
ECE 291 HW 3 Problem 1

Determines the mass of CO₂ produced per MJ in various reactions

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%Johnathan Schaff
clc; clear; close all;
n = [2,3,4,5,6,7,8];%mole ratios from CO2 to (ethane:octane)
moleWeight = [30,44,58,72,86,100,114];%ethane to octane
moleWeightCO2 = 44;
energyWeight = [51.9, 50.3, 49.5, 48.7, 48.1,48.1, 46.8];%ethane to
octane
solutionVector = ((moleWeight.*energyWeight)./n)/moleWeightCO2;
for j = 1: length(solutionVector)
    fprintf('n | %d Mass per MegaJoule | %f grams\n',n(j), (1000/
solutionVector(j)));
end

n | 2 Mass per MegaJoule | 56.518947 grams
n | 3 Mass per MegaJoule | 59.642147 grams
n | 4 Mass per MegaJoule | 61.302682 grams
n | 5 Mass per MegaJoule | 62.742414 grams
n | 6 Mass per MegaJoule | 63.820529 grams
n | 7 Mass per MegaJoule | 64.033264 grams
n | 8 Mass per MegaJoule | 65.976908 grams
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