

# Carnot Number

Input/Output Temperature Difference in Hydroelectric Power Stations  
Seasonal Temperature Difference in Ocean/River Systems

## Human Relevance

### Grand Coulee Dam



- Hydro electric stations are the oldest form of grid scale power plant
- Largest power plant in the US ~ 7GW
- ~2× larger than the second largest power station (Palo Verde - Nuclear)
- Roosevelt Reservoir heats up the rivers waters
  - Much larger impact than factories or traditional power plants which use rivers/streams for cooling
- Salmon are particularly sensitive to the temperature of the river
  - >15°C strains the fish, >20°C can be fatal to the fish
- In 2019 ~  $\frac{1}{2}$  of all 500,00 sockeye salmon returning to the Columbia River basin died because of abnormally high temperatures
  - [https://www.oregonlive.com/environment/2015/07/hot\\_water\\_killing\\_half\\_of\\_colu.html](https://www.oregonlive.com/environment/2015/07/hot_water_killing_half_of_colu.html)

# Approximating The “Carnot Number”

## Heat Drop Across Hydro-plant

### Definitions

$$\begin{aligned}\eta &\equiv \text{Turbine efficiency of power plant} \Rightarrow (1 - \eta) \approx f \cdot 10^{-1} \\ g &\equiv \text{Gravitational acceleration} \approx 10 \frac{m}{s^2} \\ h &\equiv \text{Height Difference between upper and lower reservoirs} \approx 10^2 m \\ c_p &\equiv \text{Specific heat of water} \approx f \cdot 10^3 \frac{J}{kg \cdot ^\circ C} \\ \dot{m} &\equiv \text{Flow rate through plant} \\ t &\equiv \text{time} \\ T &\equiv \text{Temperature}\end{aligned}$$

### Governing Eqns.

$$\begin{aligned}\text{Power into Heat: } P_H &= (1 - \eta) \dot{m} g h \\ \text{Work into heat: } W_H &= \int_0^{\Delta t} P_H d\tau = (1 - \eta) \dot{m} g h \Delta t \\ \text{Heat to raise the temp. of water: } \Delta Q &= c_p (\dot{m} \Delta t) \Delta T \\ (1 - \eta) \dot{m} g h \Delta t &= c_p (\dot{m} \Delta t) \Delta T \\ \Rightarrow \Delta T &= \frac{(1 - \eta) g h}{c_p} = \frac{f \cdot 10^{-1} \cdot 10 \frac{m}{s^2} \cdot 10^2 m}{f \cdot 10^3 \frac{J}{kg \cdot ^\circ C}} = 10^{-1} ^\circ C\end{aligned}$$

Checking the units:

## Heat Change Over The Year

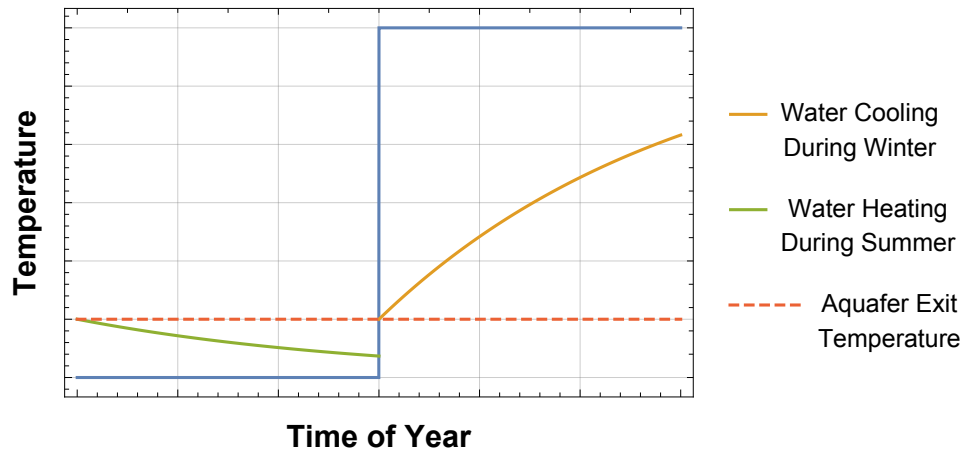
### Intuition

We know that river's in the north build up ice in the winter ( $\approx 10^0 ^\circ C$ ) to being comfortable enough to swim in during the summer ( $\approx 10^1 ^\circ C - f \cdot 10^1 ^\circ C$ ).

Thus the total temperature change is either  $10^1 ^\circ C - f \cdot 10^1 ^\circ C$ .

To make a more mathematical guess, consider the temperature variation over the seasons to be a square wave rather than a sinusoid (Its either summer, or winter). Assume that the water in a dam's reservoir originated entirely from underground aquifers, which were held at a constant temperature.

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## Saving Graphic

## Definitions

$h \equiv$  Heat transfer coefficient of Air  $\approx 10 \frac{W}{m^2 K}$   
 $c_p \equiv$  Specific heat of water  $\approx f \cdot 10^3 \frac{J}{kg \cdot ^\circ C}$   
 $m \equiv$  Mass of body being cooled  $\approx$  Lake Roosevelt  $10^{13} \text{ kg}$   
 $A \equiv$  Exposed surface area of body being cooled  $\approx$  Lake Roosevelt  $f \cdot 10^8 m^2$   
 $T_a \equiv$  Temperature of air  $\approx f \cdot 10^\circ C$  during the summer,  $0^\circ C$  during the winter  
 $T_{aq} \equiv$  Temperature of water in aquafer  $\approx f^\circ C$  year round  
 $T_w \equiv$  Temperature of water in reservoir  
 $Q \equiv$  Heat

## Governing Eqns.

Newtonian cooling:  $\frac{\partial Q}{\partial t} = h A (T_a - T_w(t))$

Heat to change the temp. of water:  $\partial Q = c_p m \partial T_w$

$$\frac{\partial T}{\partial t} = \frac{hA}{c_p m} (T_a - T_w(t))$$

$$T_w(t) = T_a - e^{-\frac{Aht}{cm}} (T_{aq} - T_a)$$

### Summer

$$T_w(t) = f \cdot 10^\circ C - \exp\left(-\frac{f \cdot 10^8 m^2 \cdot 10 \frac{W}{m^2 K} \cdot t}{f \cdot 10^3 \frac{J}{kg \cdot ^\circ C} \cdot 10^{13} \text{ kg}}\right) (f^\circ C - f \cdot 10^\circ C)$$

$$T_w(t) = f \cdot 10^\circ C - \exp(-t \cdot 10^{-7}) (f \cdot 10^\circ C)$$

After 6 months  $t \approx f \cdot 10^7 \Rightarrow$

$$T_w(t) = f \cdot 10^\circ C - \exp(-f) (f \cdot 10^\circ C) = f \cdot 10^\circ C - f \cdot 10^{-2} \cdot (f \cdot 10^\circ C) = f \cdot 10^\circ C$$

### Winter

$$T_w(t) = 0^\circ\text{C} - \exp\left(-\frac{f \cdot 10^8 \text{ m}^2 \cdot 10 \frac{\text{W}}{\text{m}^2 \text{ K}} \cdot t}{f \cdot 10^3 \frac{\text{J}}{\text{kg} \cdot ^\circ\text{C}} \cdot 10^{13} \text{ kg}}\right) (f^\circ\text{C} - 0^\circ\text{C})$$

$$T_w(t) = -\exp(-t \cdot 10^{-7}) (f^\circ\text{C})$$

After 6 months  $t \approx f \cdot 10^7 \Rightarrow$

$$T_w(t) = f \cdot 10^{-2} \cdot (f^\circ\text{C}) = 10^{-1}^\circ\text{C}$$

Checking the units:

Approximate Carnot Number

$$\frac{10^{-1}^\circ\text{C}}{f \cdot 10^\circ\text{C}} = f \cdot 10^{-2}$$

## What The Data Says

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### Columbia River - Grand Coulee Dam

Importing Data

Below Coulee

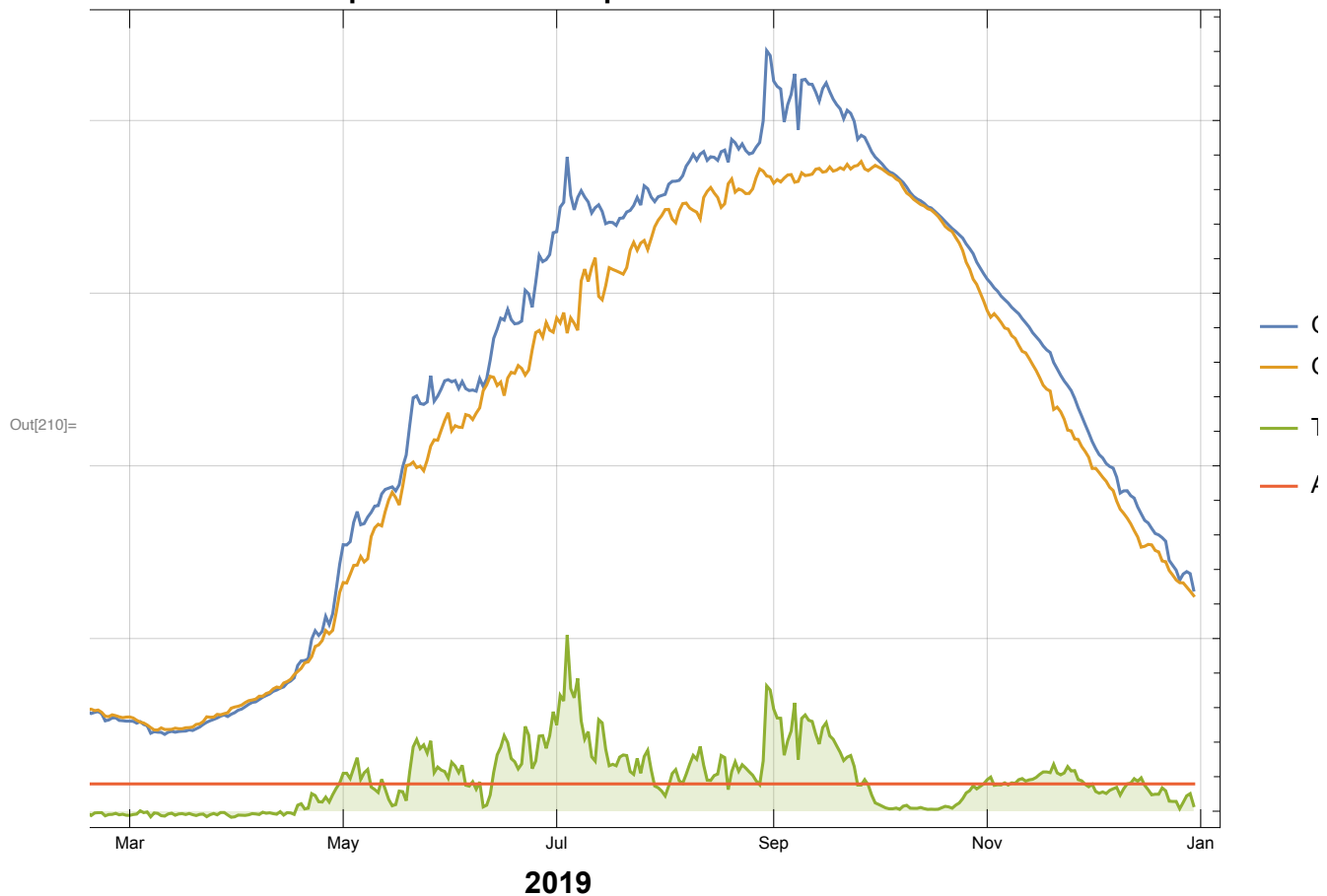
Above Coulee

Difference Across

Saving Data

## Plotting Data

### Columbia River Temperature Drop Across Grand Coulee Dam



## Saving Graphic

## Data Source

"Columbia Basin Conditions for Temperature, Dissolved Gas Percent and Spill Percent"

⇒ [http://www.cbr.washington.edu/dart/query/basin\\_conditions](http://www.cbr.washington.edu/dart/query/basin_conditions)

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