

**1.**

$$M = 100 \text{ g} \quad , \quad M_0 = 54 \frac{\text{g}}{\text{mol}}$$

$$\frac{(100 \text{ g})}{\left(54 \frac{\text{g}}{\text{mol}}\right)} = 1.85 \text{ mol}$$

$$(1.85 \text{ mol}) * \left(6.022E23 \frac{\text{monomers}}{\text{mol}}\right) = 1.11E24 \text{ monomers}$$

A crosslinked system will have all monomers covalently bonded together (networked) via their crosslinks

**1. 1.1E24 monomers covalently bonded**

2.

 $M_n$ : Number average molecular weight $M_w$ : Weight or mass average molecular weight $M_z$ : Z-average molecular weight $M_{z+1}$ : Z+1-average molecular weight $M_v$ : Viscosity average: a from Mark-Houwink equation  $[\eta] = KM^a$  $M_w/M_n > 1$  is a measure of dispersity,  $\mathcal{D}$   
 $0.5 < a < 0.8$ 

$$M = \frac{\sum N_i M_i^{n+1}}{\sum N_i M_i^n}$$

where:  $n = 1$  gives  $M = M_w$   
 $n = 2$  gives  $M = M_z$   
 $n = 3$  gives  $M = M_{z+1}$   
 $n = 0$  gives  $M = M_n$

1.00-1.10: very narrow;  
 1.10-1.25: narrow;  
 1.25-2.5: broad;  
 2.5-10.0: very broad

[1]

$$\left( \frac{\sum n_i M_i^{1+a}}{\sum n_i M_i} \right)^{1/a}$$

Intrinsic viscosity

MW	#
5.50E+03	1.09E+19
2.00E+04	3.01E+18
5.00E+04	1.20E+18
1.00E+05	6.02E+17
2.00E+05	3.01E+17
4.00E+05	1.51E+17
4.13E+05	1.46E+17
5.00E+05	1.20E+17
6.50E+05	9.26E+16
7.00E+05	8.60E+16
8.00E+05	7.53E+16

$n \cdot M$	$n \cdot M^2$	$n \cdot M^3$	$n \cdot M^4$	$n \cdot M^{1+a}$
6.02E+22	3.31E+26	1.82E+30	1.00E+34	2.50E+25
6.02E+22	1.20E+27	2.41E+31	4.82E+35	6.17E+25
6.02E+22	3.01E+27	1.51E+32	7.53E+36	1.17E+26
6.02E+22	6.02E+27	6.02E+32	6.02E+37	1.90E+26
6.02E+22	1.20E+28	2.41E+33	4.82E+38	3.09E+26
6.02E+22	2.41E+28	9.64E+33	3.85E+39	5.03E+26
6.02E+22	2.49E+28	1.03E+34	4.24E+39	5.14E+26
6.02E+22	3.01E+28	1.51E+34	7.53E+39	5.88E+26
6.02E+22	3.91E+28	2.54E+34	1.65E+40	7.06E+26
6.02E+22	4.22E+28	2.95E+34	2.07E+40	7.44E+26
6.02E+22	4.82E+28	3.85E+34	3.08E+40	8.16E+26

Sum	1.67E+19
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6.62E+23	2.31E+29	1.32E+35	8.42E+40	4.57E+27
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Mn	3.96E+04
Mw	3.49E+05
Mz	5.69E+05
Mz+1	6.40E+05
Mv	3.05E+05
$\mathcal{D}$	8.82

3.

$$\text{Weight fraction } w_x = \frac{\text{Total mass of molecules with degree of polymerization } x}{\text{Total mass of all the molecules}}$$

$$\text{Neglecting end groups } w_x = \frac{N_x(x\overline{M}_0)}{N_0\overline{M}_0} = \frac{xN_x}{N_0}$$

$$\text{Replacing } N_x \quad N_x = N_0(1-p)^2 p^{x-1}$$

$$\overline{M}_0 = \frac{\text{Molar mass of the repeat unit}}{\text{Number of monomer unit in the repeat}}$$

Average molecular weight of the repeating units

$$w_x = x(1-p)^2 p^{x-1}$$

[2]

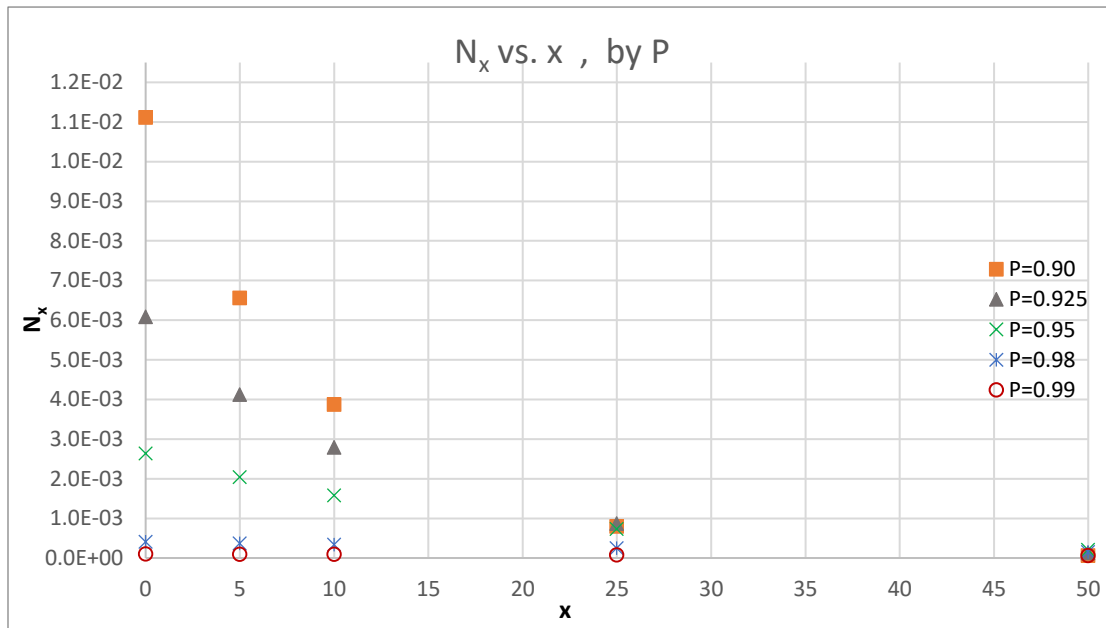


FIG. 1. Statistically derived mole fraction for a step-growth polymerization is plotted as a function of x, and varied by extent of reaction.

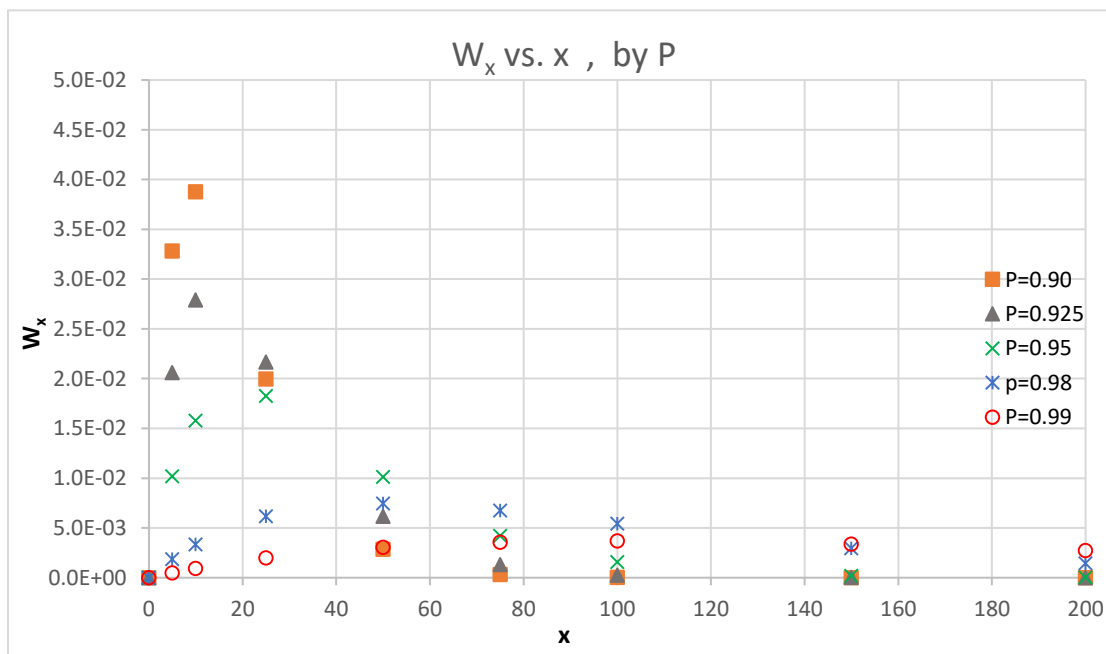


FIG. 2. Statistically derived weight fraction for a step-growth polymerization is plotted as a function of x, varied by extent of reaction.

# References

- [1] Korley, Wang. (2019). *MSEG608 – Structures and Properties of Materials*. University of Delaware. Lecture 16, Slide 17.
- [2] Korley, Wang. (2019). *MSEG608 – Structures and Properties of Materials*. University of Delaware. Lecture 17, Slide 23.

# Supplementary

2. If there is question about the calculations used, the full excel file can be found here:

<https://github.com/zswain/MSEG608> as "ZachSwain\_MSEG608-HW8.xlsx"

3. These calculations were done in the same excel worksheet as detailed above.

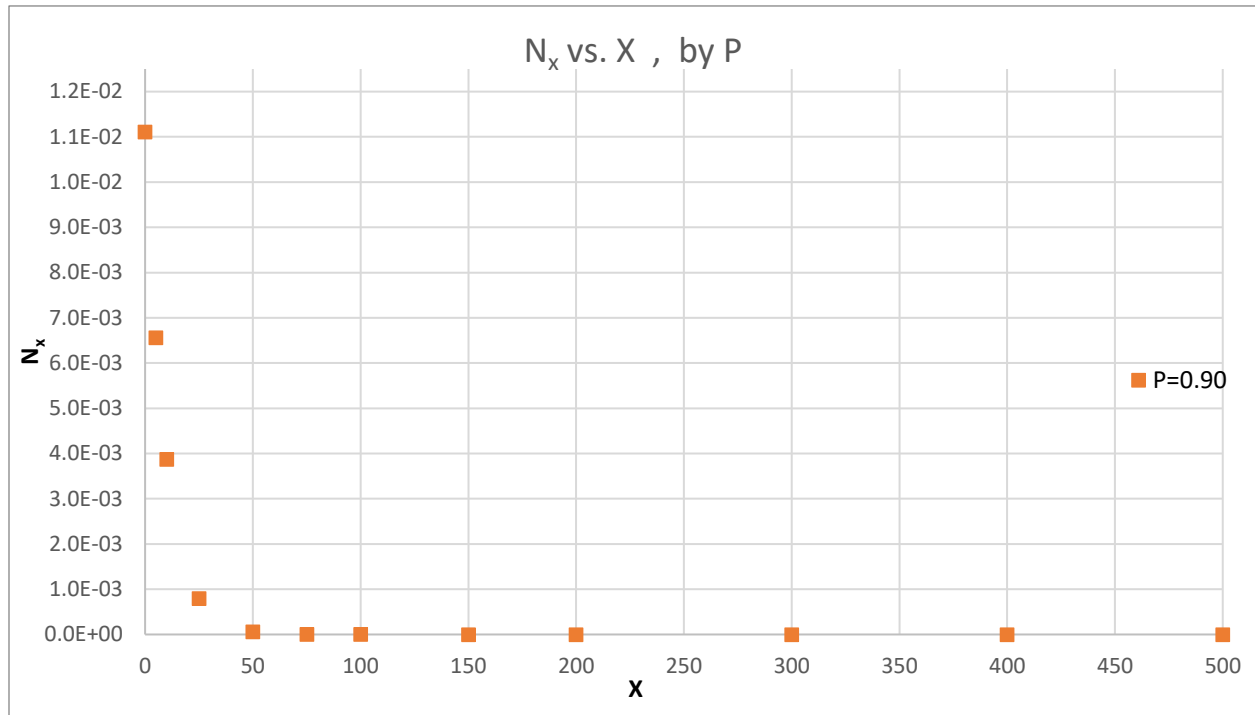
Mol  
frac

P\X	0	5	10	25	50	75	100	150	200	300	400	500
0.9	1.11E-02	6.56E-03	3.87E-03	7.98E-04	5.73E-05	4.11E-06	2.95E-07	1.52E-09	7.84E-12	2.08E-16	5.53E-21	1.47E-25
0.925	6.08E-03	4.12E-03	2.79E-03	8.66E-04	1.23E-04	1.76E-05	2.50E-06	5.07E-08	1.03E-09	4.23E-13	1.74E-16	7.16E-20
0.95	2.63E-03	2.04E-03	1.58E-03	7.30E-04	2.02E-04	5.62E-05	1.56E-05	1.20E-06	9.22E-08	5.46E-10	3.23E-12	1.91E-14
0.98	4.08E-04	3.69E-04	3.33E-04	2.46E-04	1.49E-04	8.97E-05	5.41E-05	1.97E-05	7.18E-06	9.52E-07	1.26E-07	1.67E-08
0.99	1.01E-04	9.61E-05	9.14E-05	7.86E-05	6.11E-05	4.75E-05	3.70E-05	2.24E-05	1.35E-05	4.95E-06	1.81E-06	6.64E-07

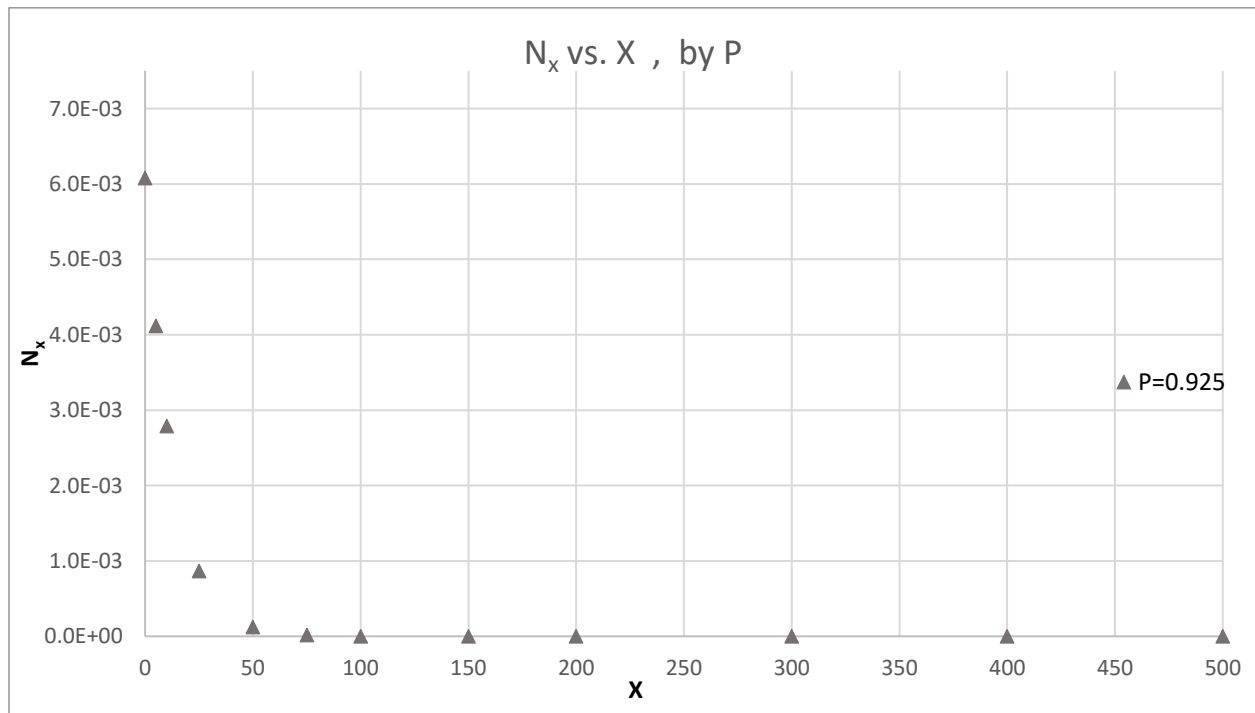
Weight frac

P\X	0	5	10	25	50	75	100	150	200	300	400	500
0.9	0	3.28E-02	3.87E-02	1.99E-02	2.86E-03	3.08E-04	2.95E-05	2.28E-07	1.57E-09	6.25E-14	2.21E-18	7.34E-23
0.925	0	2.06E-02	2.79E-02	2.17E-02	6.17E-03	1.32E-03	2.50E-04	7.61E-06	2.06E-07	1.27E-10	6.96E-14	3.58E-17
0.95	0	1.02E-02	1.58E-02	1.82E-02	1.01E-02	4.21E-03	1.56E-03	1.80E-04	1.84E-05	1.64E-07	1.29E-09	9.57E-12
0.98	0	1.84E-03	3.33E-03	6.16E-03	7.43E-03	6.73E-03	5.41E-03	2.96E-03	1.44E-03	2.86E-04	5.05E-05	8.37E-06
0.99	0	4.80E-04	9.14E-04	1.96E-03	3.06E-03	3.57E-03	3.70E-03	3.36E-03	2.71E-03	1.49E-03	7.25E-04	3.32E-04

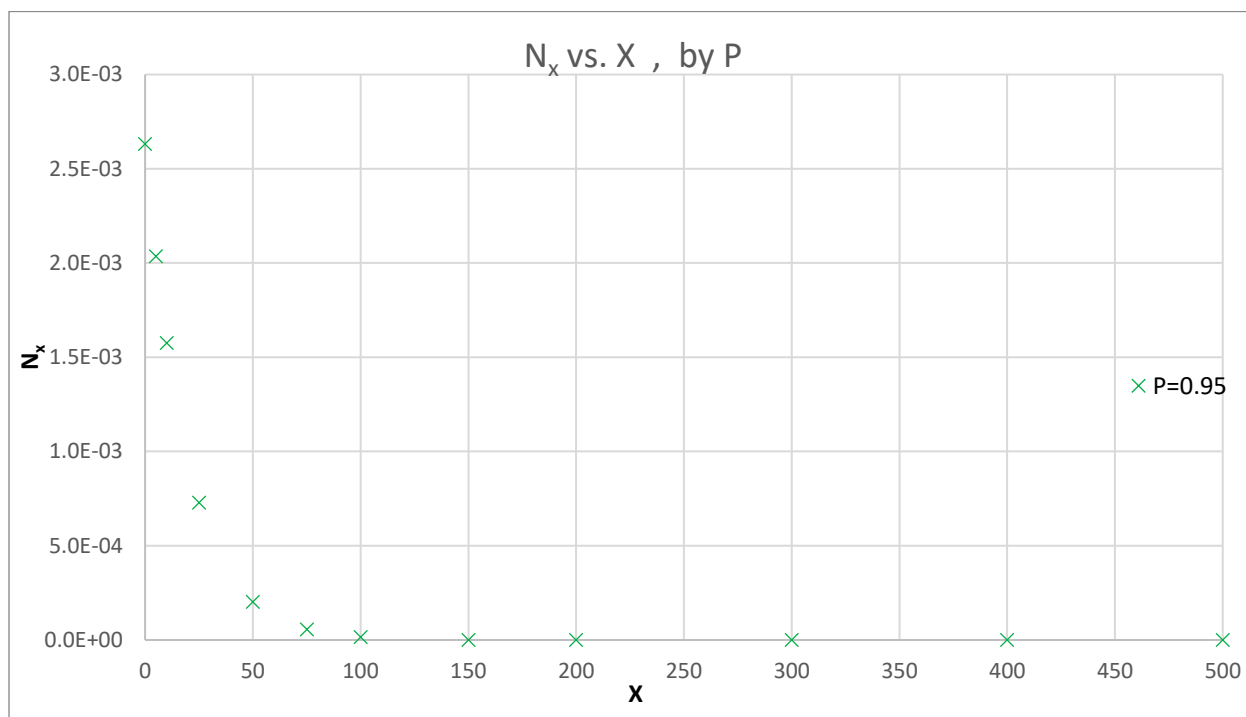
### Mole Fraction



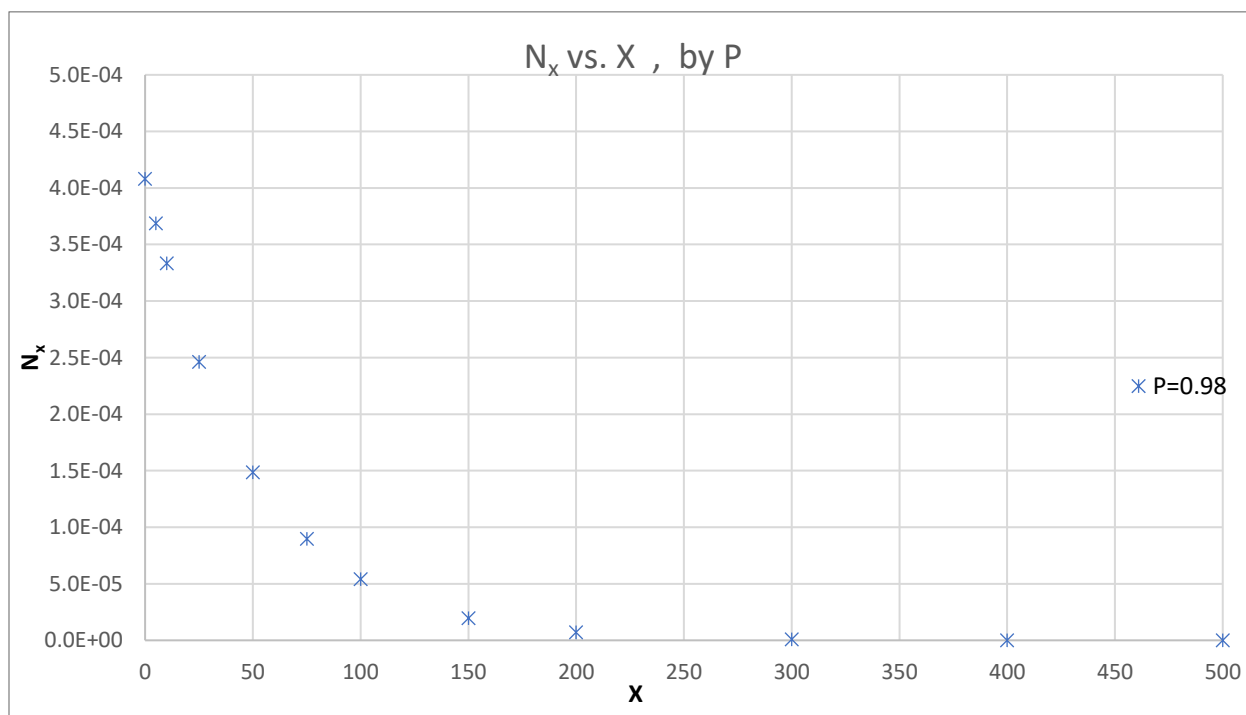
**FIG. 3.** Statistically derived mole fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.9$



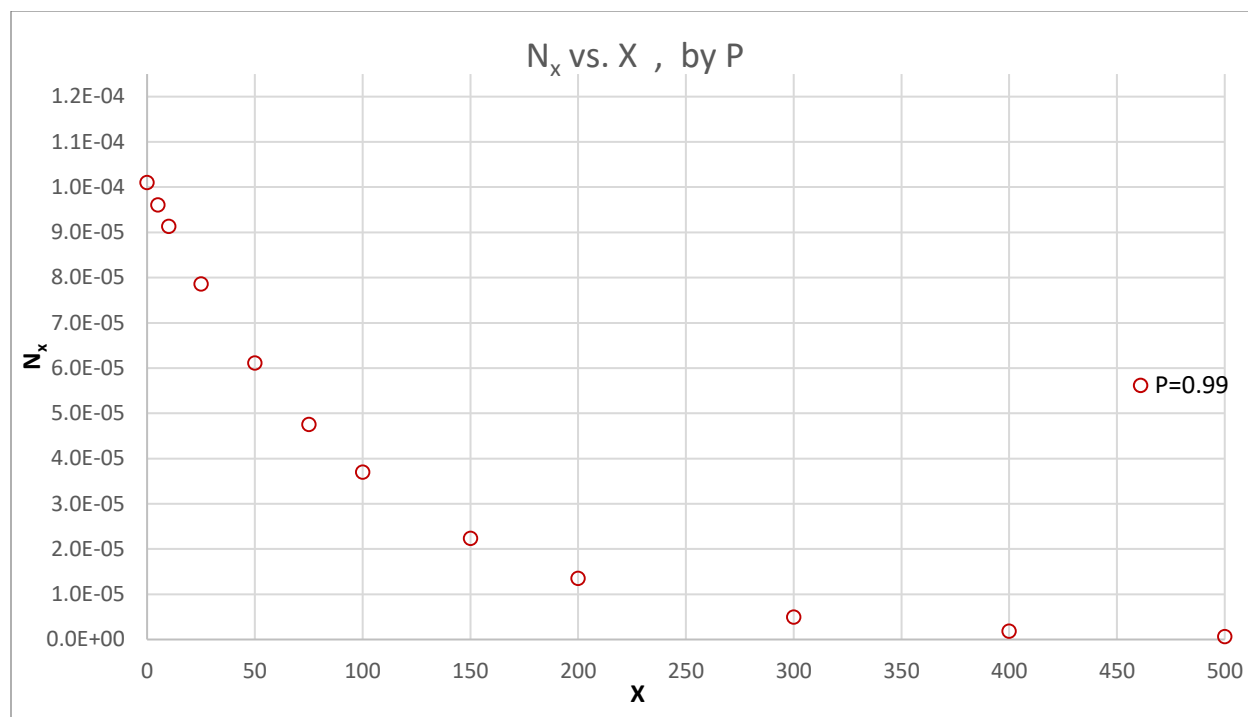
**FIG. 4.** Statistically derived mole fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.925$



**FIG. 5.** Statistically derived mole fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.95$



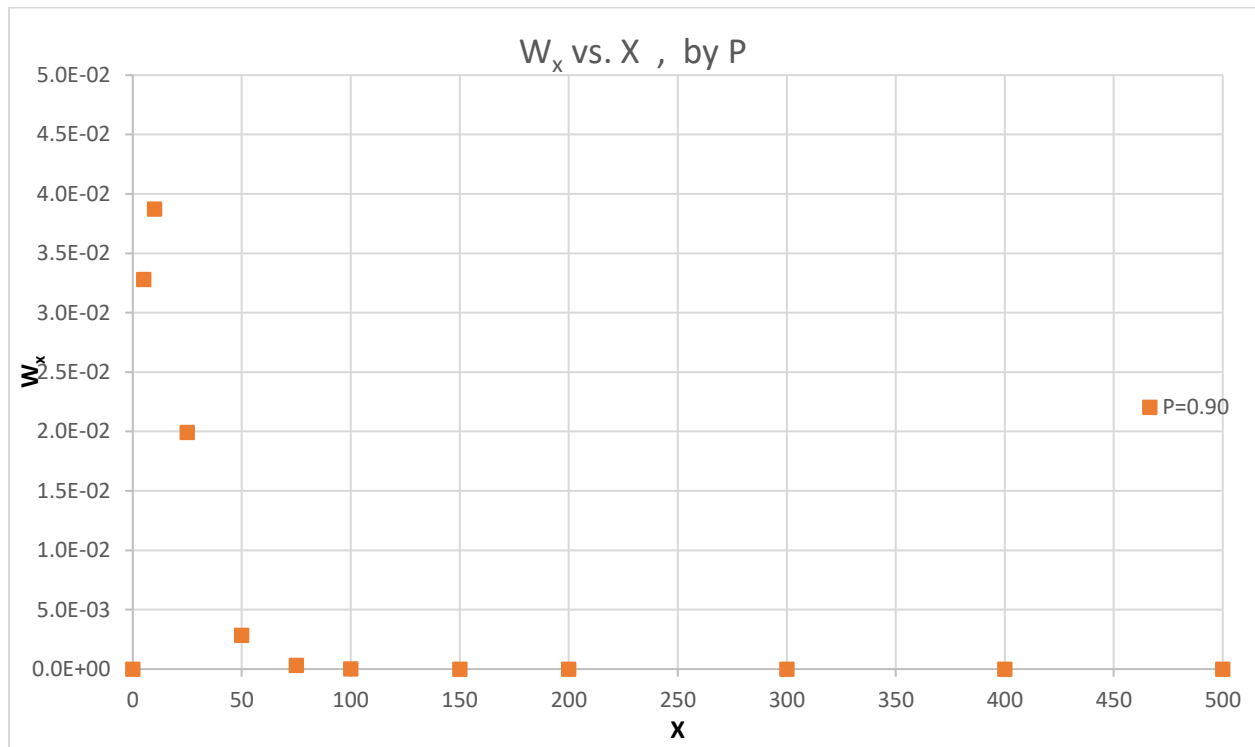
**FIG. 6.** Statistically derived mole fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.98$



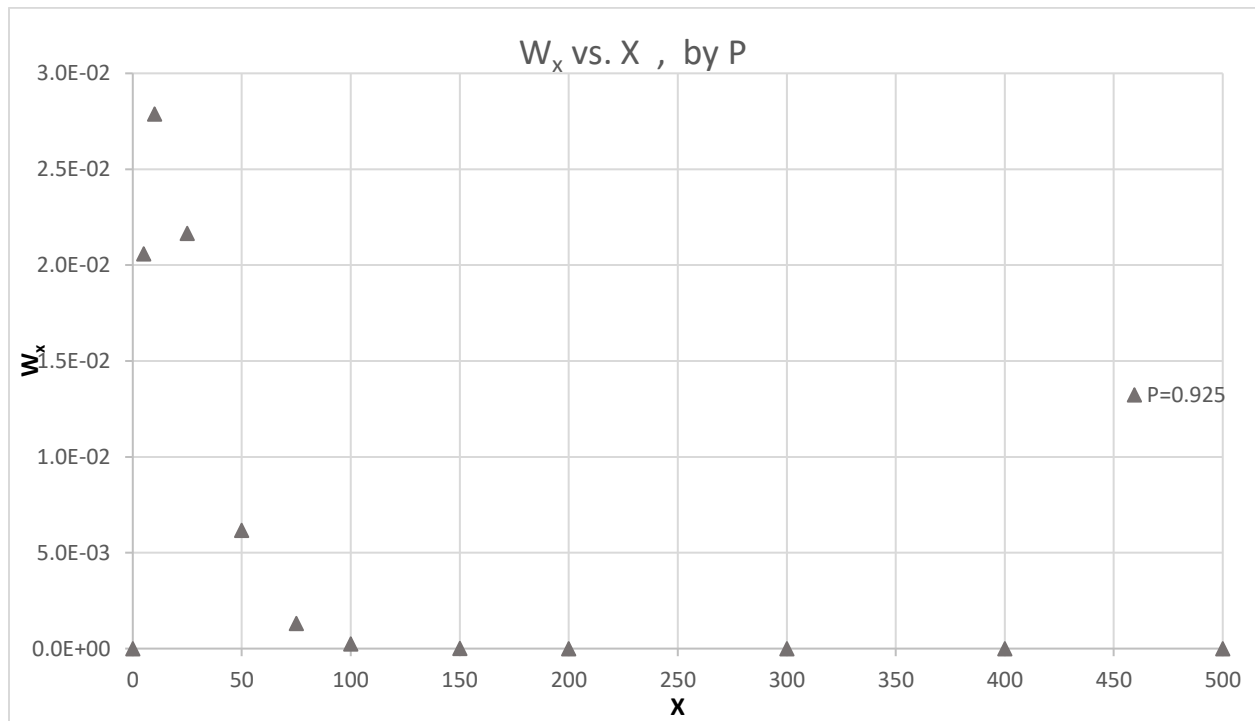
**FIG. 7.** Statistically derived mole fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.99$



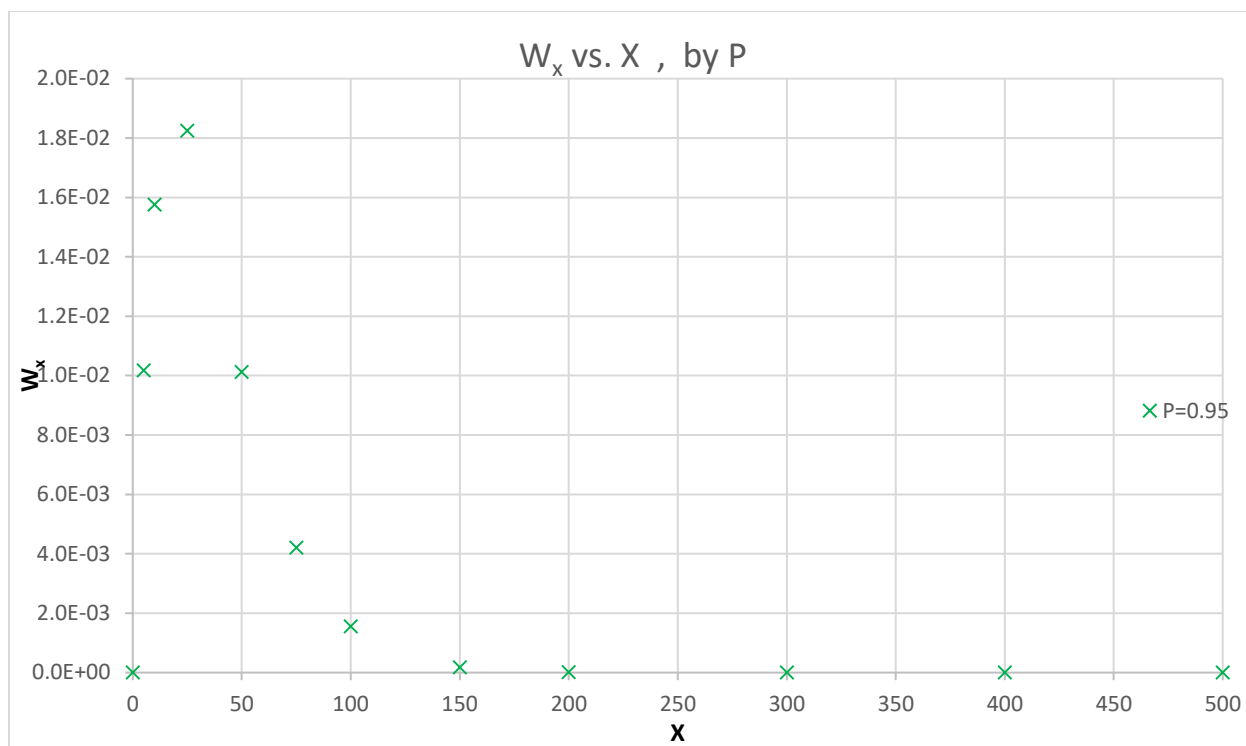
### Weight Fraction



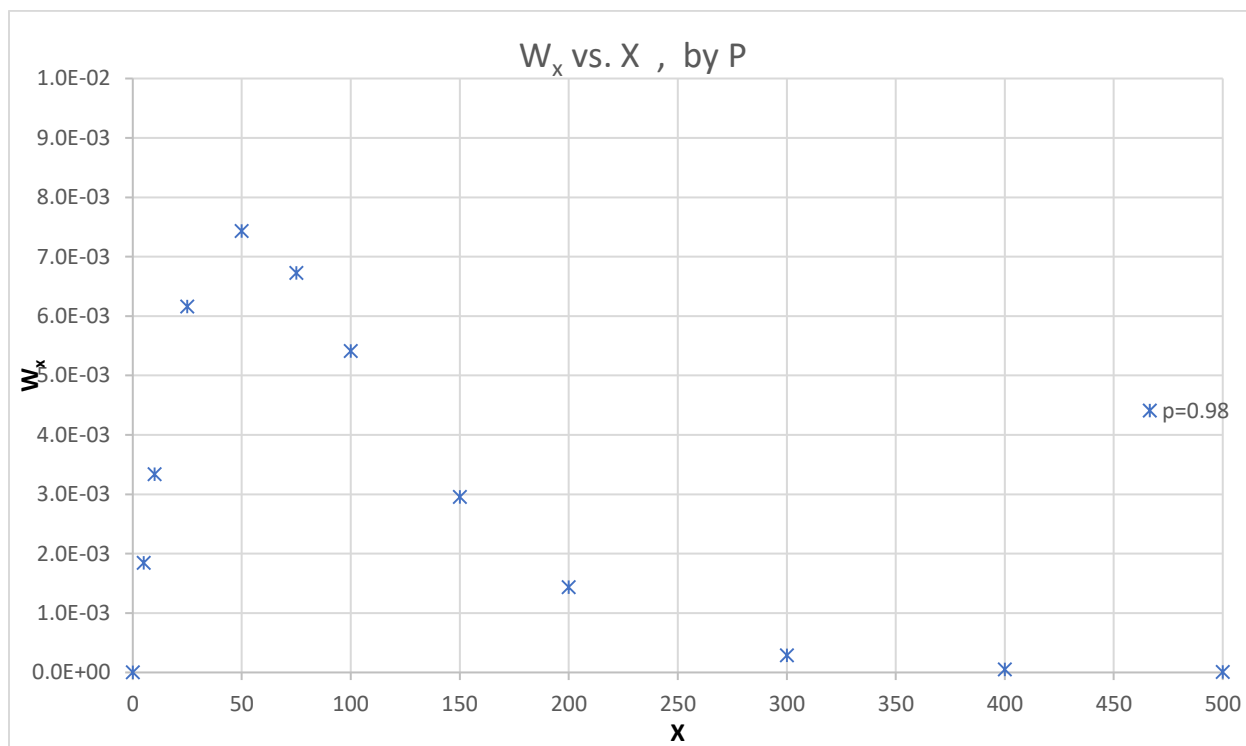
**FIG. 8.** Statistically derived weight fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.9$



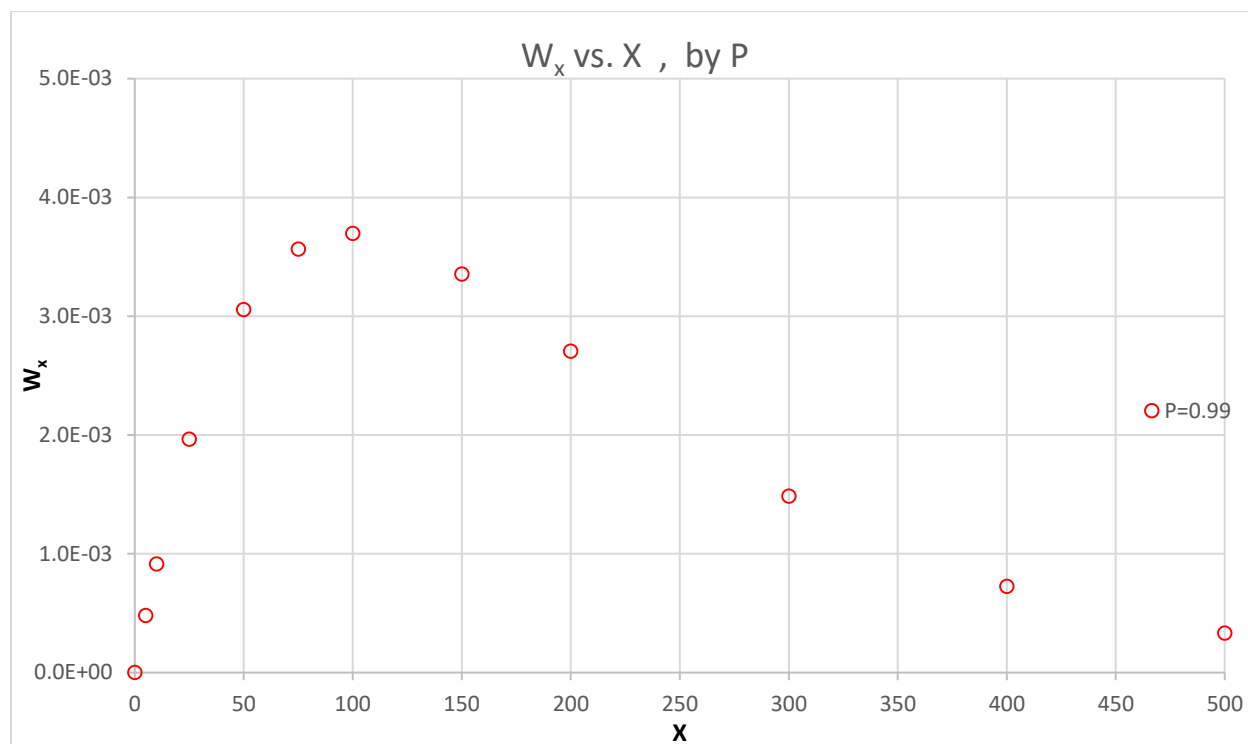
**FIG. 9.** Statistically derived weight fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.925$



**FIG. 10.** Statistically derived weight fraction for a step-growth polymerization is plotted as a function of x, for P=0.95



**FIG. 11.** Statistically derived weight fraction for a step-growth polymerization is plotted as a function of x, for P=0.98



**FIG. 12.** Statistically derived weight fraction for a step-growth polymerization is plotted as a function of  $x$ , for  $P=0.99$