Explain this equation:			
	Thus the net rate of		
CO <sub>2</sub> assimilation is	That the net late of		
$A = V_c - 0.5 \ V_o - R_d^{1}$	(1)		
where $V_c$ is the rate of carboxylate of oxygenation. The symbol $R_d$ reption from mitochondria in the light associated with the PCO cycle.	presents CO <sub>2</sub> evolu-		
Can you explain the following sen	itence? What proces	ss does represent?	
For each carboxylation, $\phi$ oxy	genations occur.		
What is meant by 'RUBISCO limite	ed' or 'RUBP limited'	photosynthesis?	
NA/le at its up a set by all attends to a sec			
What is meant by electron transp	ort (J) limited photo	isyntnesis?	

How does the following equation (from Carl's lecture) summarize the Farquhar model (and its extensions)?

$$A = \min \left\{ w_c, w_j, w_p \right\} (1 - \Gamma^* / C) - R_d$$

Why did Farquhar et al measure photosynthesis under different Oxygen concentrations?

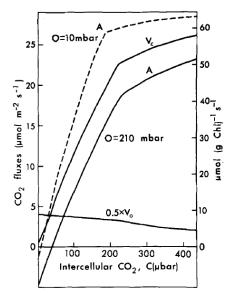


Fig. 7.  $\rm CO_2$  fluxes versus intercellular  $p(\rm CO_2)$ ,  $\rm C(\mu bar)$ . The solid lines at 25C and 1000  $\rm \mu mol$  photons m<sup>-2</sup> s<sup>-1</sup> represent the situation in ambient (210 mbar)  $p(\rm O_2)$ , with  $V_c$ , A and  $0.5 \cdot V_o$  denoting the rates of carboxylation, the net rate of assimilation of  $\rm CO_2$  and the rate of release of photorespired  $\rm CO_2$ . The dashed line represents the rate of  $\rm CO_2$  assimilation in 10 mbar  $p(\rm O_2)$ 

What is the significance of this statement from Farquhar et al 1980?
The ratio of the solubilities of $O_2$ and $CO_2$ increase with temperature and Ku and Edwards (1977) have suggested that photorespiration increases more rapidly with temperature than does carboxylation for this reason.
How does Long 1991 expand these ideas?