

1 x = 1.1 cm = 0.011 m

From Spring compressed or leaving

Bobby = 1 MAXB2 + Mgh = = = mv2

Rhoda = nAxx + ngh = = = mv2

Say this happens in a second (time some because freetail)

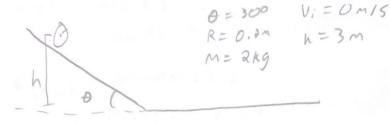
BOBBY = haxB = = = 1 M. (2.2-0.27)2

Rhota JNAXR = JM(2.2)

$$\frac{B}{A} = \frac{1}{2} \frac{h \Delta x B^2}{2} = \frac{1}{2} \frac{m (1.93)^4}{2m (2.93)^3}$$

$$= \frac{0.011^{2}}{A \times R^{2}} = \frac{1.43^{2}}{12.23^{2}}$$

1XR = \(\frac{0.0112}{\left(\frac{1.43^2}{2.26^2}\right)} = 0.012539 m. \(\frac{100cm}{100} = \left(\frac{1.25389cm}{100} \)



$$M\alpha = Mgsin \theta - F$$

$$f \cdot R \cdot sket = I \propto = I \frac{a}{R}$$

$$\Rightarrow f = I \frac{a}{R}a$$

$$M \cdot \alpha = Mg \sin \theta - I \frac{\alpha}{R} \alpha$$

$$M\alpha = Mg \sin \theta - \frac{1}{2} MR^{2} \alpha$$

$$M\alpha = Mg \sin \theta - \frac{1}{2} M\alpha$$

$$\alpha + \frac{1}{2} \alpha = g \sin \theta$$

$$\alpha = \frac{10 \cdot \sin(30)}{3} - \frac{3.33}{3} M/5^{2} \alpha$$

(6.)
$$Mgh = \frac{1}{2}Mv^{2}$$

$$V = \sqrt{2gh} = \sqrt{2.10.3} = 7.74597 \text{ M/S}$$

$$V = \Gamma W \qquad \frac{1}{2} = W = \frac{7.74597}{0.a} = \frac{38.7298}{0.a} \text{ Cad/s}$$

$$RE_{rot} = \frac{1}{a} I W^{2} = \frac{1}{a} (\frac{1}{a} M R^{2}) \cdot W^{2} = \frac{1}{a} (\frac{1}{a} \cdot a \cdot 0.2^{2}) \cdot 38.7298^{2}$$

$$= 30 \text{ J}$$

(a) From B > C Isovolumetric
$$PV = nRT \Rightarrow \frac{P_1}{T_1} = \frac{Pa}{Ta}$$

$$\frac{P_C}{T_C} = \frac{PB}{TB} \qquad lata = 10^5 Pa \qquad \frac{10^5}{300 K} = \frac{4.10^5}{TB}$$

$$T_B = \frac{4.10^5}{\frac{10^5}{3000}} = 1200 \text{ K}$$

From A >B I so Baric
$$PV = 0RT = \frac{10^{5}}{1200K}$$
 $\frac{V_B}{T_B} = \frac{V_A}{T_A} \frac{11 = 10^{-3} \text{ m}^3}{1200K} = \frac{0.5 \cdot 10^{-3}}{T_A} = \frac{0.5 \times 10^{-5}}{1200} = \frac{300K}{1200}$

Frank Ceraolo

CAA
$$\Delta E_{int} = \Lambda CV \Delta T = 0$$

 $V = -\rho \Delta V$
 $V = -\frac{5}{2} \cdot 10^{5} \cdot (0.5 \cdot 10^{-3}) \cdot (0.5 \cdot 10^{-$

(Ci) AEint = 1 CUST for all processes At Point C: PV= ORT (105). (2.10-3) = 1.8.314.300 => n = 0.080186 moles

A-B AE=nCVAT=0,080186. 3.8.314. (1200-300) = 900J

BAC AEINY = nCVAT = 0.080186 3.8.314. (300-1200) = -900 J

C>A AEin+=1000A7=0.080186 13 .8.314.(300-300)=05

(1) A>B (150 baric)
$$Q = n C \rho \Delta T$$
 $Q = 0.080186. \frac{5}{3}.8.314.(1200 - 300) = 1500 J$
 $B>C (iso volumetric) $Q = n C v \Delta T$
 $Q = 0.080186. \frac{3}{3}.8.314.(350 - 1200) = -900 J$
 $C \rightarrow A$ Since $\Delta E_{int} = 0$ $Q = -w$
 $Q = -375 J$$

(6.)
$$T = V$$

$$8 = V$$

$$V = V$$

(Ci) U'=
$$\frac{U-V}{1-\frac{uv}{ca}}$$
 Sframe = Probe Object = interceptor

$$U' = \frac{-0.7C - 0.3C}{1 - \frac{0.7C \cdot 0.3C}{C^2}} = -21.479342108$$

$$U' = \left(-0.826446C\right)$$