Density Math Practice

$$D = \frac{m}{V}$$

 $1.\,$ A rectangular block has a mass of 300 grams and a volume of 145 mL. Calculate the density of the block using the density formula.

Solution:

$$Density = \frac{Mass}{Volume} = \frac{300\,\mathrm{g}}{145\,\mathrm{mL}} \approx 2.07\,\mathrm{g/mL}$$

2. A substance has a density of $0.6~\mathrm{g/mL}$ and occupies a volume of $60~\mathrm{mL}$. Calculate the mass of this substance using the density formula.

Solution:

$$Mass = Density \times Volume = 0.6 \text{ g/mL} \times 60 \text{ mL} = 36 \text{ grams}$$

3. A gas has a density of $0.4~\mathrm{g/mL}$ and a mass of 160 grams. Calculate the volume of the gas using the density formula.

Solution:

$$Volume = \frac{Mass}{Density} = \frac{160\,\mathrm{g}}{0.4\,\mathrm{g/mL}} = 400\,\mathrm{mL}$$

$$D = \frac{m}{V}$$

4. An irregularly shaped object has a mass of 210 grams and a volume of 120 mL. Calculate the density of the object using the density formula.

Solution:

$$Density = \frac{Mass}{Volume} = \frac{210\,\mathrm{g}}{120\,\mathrm{mL}} \approx 1.75\,\mathrm{g/mL}$$

5. Another substance, with a density of 1.2 g/mL, occupies a volume of 85 mL. Calculate the mass of this substance using the density formula.

Solution:

$$Mass = Density \times Volume = 1.2 g/mL \times 85 mL = 102 grams$$

6. A metal sphere has a density of $4.3~\mathrm{g/mL}$ and a mass of $200~\mathrm{grams}$. Calculate the volume of the metal sphere using the density formula.

Solution:

$$Volume = \frac{Mass}{Density} = \frac{200\,\mathrm{g}}{4.3\,\mathrm{g/mL}} \approx 46.51\,\mathrm{mL}$$