MSE6701H Multiscale Materials Modeling and Simulation First Principles Calculations

A brief Introduction & Density Functional Theory

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Outline for today

- Fundamentals of Quantum Mechanics
- 2 Born-Oppenheimer Approximation
- 3 Density Functional Theory
- 4 Homework
- Further readings

All is about energy

Properties/processes of materials depend on the total energy (change):

Ground state

- Cohesive energy
- Equilibrium lattice constant (density)
- Elastic moduli and elastic constants
- Phonon frequencies
- Magnetic ordering
- Thermodynamic stability (phase diagram)
- Static electric/magnetic susceptibility
- Phase transitions
- Vacancy formation
- Diffusion
- Chemical reaction
- Interatomic forces

Excited state

- Specit heat
- Photoemission
- Optical absorption
- Energy gaps
- Transport property

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Fundamentals of Quantum Mechanics

Schrödinger Equations

$$i\hbar \frac{\partial}{\partial t}\Psi = \hat{\mathbf{H}}\Psi$$

 Ψ : the wave function;

the **probability amplitude** for different configurations of the system at different times;

 $i\hbar \frac{\partial}{\partial t}$: energy operator;

Ĥ: Hamiltonian operator;

Quantum Mechanics

Time dependent Schrödinger equation

$$i\hbar \frac{\partial}{\partial t} \Psi = \hat{\mathbf{H}} \Psi$$

$$\begin{split} \hat{\mathbf{H}} &= \hat{\mathbf{T}} + \hat{\mathbf{V}} \\ \hat{\mathbf{T}} &= \hat{\mathbf{T}}_e + \hat{\mathbf{T}}_n, \quad \hat{\mathbf{V}} &= \hat{\mathbf{V}}_{e-e} + \hat{\mathbf{V}}_{e-n} + \hat{\mathbf{V}}_{n-n} + \hat{\mathbf{V}}_{\text{ext}} \end{split}$$

Time independent or stationary Schrödinger equation

$$\hat{\mathbf{H}}\Psi = E\Psi$$

$$\hat{\mathbf{H}} = -\sum_{i} \frac{\hbar^{2}}{2m_{e}} \nabla_{i}^{2} - \frac{1}{4\pi\epsilon_{0}} \sum_{i,l} \frac{Z_{l}e^{2}}{|\mathbf{r}_{i} - \mathbf{R}_{l}|} + \frac{1}{8\pi\epsilon_{0}} \sum_{i \neq j} \frac{e^{2}}{|\mathbf{r}_{j} - \mathbf{r}_{i}|}
- \sum_{l} \frac{\hbar^{2}}{2M_{l}} \nabla_{l}^{2} + \frac{1}{8\pi\epsilon_{0}} \sum_{l \neq J} \frac{Z_{l}Z_{J}e^{2}}{|\mathbf{R}_{J} - \mathbf{R}_{l}|}
\Psi = \Psi(r_{1}, r_{2}, \dots, r_{i}, \dots, R_{1}, R_{2}, \dots, R_{l}, \dots)$$

Solvable Quantum Systems: two examples

Free particle

The Schrödinger equation

$$-\frac{\hbar^2}{2m} \bigtriangledown^2 \psi(\mathbf{r},t) = i\hbar \frac{\partial}{\partial t} \psi(\mathbf{r},t)$$

Solution for a particular momentum

$$\psi(\mathbf{r},t) = e^{i(\mathbf{k}\mathbf{r} - \omega t)}$$

with the constraint

$$\frac{\hbar^2 k^2}{2m} = \hbar \omega.$$

$$\begin{split} \langle \mathbf{p} \rangle &= \langle \psi | - i\hbar \bigtriangledown | \psi \rangle = \hbar \mathbf{k} \\ \langle E \rangle &= \langle \psi | i\hbar \frac{\partial}{\partial t} | \psi \rangle = \hbar \omega \end{split}$$

1D infinite potential well

Potential:



Schrödinger equation:

$$\mathrm{i}\hbar\frac{\partial}{\partial t}\psi(x,t) = -\frac{\hbar^2}{2m}\frac{\partial^2}{\partial x^2}\psi(x,t) + V(x)\psi(x,t),$$

Solution:
$$\psi_n(x,t) = \begin{cases} A \sin(k_n x) \mathrm{e}^{-\mathrm{i}\omega_n t}, & 0 < x < L, \\ 0, & \text{otherwise,} \end{cases}$$

Constraint:
$$k_n = \frac{n\pi}{L}$$
, where $n = \{1, 2, 3, 4, ...\}$,

Energy levels: $E_n = \frac{\hbar^2 k_n^2}{2m} = \frac{n^2 h^2}{8mL^2}$

Born-Oppenheimer/Adiabatic Approximation

Wave function for general system:

$$\Psi_{\text{total}} = \Psi(\mathbf{r}_1, \cdots, \mathbf{r}_i, \cdots, \mathbf{R}_1, \cdots, \mathbf{R}_I, \cdots)$$

BO Approximation:

$$\Psi_{\text{total}} = \psi(\mathbf{r}; \mathbf{R}) \cdot \phi(\mathbf{R})$$

First step of BO Approximation:

$$\hat{\mathbf{H}}_e\psi(\mathbf{r},\mathbf{R})=E_e\psi(\mathbf{r},\mathbf{R})$$

where $\hat{\mathbf{H}}_e = \hat{\mathbf{T}}_e + \hat{\mathbf{V}}_{e-e} + \hat{\mathbf{V}}_{e-n}$

Second step of BO Approximation:

$$[\hat{\mathbf{T}}_n + E_e(R) + \hat{\mathbf{V}}_{n-n}]\phi(\mathbf{R}) = E\phi(\mathbf{R})$$

M Born and J R Oppenheimer, Ann. Physik 84, 457 (1927) http://en.wikipedia.org/wiki/Born%E2%80%93Oppenheimer_approximation

Born-Oppenheimer/Adiabatic Approximation

Neglect quantum effect in ionic system \rightarrow replacing Schrödinger Eq. by Newton Eq.

$$M_I \frac{\partial^2 \vec{R}_I(t)}{\partial t^2} = - \bigtriangledown_I E_0(\vec{R})$$

$$E_0(\vec{R}) = E_e(\vec{R}) + V_{nn}(\vec{R})$$

Force $-\nabla_I E_0(\vec{R})$ contains contributions from the direct ion-ion interaction and a term from the gradient of the electronic total energy.

Hellmann-Feynman theorem

$$\nabla_I E_e(\vec{R}) = \langle \psi(\vec{R}) | \nabla_I \hat{H}_e(\vec{R}) | \psi(\vec{R}) \rangle$$

Couples the electronic Schrödinger Eq. with the ionic Newtonian Eq.. Car-Parrinello

http://en.wikipedia.org/wiki/Hellmann%E2%80%93Feynman_theorem

Schrödinger for electrons

$$\hat{\mathbf{H}}\psi(\mathbf{r}) = E\psi(\mathbf{r})$$

$$\hat{\mathbf{H}} = \hat{T}_{e} + \hat{V}_{e-e} + \mathbf{v}(r)
= -\sum_{i} \frac{\hbar^{2}}{2m_{e}} \nabla_{i}^{2} + \frac{1}{8\pi\epsilon_{0}} \sum_{i \neq j} \frac{e^{2}}{|\mathbf{r}_{i} - \mathbf{r}_{i}|} - \frac{1}{4\pi\epsilon_{0}} \sum_{i,l} \frac{Z_{l}e^{2}}{|\mathbf{r}_{i} - \mathbf{R}_{l}|}$$

Usual procedure to solve SE:

$$\mathbf{v}(\mathbf{r}) \quad \underline{\text{SE}} \quad \psi(\mathbf{r}) \quad \underline{\langle \psi | \cdots | \psi \rangle} \quad \text{observables.}$$

- Define the system by choosing $v(\mathbf{r})$;
- Plug into SE and solves ψ ;
- \bullet Get observables by operations over ψ .

Chemical approaches: Hartree approximation

Postulation

 ψ can be written as a simple product of one-electron wave functions

$$\psi(\mathbf{r}) = \prod_{i} \phi_{i}(\mathbf{r}_{i})$$

$$\left[-\frac{\hbar^{2}}{2m} \nabla^{2} + V_{\text{eff}}^{(i)}(\mathbf{R}, \mathbf{r}) \right] \phi_{i}(\mathbf{r}_{i}) = \epsilon_{i} \phi_{i}(\mathbf{r}_{i})$$

$$V_{\text{eff}}^{(i)}(\mathbf{R}, \mathbf{r}) = \int \frac{\sum_{j \neq i} \rho_{j}(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V(\mathbf{R}, \mathbf{r})$$

$$\rho_{i}(\mathbf{r}) = |\phi_{i}(\mathbf{r})|^{2}$$

Problem

Electrons are Fermions and indistinguishable, follow Pauli exclusion principles.

Chemical approaches: Hartree-Fock approximation (SCF)

Introduce Pauli exclusion principle

$$\psi(\mathbf{r}) = \mathrm{SD}\{\phi_i(r_i, \sigma_i)\}\$$

$$\left[-\frac{\hbar^2}{2m} \bigtriangledown^2 + \int \frac{\sum_{j \neq i} \rho_j(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V(\mathbf{R}, \mathbf{r}) \right] \phi_i$$

$$-\sum_i \left[\sum_{\mathbf{r}'} \int \frac{\phi_j^* \phi_i}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' \right] \phi_j = \sum_i \lambda_{ij} \phi_j$$

Problem

The implied mean field approximation neglects electron correlation for the electrons of opposite spin.

Density Functional Theory: Preliminaries

Function

Function is a relation that uniquely associates members of one set with members of another set. $f: x \longrightarrow f(x)$

Functional

Functional is a real-valued function on a vector space V, usually of functions. $V = \{f : [0,1] \longrightarrow \Re : f \text{ is contineous}\}$

A functional is a function of functions. It maps a function to a number.

Extremum

 x_0 minimizes/maximizes f(x) if $\frac{df(x)}{dx}|_{x=x_0} = 0$.

 $f_0(x)$ minimizes/maximizes F[f(x)] if $\frac{\delta F[f(x)]}{\delta f(x)}|_{f=f_0}=0$.

Ref: http://mathworld.wolfram.com

K. Kang, C. Weinberger, and W. Cai, A Short Essay on Variational Calculus, Stanford University, 2006.

<u>Density</u> Functional Theory (DFT)

$$\mathbf{n}(\mathbf{r}) = N \int d^3 r_2 \int d^3 r_3 \cdots \int d^3 r_N \psi^*(\mathbf{r}, \mathbf{r}_2, \cdots, \mathbf{r}_N) \psi(\mathbf{r}, \mathbf{r}_2, \cdots, \mathbf{r}_N)$$

DFT procedure to solve SE:

$$n(\mathbf{r}) \implies \psi(\mathbf{r}) \implies v(\mathbf{r}).$$

$$V_{ext}(r) \quad \stackrel{HK}{\longleftarrow} \quad n(r)$$

$$\downarrow \qquad \qquad \uparrow$$

$$\Psi_i(\{r\}) \quad \Longrightarrow \quad \Psi_0(\{r\})$$

P. Hohenberg and W. Kohn, Phys. Rev. 136:B864, 1964.











(1938.4 -)

(1934.10-2017.12) (1923.3-2016.4)

Hohenberg-Kohn theorem

First (existence) theorem

For any system of interacting particles in $v_{\text{ext}}(\mathbf{r})$, $n_0(\mathbf{r})$ is uniquely determined, i.e., there is a one-to-one correspondence between $v_{\text{ext}}(\mathbf{r})$ and $n_0(\mathbf{r})$.

Second (variational) theorem

An energy functional E[n] can be defined. The exact ground state is the global minimum value of this functional, i.e, $E_v[n_0] \leq E_v[n]$.

Interpretation

- Unique mapping of $v(\mathbf{r}) \Leftrightarrow n_0$;
- All properties are given by $n_0(\mathbf{r})$;
- Each property is a functional of n₀(r), f[n₀];
- A functional $f[n_0]$ maps a function to a result: $n_0(\mathbf{r}) \to f$.

Advantages

- reduces problem dim. from 3N to 3;
- n(r) is an experimental observable;
- Easy to visualize.

Hohenberg-Kohn theorem

Hamiltonian of electrons

$$\hat{\mathbf{H}} = -\sum_{i} \frac{\hbar^2}{2m_e} \nabla_i^2 + \frac{1}{8\pi\epsilon_0} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_j - \mathbf{r}_i|} - \frac{1}{4\pi\epsilon_0} \sum_{i,l} \frac{Z_l e^2}{|\mathbf{r}_i - \mathbf{R}_l|}$$

Specially for energy

- ψ_0 produces n_0 , minimizes $E \Rightarrow n_0$ reproduces ψ_0 , minimizes E;
- Total energy functional:

$$E[n] = F[n] + V[n] = T[n] + U[n] + V[n].$$

T[n] and U[n]: universal (system independent); V[n]: system dependent, but once $v(\mathbf{r})$ is known, $V[n] = \int n(\mathbf{r})v(\mathbf{r})d^3r$.

For a given system $(v(\mathbf{r}))$, all one has to do is to minimize E[n] with respect to n.

Hohenberg-Kohn theorem

pros

simple; exact.

cons

T[n] and U[n] are unknown.

Borrow ideas from Thomas-Fermi approximation?

$$U[n] \approx U_H[n] = \frac{e^2}{2} \int d^3r \int d^3r' \frac{n(\mathbf{r})n(\mathbf{r'})}{|\mathbf{r} - \mathbf{r'}|},$$

$$T[n] \approx T_{\mathrm{single \ particle}}^{\mathrm{LDA}}[n] = \int t_s^{hom}(n(\mathbf{r})) d^3r.$$

$$t_s^{hom}(n) = \frac{3\hbar^2 (3\pi)^{2/3} n^{5/3}}{10m}$$

Unfortunately: unstable molecules.

Reasons: 1) neglect of correlations; 2) LDA in T[n].

Decompose T[n] and U[n]:

$$T[n] = T_s[n] + T_c[n]$$
$$U[n] = U_H[n] + U_x[n]$$

Rewrite the exact energy functional:

$$E[n] = T_s[n] + U_H[n] + E_{xc}[n] + V[n]$$

Energy minimization:

$$\frac{\delta E[n]}{\delta n(\mathbf{r})} = \frac{\delta T_s[n]}{\delta n(\mathbf{r})} + \frac{\delta V[n]}{\delta n(\mathbf{r})} + \frac{\delta U_H[n]}{\delta n(\mathbf{r})} + \frac{\delta E_{xc}[n]}{\delta n(\mathbf{r})}$$

$$= \frac{\delta T_s[n]}{\delta n(\mathbf{r})} + v(\mathbf{r}) + v_H(\mathbf{r}) + v_{xc}(\mathbf{r})$$

$$= 0.$$

Noninteracting particles moving in the potential field of $v_s(\mathbf{r})$:

$$0 = \frac{\delta E[n]}{\delta n(\mathbf{r})} = \frac{\delta T_s[n]}{\delta n(\mathbf{r})} + \frac{\delta V_s[n]}{\delta n(\mathbf{r})} = \frac{\delta T_s[n]}{\delta n(\mathbf{r})} + v_s(\mathbf{r})$$

Effective external field:

$$v_s(\mathbf{r}) \equiv v_{eff}(\mathbf{r}) = v(\mathbf{r}) + v_H(\mathbf{r}) + v_{xc}(\mathbf{r})$$

Auxiliary Schrödinger Equation:

$$\left[-\frac{\hbar^2\nabla^2}{2m} + v_s(\mathbf{r})\right]\phi_i(\mathbf{r}) = \epsilon_i\phi_i(\mathbf{r})$$

$$n(\mathbf{r}) \equiv n_s(\mathbf{r}) = \sum_{i=1}^{N} f_i \cdot |\phi_i(\mathbf{r})|^2.$$

$$T_{s}[n] = -\frac{\hbar^{2}}{2m} \sum_{i} \int \phi_{i}^{*}(\mathbf{r}) \bigtriangledown^{2} \phi_{i}(\mathbf{r}) d^{3}r, \qquad T_{s}[n] + V_{s}[n] = \sum_{i} f_{i} \cdot \epsilon_{i}.$$

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Total energy:

$$E_{0} = T_{s}[n] + U_{H}[n] + E_{xc}[n] + V[n]$$

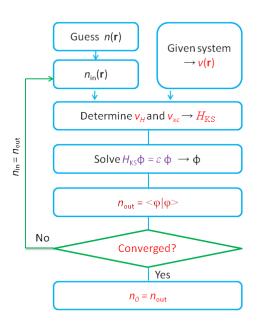
$$= T_{s}[n] + U_{H}[n] + E_{xc}[n] + \int v(\mathbf{r})n(\mathbf{r})d\mathbf{r}$$

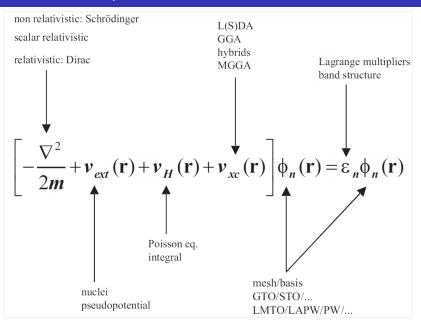
$$= T_{s}[n] + U_{H}[n] + E_{xc}[n] + \int [v_{s}(\mathbf{r}) - v_{H}(\mathbf{r}) - v_{xc}(\mathbf{r})]n(\mathbf{r})d\mathbf{r}$$

$$= T_{s}[n] + U_{H}[n] + E_{xc}[n] + V_{s}[n] - \int v_{H}(\mathbf{r})n(\mathbf{r})d\mathbf{r} - \int v_{xc}(\mathbf{r})n(\mathbf{r})d\mathbf{r}$$

$$= T_{s}[n] + V_{s}[n] + U_{H}[n] - \int v_{H}(\mathbf{r})n(\mathbf{r})d\mathbf{r} + E_{xc}[n] - \int v_{xc}(\mathbf{r})n(\mathbf{r})d\mathbf{r}$$

$$= \sum_{i} f_{i}\epsilon_{i} - \frac{e^{2}}{8\pi\epsilon_{0}} \int \int \frac{n_{0}(\mathbf{r})n_{0}(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r} d\mathbf{r}' + E_{xc}[n] - \int \frac{\delta E_{xc}[n]}{\delta n}n(\mathbf{r})d\mathbf{r}$$





DFT in practice: approximations

Local Density Approximation (LDA)

$$E_{x}^{\text{LDA}} = -\frac{3e^{2}}{4} \left(\frac{3}{\pi}\right)^{1/3} \int n(\mathbf{r})^{4/3} d^{3}r$$

$$E_{xc}[n] \simeq E_{xc}^{\mathrm{LDA}}[n] = \int e_{xc}^{\mathrm{hom}}(n) d^3 r|_{n \to n(\mathrm{r})} = \int e_{xc}^{\mathrm{hom}}[n(\mathbf{r})] d^3 r.$$

Generalized Gradient Approximation (GGA)

$$E_{xc}^{GGA}[n] = \int f(n(\mathbf{r}), \nabla n(\mathbf{r}))d^3r.$$

Chemistry: BLYP Physics: PBE

Spin-polarized

 $n = n_{\alpha} + n_{\beta}$.

$$E_{xc}^{\mathrm{LSDA}}[n_{\alpha},n_{\beta}] = \int f(n_{\alpha},n_{\beta})n(\mathbf{r})d^{3}r.$$

Homework

- Go through all the slides and deduce all equations by yourself.
- Read through: Bloch theorem.
- Think: If you evaluate the kinetic energy of a Kohn-Sham system, does it equal to the physical kinetic energy? why?
- Hand in: Solve the free particle and 1-D infinite potential well problems illustrated in page 6.
- **Mand in:** Derive the Kohn-Sham equations (page 17) by yourself.
- **Hand in:** When we say F[n] is a *universal* functional, what do we mean exactly? Since $v_{ext}(r)$ itself is a functional of n(r), is the total energy functional universal?

Due: Nov 9th, 2022. You can refer to any kind of literature, but specify your references explicitly.

References and further readings

- K. Capelle, A Bird's-Eye View of DFT, http://arxiv.org/abs/cond-mat/0211443.
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- R. O. Jones and O. Gunnarsson, Review of Modern Physics 61(3):689, 1989.
- J. Hafner, Atomic-scale computational materials science, Acta Materialia 48:71-92, 2000.
- R.M. Dreizler and E.K.U. Gross, Density Functional Theory, Springer-Verlag (1990), ISBN 3-540-51993-9.