

First principles calculations

hand-on experiments

Ling-Ti Kong

E-Mail: konglt@sjtu.edu.cn

School of Materials Science and Engineering, Shanghai Jiao Tong University

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Outline

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- 2 Equilibrium lattice constant determination
 - FCC Cu
 - Lattice constant of Fe
- 3 Band structure calculation
 - FCC Cu
 - Band structure of bcc Fe
 - Band structure of Graphene
 - Band structure of Polyethylene
- 4 Equilibrium bond length determination
 - Ground state of H atom
 - Equilibrium bond length and bond energy of H
- 5 Surface energy
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Murnaghan's Equation of states (EOS)

$$E(V) = \frac{B_0 V}{B'_0} \left[\frac{(V_0/V)^{B'_0}}{B'_0} + 1 \right] - \frac{B_0 V_0}{B'_0 - 1} + E_0$$

E : Total energy

E_0 : Minimum total energy

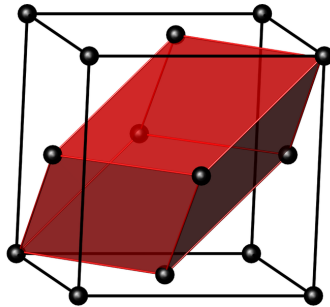
V : Volume

V_0 : Equilibrium volume

B_0 : Bulk modulus under equilibrium, $V \left. \frac{\partial^2 E}{\partial V^2} \right|_{V_0}$

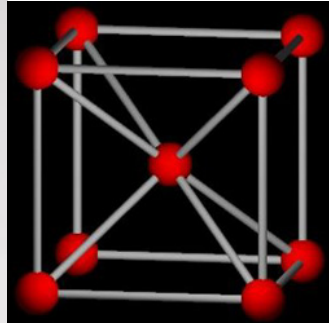
B'_0 : $\left. \frac{\partial B}{\partial P} \right|_{P_0}$

Experimental data for fcc Cu



Lattice constant: 3.615 Å Bulk Modulus: 140 GPa

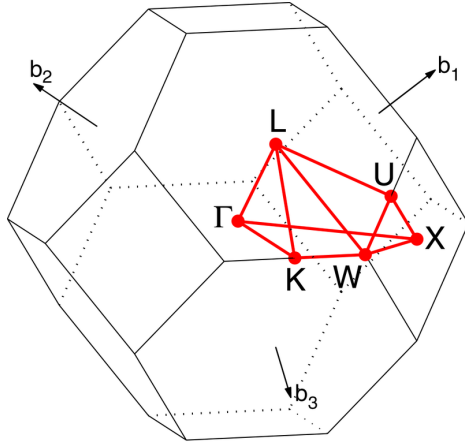
Experimental data for bcc Fe



Lattice constant: 2.8665 Å
Bulk Modulus: 170 GPa

<http://www.periodictable.com/Elements/026/data.html>

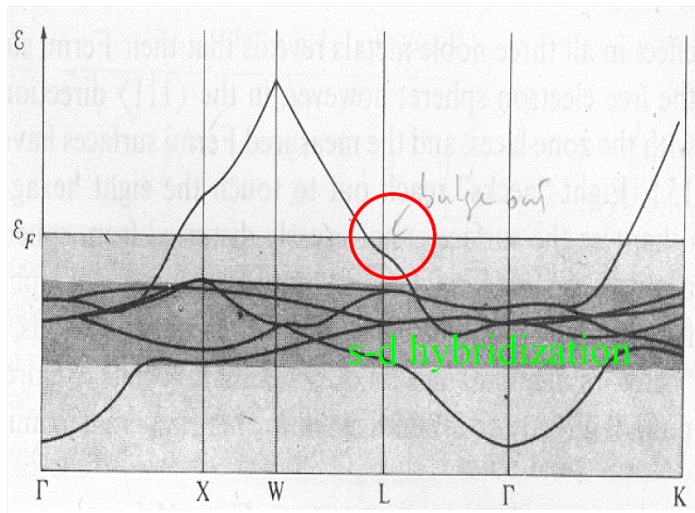
First Brillouin zone of fcc lattice



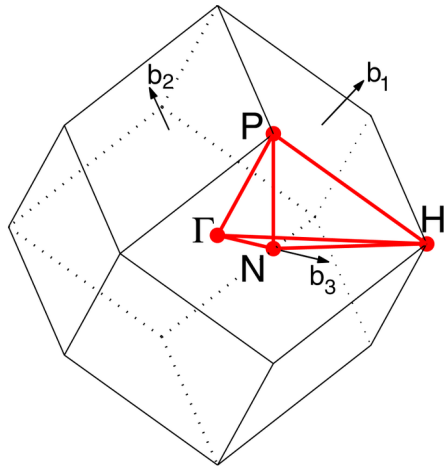
FCC path: Γ -X-W-K- Γ -L-U-W-L-K|U-X

[Setyawan & Curtarolo, DOI: 10.1016/j.commatsci.2010.05.010]

Band structure of fcc Cu

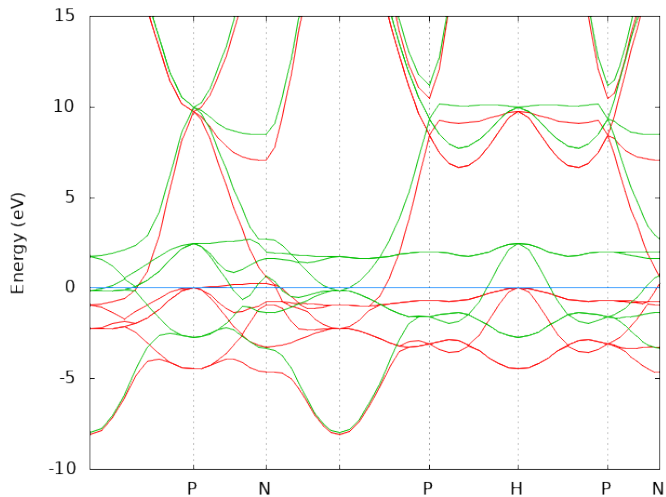


First Brillouin zone of bcc lattice

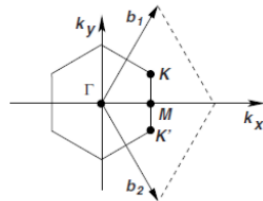
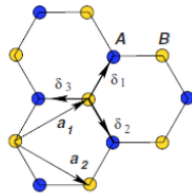
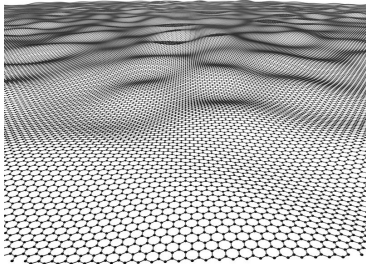


BCC path: Γ - H - N - Γ - P - H | P - N

Band structure of bcc Fe



Graphene

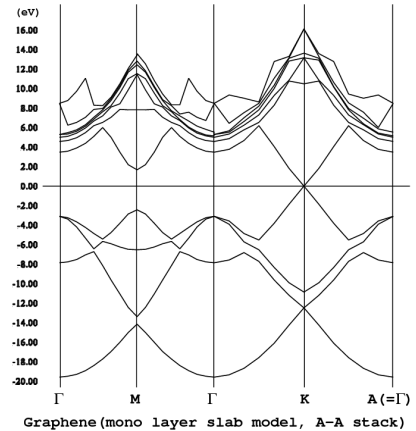
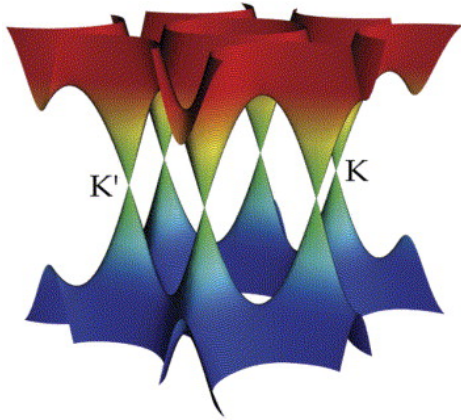


Source:

<http://news.discovery.com/tech/cooling-down-electronics-with-graphene.html>

http://thep-center.org/src/article_edu_e.php?article_edu_id=17

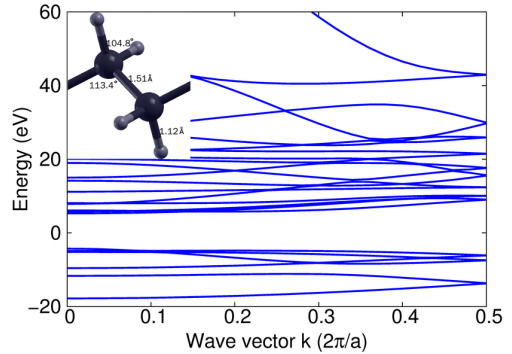
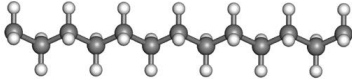
Band structure of graphene



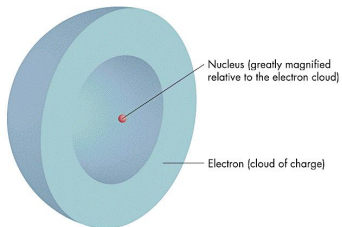
Source:
<http://www.sciencedirect.com/science/article/pii/S1369702106717886>

<http://www.bandstructure.jp/Table/BAND/Graphene.html>

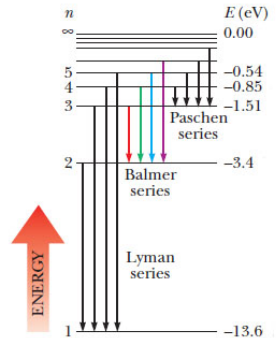
Polyethylene



Hydrogen atom



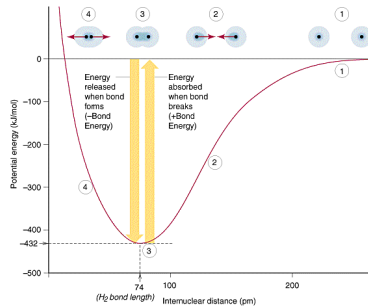
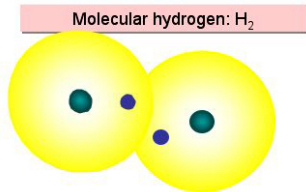
Cross section of a hydrogen atom



An energy level diagram for hydrogen.

$$E_n = -\frac{2\pi^2 m e^4}{n^2 h^2} = -\frac{13.6 \text{ eV}}{n^2}$$

Hydrogen molecule



Bond length $b_0 = 0.74 \text{ \AA}$

Bond energy $E_b^0 = -4.48 \text{ eV} = -0.33 \text{ Ry}$

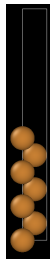
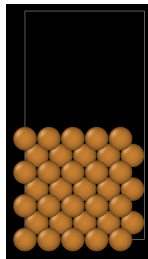
Surface energy calculation

- Bulk energy calculation (reference, E_{bulk}) ;
- Static calculation for clean surface ($E_{unrelaxed}$) ;

$$\gamma_{unrelaxed} = \frac{E_{unrelaxed} - N \cdot E_{bulk}}{2A}$$

- Relaxing one side of the surface while keeping the other side fixed;
- Static calculation for the relaxed structure ($E_{relaxed}$) ;

$$\gamma = \gamma_{unrelaxed} + \frac{E_{relaxed} - E_{unrelaxed}}{A}.$$



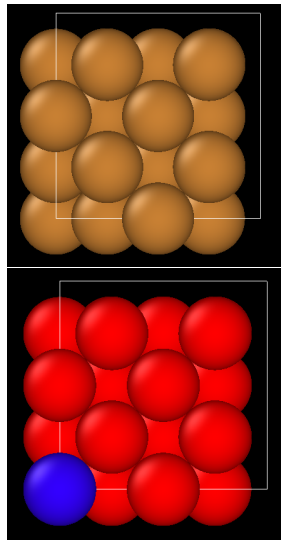
N : number of atoms in surface model. E_{bulk} in unit of eV/atom.

Vacancy formation energy calculation

- Bulk energy calculation (reference, E_{bulk}) ;
- Relaxation of supercell with a vacancy, and static calculation of the relaxed structure ($E_{relaxed}$) ;

$$E_v^f = E_{relaxed} - \frac{N-1}{N} E_{bulk}.$$

N : number of atoms in perfect crystal.



Homework

- Go over all the examples and finish all the remaining calculations by yourself.
- Perform DFT calculations for Al with its FCC primitive cell:
 - ① Carry out convergence tests with respect to k -mesh size and E_{cut} .
 - ② Determine the a_0 and B_0 of FCC Al by scanning a range of lattice parameters with Gaussian, Fermi-Dirac, and Methfessel-Paxton smearings, respectively.
 - ③ Calculate the surface energy for Al(001), Al(110), Al(111), and vacancy formation energy for Al.
 - ④ Compare your results to experimental data, the MD results, and DFT results in literatures.
 - ⑤ Finish an experimental report.

Due: Dec 7th, 2022.

<http://www.periodictable.com/Elements/029/data.html>