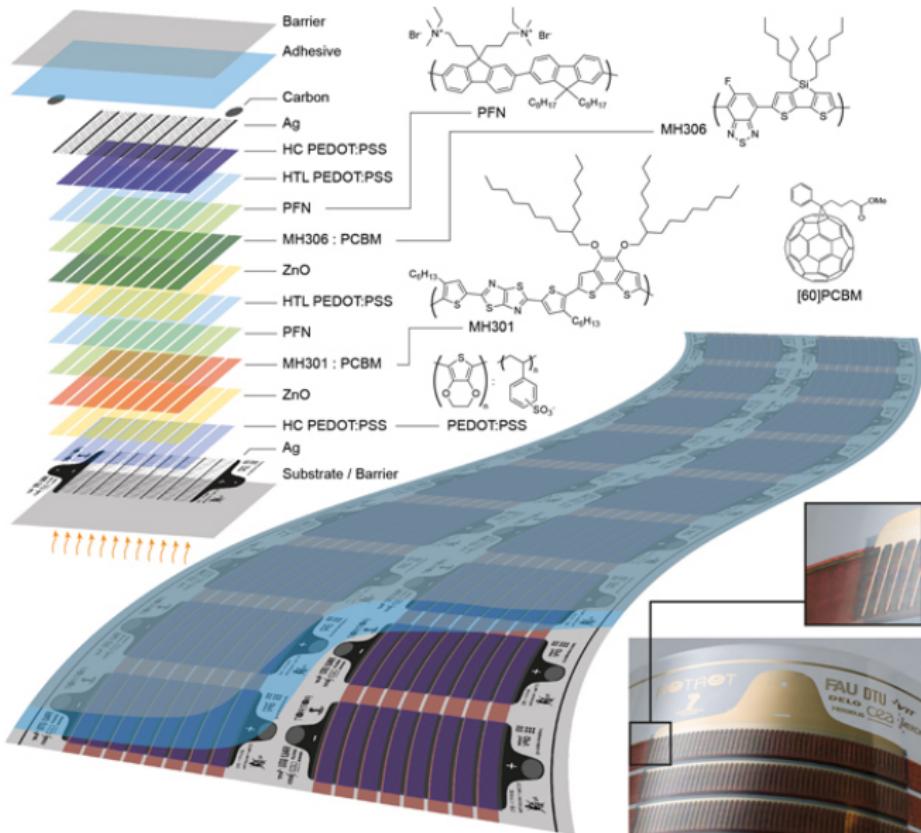


Real world examples: Solar cells

Largest PCE progress: from 4 to 18% compared to 13-26 for Si and 15-25 for perovskites

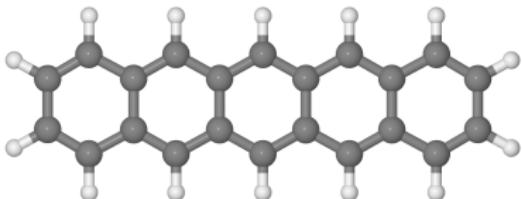


How is it made: Solar cells – R2R roll-to-roll processing

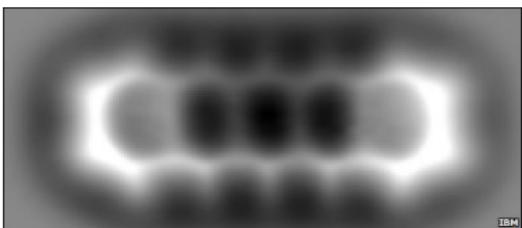
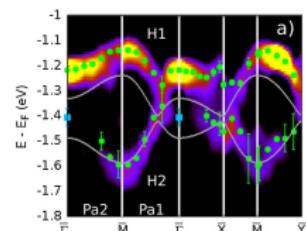
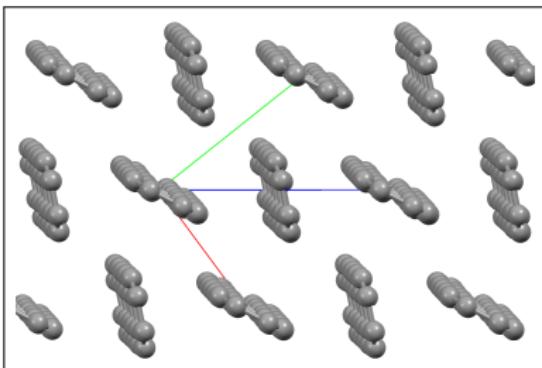


Historical example: A century-long story of pentacene

Charge carrier mobility $\sim 10 \frac{\text{cm}^2}{\text{V}\cdot\text{s}}$
($>$ a-Si, 1/100 of c-Si)

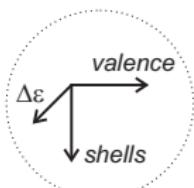


- Synthesized in 1920s
- Crystal resolved in 1960s
- Used in OFET since 1980s
- Good-quality crystals in 2000s
- Bands resolved in \sim 2010
- Atoms resolved in 2009
- Solution-processable TIPS-P 2007



What is organic semiconductor

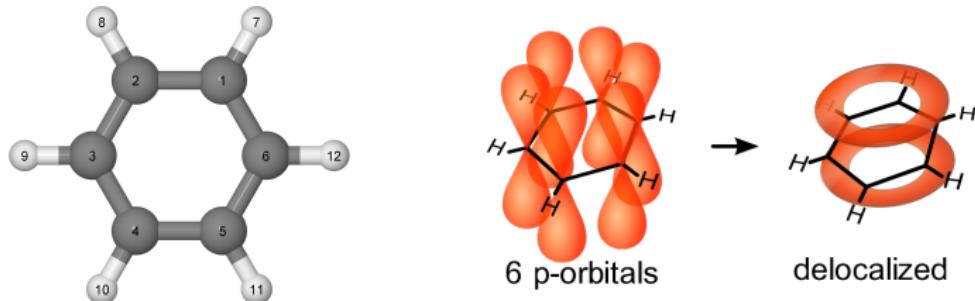
1. **Organic** – built from C with terminal H, possibly with isovalent substitutions (N for CH, O/S for CH₂, F/Cl for H)
2. **π -conjugated** – π -electrons on frontier orbitals (next slide)



(1)	(2)			(3)	(4)	(5)	(6)	(7)	(8)
H									He
Li	Be			B	C	N	O	F	Ne
Na	Mg		d-shell	Al	Si	P	S	Cl	Ar
K	Ca		f-shell	Ga	Ge	As	Se	Br	Kr
Rb	Sr		Sc V Mn Co Cu Ti Cr Fe Ni Zn Y Nb Tc Rh Ag Zr Mo Ru Pd Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La-Yb	Lu Ta Re Ir Au Hf W Os Pt Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac-No							strong relativistic effects

Number of organic semiconductors is comparable to number of inorganic ones

What is π -conjugated system: example of benzene



- each carbon has $3\ sp^2$ AOs connected by σ -bonds and 1 $pp\pi$ -connected AO
 - energy of bonding MO is $-t$ per electron
 - $t_{sp2} = 3.26 \frac{\hbar^2}{md^2} \gg t_{pp\pi} = 0.63 \frac{\hbar^2}{md^2}$ [Harrison]
- \implies
- π -conjugated electronic system is separated from sp^n -MOs (by 10 eV in terms of NBO energies)
 - primary bonding is sp^n , modulated by $pp\pi$ (secondary bond)

What elements can participate in π -conjugation

The diagram illustrates the periodic table with several highlighted regions:

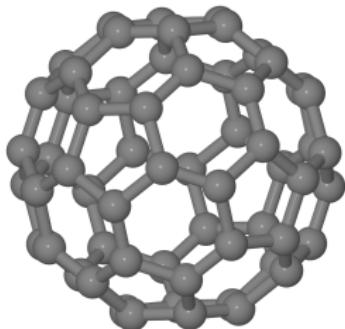
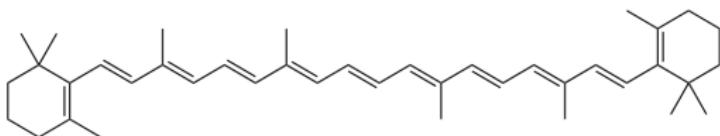
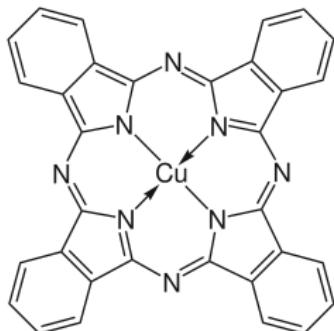
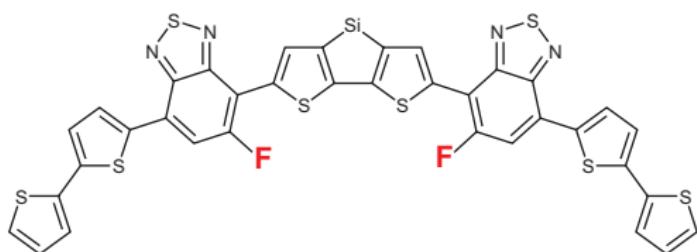
- Valence shells:** A dashed circle labeled "valence" encompasses groups 13-18 (B, C, N, O, F, Ne), group 1 (H), and group 2 (Li, Be).
- d-shell:** A dashed circle labeled "d-shell" encompasses Sc, V, Mn, Co, Cu, Ti, Cr, Fe, Ni, Zn.
- f-shell:** A dashed circle labeled "f-shell" encompasses Y, Nb, Tc, Rh, Ag, Zr, Mo, Ru, Pd, Cd.
- Strong relativistic effects:** A dashed circle labeled "strong relativistic effects" encompasses Lu, Ta, Re, Ir, Au, Hf, W, Os, Pt, Hg, Tl, Pb, Bi, Po, At, Rn.

Numbered circles (1-8) are placed above the first two columns (H, Li/Be) and the last two columns (He/F/Ne). The first two columns are grouped under ① and ②, while the last two columns are grouped under ③ through ⑧.

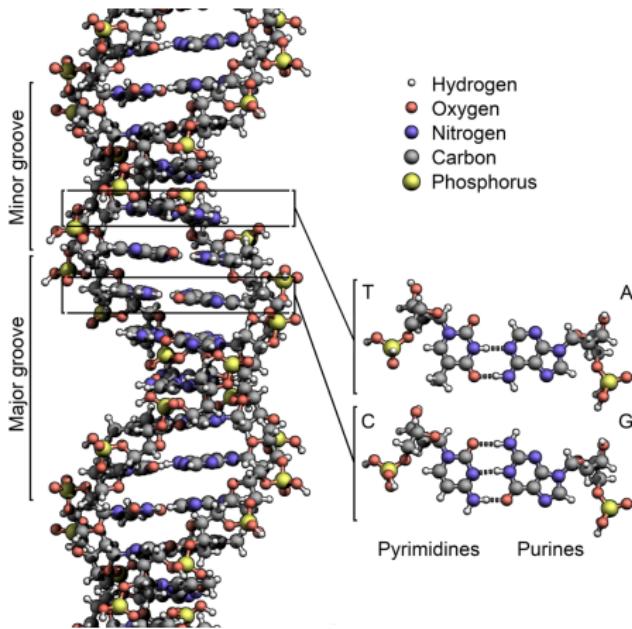
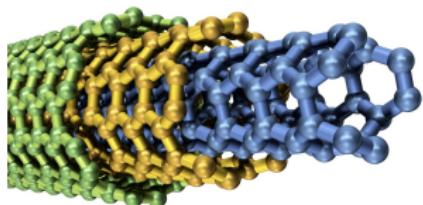
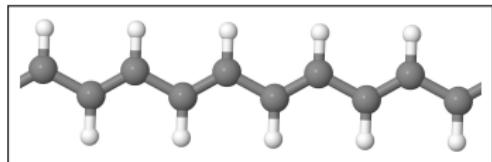
①	②								
H							He		
Li	Be								
Na	Mg								
K	Ca								
Rb	Sr	f-shell	Sc V Mn Co Cu Ti Cr Fe Ni Zn	Ga	Ge	P	S	Cl	Ar
Cs	Ba		Y Nb Tc Rh Ag Zr Mo Ru Pd Cd	In	Sn	Sb	Te	I	Xe
Fr	Ra	La-Yb	Lu Ta Re Ir Au Hf W Os Pt Hg	Tl	Pb	Bi	Po	At	Rn
		Ac-No		strong relativistic effects					

- C, N, B as building blocks
- O, F, S, Cl via LP and terminal contacts
- d -elements via $pd\pi$
- any electronic system in resonance

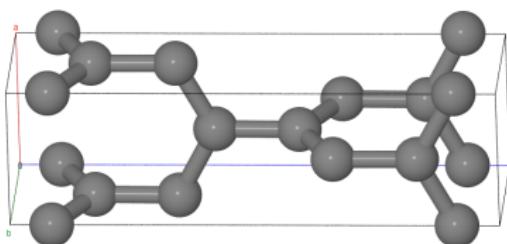
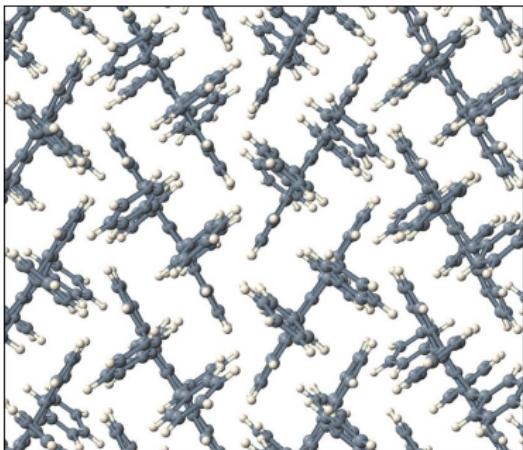
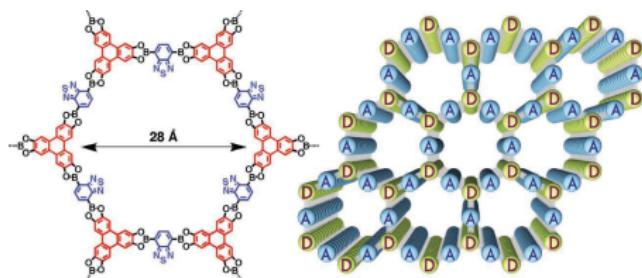
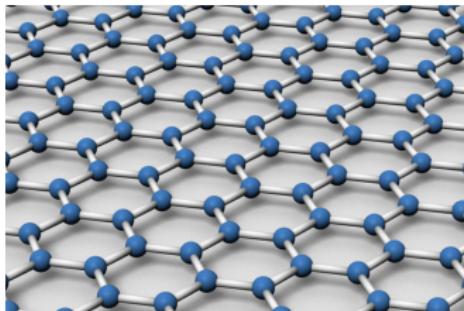
Examples of π -conjugated systems: molecules



Examples of π -conjugated systems: polymers and 1D



Examples of π -conjugated systems: 2D and 3D



Currently we explore only a small set of possible structural forms

Applications: Organic electronics

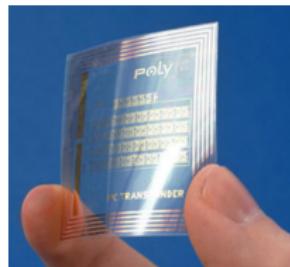
Solar cells



Light emitters

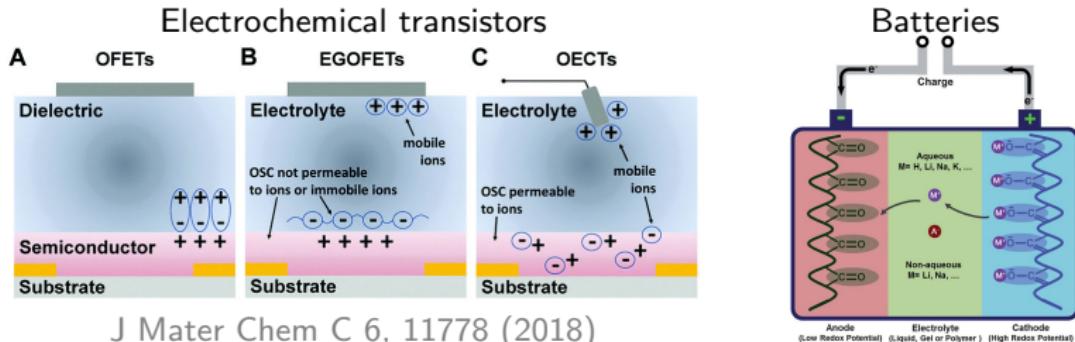


Field effect transistors

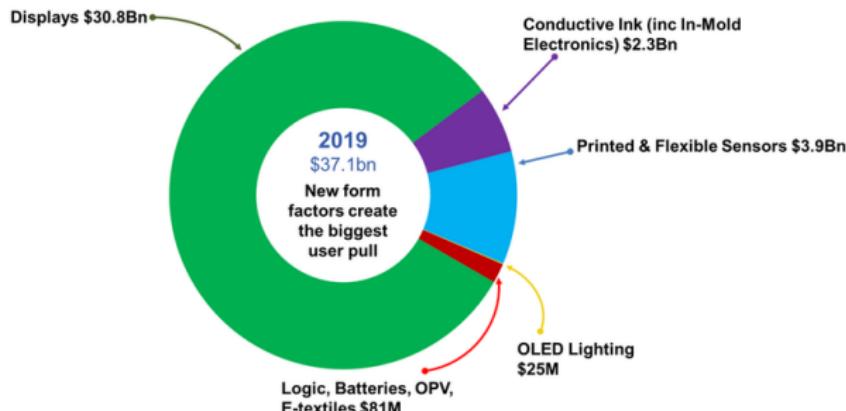


- Unlimited possibilities of nanoscale molecular engineering
- Cost effective solution – perfect for consumer markets
- Ease of production – dominates printed electronics market
- Ease of recycling – green technology
- Light weight and flexibility – more versatile in use

Other applications



2019 Market Snapshot



More examples and applications

- UV-protection of DNA
- Vision, bioluminescence, sunlight harvesting
- Fluorescent probes for bioimaging
- Charge transport layers
- Capacitors (PEDOT/PSS) and supercapacitors
- Chemical sensors
- Catalysts (transition metal complexes, e.g. water splitting)
- Membranes for water desalination
- THz generators, electrochromic devices, photodiodes, polyelectrolytes, explosives, piezo-, pyro-, ferroelectrics, ...

Any electronic device can be made all-organic

Discussion

“Debunking” myths

1. Are organic semiconductors cheap?
2. Are organic semiconductors environmentally friendly?
3. Are organic semiconductors poor-performers?

Looking for applications

4. What organic materials have industry-scale applications
5. Organic batteries: major problems and possible solutions
6. Short operational lifetime: when it is problem and when is not

Lessons for “startups”

7. How Konarka bankrupted
8. How solar cell research moved to Asia

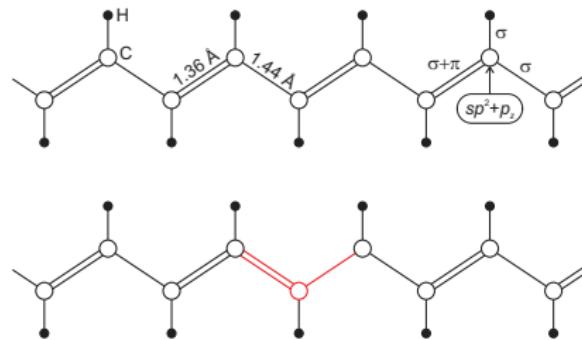
Why π -conjugated molecules are special?

Strong mode-specific electron-phonon coupling

What is electron-phonon coupling?

- piezoelectric coupling: mechanical stress \leftrightarrow electric field
- e-phonon coupling: molecular deformations \leftrightarrow electronic levels

Why electron-phonon coupling in π -conjugated systems is special?



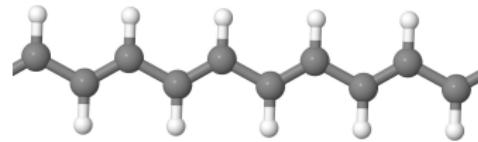
⇒ Unique quasiparticles – solitons in transpolyacetylene

Why π -conjugated molecules are special?

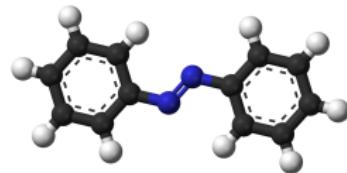
Strong electron-phonon coupling in combination with soft structure

⇒ *Electronic properties strongly depend on material morphology, and electronic dynamics is strongly bound to molecular dynamics*

Peierls transition in polyacetylene
(C–C bond stretching mode)



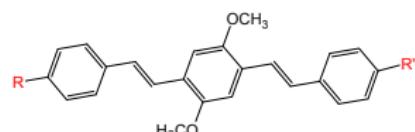
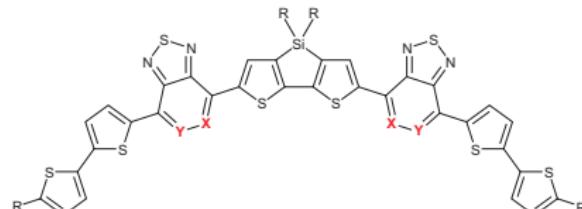
Photoisomerization of azobenzene
(librations of non-rigid dihedrals)



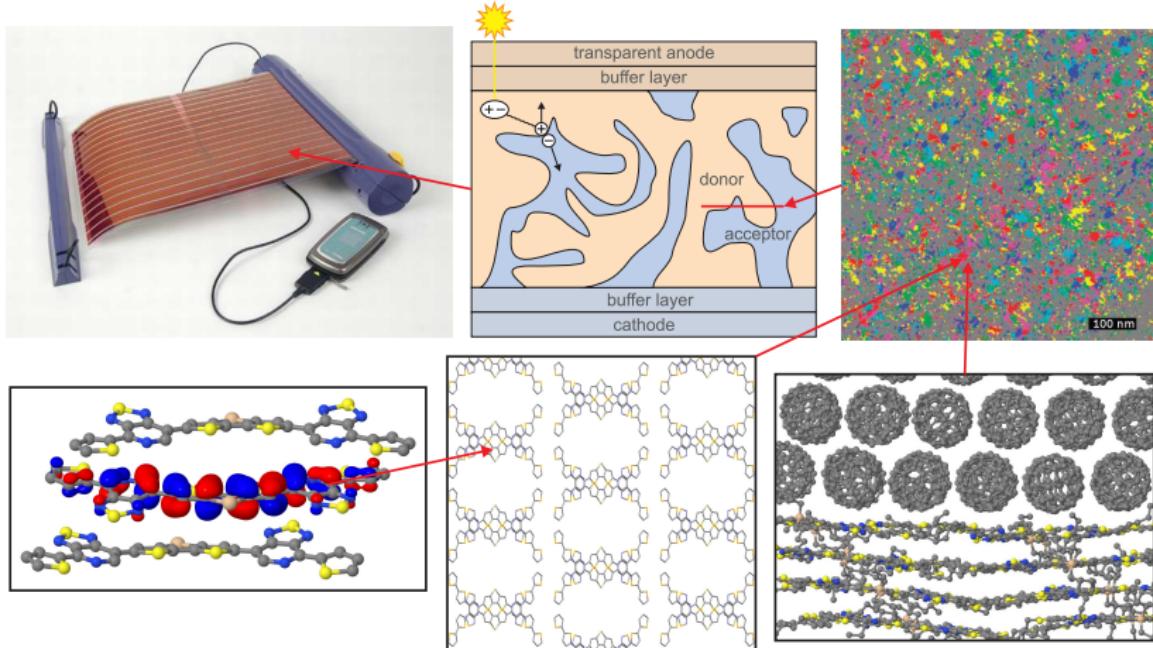
also polaron formation, vibronic progression in spectra

In fact, it is very useful in applications:

- Change structure ⇒ tune electronic properties
- Affect electronic system ⇒ change structure



Scale problem in organics: solar cell example



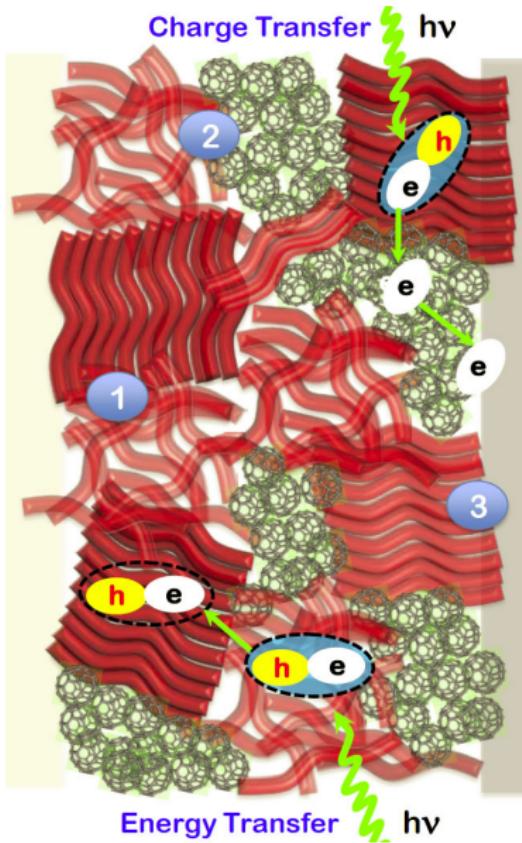
Understanding scales: solar cell example

Spatial scales:

- molecule ($\lesssim 1$ nm)
- single phase (~ 10 nm)
- interfaces (intra and inter)
- functional layer ($\gtrsim 100$ nm)

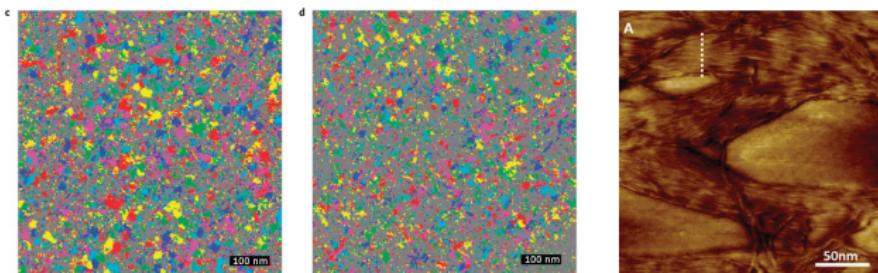
Time scales:

- ultrafast intramolecular (fs)
- intermolecular transfer (ps)
- electronic transport (ns)
- transients, degradation ($>ns$)



Understanding scales: computational perspective

Functional properties of organic semiconductors are often determined by structure on scales up to tens of nm
⇒ atomistic description is needed for up to 10^6 atoms



Challenges

- realistic mesoscale structure (coarse-grained MD)
- accurate intermolecular geometry (best DFT-D)
- accurate electronic structure (best range-separated hybrids)
- accurate charge dynamics (best NAMD)

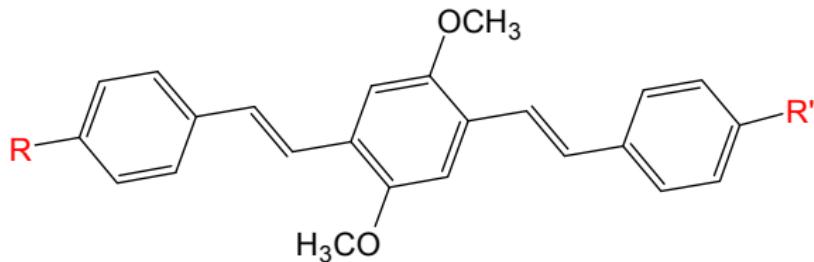
No direct simulation – only multiscale modeling and “machine learning”

Discussion

1. Are there other classes of materials with “split” electronic structure (σ -bonds and π -system)?
2. Are there other classes of materials with similar electron-phonon interaction effects?

Structural motifs

- Locally 2D due to nature of π -conjugation
- Bond length alternation (BLA) pattern
- Rigid fused rings + floppy dihedrals + vdW contacts
- Usually functionalized (tuning, soluble side-chains, transition metals, bridges)

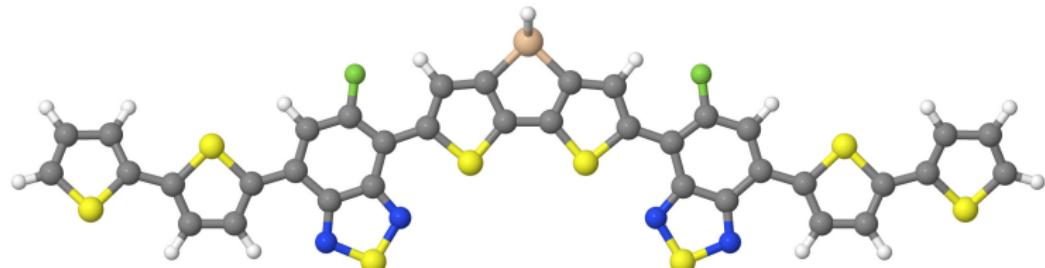


Structural classes

- Graphene, carbon nanotubes
- Conjugated polymers
- Oligomers
- Small-molecule crystals
- All-organic frameworks
- Metal organic frameworks
- Biopolymers with aromatic fragments and macromolecules
- Blends
- Strongly correlated systems

Rational design: Quasi-1D π -conjugated systems

(majority of materials used in organic electronics)



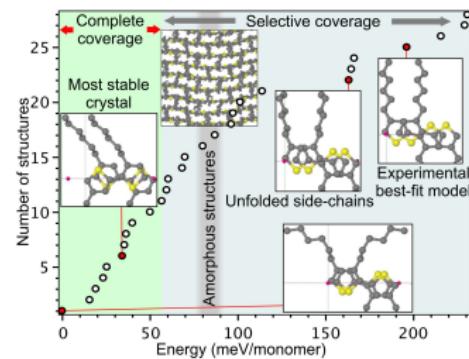
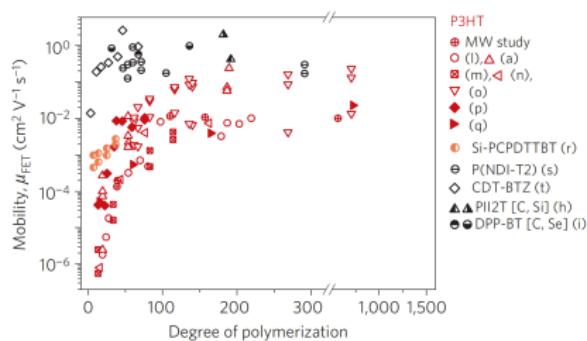
- Have block structure with few interconnections per block
- Each block is rigid, limited number of local structural patterns

⇒ *Success of simple force fields*

- The π -conjugated system of each block is closed-shell
- Inter-block couplings $\sim 1 \text{ eV} \ll$ bandgap of blocks
- Intermolecular couplings $\sim 0.1 \text{ eV} \ll$ bandgap of molecules

⇒ *Can be rationally designed by block assembly approach*

Rational design: Improving morphology



- Isovalent substitutions and “side chain engineering”
- Processing conditions, e.g. additives

Challenges

- Fast degradation and aging – mechanisms to be studied
- Large batch-to-batch variations – technology
- Mediocre performance (5% of elements) – hybrids
- Limited charge carrier mobility – improving
- Doping is nontrivial – solvable
- Too complex to characterize/describe – consider it as excellent research opportunity for you

Research at Skoltech and in Russia

Theoreticians:

- Sergei Tretiak, Andriy Zhugayevych – see also [this page](#)

Experiment:

- Keith Stevenson (CEST CREI)
- Albert Nasibulin (CPQM CREI) – carbon nanotubes
- Pavel Troshin (IPCP)
- Sergey Ponomarenko (ISPM)
- Dmitry Paraschuk (MSU)

Events:

- IFSOE (International Fall School on Organic Electronics)

Resources

- Wikipedia
- List of references
- A Koehler, H Bassler, Electronic Processes in Organic Semiconductors: An Introduction (Wiley, 2015) *in library*
More specifically Sections 1.1-1.4
- S R Forrest, Organic Electronics: Foundations to Applications (OUP, 2020)
More specifically Chapter 1