

KADI SARVA VISHVAVIDHYALAYA
B.E. SEM III

Subject Code: ME-304/AE-304

Subject Name: Material Science & Metallurgy

Date: 16/04/2015

Time: 10.30a.m.-1.30p.m.

Total Marks: 70

Instructions:

1. Answer each section in separate Answer Sheet.
2. Use of Scientific calculator is permitted.
3. All questions are **compulsory**.
4. Indicate **clearly**, the options you attempt along with its respective question number.
5. Use the last page of main supplementary of **rough work**.

Section - I

- Q 1** (A) Give differences between metals and nonmetals. [5]
(B) State the four major materials groups for materials classification, bring out the basis of its classification and mention the important characteristics of each group. [5]
(C) Compare and Contrast: Destructive test with Non Destructive test. [5]

OR

- (C) Explain selection criteria for engineering materials.
Q 2 (A) Differentiate: Macrostructure and Microstructure Examination. [5]
(B) State composition, specific properties and applications of Grey Cast Iron. [5]

OR

- Q 2** (A) State the types of solid solution and explain Hume Rothery's rule for the formation of solid solution. [5]
(B) Which are advantages and limitations of powder metallurgy? [5]
Q 3 (A) Specify, with reasons alloy suitable for the manufacture of: Aerospace application, Glass Cutter. [5]
(B) What is solid solution? Explain types of solid solution. [5]

OR

- Q 3** (A) Explain Lever arm principle for solid solution. [5]
(B) Explain Tempering process. [5]

Section - II

- Q 4** (A) Draw and explain TTT Diagram [5]
(B) Enlist types of corrosion. Explain any two type of corrosion with sketch. [5]
(C) Explain the effects on steel by alloying elements Cobalt, Tungsten and Nickle. [5]

OR

- (C) Explain in brief: Sintering Process. [5]
Q 5 (A) Explain Radiography Testing with advantages and limitations. [5]
(B) Compare and contrast Austempering and Martempering. [5]

OR

- Q 5** A) Write short note on Nitriding process. [5]
(B) Explain any one organic coating process. [5]

- Q 6** (A) Explain White metal along with its application. [5]
(B) Explain Pitting and Crevice corrosion.

OR

- Q 6** (A) Show Eutectic and Eutectoid point on Fe – C Diagram. Explain its importance. [5]
(B) Explain Phosphorous Print. [5]

----ALL THE BEST----

KADI SARVA VISHWAVIDHYALAYA

B.E. Semester: III MECHANICAL ENGINEERING

Subject Code: - ME-304 Subject Name: - MATERIAL SCIENCE & METALLURGY

Date: - 28/11/13

Time: -10:00 am TO 1:00 pm

Total Marks: - 70

Instruction:

1. Answer each section in separate answer sheet.
2. Use of scientific calculator is permitted.
3. All questions are **Compulsory**.
4. Indicate **Clearly**, the option you attempt along with its respective question number.
5. Use the last page of main supplementary of **rough work**.

Section – I

Q-1 (All compulsory)

- (A) State the importance of study of materials science briefly. Explain Engineering requirements of materials [5]
- (B) What is Gibb's phase rule? Explain its importance. [5]
- (C) Enlist the properties of pure aluminum and mention the composition, properties and application of any one aluminum alloy. [5]

OR

- (C) What is the purpose of Alloying? Give effects of nickel as an alloying element. [5]

Q-2 Answer the following questions.

- (A) Name the different annealing processes. Is spheroidising different from annealing? Explain. [5]
- (B) Draw TTT diagram for eutectoid steel. Explain it briefly by considering few cooling rates. [5]

OR

- (A) Which are various surface hardening processes? Explain induction hardening process. [5]
- (B) Classify types of cast iron. Discuss any one. [5]

Q-3 Answer the following questions.

- (A) Explain with neat sketches the arrangement of atoms in B.C.C, F.C.C. and H.C.P. lattice. Define unit cell. Show that a F.C.C. structure is always more close packed than B.C.C. structure. [5]
- (B) What is "Wrought iron"? enlist the properties and uses of It. [5]

OR

- (A) Explain selection criteria for engineering materials. [5]
- (B) Define (1) Creep, (2) Malleability (3) Hardness (4) Machinability (5) Fatigue [5]

Section – II

Q-4 (All compulsory)

- (A) With the aid of an iron-iron carbide equilibrium diagram show and explain eutectic, peritectic and eutectoid transformation. Also mention the significance of these transformations. [5]
 - (B) Enlist the properties of pure copper and mention the composition, properties and application of phosphorus bronze. [5]
 - (C) Explain allotropic transformation of iron. [5]
- OR**
- (C) Give composition properties and uses of malleable cast iron [5]

Q-5 Answer the following questions.

- (A) State different types of corrosion and explain their probable causes of occurrence. Enlist common methods to protect corrosion. [5]
 - (B) Explain Cathodic protection against corrosion. [5]
- OR**
- (A) How will you classify brasses based on the composition of zinc Explain the properties & application of the main type of brasses. [5]
 - (B) Explain flame-hardening process in brief [5]

Q-6 Answer the following questions.

- (A) What is powder metallurgy? Describe various steps involved in powder metallurgy with each step controlling properties of final sintered component. [5]
 - (B) Which are merits, demerits and application of powder metallurgy? [5]
- OR**
- (A) Explain Ultrasonic testing with advantages and limitations. [5]
 - (B) What is non destructive test? Explain X ray Radiography. [5]

ALL THE BEST

KADI SARVA VISHVAVIDHYALAYA
B.E. SEM III

Subject Code: ME-304

Subject Name: Material Science & Metallurgy

Date: 28/04/2014

Time: 10.30a.m.-1.30p.m.

Total Marks: 70

Instructions:

1. Answer each section in separate Answer Sheet.
2. Use of Scientific calculator is permitted.
3. All questions are **compulsory**.
4. Indicate **clearly**, the options you attempt along with its respective question number.

Section - I

- Q 1** (A) Define 1) Toughness 2) Hardness 3) Hardenability 4) Malleability 5) Creep [5]
(B) Explain selection criteria for engineering materials. [5]
(C) Draw neat sketch of metallurgical microscope. Explain its construction. [5]

OR

- (C) Explain properties and application of Wrought iron.
Q 2 (A) Define Allotropy. Explain allotropy of Iron. [5]
(B) State composition, specific properties and applications of White Cast Iron. [5]

OR

- Q 2** (A) What is metallography? What useful information can be obtained from it? [5]
(B) Which are advantages and limitations of powder metallurgy? [5]
Q 3 (A) Specify, with reasons alloy suitable for the manufacture of: Bolts and Nuts; Lathe Bed ; Milling Cutter. [5]
(B) What is solid solution? Explain types of solid solution. [5]

OR

- Q 3** (A) What is phase diagram? Explain Lever rule. [5]
(B) Explain the difference between Annealing and Normalizing. [5]

Section - II

- Q 4** (A) Draw Iron-Carbon Diagram and explain eutectic reaction in it. [5]
(B) Explain the effects on steel by alloying elements Silicon, Sulphur and Phosphorous. [5]
(C) Explain cathodic protection against corrosion. [5]

OR

- (C) Explain any two methods for production of metal powders. [5]
Q 5 (A) Explain Ultrasonic testing with advantages and limitations. [5]
(B) Which are various Surface Hardening processes? Explain Induction Hardening process with sketch. [5]

OR

- Q 5** (A) Write short note on Metallic coatings. [5]
(B) Compare and Contrast: Destructive test with Non Destructive test [5]
Q 6 (A) State various methods of prevention of corrosion and briefly explain any two. [5]
(B) Define "corrosion". Explain different types of corrosion.

OR

- Q 6** (A) Explain TTT diagram in detail. [5]
(B) Explain the method of Sulphur Printing for steels and state the inferences that could be drawn out by the technique. [5]

----ALL THE BEST----

KADI SARVA VISHWAVIDHYALAYA

B.E. SEM III

Subject Code: ME-304

Subject Name: Material Science & Metallurgy

Date: 17/11/2014

Time: 10.30a.m.-1.30p.m.

Total Marks: 70

Instructions:

1. Answer each section in separate Answer Sheet.
2. All questions are **compulsory**.
3. Indicate **clearly**, the options you attempt along with its respective question number.
4. Use the last page of main supplementary of **rough work**.

Section - I

- Q 1** (A) Define utmost applicable mechanical property for any five components mentioned below : [5]
1. Milling Cutter 2. Forging Die 3. Spring 4. Boiler Tube 5. Telephone wire 6. Lathe machine Bed 7. Fasteners 8. Machine Structure
- (B) Describe method used for finding the Sulfur Percentage in Steel. [5]
- (C) What is micro examination of metal? What are the various steps required for such an examination. [5]

OR

- (C) Differentiate between white cast iron and grey cast iron.
- Q 2** (A) What are limitations of Phase Diagram? Explain Lever Rule. [5]
- (B) Explain process of Nucleation in Solidification of metal. [5]

OR

- Q 2** (A) Describe Ductile fracture. [5]
- (B) Explain with neat sketches the arrangement of atoms in B.C.C., F.C.C. and H.C.P. lattice. Define Unit Cell. [5]
- Q 3** (A) Specify, with reasons alloy suitable for the manufacture of: Cutting Tool of Lathe, Gear, Jaw Crusher. [5]
- (B) State the types of solid solution and explain Hume Rothery's rule for the formation of solid solution. [5]

OR

- Q 3** (A) Discuss and differentiate between Austempering and Martempering. [5]
- (B) What are the methods of carburizing? Explain Pack carburizing in brief. [5]

Section - II

- Q 4** (A) Draw and Explain Time Temperature Transformation diagram. [5]
- (B) Explain the effects on steel by alloying elements Tungsten, Cobalt and Molybdenum. [5]
- (C) Describe: 1) Pearlite 2) Austenite 3) Cementite 4) Ledeburite 5) Bainite. [5]

OR

- (C) Explain any two methods for production of metal powders. [5]
- Q 5** (A) Explain ultrasonic testing with advantage, limitations and applications. [5]

(B) What is Critical Cooling Rate? Explain Cooling curves with respect to TTT diagram. [5]

OR

Q 5 (A) Draw Iron-Carbon Diagram and explain eutectic reaction in it. [5]

(B) What is nondestructive test? Explain X ray Radiography. [5]

Q 6 (A) State various methods of prevention of corrosion and briefly explain any two. [5]

(B) Define "corrosion". Explain different types of corrosion.

OR

Q 6 (A) Explain Bearing Materials along with its properties and types. [5]

(B) Describe method for manufacturing of Self Lubricating Bearing. [5]

----ALL THE BEST----

Section - II