#### Introduction to Laser Theory

#### John Suárez Princeton REU Program Summer 2003



#### What we'll talk about . . .

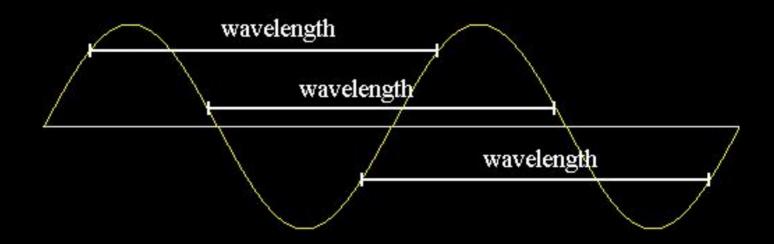
- <u>Light amplification by stimulated emission of radiation</u>
- Electronic energy levels
- Energy transfer processes
- Classifications of lasers and properties of laser beams
- Detecting and characterizing very fast laser pulses

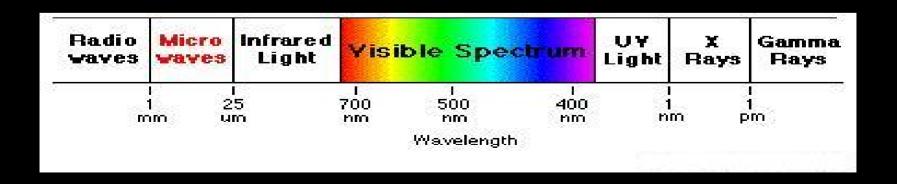
#### The Invention of the Laser



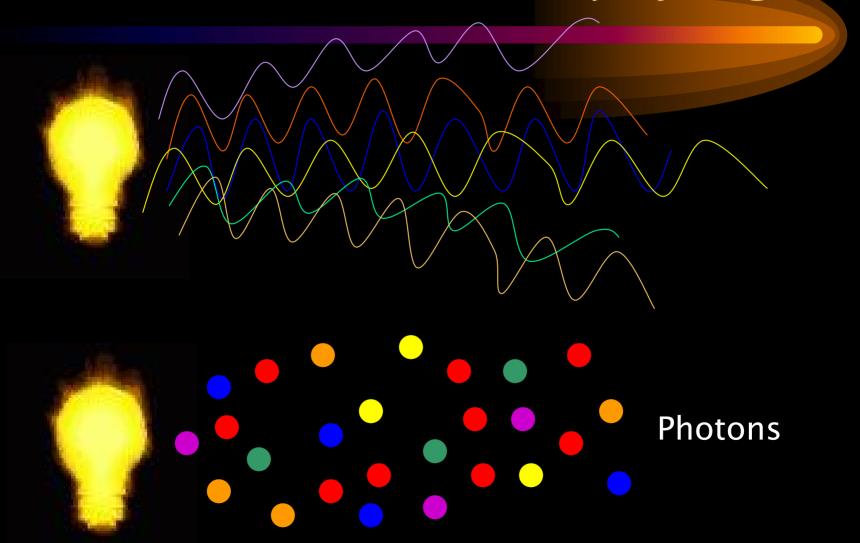
- Invented in 1958 by Charles Townes and Arthur Schawlow of Bell Laboratories
- Was based on Einstein's idea of the "particle-wave duality" of light, more than 30 years earlier
- Originally called MASER (m="microwave")

#### The Electromagnetic Spectrum



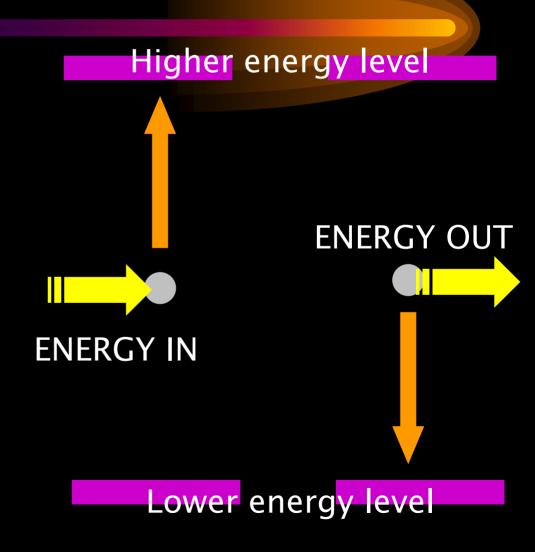


### Particle-Wave Duality of Light



# Energy is "quantized"

- To raise an electron from one energy level to another, "input" energy is required
- When falling from one energy level to another, there will be an energy "output"
- Theoretically: infinite number of energy levels.



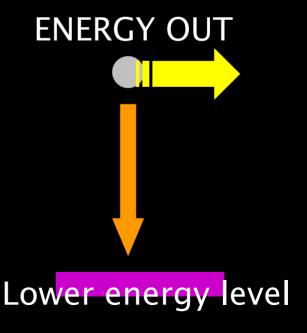
#### Spontaneous Emission

Electron initially in level 2 "falls" to level 1 and gives off energy (just happens spontaneously)

Energy is emitted in the form of a photon:

$$E = hf$$

Higher energy level



#### Stimulated Emission

Same idea as spontaneous emission except we MAKE it happen by sending in an EM wave of frequency  $f_0$ 

A photon is given off with the energy

$$E = hf_0$$

Higher energy level

ENERGY IN

(frequency =  $f_0$ )

ENERGY OUT

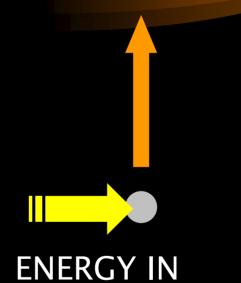
(frequency =  $f_0$ )

Lower energy level

#### Stimulated Absorption

Electron is initially in level 1. We send in an EM wave and the electron goes UP from level 1 to level 2.

Higher energy level



Lower energy level

#### Energy Transfer Processes

#### Spontaneous emission

 electron "naturally" falls down from level 2 to level 1

#### Send in an electromagnetic (EM) wave:

- Stimulated emission
  - electron can be knocked down from level 2 to level 1
- Stimulated absorption
  - electron can be raised from level 1 to level 2

# Energy Transfer Processes: Quantitatively

- Define  $N_i$ , the <u>population</u> of level i
- $N_i(t)$  = the # of electrons, per unit of volume, occupying energy level i at time t

# Energy Transfer Processes: Quantitatively

For spontaneous emission:

$$\frac{dN_2}{dt} = -AN_2$$

For stimulated emission:

$$W_{2 \text{tol}} = \sigma_{2 \text{tol}} \Phi_{\text{photon}}$$

$$\frac{dN_2}{dt} = -W_{2 \text{ to } 1} N_2$$

$$W_{1 \text{to} 2} = \sigma_{1 \text{to} 2} \Phi_{\text{photon}}$$

For stimulated absorption:

$$g_2 W_{2 \text{ to 1}} = g_1 W_{1 \text{ to 2}}$$

$$\frac{dN_2}{dt} = +W_{1 \text{ to } 2}N_1$$

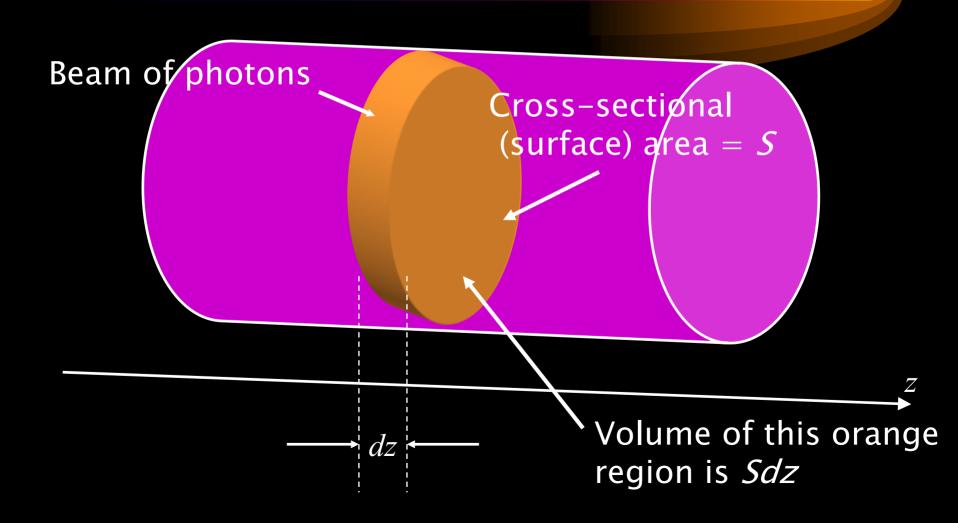


Beam of photons

Beam has a cross-sectional (surface) area of S

Material

As the beam travels, it will cause stimulated emission and absorption within the material.



How would we express the <u>incremental change in photon</u> flux  $(d\Phi)$  in the material due to the photon beam?

$$Sd\Phi = \begin{pmatrix} \text{change in } \# \text{ of photons} \\ \text{due to stim. emission} \\ \text{(per unit of time)} \end{pmatrix} - \begin{pmatrix} \text{change in } \# \text{ of photons} \\ \text{due to stim. absorption} \\ \text{(per unit of time)} \end{pmatrix} Sdz$$

$$Sd\Phi = \left( \left| \frac{dN_2}{dt} \right|_{\text{stim. emiss.}} - \left| \frac{dN_2}{dt} \right|_{\text{stim. absorp.}} \right) Sdz$$

$$W_{2 \text{ to } 1} = \sigma_{2 \text{ to } 1} \Phi_{\text{photon}}$$

$$W_{1 \text{ to } 2} = \sigma_{1 \text{ to } 2} \Phi_{\text{photon}}$$

$$g_2 W_{2 \text{ to } 1} = g_1 W_{1 \text{ to } 2}$$

$$Sd\Phi = \left( \frac{|dN_2|}{dt} \right|_{\text{stim. emiss.}} - \left( \frac{|dN_2|}{dt} \right|_{\text{stim. absorp.}} Sdz$$

$$Sd\Phi = (W_{2 \text{ to } 1}N_2 - W_{1 \text{ to } 2}N_1)Sdz$$

$$\frac{d\Phi}{dz} = \left(W_{2 \text{ to } 1} N_2 - W_{1 \text{ to } 2} N_1\right)$$

$$\frac{d\Phi}{dz} = \left[ \sigma_{2 \text{ to } 1} (N_2 - \frac{g_2}{g_1} N_1) \right] \Phi$$

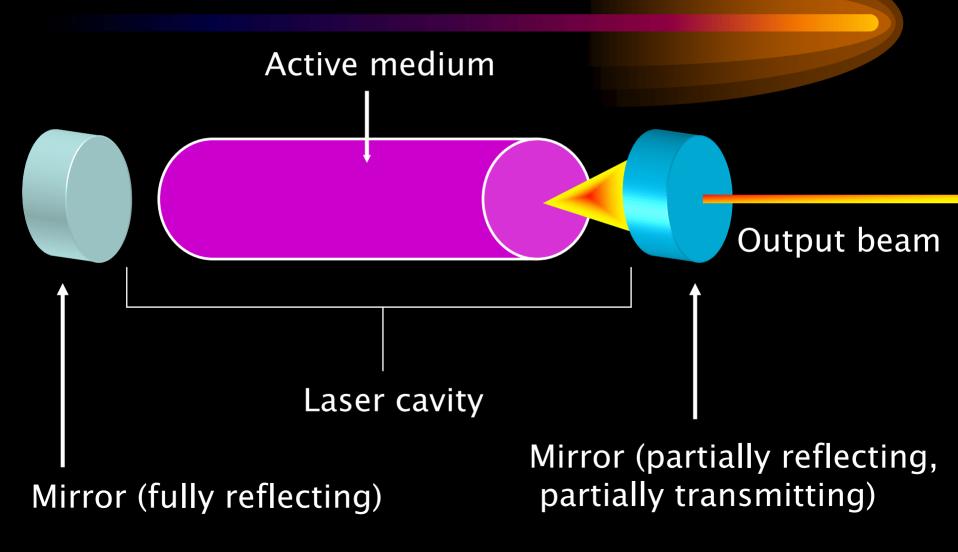
$$\frac{d\Phi}{dz} = \begin{bmatrix} \sigma_{2 \text{ to } 1} (N_2 - \frac{g_2}{g_1} N_1) \end{bmatrix} \Phi \quad \frac{d\Phi}{dz} > 0 \Rightarrow \text{amplifier}$$
"Population inversion"

In thermal equilibrium: material acts as an absorber

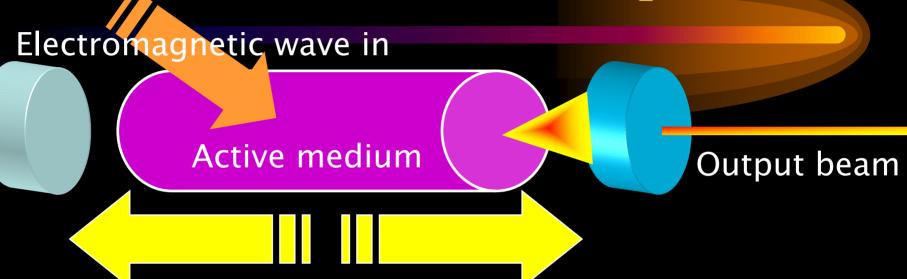
Some materials: amplifiers when not in thermal equilibrium



### Components of a Laser



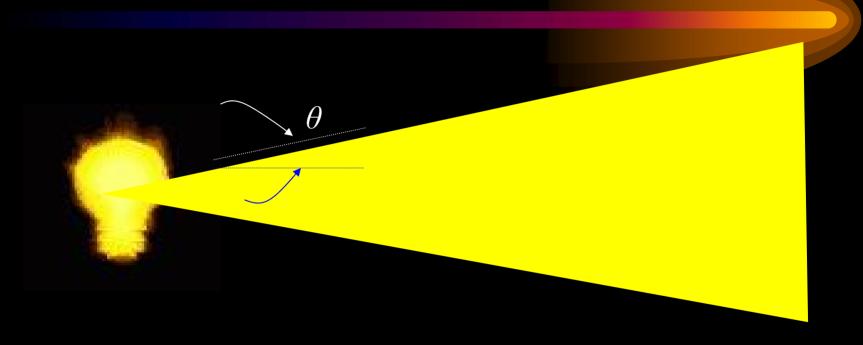
#### Basic Laser Operation



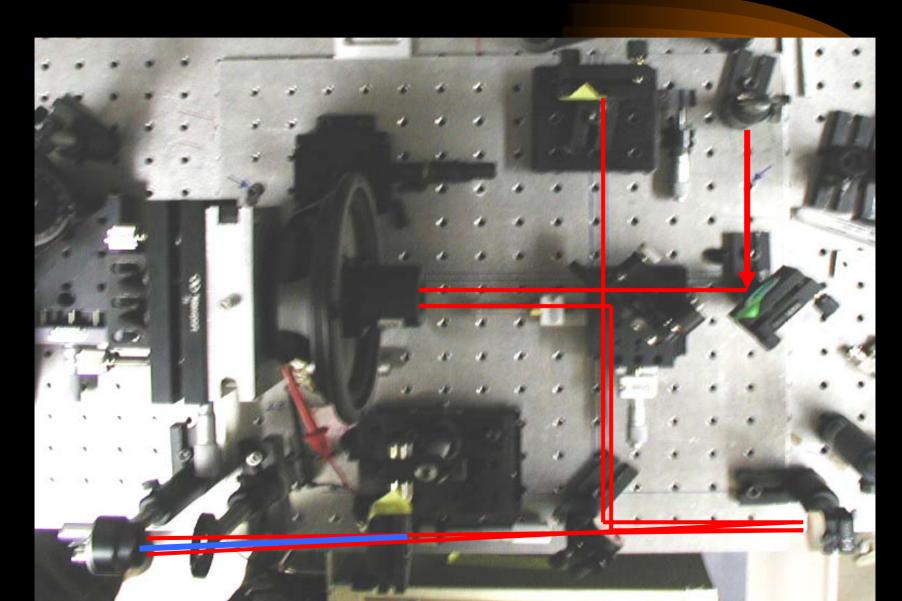
EM wave is repeatedly reflected within the laser cavity

If the frequency (f) of the output beam is . . . microwave region (1 GHz – 30 THz) . . . MASER optical region (430 THz – 750 THz ) . . . LASER

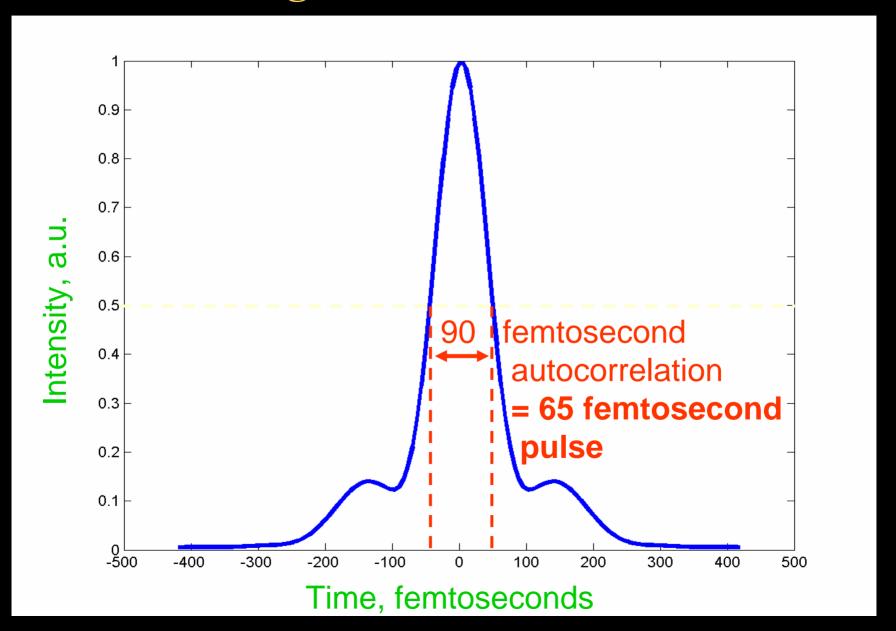
### Properties of Laser Beams



# Characterizing Fast Laser Pulses: Autocorrelation



#### Characterizing Fast Pulses: Autocorrelation



#### Summary

- The electromagnetic spectrum
- Energy level diagram representation
- Spontaneous and stimulated emission, absorption
- The laser concept: rate equations, population inversion, amplification
- Properties of laser beams
- Autocorrelation: characterizing femtosecond laser pulses

#### Thank You

- Dr. Warren, Wolfgang, and the entire Warren group
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