Physics 112 - Homework 5 Zooey Nguyen zooeyn@ucla.edu May 18, 2021

Physics 112 Homework 5

Question 1.

Number of photons in cavity of volume $V = L^3$.

$$\begin{split} N &= \sum_{n} \langle s_{n} \rangle \\ &= \sum_{n} \frac{1}{e^{E_{n}/\tau} - 1} \\ &= \sum_{n} \frac{1}{e^{\hbar\omega_{n}/\tau} - 1} \\ &= \sum_{n_{e}, n_{y}, n_{z}} \frac{1}{e^{\hbar\omega_{n}/\tau} - 1} \\ &= \sum_{n_{e}, n_{y}, n_{z}} \frac{1}{e^{\hbar\omega_{n}/\tau} - 1} \, \mathrm{d}n \\ &= \frac{3}{8} \int_{0}^{\infty} \frac{4\pi n^{2}}{e^{\hbar\omega_{n}/\tau} - 1} \, \mathrm{d}n \\ &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} \int_{0}^{\infty} \frac{x^{2}}{e^{x} - 1} \, \mathrm{d}x \\ &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} \left[-x^{2}e^{-x}|_{0}^{\infty} + \int_{0}^{\infty} 2xe^{-2x} \, \mathrm{d}x\right] \\ &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} \left[-x^{2}e^{-x} - 2xe^{-x} - 2e^{-x}\right]_{0}^{\infty} \\ &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} \left[-e^{-x}(x^{2} + 2x + 2)\right]_{0}^{\infty} \\ &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} \left[-e^{-x}(x^{2} + 2x + 2)\right]_{0}^{\infty} \\ &= \lim_{x \to 0} -\frac{2x + 2}{e^{x}} \\ &= \log n \\ N &= \frac{3\pi}{2} \left(\frac{\tau L}{\hbar\pi c}\right)^{3} (2) \\ &= \frac{3L^{3}}{\pi^{2}} \left(\frac{\tau}{\hbar c}\right)^{3} \\ &= \left[\frac{3V}{\pi^{2}} \left(\frac{\tau}{\hbar c}\right)^{3}\right] \end{split}$$

I'm off by a little bit of the coefficient... maybe it had to do with the number of polarisations.

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Question 2.

Pressure of a photon gas and so on.

$$\begin{split} U &= \sum_{j} s_{j} \hbar \omega_{j} \\ p &= -\frac{\partial U}{\partial V} = -\sum_{j} \frac{\partial s_{j} \hbar \omega_{j}}{\partial V} = \boxed{-\sum_{j} s_{j} \hbar \frac{\mathrm{d}\omega_{j}}{\mathrm{d}V}} \\ \frac{\mathrm{d}\omega_{j}}{\mathrm{d}V} &= \frac{\mathrm{d}}{\mathrm{d}V} \frac{n\pi c}{L} = \frac{\mathrm{d}}{\mathrm{d}V} \frac{j\pi c}{L} = \frac{\mathrm{d}}{\mathrm{d}V} \frac{j\pi c}{V^{1/3}} = -\frac{j\pi c}{3V^{4/3}} = -\frac{j\pi c}{L} \frac{1}{3V} = \boxed{-\frac{\omega_{j}}{3V}} \\ p &= -\sum_{j} s_{j} \hbar \frac{\mathrm{d}\omega_{j}}{\mathrm{d}V} = \sum_{j} s_{j} \hbar \frac{\omega_{j}}{3V} = \frac{1}{3V} \sum_{j} s_{j} \hbar \omega_{j} = \boxed{\frac{U}{3V}} \end{split}$$

Kinetic pressure p_k and thermal radiation pressure p of gas of H atoms.

$$p_k = \frac{N}{V}\tau = (1 \text{ mol/cm}^3)(6.022 \times 10^{23} / \text{mol})(1 \times 10^6 \text{ cm}^3 / \text{m}^3)\tau$$

$$= \left[1.66 \times 10^{14} \text{ N/m}^2\right]$$

$$p = \frac{U}{3V} = \frac{1}{3} \frac{8\pi^2(\tau)^4}{15h^3c^3} = \left[4.03 \times 10^{13} \text{ N/m}^2\right]$$

Temperature for which the pressures are equal.

$$(6.022 \times 10^{29} / \text{m}^3)\tau = \frac{8\pi^2 \tau^4}{45h^3 c^3}$$
$$\tau = hc \left(\frac{45}{8\pi^2} (6.022 \times 10^{29} / \text{m}^3)\right)^{1/3}$$
$$= 1.39 \times 10^{-15} \text{ J}$$
$$T = \boxed{1.01 \times 10^8 \text{ K}}$$

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Question 3.

Heat capacity of one-dimensional EM wave/photons. Note the solution to the one-dimensional wavefunction is $E = E_0 \sin \frac{n\pi x}{L} \cos \frac{n\pi vt}{L}$.

$$C_{v} = \frac{\partial U}{\partial \tau}$$

$$U = \sum_{j} \frac{\hbar \omega_{j}}{e^{\hbar \omega_{j}/\tau} - 1}$$

$$= \sum_{n} \frac{\hbar n\pi v/L}{e^{\hbar n\pi v/L\tau} - 1}$$

$$= \int_{0}^{\infty} \frac{\hbar n\pi v/L}{e^{\hbar n\pi v/L\tau} - 1} dn$$

$$= \frac{L\tau^{2}}{\hbar \pi v} \int_{0}^{\infty} \frac{x}{e^{x} - 1} dx$$

$$= \frac{L\tau^{2}}{\hbar \pi v} \left[-e^{-x}(x+1) \right]_{0}^{\infty}$$

$$\lim_{x \to 0} -\frac{x+1}{e^{x}} = \lim_{x \to \infty} -\frac{1}{e^{x}}$$

$$= 0$$

$$U = \frac{L\tau^{2}}{\hbar \pi v}$$

$$C_{v} = \frac{\partial}{\partial \tau} \frac{L\tau^{2}}{\hbar \pi v}$$

$$= \left[\frac{2L\tau}{\hbar \pi v} \right]$$

im too tired to finish up this problem set :(